Biochemistry Team 434

## Kidney stones

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## What are kidney stones?

Renal calculi (kidney stones) are formed in <u>renal tubules</u>, ureter or bladder

Composed of metabolic products <u>present</u> in glomerular filtrate

These products are in <u>high</u> conc. Near or above maximum solubility.

solubility.



High conc. of metabolic products in glomerular filtrate

(when their conc. Increases against normal fluid volume)

Urinary stagnation

= solid state of urine

(blockage>>accumulation of urine will take place>>urine becomes concentrated>>stone formation) due to:

Low urinary volume (with normal renal function) due to restricted fluid intake (e.g. person doesn't drink water) Increased fluid loss from the body (dehydration)

Increased excretion of metabolic products forming stones (in this case your fluid volume is OK, but the problem is with increased metabolites) High plasma volume (high filtrate level>>more metabolites) Low tubular reabsorption from filtrate (metabolites remain>>increases their concentration)

due to: Obstruction of urinary flow (medications or diet-dependent factor.some diet will make your urine alkaline whilst others makes it acidic.this is also applied upon medications )



Deficiency of stoneforming inhibitors in urine

#### ex. Citrates

(normally present chemicals inhibiting stone formation>>deficiency in these substanses>>stone) Changes in urine pH due to: Bacterial infection (mostly to alkaline) Precipitation of salts at different pH

(some prefer alkaline medium & others prefer acidic one)

Hypocitraturia: a low amount of citrate in the urine, is an important risk factor for kidney stone formation.

Citrate, pyrophosphate, glycoproteins (these are all soluble si when ca binds to one of them it won't be precipitate)<u>inhibit</u> growth of calcium phosphate and calcium oxalate crystals

In type I renal tubular acidosis, hypocitraturia leads to renal stones







#### **Calcium salt stones** 80% of kidney stones contain calcium

The type of salt depends on Urine pH (as said before some salts are insoluble at certain pH while others are soluble at such pH) Availability of oxalate (the high you have oxalate the more the chance to get ca oxalate stone-type)

General appearance: White, hard, radio-opaque (dark=معتم) Calcium PO<sub>4</sub>: <u>staghorn (مرجانية الشكل</u>) in renal pelvis (large) Calcium oxalate: present in <u>ureter</u> (small)

Note that : oxalate is much important constituent than Ca because some times we would have Ca stones with normal levels of Ca so calcium level can't be always used as an indicator. oxalate is more important

#### cm [1111]1111]1111]1111]1111]1111

#### Treatment:

Treatment of primary causes such as infection, hypercalcemia, hyperoxaluria Oxalate-restricted diet (no oxalate in diet, but at the same time No need to lower the ca level) Increased fluid intake (to decrease the concentration of some solutes) Acidification of urine (by dietary changes) Calcium salt stones are formed in alkaline urine

(give nitrogen-containing substances such as: protein, milk, beans)



You don't have to memories the numbers

## **Uric acid stones**

About 8% of renal stones <u>contain</u> uric acid

May be associated with hyperuricemia (with or without gout)

Form in acidic urine General appearance:

Small, friable, yellowish May form staghorn Radiolucent

(transparent=شفاف) (plain xrays cannot detect) Visualized by ultrasound or i.v. pyelogram





Treatment: Purine-restricted diet Alkalinization of urine (by dietary changes) Increased fluid intake



## Mg ammonium PO<sub>4</sub> stones

(mostly staghorn appearance)

About 10% of all renal stones contain Mg amm. PO<sub>4</sub>

Also called struvite kidney stones

Associated with chronic urinary tract infection (most common cause)

Microorganisms (such as from Proteus genus) that metabolize urea into ammonia

Causing urine pH to become alkaline and stone formation

(most infections make the urine Alkaline)

Commonly associated with staghorn calculi

75% of staghorn stones are of struvite type

Treatment:

Treatment of infection

Urine acidification

Increased fluid intake

## **Cystine stones**

Cystine is an amino acid A rare type of kidney stone Due to homozygous cystinuria Form in acidic urine Soluble in alkaline urine Faint radio-opaque



#### Treatment:

Increased fluid intake Alkalinization of urine (by dietary changes) <u>Penicillamine</u> (binds to cysteine to form a compound more soluble than cystine)

# Laboratory investigations of kidney stones



### Summary :



Stones of calcium salts: -80% of patients with nephrolithiasis form calcium stone. - white,hard&radio-opaque. Ca-Oxalate:smaller, lodge in ureter. - Ca-Phosphate: staghorn, in renal pelvis (big). Causes: - Hypercalciuria. Hyperoxaluria. Treatment: -Treatment of primary condition (i.e. Infection, hypercalcemia, hyperoxaluria) - Reducing oxalates in diet. - Increase fluid intake. - Acidification of urine. Mg ammonium phosphate (struvite) stone: -~ 10% of all renal stones. - With chronic urinary tract infection (by urease splitting oraganisms as Proteus species). - Alkaline urine pH (>7.0). Treatment: - Aggressive prevention &treatment of the cause (urinary tract infection). - Urine acidification. - Increase fluid intake. - It may require complete stone removal (percutaneous

nephrolithotomy).

1-Calcium stones represent : A) 40% of kidney stones B) 65%

MCQs

2-which one of the following is visualized by ultrasonography?

A) Ca stones

**C) 80%** 

- B) Uric acid stones
- C) Cystine stones

**3-Acidification of urine will treat :** 

- A) Uric stones
- B) Ca stones
- C) Cystine stones
- D) a+c

4-Which one of the following is not true about Ca oxalate?

- A) Small
- **B)** Found in ureter
- C) Staghorn



1-What is the other name of Mg ammonium phosphate stone?

2-What is the most imp inhibitor of stone formation?

•••••

MCQs: 1-c 2-b 3-b 4-c

SAQs:1) Struvite stone2) Citrates

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Done by biochemistry team.

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