



MED444

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- Boys' slides only

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Body fluids & electrolytes



1- Identify and describe daily intake and output of water and maintenance of water balance.

2- List and describe of body fluid compartments as intra-cellular fluid(ICF), Extra-cellular fluid (ECF), interstitial fluid, trans-cellular fluid and total body water (TBW).

3- Describe the composition of each fluid compartment, in terms of volume and ions and represent them in graphic forms.

4-Physiology factor influencing body fluid: age, sex, adipose tissue,etc. Pathological factors: Dehydration, fluid infusion.



Body Water Content

In general Human body contain 50	0-70% or 40%-70% water.
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- kg of water = Liter of water

Example:

- -70 kg man has 42L of water
- So ---> he has 42kg of water

PERCENTAGE OF WATER IN THE BODY

Click each of the people below to determine the approximate percentage of water in their bodies.



	Percentage of body water	clarification
Infants	70 - 73% or more	low body fat,low bone mass
Healthy males adult	60%	Low body fat larger amount of skeletal muscle
Healthy females adult	40-50%	Higher body fat <mark>Smaller</mark> amount of skeletal muscle
Older age	45%	Total body water declines throughout life.
Obese	45%	In general Higher body fat Means less body water

Daily intake of water

The amount of water taken = the amount of water lost

Table 25-1 Daily Intake and Output of Water (ml/day)			
	Normal	Prolonged, Heavy Exercise	
Intake			
Fluids ingested	2100	?	
From metabolism	200	200	
Total intake	2300	?	
Output			
Insensible—skin	350	350	
Insensible—lungs	350	650	
Sweat	100	5000	
Feces	100	100	
Urine	1400	500	
Total output	2300	6600	

These numbers are average numbers so if there is a little changes in exams no problem. For E.g: The urine output in normal status equals: 1500 200 100 350

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Regulation of Water Intake

In steady state water intake = water loss

• Regulation of water intake depends on:



Factors that affect the TBW

1) Physiological factors:

- Age
- Sex
- Body fat
- Climate
- Physical activity

2) Pathological factors:

- Vomiting (القيء)
- Diarrhea (اسهال)
- Diseases with excessive loss of water (Diabetes mellitus DM, excessive sweating)
- Blood loss



Fluid Compartments :

Water occupies two main fluid compartments:

- 1) Intracellular fluid (ICF)
- Extracellular fluid (ECF)
 -Plasma
 - -Interstitial fluid (IF)





1) Intracellular fluid (ICF)

- Location: Inside the cell.
- Amount: 2/3 of Total body water (TBW).
- High concentration of protein
 لأن البروتينات حجمها كبير فهي لا تنتقل للخارج بسهولة عبر غشاء الخلية

2) Extracellular fluid (ECF)

Location: outside the cell.

Amount: 1/3 of Total body water (TBW).

1) Plasma:

- Location: Fluid circulating in the blood vessels.
- Amount: 1/4 of ECF

2) Interstitial fluid:

- Location: Fluid bathing the cell.
- This fluid come from: Ultra filtration of plasma.
- Amount: 3/4 of ECF

Plasma & interstitial fluid are almost having the same composition except for high protein concentration in plasma.

لان لما تصير عملية الفلتره البروتينات حجمها كبير ولا تستطيع الانتقال عبر غشاء اوعية الدم فتبقى في البلاز ما(الدم)

3) Transcellular fluid compartment:

• Amount: small amount

• Location:

CSF سوائل حول الدماغ والحبل الشوكي GIT fluid سوائل تفرز في القناة الهضمية biliary fluid المحارة الصغر اوية الذي يخرج من الكبد أو المرارة synovial fluid سائل داخل المفاصل Intrapleural fluid سائل في الغشاء حول الرئتين intraperitoneal fluid السوائل بداخل غشاء يغلف المعدة السوائل بداخل غشاء يغلف المعدم الما السوائل بداخل الما والما



Composition of Body Fluids



Concentration

In general:-	In Biological solutions:	- In Electrolytes:-
Molarity= moles/Liter (M/L)	●Millimoles per liter (mM/L)	•Expressed in milliequivalents per liter (mEq/L), a measure of the number of
Osmolarity = Osmoles/Liter (Osm/L)	•Milliosmoles per (mOsm/L) •1mM=1/1000 M	 electrical charges in one liter of solution. mEq/L = (concentration of ion in [mg/L]/the atomic weight of ion) x number
Osmolality = Osmoles/kilogram (Osm/kg)	•1mOsm=1/1000 Osm	of electrical charges on one ion. •For single charged ions, 1 mEq = 1 mOsm
Moles vs Osmoles:- Moles are a certain solutes n need to be a solute). While (solute) are the number of me Osmotic pressure (1 Mole of Osmoles)	umber (doesn't actually Osmoles (needs to be in a oles that contribute to NaCl dissolves into 2	•For bivalent ions, 1 mEq = 1/2 mOsm







Constituents of ECF & ICF

	Plasma (m0sm/liter of H ₁ D)	Interstitio	Intracellula
Na*	142	139	14
K-	4.2	4.0	140
Ca ⁺⁺	1.3	1.2	0
Mg*	0.8	0.7	20
CI-	108	108	4
HCO,	24	28.3	10
HPO, H,PO,	2	2	11
so,	0.5	0.5	1
Phosphocreatine			45
Carnosine			14
Amino acids	2	2	8
Creatine	0.2	0.2	9
Lactate	1.2	12	15
Adenosine triphosphate			5
lexose monophosphate			37
Glucose	5.6	5.6	our .
Protein	1.2	0.2	4
Urea	4	4	4
Others	4.8	3.9	10
Total mOsm/liter	301.8	300.8	301.2
Corrected osmolar activity (mOsm/liter)	282.0	281.0	281.0
Total osmotic pressure	5443	5423	5423

ECF & ICF

Each fluid compartment of the body has a distinctive pattern of electrolytes.

Extracellular fluids are similar (except for the high protein content of plasma relative to interstitial fluids)

ECF

•Extracellular fluid has low Potassium and Phosphate.

-Sodium is the chief cation -Chloride is the major anion ICF

•Intracellular fluid has low sodium and chloride.

-Potassium is the chief cation

-Phosphate is the chief anion

Each compartment must have almost the same concentration of negative charges (anions) as the positive charges (Cations).

(Electroneutrality)

Abnormalities of ECF & ICF



•Hypokalemia: Abnormal decrease (1-2 mEq/L) of Potassium (K) concentration in ECF.

•Hyperkalemia: abnormal increase (60-100% above normal) of potassium(K) Concentration in ECF.

Hypernatremia: Abnormal increase of Na concentration in ECF.

Hyponatremia: Abnormal decrease of Na concentration in the ECF.

Normal range of Na and K:-

Na (135-145 mEq/L) K (3.5-4.5 mEq/L)

The slides are hyperfocusing on ECF (specifically plasma/blood) values and Abnormalities because they are easier to measure and take compared to ICF

Extracellular fluid	Intracellular fluid
Na+ 142 mEq/L	10 mEq/L
K ⁺ 4 mEq/L	140 mEq/L
Ca ⁺⁺ 2.4 mEq/L	0.0001 mEq/L
Mg + + 1.2 mEq/L	58 mEq/L
CI 103 mEq/L	4 mEq/L
HCO3 ⁻ 28 mEq/L	10 mEq/L
Phosphates 4 mEg/L	75 mEq/L
SO4	2 mEq/L
Glucose 90 mg/dl	- 0 to 20 mg/dl
Amino acids 30 mg/dl	200 mg/dl ?
Cholesterol)	
Phospholipids 0.5 gm/dl	2 to 95 gm/dl
Neutral fat	
PO2 35 mm Hg	20 mm Hg ?
PCO2 46 mm Hg	50 mm Hg ?
pH 7.4	7.0
Proteins 2 gm/dl 2 gm/dl (5 mEq/L)	16gm/dl (40 mEq/L)



Regulation of fluid exchange



Intracellular:-

Cell membrane

1- Highly permeable to water (I disagree, unless we include the water pumps {aquaporins})

2-Relatively impermeable to small ions (only water is moving)

Extracellular

1-Osmotic effect of electrolytes (e.g. Na,K,Cl)

Overall this boys' slide seems to be with no actual substance (Low-yield)



1-Composed of ultrafiltration of plasma			
A) Intracellular fluid	B) Interstitial fluid	C) Transcellular Fluid	D) Both A and B
2-What is the chief cation in the ECF			
A) phosphate	B) sodium	C) chloride	D) potassium
3-When someone is thirsty the plasma osmolality			
A) increases	B) declines	C) doubled	D) tripled
4-Which of these fluids have high concentration of protein?			
A) Intracellular fluid	B) Plasma	C) Interstitial fluid	D) Both A and B
5-Which of the following is considered the "Internal Environment" of the body?			
A) Capillary walls	B) Extracellular fluid	C) interstitial fluid	D) Transcellular fluid

1-B,2-B,3-A,4-D,5-B

