

نوع الزندي .

# **GLOBALAR** ⇒ **HEMOPROTEINS**

what's the meaning of Globular ?  
 - شكل كروي  
 - have 3 Dimention  
 - not more than one sub unit.

random note/ remember that hemeprotein in cytochrome can accept and release electrones (so iron in Hemoprotein cytochrome can be in oxidise and reduced state) but not in globular Hemoprotein

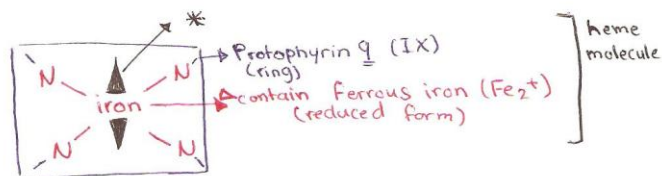
الجزء من (hemeprotein) الذي يرتبط مع Protein  
 covalant bond ↓

- **Hemeproteins** are a group of specialized proteins that contain heme as a tightly bound prosthetic group. ⇒
- **Heme** is a complex of **protoporphyrin IX** and **ferrous iron** ( $Fe^{2+}$ ).  
 ① ②
- The iron is held in the **center** of the heme molecule by **bonds** to the **four nitrogens** of the porphyrin ring.
- \* The heme  $Fe^{2+}$  can form **two additional bonds**, one on each side of the **planar porphyrin ring**. ⇒
- In myoglobin and hemoglobin, one of these positions is coordinated to the side chain of a histidine residue of the globin molecule, whereas the other position is available to bind oxygen

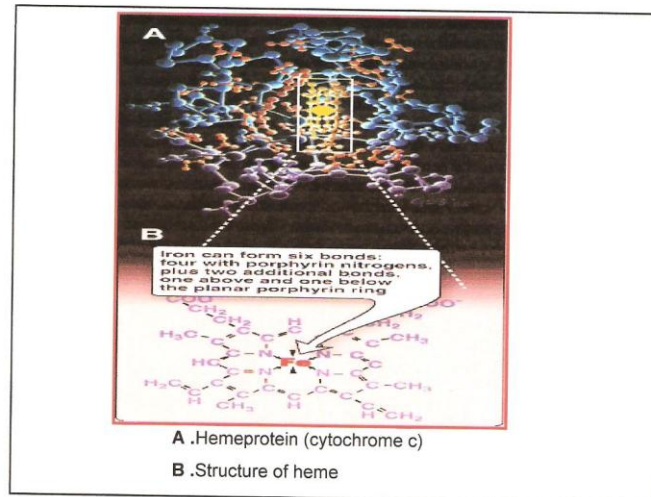
معنى آخر :-  
 Hemoprotein is Protein contain (heme prosthetically (covalantly) bond to it)

these 2 additional bonds attach to histidine in myoglobin and hemoglobin

random note |  $Fe^{2+}$  is reduced and globin or protein Part of hemoglobin prevent the oxidation of it.



So heme molecule containing - iron (in the center) attach to porphyrin ring by 4 Nitrogens bonds  
 - two additional bonds one of each side of (Planar porphyrin ring)



Myoglobin: is globular heme-protein containing heme that attached to protein prosthetically.

Myoglobin  $\Rightarrow$  one subunit = tertiary globular protein

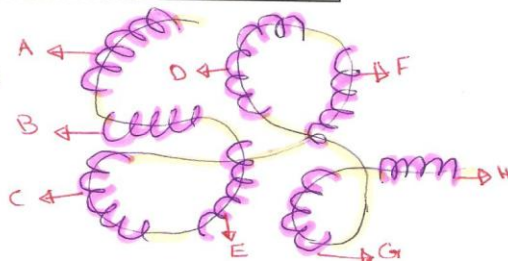
#### Structure and function:

- Myoglobin, a hemeprotein present in heart and skeletal muscle, functions both as a reservoir for oxygen, and as an oxygen carrier that increases the rate of transport of oxygen within the muscle cell.
- Myoglobin consists of a single polypeptide chain that is structurally similar to the individual subunit polypeptide chains of the hemoglobin molecule.
- Myoglobin is a compact molecule, with approximately 80 % of its polypeptide chain folded into 8 stretches of  $\alpha$ -helix. These  $\alpha$ -helical regions, labeled A to H

$\Rightarrow$  most of Myoglobin  $\alpha$ -helical structure

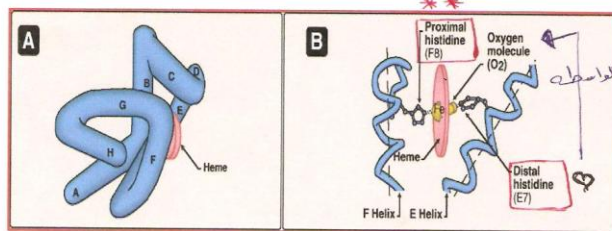
- it's one polypeptide chain folded into 8 stretches of  $\alpha$  helix (80%)

this  $\alpha$ -helical regions labeled A to H (8)



Amino acids  $\left\{ \begin{array}{l} \text{Polar} \rightarrow \text{surface of molecule} \\ \text{non polar} \rightarrow \text{interior (anterior of myoglobin)} \end{array} \right.$   
 ولكن ملاحظة استثناء  $\Rightarrow$  except for histidine (inside the crevice) but they are polar

- The interior of the myoglobin molecule is composed almost entirely of nonpolar amino acids.
- In contrast, polar amino acids are located almost exclusively on the surface of the molecule.
- The heme group of myoglobin is located in a crevice in the molecule between helix E and helix F, which is lined with nonpolar amino acids. Notable exceptions are two histidine residues.



A. Model of myoglobin showing helices A to H.  
 B. Schematic diagram of the oxygen-binding site of myoglobin

remember :: heme is complex of protoporphyrin 9 and ferrous iron ( $\text{Fe}_2^+$ )

$\text{Fe}_2^+$  have 2 additional bonds (one on each side of planar porphyrine ring) attach to histidine

\*\* one bond which is proximal histidine F8 (in F helix)  $\Leftarrow$

the other bond which is Distal histidine (E7)

(O<sub>2</sub>)  $\Leftarrow$   $\text{Fe}_2^+$

For You:  $F8 \rightarrow$  الكبيرين  $\rightarrow$  directly  $=)$   
 $E7 \rightarrow$  الصغيرين  $\rightarrow$  indirectly

• One, the proximal histidine (F8), binds directly to the iron of heme.

• The second, or distal histidine (E7), does not directly interact with the heme group, but helps stabilize the binding of oxygen to the ferrous iron.

الوظيفة  
 تثبيت وجود الـ  $O_2$  في الـ  $Fe^{2+}$   
 وبالتالي تقليل تساهي  $CO$  على مكان  $O_2$

random  
note

$\Rightarrow$  • The protein, or globin, portion of myoglobin prevents the oxidation of iron of heme.

**Hemoglobin**  $\Rightarrow$  more than one  
 = Quaternary

sub unit = 4 subunit  
 ليس لاقدمع من 4 subunit  
 ولكن أكثر من  
 one sub-unit

#### Structure and function:

- Hemoglobin is found exclusively in red blood cells, where its main function is to transport oxygen from the lungs to the capillaries of the tissues.
- Hemoglobin A, the major hemoglobin in adults, is composed of four polypeptide chains - two  $\alpha$  chains and two  $\beta$  chains - held together by noncovalent interactions.
- Each subunit has stretches of  $\alpha$ -helical structure, and a heme-binding pocket similar to that described for myoglobin.

Physiology

- However, the tetrameric hemoglobin molecule is structurally and functionally more complex than myoglobin.
- For example, hemoglobin can transport  $H^+$  and  $CO_2$  from the tissues to the lungs, and can carry four molecules of  $O_2$  from the lungs to the cells of the body.
- Furthermore, the oxygen-binding properties of hemoglobin are regulated by interaction with allosteric effectors.

هذه عبارة عن مواد تتفاعل مع  $O_2$  و  $H^+$  و  $CO_2$  في الدم

2,3 BPG,  $CO_2$

Form	Chain composition	Fraction of total hemoglobin
HbA <sup>adult</sup>	$\alpha_2\beta_2$	90%
HbF <sup>fetal</sup>	$\alpha_2\gamma_2$	<2% in adult
HbA <sub>2</sub>	$\alpha_2\delta_2$	2-5%
HbA <sub>1c</sub>	$\alpha_2\beta_2$ -glucose	3-6% ← normal

Normal adult human hemoglobins.

[Note: The  $\alpha$ -chains in these hemo-globins are identical]

HbA<sub>1c</sub> could be used as a monitor for the control of the blood glucose level during the last 2 months for diabetic patients.

إذا قلّت للسكري السكر  
لا تترك carbohydrate  
والأكل وحبوب  
السكر في الدم

Hb A<sub>1c</sub>





$a_1 B_1 \rightarrow \text{dimer 1}$   
 $a_2 B_2 \rightarrow \text{dimer 2}$

أَوَىٰ لِمَنْ عِشِينَ ؟

Covalent bond      (1) الأتو  
non covalent bond      (2) والكن ال

Weak ionic and hydrogen bonds occur between  $\alpha\beta$  dimer pairs in the deoxygenated state.

Strong interactions, primarily hydrophobic, between  $\alpha$  and  $\beta$  chains form stable  $\alpha\beta$  dimers.

Some ionic and hydrogen bonds between  $\alpha\beta$  dimers are broken in the oxygenated state.

**T**, or taut, structure of deoxyhemoglobin

**R**, or relaxed, structure of oxyhemoglobin

So the diaphragm is not movable ليس متحرك work as one unit.

Schematic diagram showing structural changes resulting from oxygenation and deoxygenation of hemoglobin .

- The hemoglobin tetramer can be envisioned as being composed of two identical dimers,  $(\alpha\beta)_1$  and  $(\alpha\beta)_2$ , in which the numbers refer to dimers one and two.
- The two polypeptide chains within each dimer are held tightly together, primarily by hydrophobic interactions
- In contrast, the two dimers are able to move with respect to each other, being held together primarily by polar bonds.
- The weaker interactions between these mobile dimers result in the two dimers occupying different relative positions in deoxyhemoglobin as compared with oxyhemoglobin

Tense form:-

because low affinity of  $O_2$  to bind with Hb

right-shift of  $O_2$ -Hb curve good at tissue level.

- rupture form:-

because of  $O_2$  high affinity binding to Hb

Left-Shift of  $O_2$ -Hb curve good at lung level.

• **T form:** The deoxy form of hemoglobin is called the "T," or taut (tense) form.

diemer I diemer II

• In the T form, the two  $\alpha\beta$  dimers interact through a network of ionic bonds that constrain the movement of the polypeptide chains. The T form is the low-oxygen-affinity form of hemoglobin.

• **R form:** The binding of oxygen to hemoglobin causes the rupture of some of the ionic bonds between the  $\alpha\beta$  dimers.

1 ab1  
2 ab2

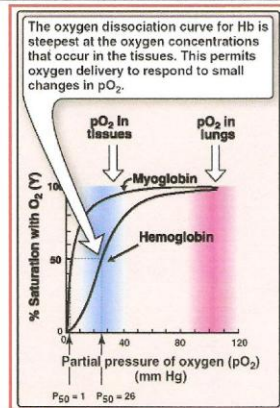
• This leads to a structure called the "R," or relaxed form, in which the polypeptide chains have more freedom of movement. The R form is the high-oxygen-affinity form of hemoglobin.

### Binding of oxygen to myoglobin and hemoglobin

- Myoglobin can bind only one molecule of oxygen ( $O_2$ ), because it contains only one heme group.
- In contrast, hemoglobin can bind four oxygen molecules—one at each of its four heme groups.
- The degree of saturation (Y) of these oxygen-binding sites on all myoglobin or hemoglobin molecules can vary between zero (all sites are empty) and 100% (all sites are full).

the curve of myoglobin-O<sub>2</sub> → hyperbolic curve  
 the O<sub>2</sub>-Hb curve → sigmoid curve

P<sub>O<sub>2</sub></sub> at tissue level (in venous end) = 40 mmHg  
 Y = 75 (low)



P<sub>O<sub>2</sub></sub> at lung level (arterial end)  
 P<sub>O<sub>2</sub></sub> = 104  
 Y = 97 (high)

Oxygen dissociation curves for myoglobin and hemoglobin

### Oxygen dissociation curve:

- A plot of Y measured at different partial pressures of oxygen (pO<sub>2</sub>) is called the oxygen dissociation curve.
- The curves for myoglobin and hemoglobin show important differences.
- This graph illustrates that myoglobin has a higher oxygen affinity at all pO<sub>2</sub> values than does hemoglobin