

# Pharmacokinetics II: Bioavailability and Distribution

Prof. Hanan Hagar Pharmacology Department

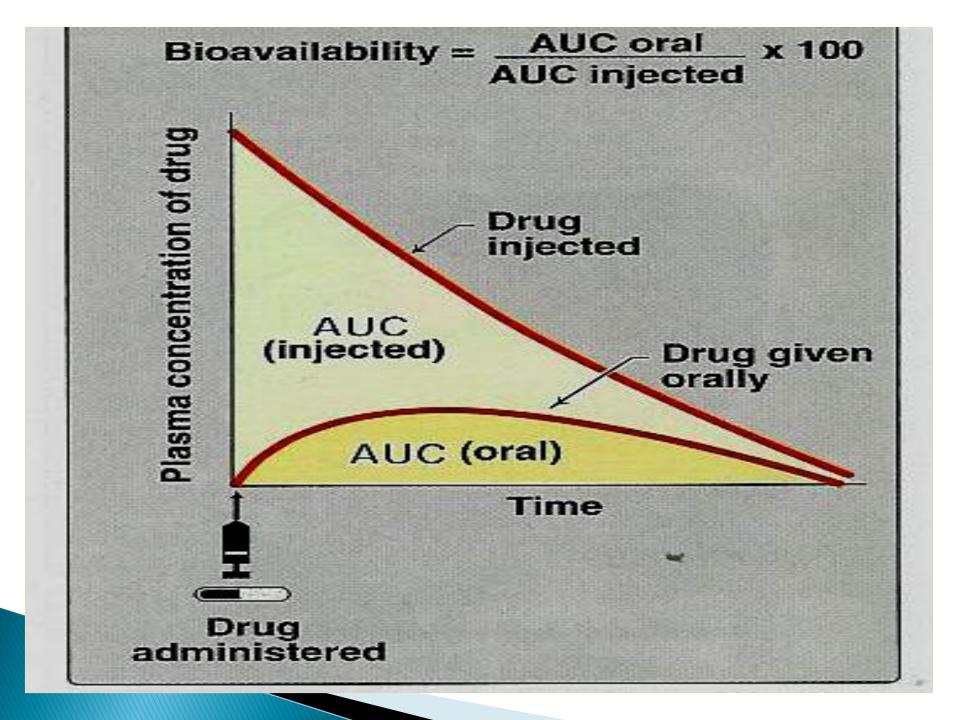
#### By the end of the lectures, students should be able to define the following:

- Major body fluid compartments
- **Concept of compartments.**
- **▶** Apparent volume of distribution (vd).
- Plasma protein binding.
- **▶** Tissue binding.

# Bioavailability

Is the amount of <u>unchanged</u> drug that enters systemic circulation after administration and becomes available to produce pharmacological actions

Bioavailability (F) = AUC (oral) X 100 AUC (I.V.)



# **Bioavailability**

- I.V. provides 100% bioavailability i.e. F= 1.
- Subcutaneous, intramuscular, oral, rectal, and other extra vascular routes of administration require that the drug be absorbed first, which can reduce bioavailability.

# Absolute bioavailability

The bioavailability of a drug after administration by any route is compared to its intravenous standard formulation.

## Relative bioavailability

- is determined when two products are compared to each other, not to an intravenous standard.
- This is commonly calculated in the drug industry to determine that the generic formulation is bioequivalent to another formulation.
- e.g Tylenol (paracetamol 500 mg) compared to panadol (paracetamol 500 mg).

# Relative bioavailability

is important to get an idea of how <u>different</u> formulations or <u>routes of administration</u> differ in their bioavailability.

dosage adjustment is required when changing formulations or routes of administration.

#### Bioequivalence

Two pharmaceutically products are bioequivalent when the rate and extent of bioavailability of active ingredients in two products are the same.

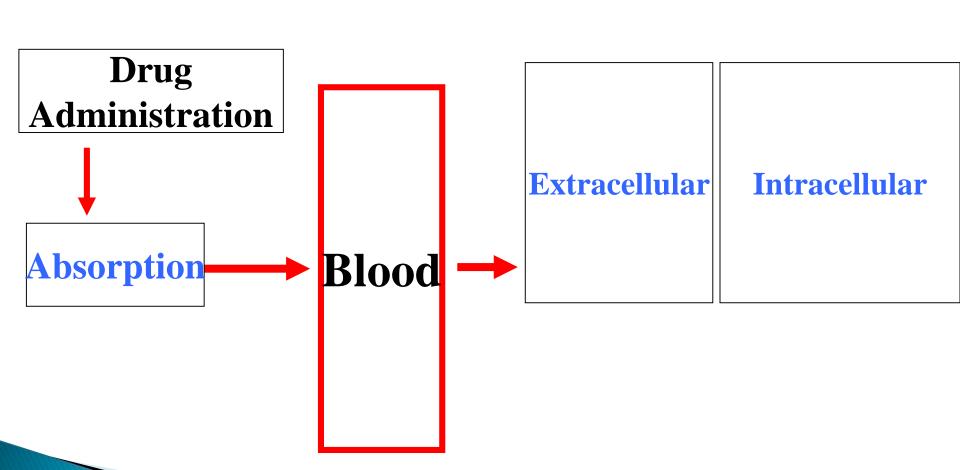
#### **Factors affecting bioavailability:**

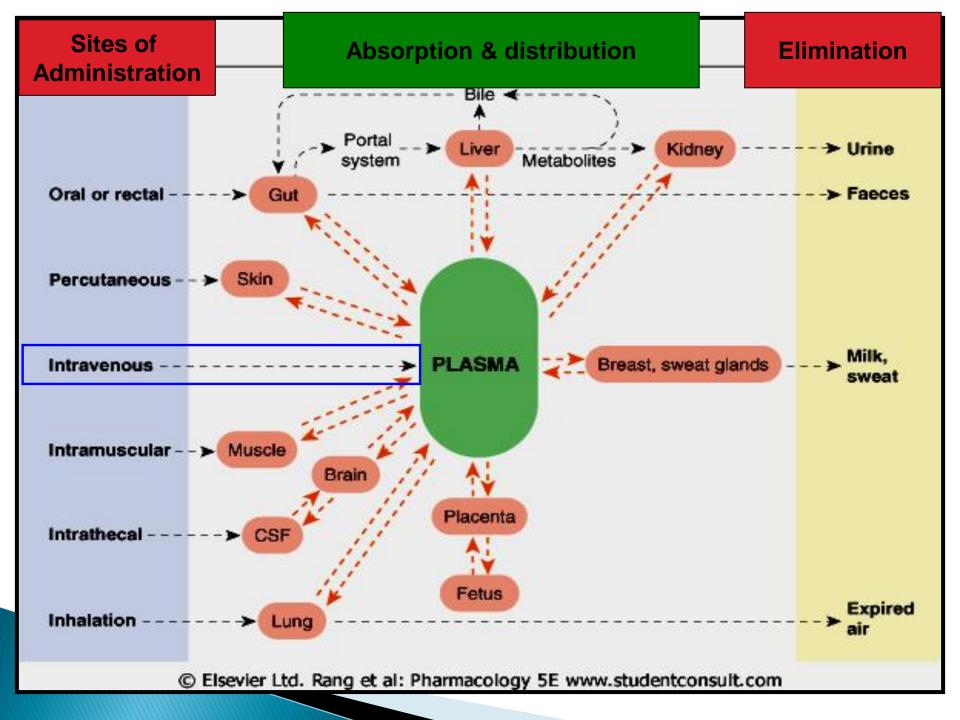
- are the same factors controlling drug absorption
- First pass effect

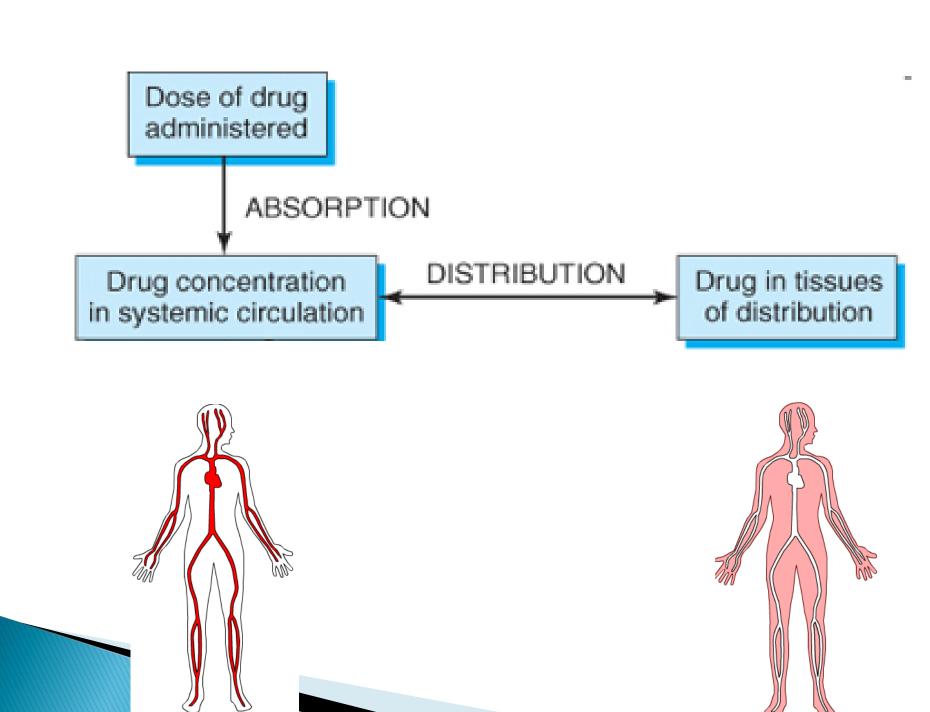
# Distribution

# Distribution

Is the process by which drugs leave blood circulation and enters the interstitium and/or the cells of the tissues.







#### Apparent Volume of Distribution (Vd)

is the ratio of drug amount in the body (dose) to the concentration of drug in blood.

Why is Vd important?

- > To calculate loading dose
- Large Vd = means long duration of action

#### The major body fluid compartments are

#### Extracellular fluids (22%)

- Plasma (5 % of body weight = 4 liters ).
- Interstitial fluid (16% = 10 liters).

#### Intracellular fluids (35 %)

fluid present inside all cells in the body (28 L).

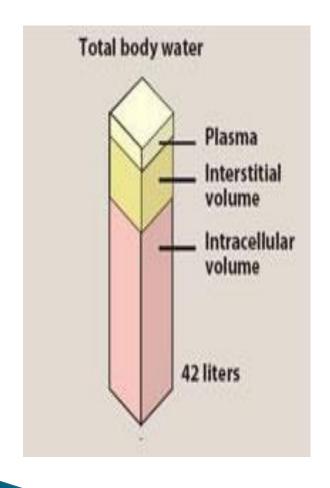
# Total body fluids (70% of body weight in 70-kg individual)

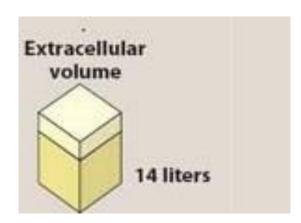
Total body Fluids (42 Liters) → Plasma volume (4 L)

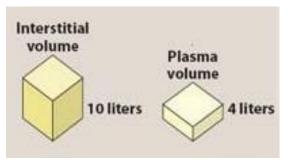
→ Interstitial volume (10 L)

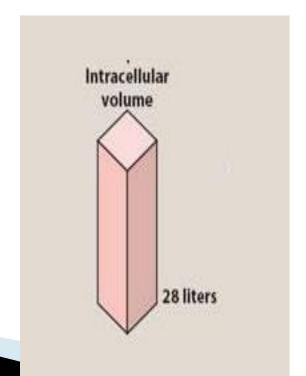
→ Intracellular volume (28 L)

### Volume of distribution









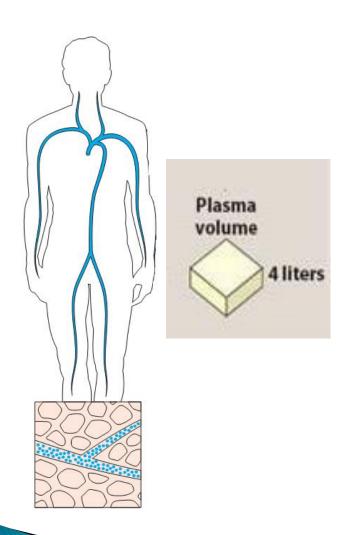
# Volumes of some compartments of the adult human body in relation to VD:

- ▶ Total body water 0.6 L/kg BW
- Intracellular water 0.4 L/kg BW
- Extracellular water 0.2 L/kg BW
- Plasma 0.04 L/kg BW

### Drugs may distribute through:

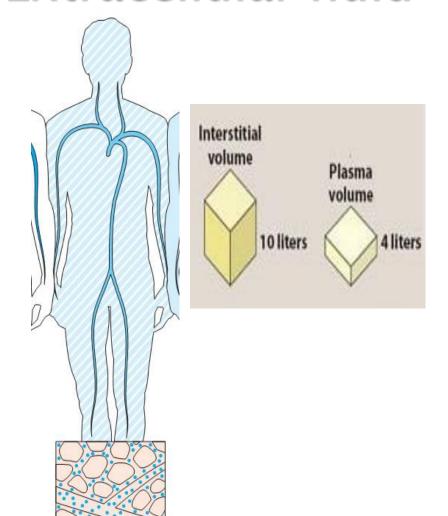
- One compartment
- Two compartments
- Multi-compartments

# Plasma compartment



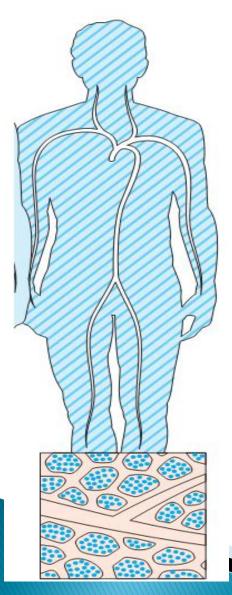
- > Vd: around 4 L.
- Very high molecular weight drugs, or drugs that bind to plasma proteins
- Can not moves across endotelial cells of capillaries
- Drugs are trapped in blood
- Example: heparin 4L

#### Extracellular fluid



- Distribute through extracellular fluids.
- Pass endothelium into interstitial fluids BUT can not cross cell membranes to intracellular fluids.
- Drugs that have a low molecular weight but are hydrophilic.
- Vd: between 4 and 14 L.
  - > E.g. atracurium 11 L

# Total body water (extracellular and intracellular)



- Diffusion to intracelullar fluid
- For lipid soluble drugs
- Vd equal to total body water.
  - Ethanol 38 L (34–41)
- Drug that binds strongly to tissues. Vd higher than total body water.
  - Digoxin:385 L

#### Volume of Distribution (Vd)

#### **Drugs with low Vd**

- distributed in extracellular compartments (plasma & interstitial fluid).
- Polar comp or lipid insoluble drugs. e.g. gentamycin, atracurium
- ▶ High MW e.g. heparin insulin.
- High plasma protein binding e.g. warfarin (anticoagulant).
  - **Po not cross BBB or placental barriers.**

#### Volume of Distribution (Vd)

### Drugs with high Vd

- Have higher concentrations in tissues than in plasma.
- Lipid soluble.
- Distributed intracellularly
- e.g. digoxin, phenytion, morphine

#### FACTORS AFFECTING DISTRIBUTION

- 1. Cardiac output and blood flow.
- 2. Physical and chemical properties of the drug.
  - Molecular weight
  - Pka.
  - Lipid solubility.
- 3. Capillary Permeability
- 4. Plasma protein binding
- 5. Tissue binding.

### **Blood flow to organs**

The greater the blood flow to tissues, the more distribution that occurs from plasma to interstitial fluids.

Drugs distribute more rapidly to <u>brain</u>, <u>liver and kidney</u> > more than skeletal muscles & fat.

### Physical and chemical properties of drug

- Most lipid soluble drugs (unionized, uncharged, nonpolar) cross biological membranes.
- Hydrophilic drugs (ionized, charged, polar) do not readily cross membranes but go through slit junctions in endothelial cells of capillaries.

# Capillary permeability

Endothelial cells of capillaries in tissues other than brain have wide slit junctions allowing easy movement, permeation and distribution.

Brain has tight junction Blood Brain Barrier (BBB).

#### **Blood brain barrier (BBB):**

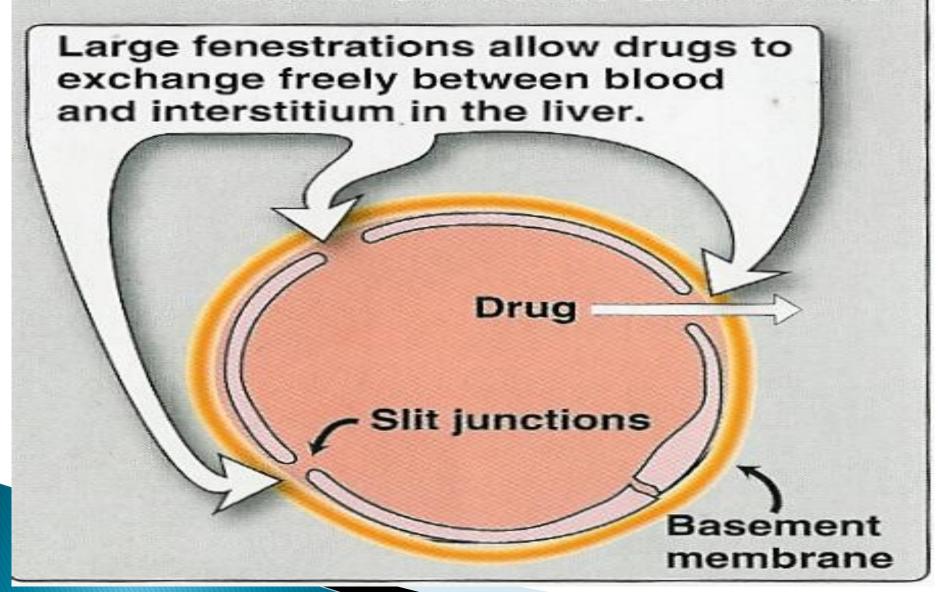
- Only lipid soluble drugs or actively transported drugs can cross BBB.
- Hydrophilic drugs (ionized or polar drugs) can not cross BBB.
- Inflammation as in <u>meningitis</u> increase permeability to hydrophilic drugs
- e.g. penicillin & gentamycin

#### Placental barrier

Lipid soluble drugs can cross placental barrier and enter the fetal blood.

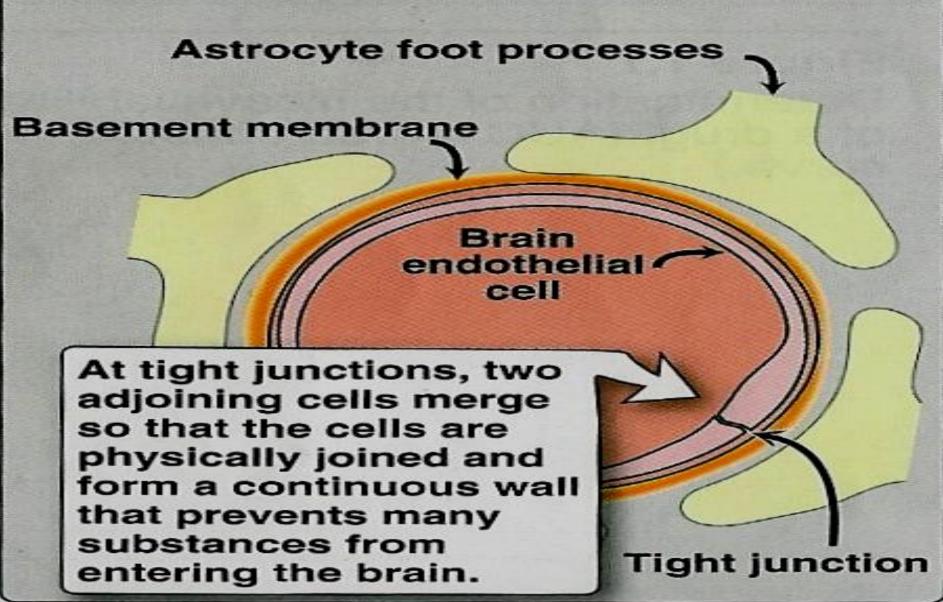


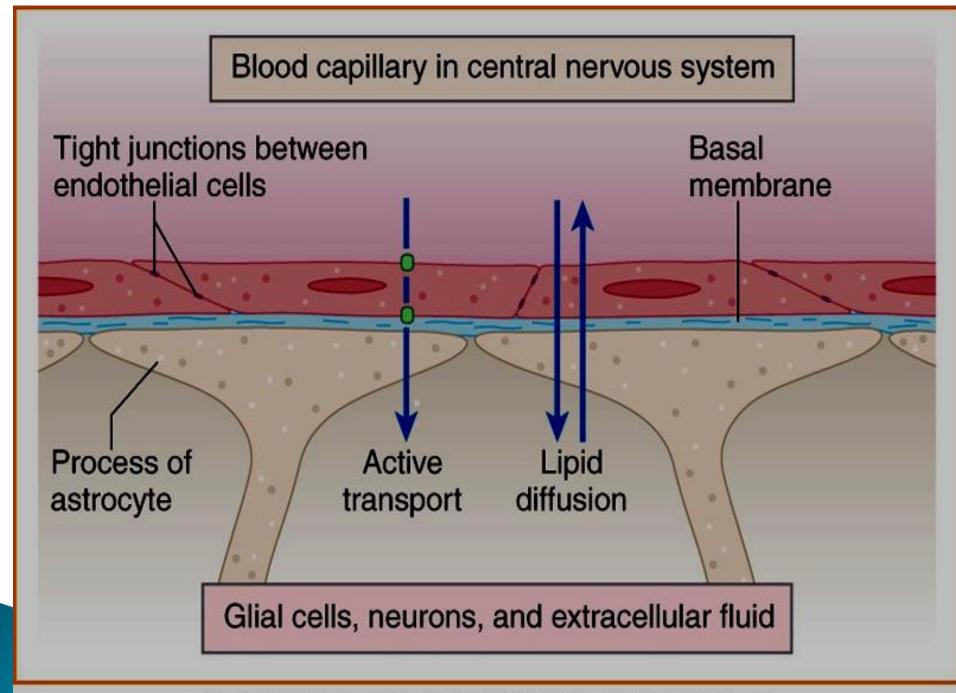
#### Structure of endothelial cells in the liver





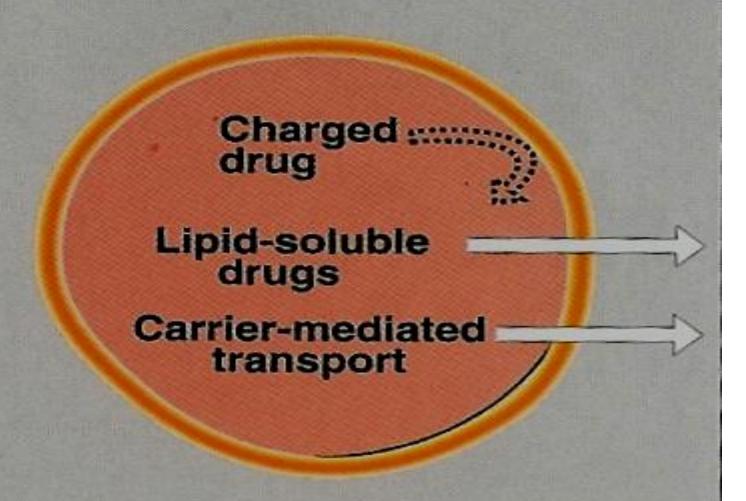
# Structure of a brain capillary







# Permeability of a brain capillary



### **Binding of Drugs**

- Plasma proteins binding.
- Tissue proteins binding.

#### Plasma protein binding:

- Extensive plasma protein binding will cause more drug to stay in the blood compartment.
- Therefore, drugs which bind strongly to plasma protein tend to have lower distribution (Vd).

#### **Plasma Proteins**

#### **Albumin**

Has affinity for acidic drugs as warfarin, phenytoin, aspirin

alpha 1 -acid glycoproteins

Has affinity for basic drugs (cationic) as diazepam, quinidine.

#### Plasma protein binding

- drugs which bind strongly to plasma protein tend to have lower distribution (Vd).
- In blood, drugs exist in two forms bound and unbound forms in equilibrium

Unbound drug (free) bound drug

# **Tissues Binding**

Drugs can bind to specific tissues and will have high volume of distribution (Vd)

Tetracycline bind to bone

#### Bound form of drug

- non diffusible form
- can not cross endothelial barrier
- can not combine with receptors
- inactive
- not available for metabolism & excretion
- has long duration of action (t  $\frac{1}{2}$ ).

# **Unbound form of drug**

- diffusible form
- cross endothelial barrier
- combine with receptors
- active
- available for metabolism& excretion

-has short duration of action (t  $\frac{1}{2}$ ).

#### **Characters & consequences of Binding**

- **Usually reversible.**
- determines volume of distribution (vd)
- Slows drug metabolism & excretion.
- ▶ Prolongs duration of drug action (t1/2).
- Result in clinically important drug interactions

#### Displacement

- ▶ Aspirin + Albumin-warfarin → →

Albumin-aspirin + free warfarin  $\longrightarrow$  bleeding.