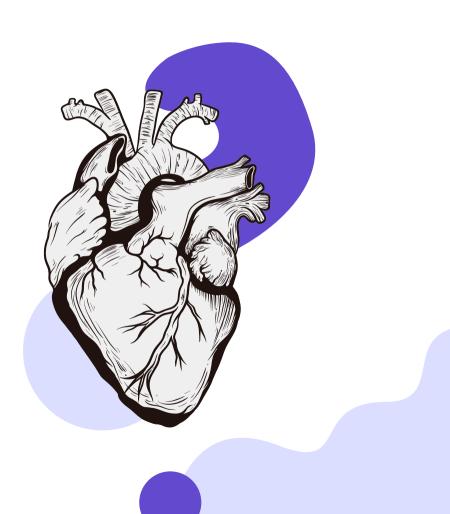


Lactic acidosis

Editing File

COLOR INDEX: MAIN TEXT (BLACK) FEMALE SLIDES (PINK) MALE SLIDES (BLUE) IMPORTANT (RED) DR'S NOTE (GREEN) EXTRA INFO (GREY)



Objective



Define metabolic acid-base disorders including lactic acidosis.

Understand the causes and clinical effects of metabolic acidosis and alkalosis.

Recall the lactate metabolism in the body.

Differentiate between the types of lactic acidosis.

Understand the clinical significance of measuring anion gap.



Discuss the causes and diagnosis of lactic acidosis in conditions such as myocardial infarction.

Metabolic acid-base disorders

Changes in bicarbonate concentration in the extracellular fluid (ECF) cause metabolic acid-base disorders.

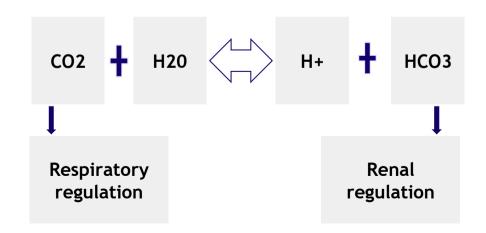
Occur due to high concentration or loss of H+ ions. Metabolic acid-base disorders

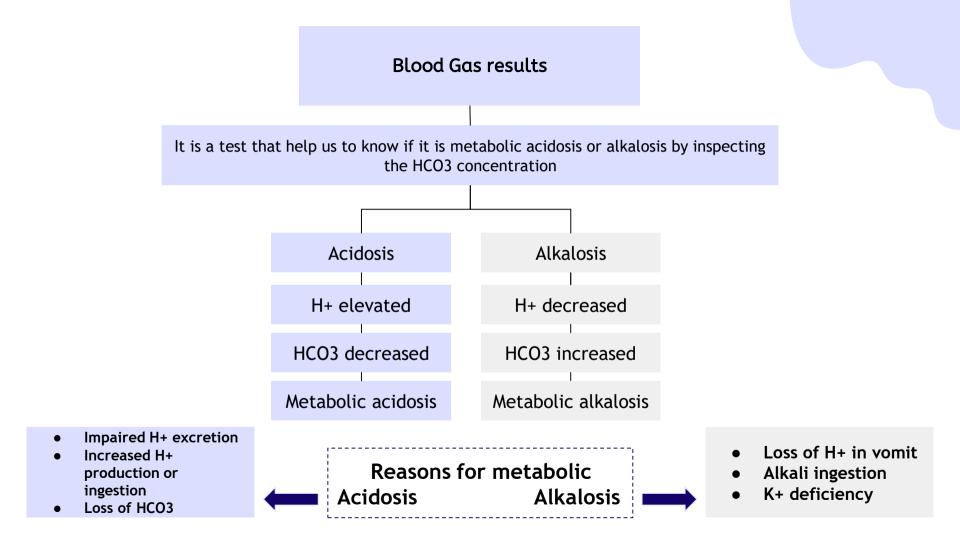
Can lead to:

- Metabolic acidosis.
- Metabolic alkalosis.

In both, the function of the body will be affected because of the change in PH

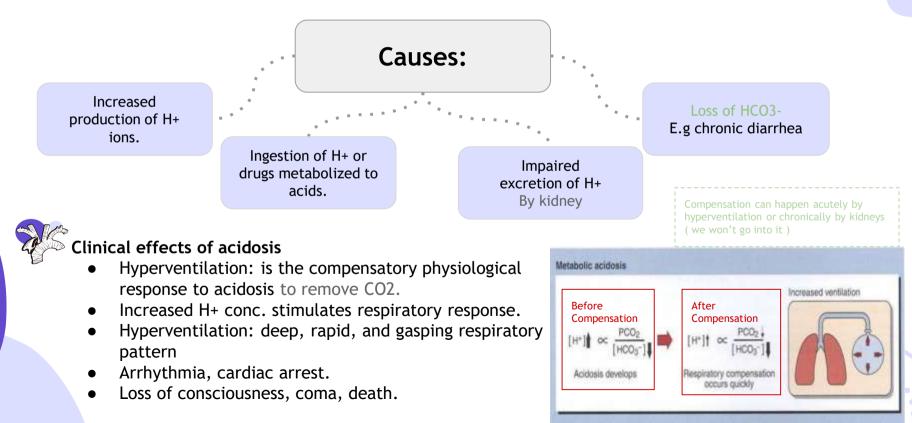
Carbonic acid-Bicarbonate buffering system





Metabolic acidosis

Reduction in bicarbonate concentration of ECF.



Anion Gap

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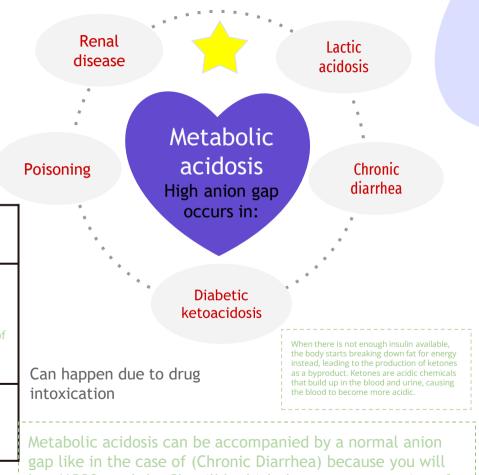
It is the difference between:

• The sum of Na+ and K+ (cations).

Helps in assessing acid-base problems.

• The sum of Cl- and HCO3- (anions).

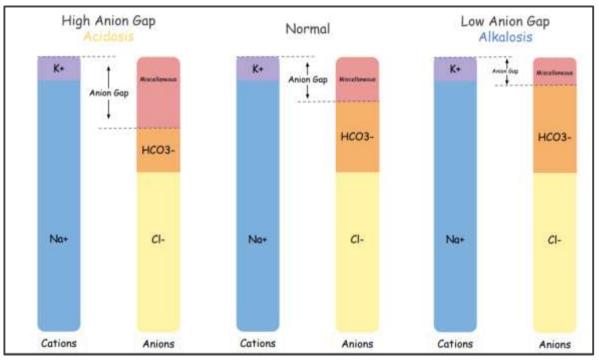
Normal anion group	3-11 mEq/L		
High anion group	>11 mEq/L (acidosis) (Can happen due to increase in unmeasured ions:po4/lactate/protein/ketos or due to loss of HCo3)		
Low anion group	<3 mEq/L (alkalosis)		



loss HCO3- and the Cl- will be high due to compensation of HCO3- This is known as hyperchloremic metabolic acidosis.

Anion Gap

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فر ضـا

In anions we only count HCO3 and Cl

Metabolic alkalosis Increase in bicarbonate conc. in ECF Causes: Ingestion of Loss of H+ ions in Sodium bicarbonate gastric fluid due to vomiting Potassium deficiency as a result of diuretic therapy In hypokalemia cells will take H+ from ECF to inside the cell to become neutralied because K+ & H+ have the same charge thus less H+ in ECF and alkalosis letabolic alkalosis linical effects of alkalosis Hypoventilation (depressed breathing) Decreased ventilation Increases PCO2 to compensate alkalosis Respiratory arrest, Confusion, coma, death [H+] ∞ Respiratory compensation Alkalosis develope occurs quickly

Lactic Acidosis

Elevated conc. of plasma lactate is called lactic acidosis

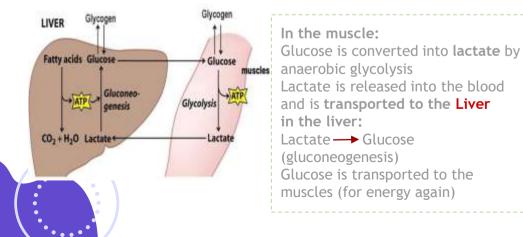


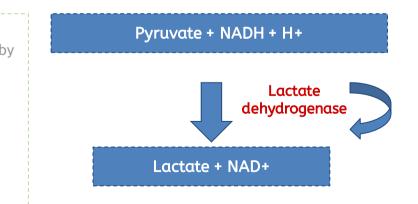
Mechanisms involved in lactic acidosis

- Excessive tissue lactate production
- Impaired Hepatic metabolism of lactate

Lactate metabolism in tissue

- The body tissues produce ~1500 moles of lactate each day
 - The lactate enters blood stream and metabolized mainly by the Liver (Cori cycle)
 - All tissues can produce lactate under anaerobic conditions
 - Pyruvate is converted to lactate by Lactate dehydrogenase enzyme
 - The skeletal muscles produce high amounts of lactate during vigorous exercise
 - Lactate is metabolized in Liver (60%) and Kidney (30%) to glucose
 - Some lactate is metabolized to CO2 and Water (Krebs cycle)





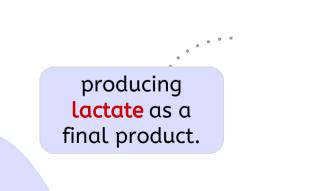
Types and causes of lactic acidosis



Type A

Due to hypoxia in tissues, which is in:

- -Myocardial infarction
- -pulmonary embolism
- -Uncontrolled hemorrhage
- -Tissue hypoperfusion (shock, cardiac arrest, acute heart failure)
- -Anaerobic muscular exercise



Causing

Impaired oxidative phosphorylation and decreased ATP synthesis.

Types and causes of lactic acidosis



To survive, Cells switch to anaerobic glycolysis for ATP synthesis And the amount of oxygen required to recover from oxygen deficiency is called oxygen debt



Туре В

- -Disorders in carbohydrate metabolism
- (Congenital lactic acidosis is due to deficiency of pyruvate dehydrogenase enzyme)
- -Chronic hepatic disease accompanied by shock or bleeding
- -Liver failure
- -Drug intoxication (poisoning)

It's important to tie between the concept of carbohydrate metabolism and Type B lactic acidosis

Diagnosis and Treatment

Diagnosis

Diagnosis done by measuring blood lactate levels: -Hyperlactemia: 2-5 mmoles/L (Risk for lactic acidosis) -Severe lactic acidosis: > 5 mmols/L

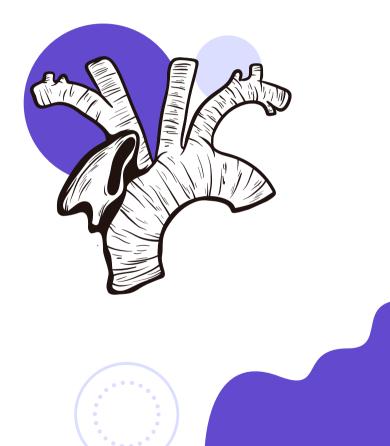
Treatment

-Correcting the underlying conditions -Restoring adequate tissue oxygen* -Avoiding Sodium Bicarbonate** 443: Dr note: * Because most of the time the cause of lactic acidosis is hypoxia

** In Acidosis in general it's known that adding sodium bicarbonate will help to get PH back to it's normal range but in lactic acidosis for unknown reasons it will make it worse by increasing lactic acidosis

Take home Messages

- **OI** Lactic acidosis can be caused by hypoxia, excessive production and impaired clearance of lactic acid.
- **02** It carries clinical significance in the diagnosis of myocardial infarction, pulmonary embolism and other metabolic conditions.



Q1:Which one is seen in metabolic acidosis with high anion gap?						
A- Chronic diarrhea	B- heart failure	C-Lactic acidosis	D- vomiting			
Q2: Hyperventilation is a clinical effect of?						
A- Acidosis	B- Alkalosis	C- Both	D- None			
Q3:Metabolic alkalosis causes which of the following ?						
A- Gain of H+	B- hypoventilation	C- Hyperventilation	D- Both A and B			

MCQ

Q1: What enzyme is deficient in type B lactic acidosis						
A- Pyruvate dehydrogenase	B-lactate dehydrogenase	C- succinate dehydrogenase	D-phosphofructokinase			
Q2: In lactic acidosis what level of blood lactate indicate severe lactic acidosis						
A- >4 mmols/L	B- <4 mmols/L	C-2-5 mmols/L	D->5 mmols/L			
Q3: Which of the following causes doesn't cause Type A lactic acidosis						
A- Uncontrolled hemorrhage	B- Myocardial infarction	C- Drug intoxication	D- Anaerobic muscular exercise			

SAQ

Q1: Define the anion gap.

A1:: [The sum of Na+ and K+ (cations)]—[The sum of Cl- and HCO3- (anions)]

Q2: What is the amount of oxygen required to recover from O2 deficiency called

A2: oxygen debt

Q3: Mention two main causes of lactic acidosis

A3: failure of circulatory system, disorders of carbohydrate metabolism





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