

Body Fluids

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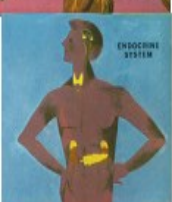
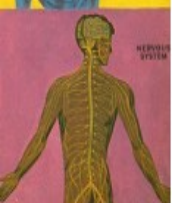
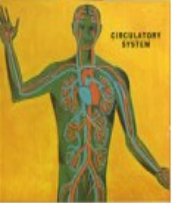
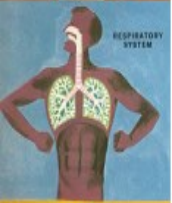
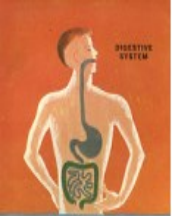
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Objectives

- State the water content of the body.
- Describe the physiologic factors that determine the body's water content.
- Discuss water balance in the body with regards to intake and output.
- Describe the physiologic mechanisms involved in water balance (the role of ADH hormone and thirst mechanism).
- Enumerate the different body fluid compartments and state their normal volume.
- Describe the composition of the different body fluid compartments.
- Define osmolarity and state the normal osmolarity of body fluids.
- Explain the effects of placing a cell in solutions with varying tonicity on the cell volume.
- Enumerate fluids used in clinical practice and state their tonicity compared to that of normal body fluids.
- Describe the effects of infusing IV fluids with different tonicities on the volume and osmolarity of the different body compartments.

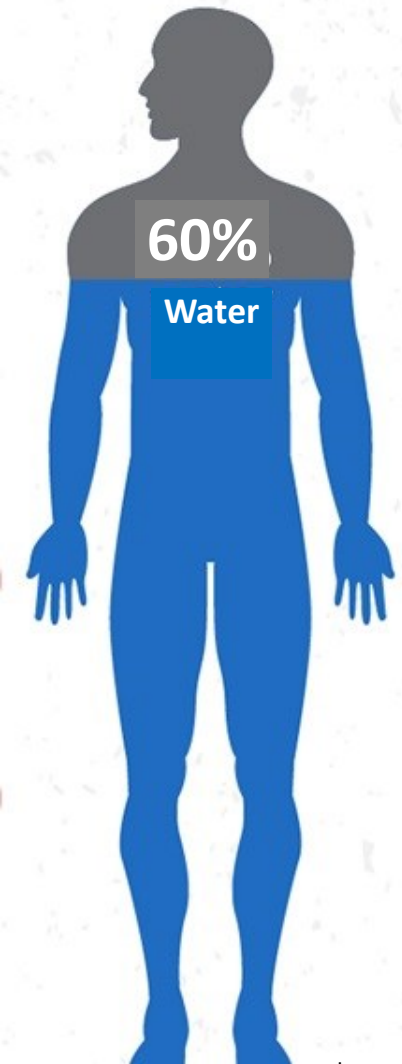
Study source for this lecture:

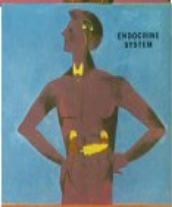
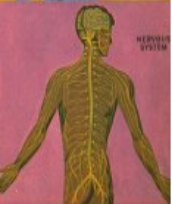
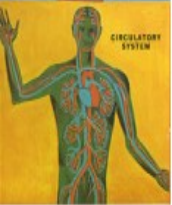
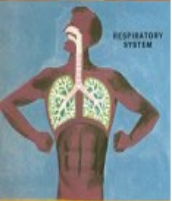
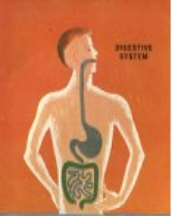
(Guyton & Hall Textbook of Medical Physiology, 13th ed, Chapter 25)



Body Fluid Content

- Total body fluids \approx 60% (50-70%) of total body weight (TBW).
- Body fluids = water + dissolved solutes.
 - Solutes
 - **Electrolytes**
(Na^+ , K^+ , Cl^- , HCO_3^- , Mg^{+2} , Ca^{+2})
 - **Non-electrolytes**
(glucose, urea, creatinine)
- Total body water \approx 60% of TBW.





**Calculate the total body water content
of a 30-year-old 70kg man?**

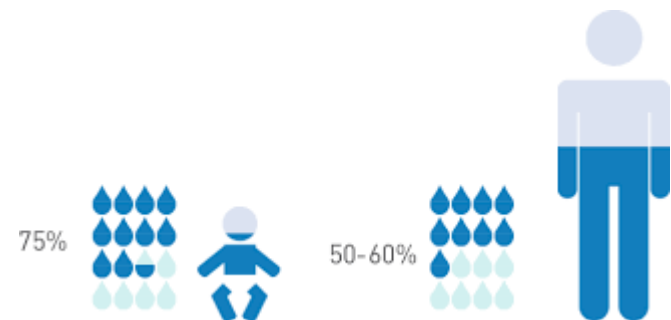
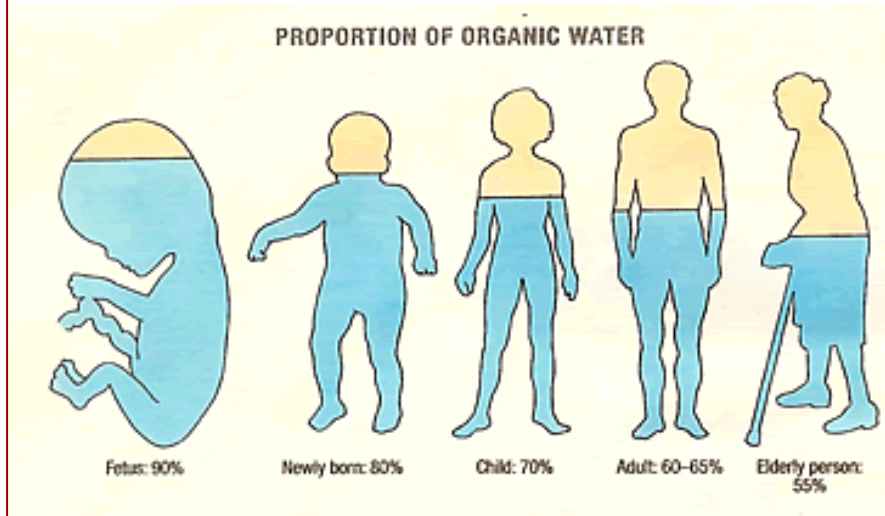
Physiologic factors affecting body water content

- *Is body fluid/water content similar between;*

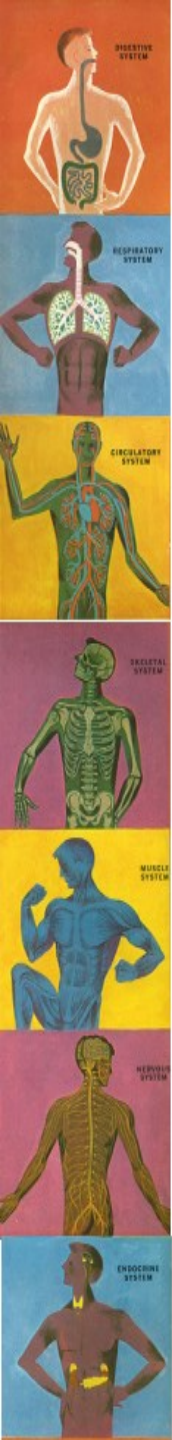
- Child & adult?

- Young & old?

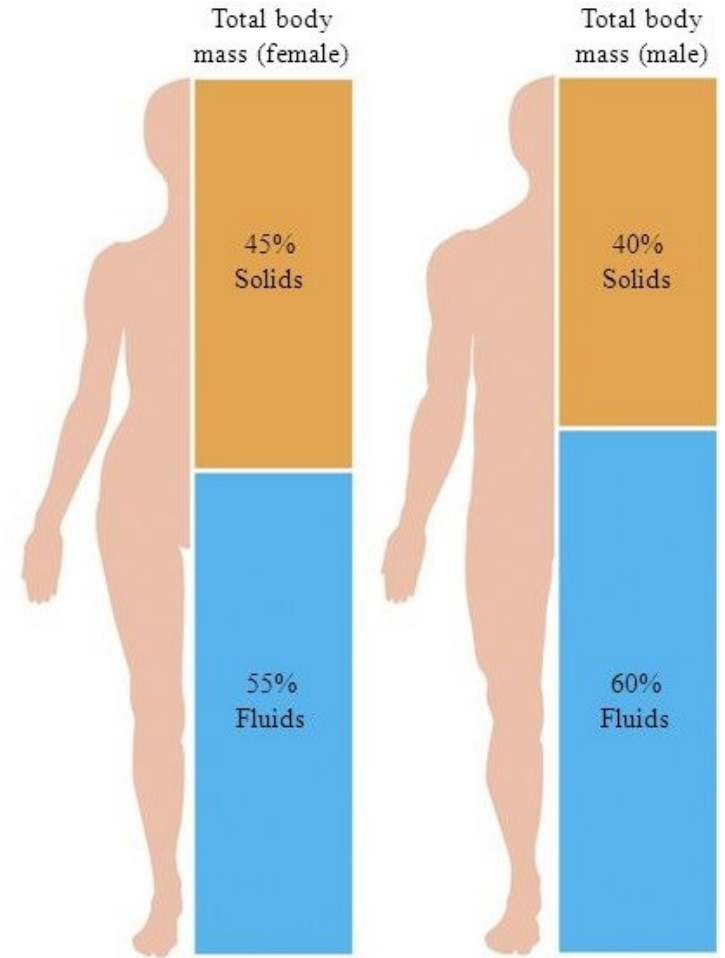
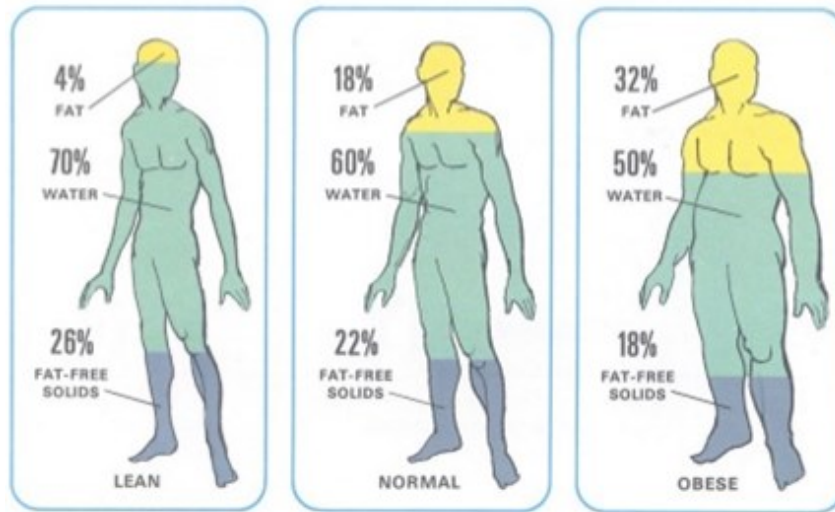
- Male & female?



Continue.. Physiologic Factors Affecting Body Fluid Content

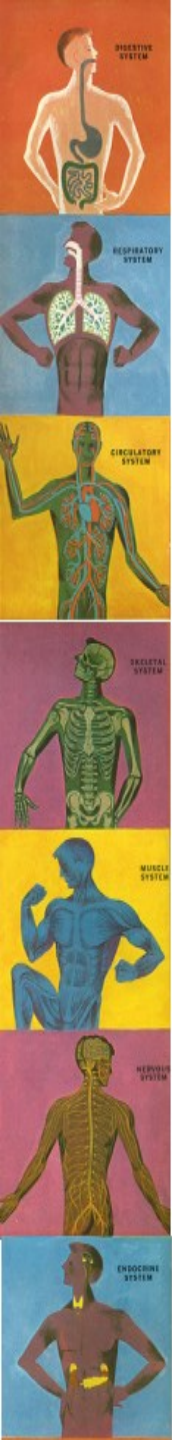
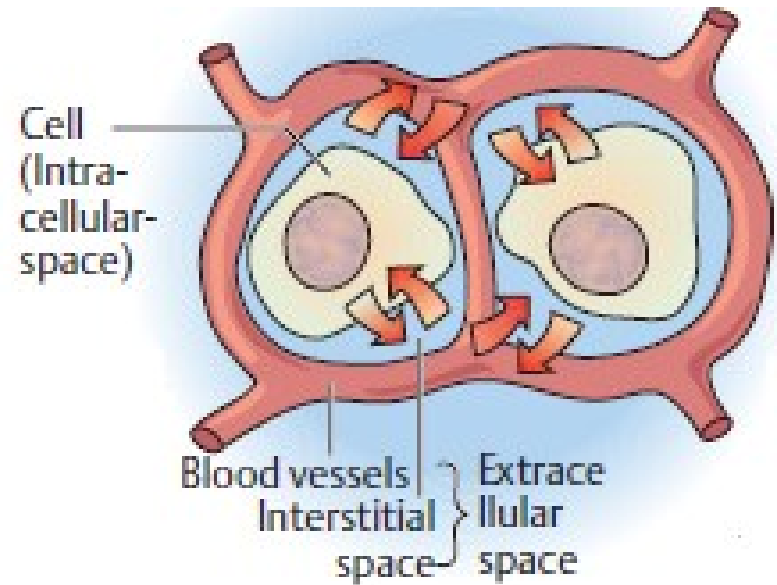


TOTAL BODY WATER VARIES IN EACH INDIVIDUAL BASED ON THE FAT CONTENT OF THE BODY

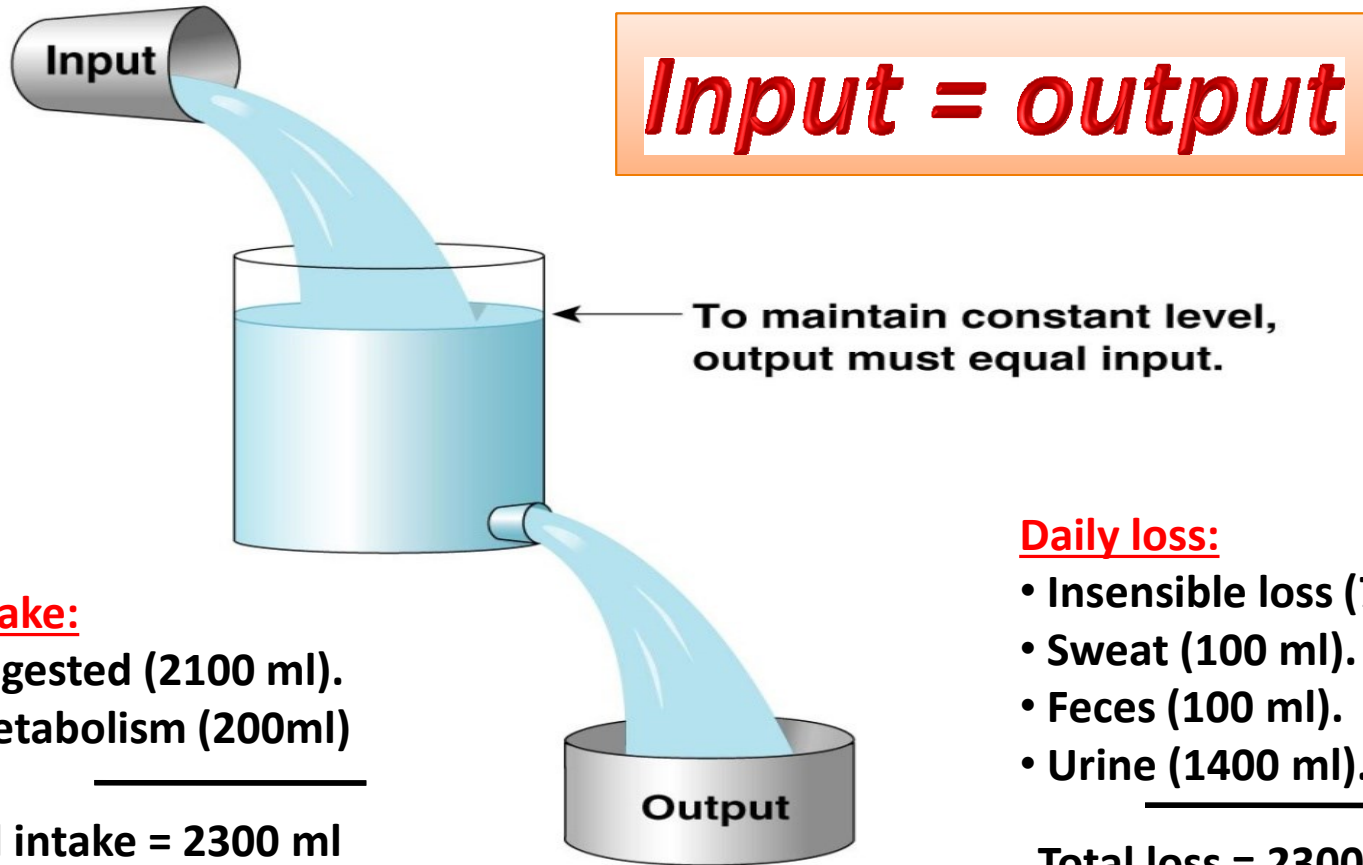


Water Balance

- Cells are largely made of water and are surrounded by a “sea” of water.
- Water is constantly being exchanged between the cells and surrounding environment.
- Keeping a constant volume of water is essential for homeostasis.



Maintenance of Fluid Balance



Water intake:

- Fluids ingested (2100 ml).
- From metabolism (200ml)

Total intake = 2300 ml

Daily loss:

- Insensible loss (700 ml).
- Sweat (100 ml).
- Feces (100 ml).
- Urine (1400 ml).

Total loss = 2300 ml

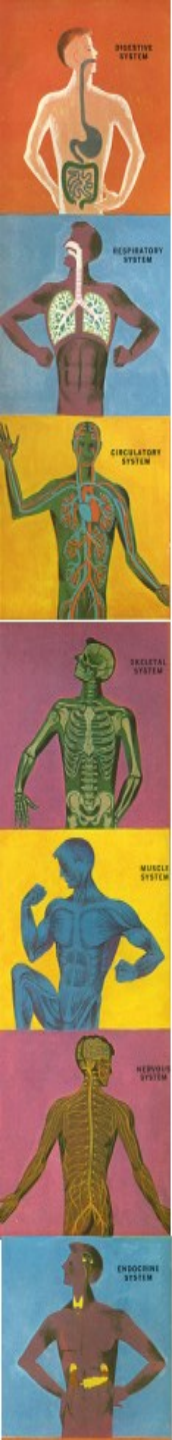
Disturbances in Fluid Balance

Water/fluid intake

- Intake is variable.
- Depends on:
 - Climate.
 - Habits.
 - Physical activity.

Water/fluid output

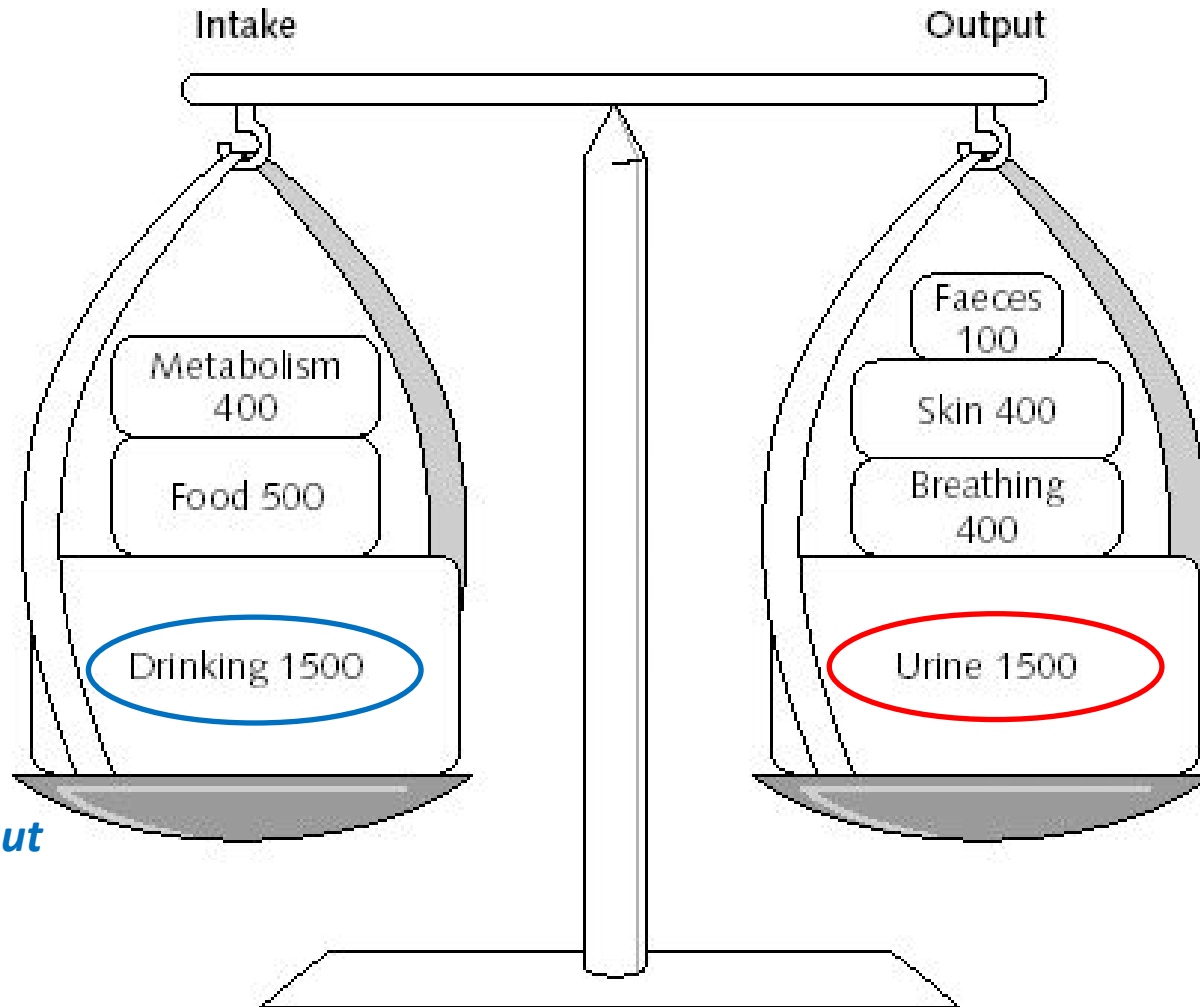
- Burns
- Exercise
- Hot weather.
- Diarrhea
- Vomiting
- Blood loss
- Diabetes



Regulation of Fluid Balance

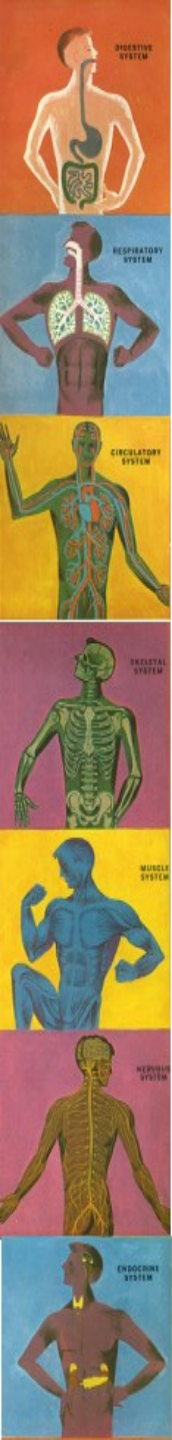
Modify intake → thirst

Modify output → Kidney excretion



Major input

Major output



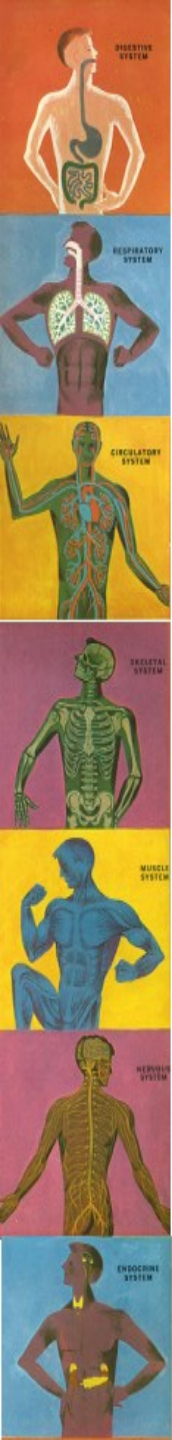
Regulation of Fluid Balance

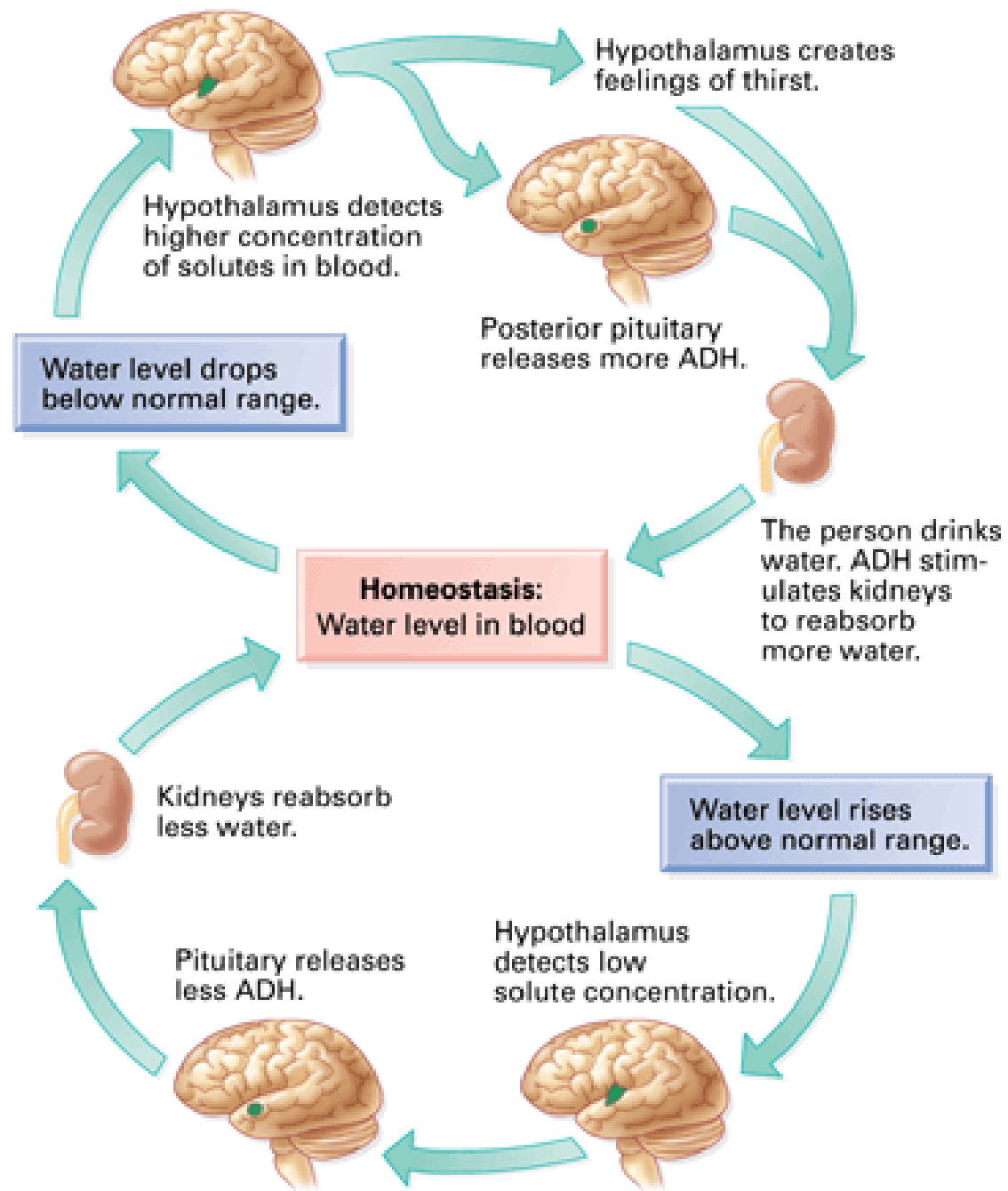
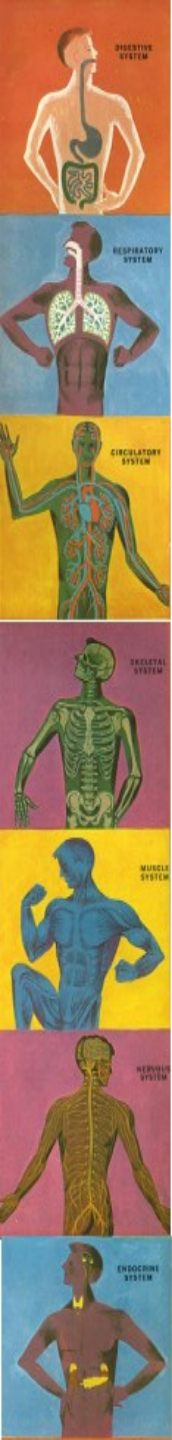
Water deficit

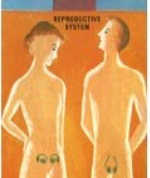
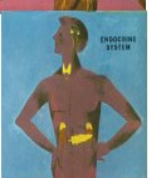
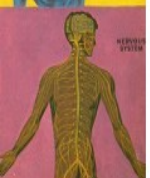
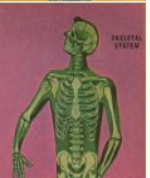
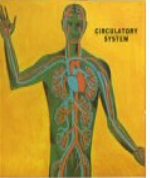
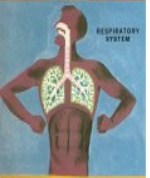
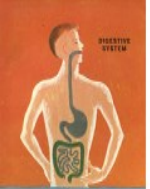
- Input < output.
- Leads to:
 - Hypovolemia
 - Dehydration
- Physiologic regulation:
 1. **Activates hypothalamic thirst centre → ↑ fluid intake**
- The hypothalamic thirst centre is activated by:
 - ↓ ECF volume (10-15%)
 - ↑ ECF osmolarity (1-2%)
- 2. **↑ ADH secretion by posterior pituitary → ↑ water reabsorption by the kidney.**

Water excess

- Input > output
- Leads to:
 - Hypervolemia.
 - Edema.
- Physiologic regulation:
 1. **↓ ADH secretion → ↓ water reabsorption → ↑ water excretion by kidney.**
 2. **Decrease thirst**





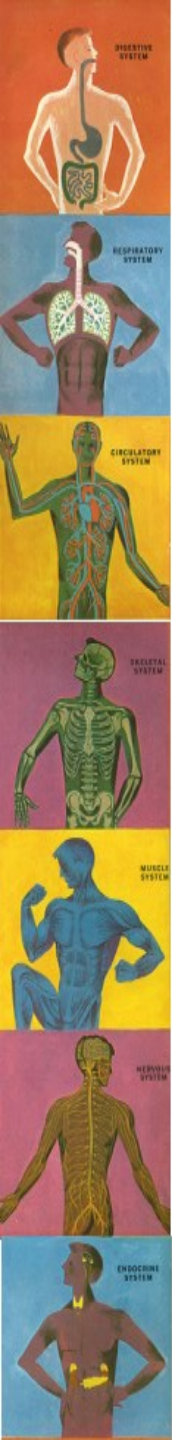
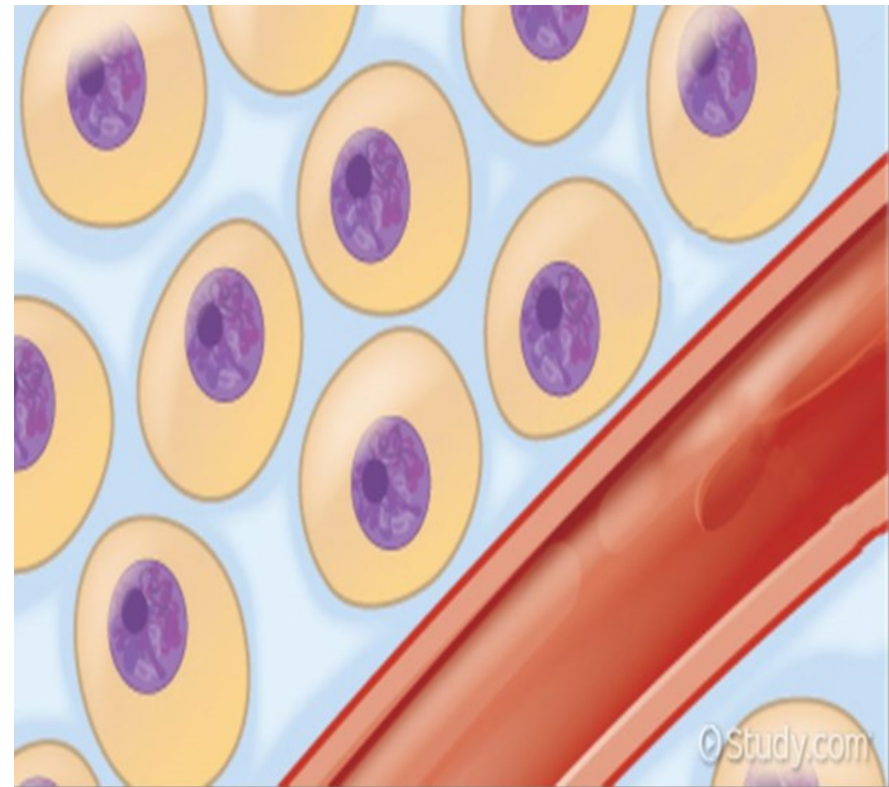
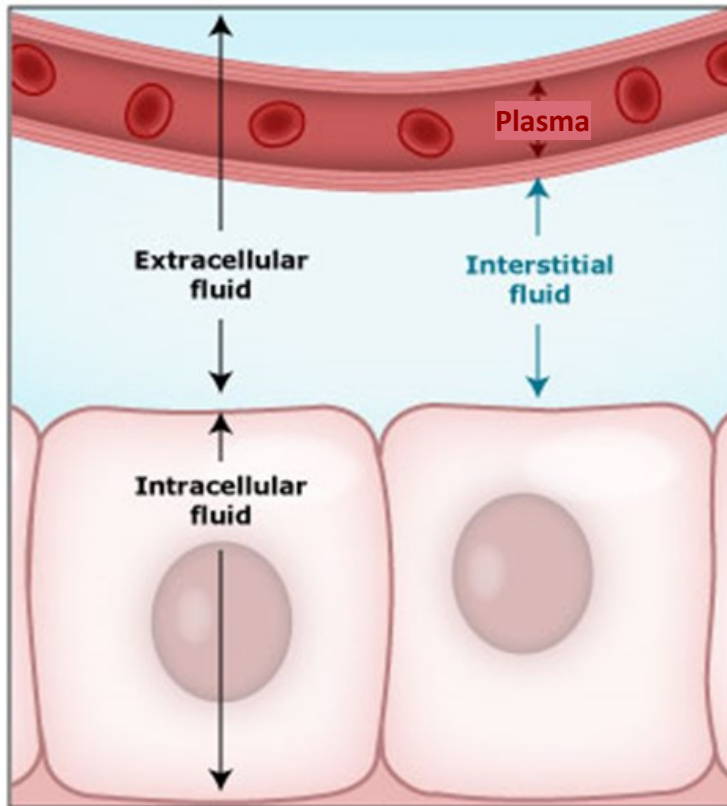


DISTRIBUTION OF BODY FLUID CONTENT

How is this water/fluid distributed in the body?

Where is this 60% of fluid/water located in the body?

Cellular Fluids



Body Fluid Compartments

Total body water = 60% TBW

* $\frac{2}{3}$ (40% of TBW)
Intracellular

$\frac{1}{3}$ (20% of TBW)
Extracellular

$\frac{3}{4}$ (15%)
Interstitial Fluid

$\frac{1}{4}$ (5%)
Plasma

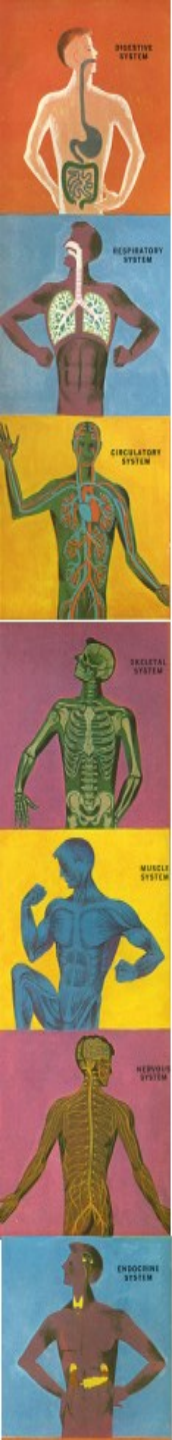
Transcellular fluid (specialised type of ECF e.g. synovial, CSF, ocular, pericardial, pleural, peritoneal fluids)= 1-2L

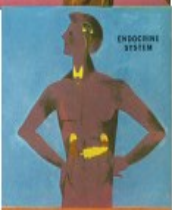
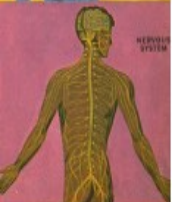
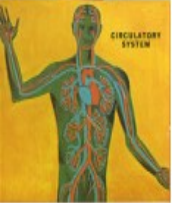
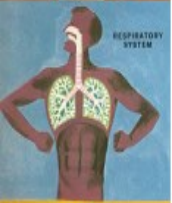
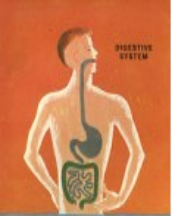
N. B. the fractions refer to its relation with the preceding compartment. While the percentage refers to its relation to Total Body Weight

Let's go back to our 30-year-old man

Calculate the total body water content of a 40-year-old 70kg man?

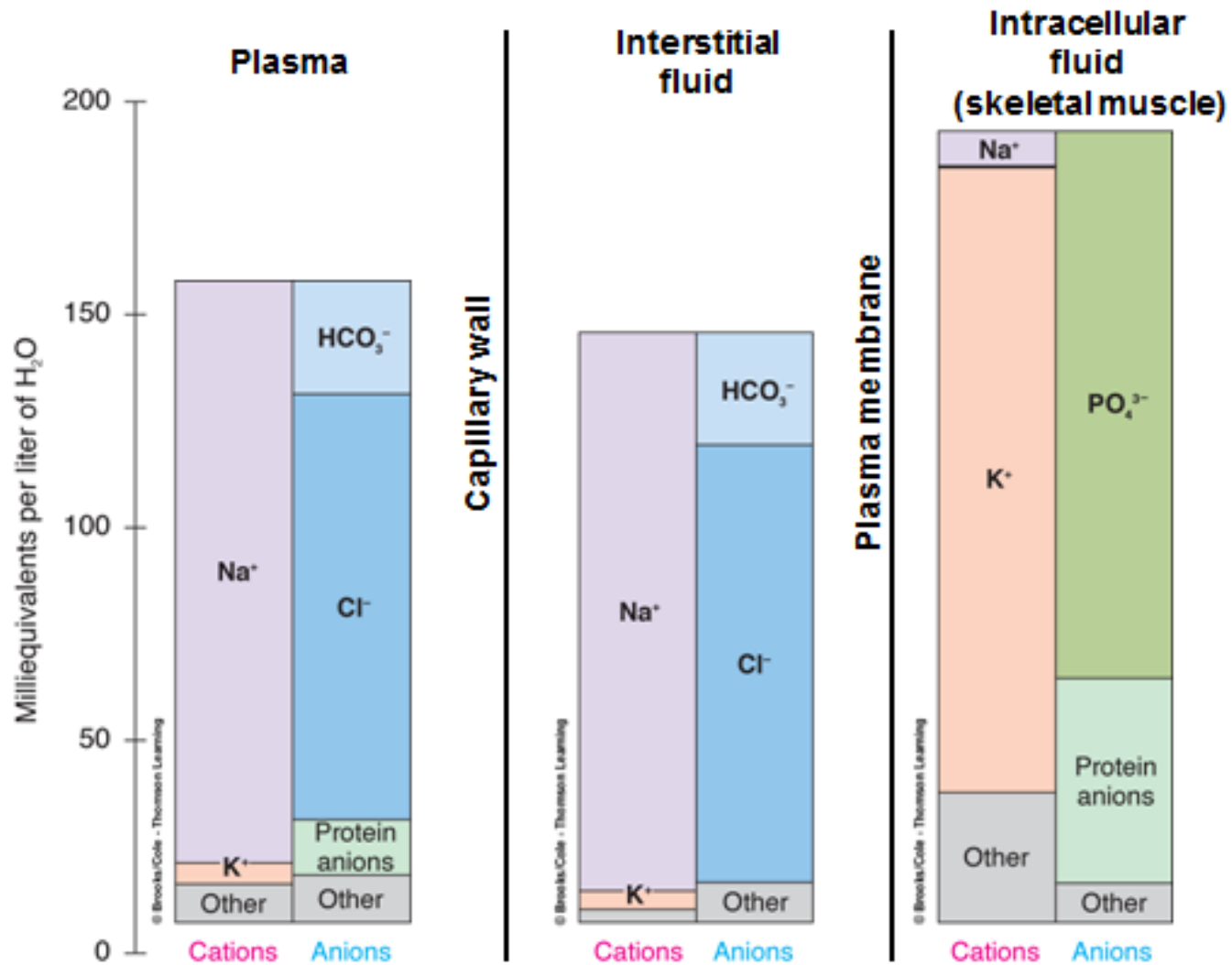
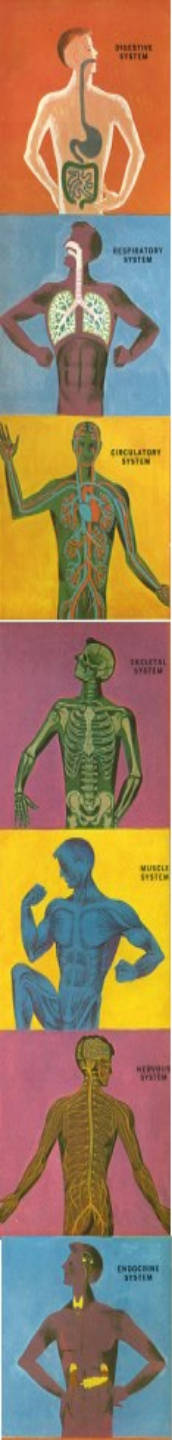
- TBW = 42 litres
- ***How many litres lie intracellularly?***
 - $42 \times \frac{2}{3} = 28\text{L}$ **OR** $70 \times \frac{40}{100} = 28\text{L}$.
- ***How many litres lie extracellularly?***
 - $42 \times \frac{1}{3} = 14\text{L}$ **OR** $70 \times \frac{20}{100} = 14\text{L}$.
- ***How many litres constitute the interstitial fluid?***
- ***How many litres are plasma?***





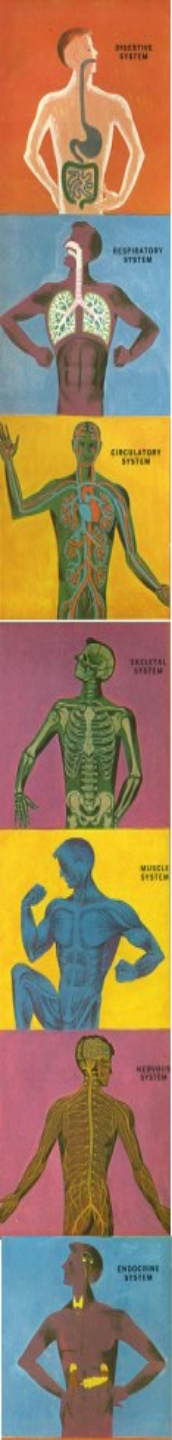
COMPOSITION OF BODY FLUIDS

Does the composition of the different body fluid compartments differ?



[Figure 15-2; Sherwood]

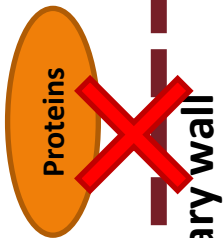
Constituents of the Different Body Fluid Compartments



Extracellular compartment

Plasma

Interstitial fluid



↑↑ Na+
↑↑ Cl-, HCO3-
↓↓ K+

Cations = anions

ECFs are constantly mixing and have the same composition except for proteins.

Intracellular compartment

↑↑ K+
↑↑ Phosphate (PO4-3)
↑ Proteins

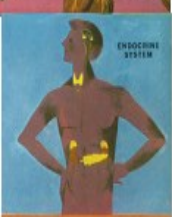
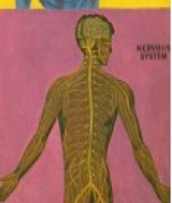
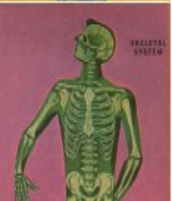
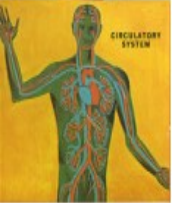
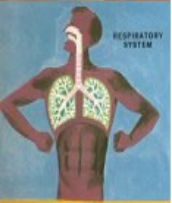
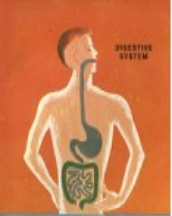
↓↓ Na+
↓↓ Cl-

Cations = anions

Osmolarity is the same ≈300 mOsm/L

What is Osmolarity?

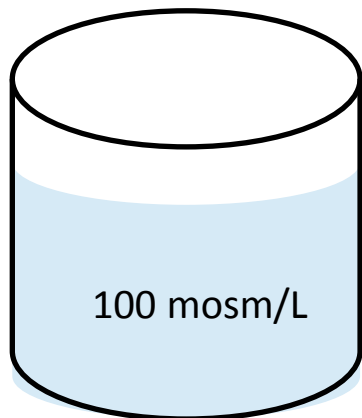
- A unit of concentration.
- It measures how concentrated or diluted a solution is.
- Defined as = the number of osmoles of a solute per litre of solution.
- The normal osmolarity of body fluids is ≈ 300 mOsm/L



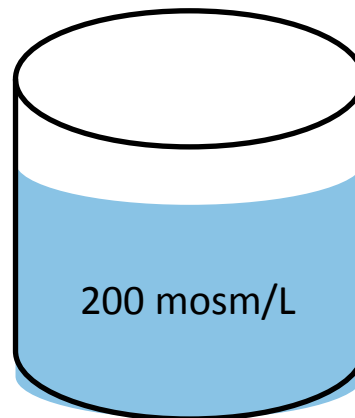
What is Tonicity?

- **Osmolarity** describes the concentration of one solution.
- **Tonicity** is used to compare between the osmolarities of two or more solutions separated by a semi-permeable membrane.

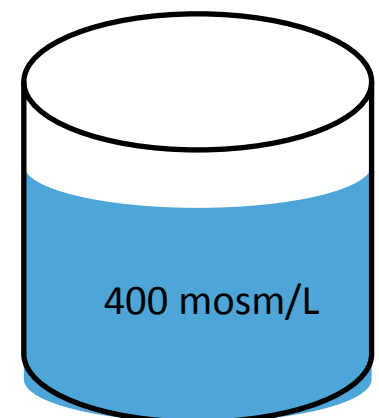
Which solution is hypertonic to B?



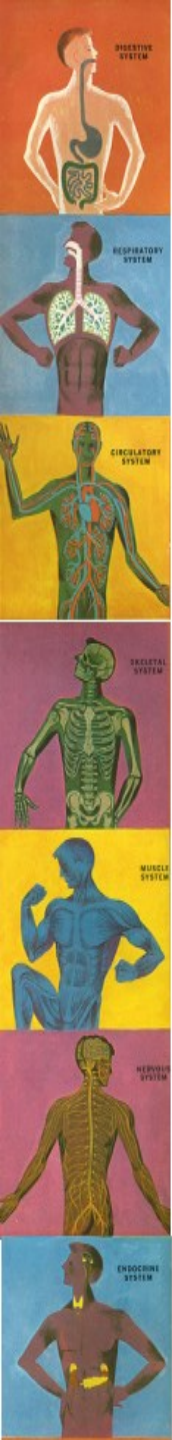
A



B

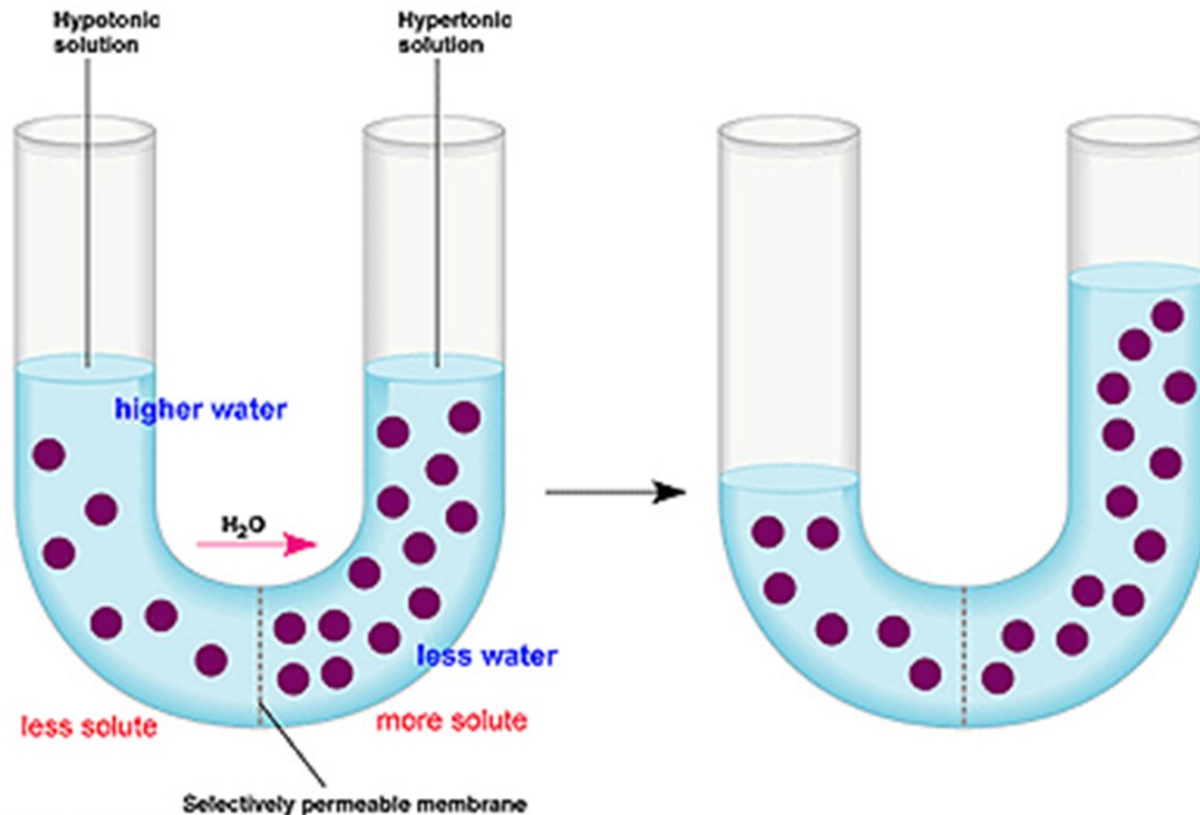


C



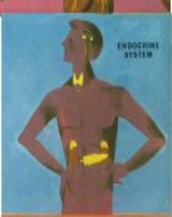
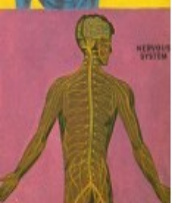
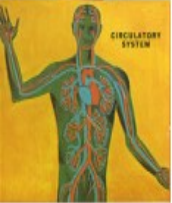
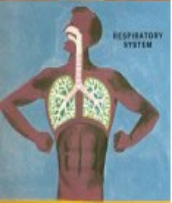
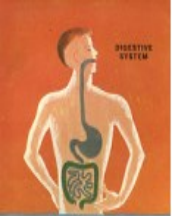
Why is it important?

The difference in tonicity between two different solutions separated by a semi-permeable membrane determines the movement of water.



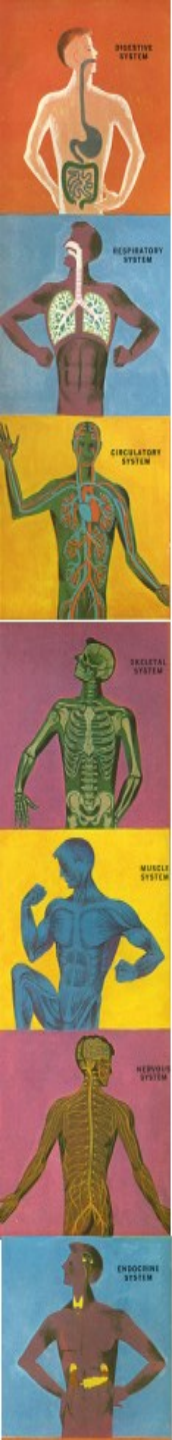
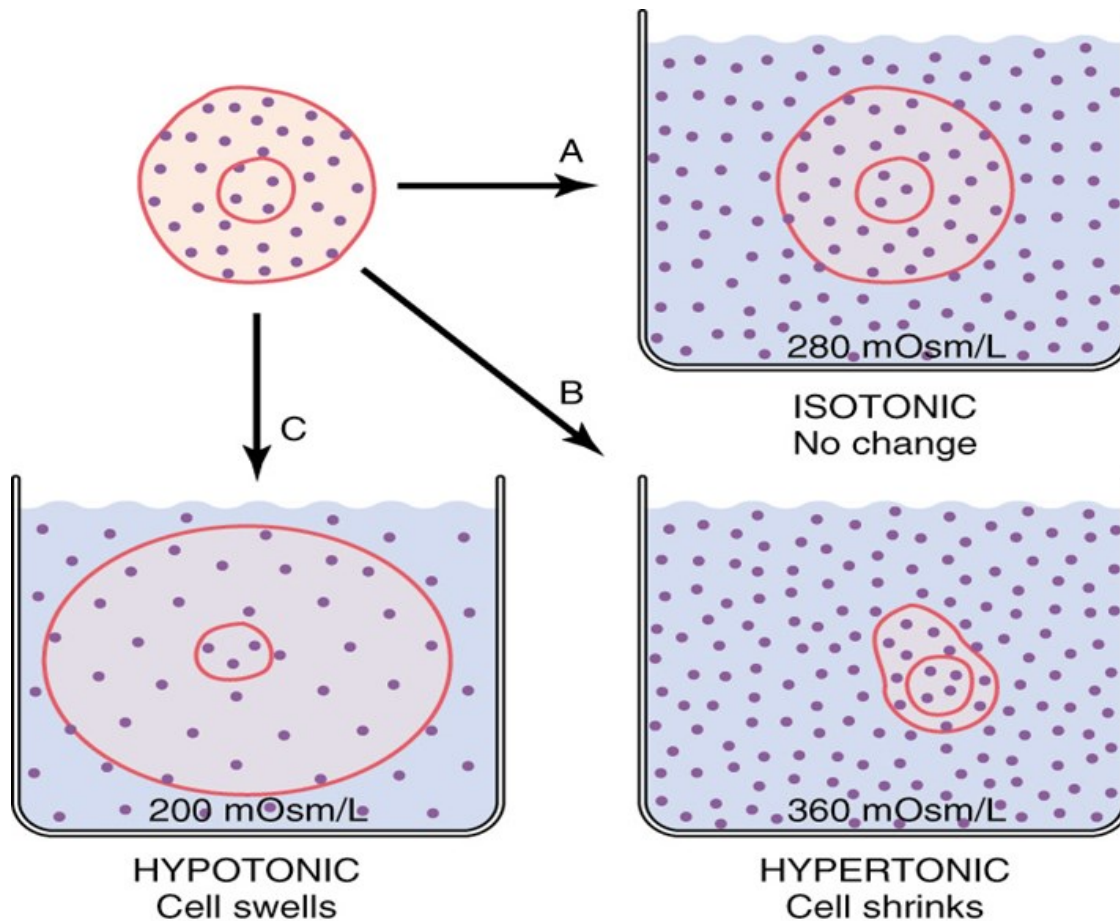
Osmosis

- It is the movement of water across a semi-permeable membrane from a region of low solute concentration to a region of high solute concentration.



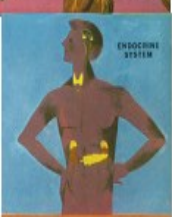
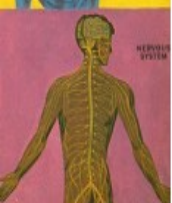
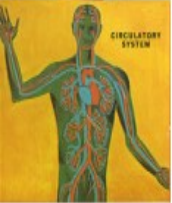
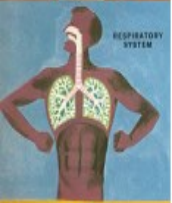
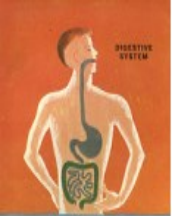
How does it apply to physiology?

The cell membrane is a semi-permeable membrane
(Allows only water to move freely across it)

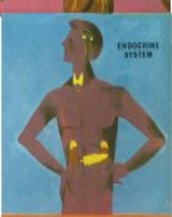
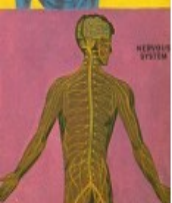
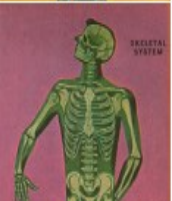
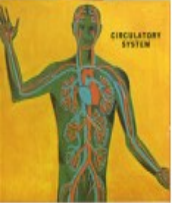
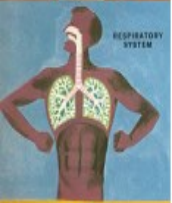
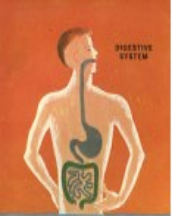
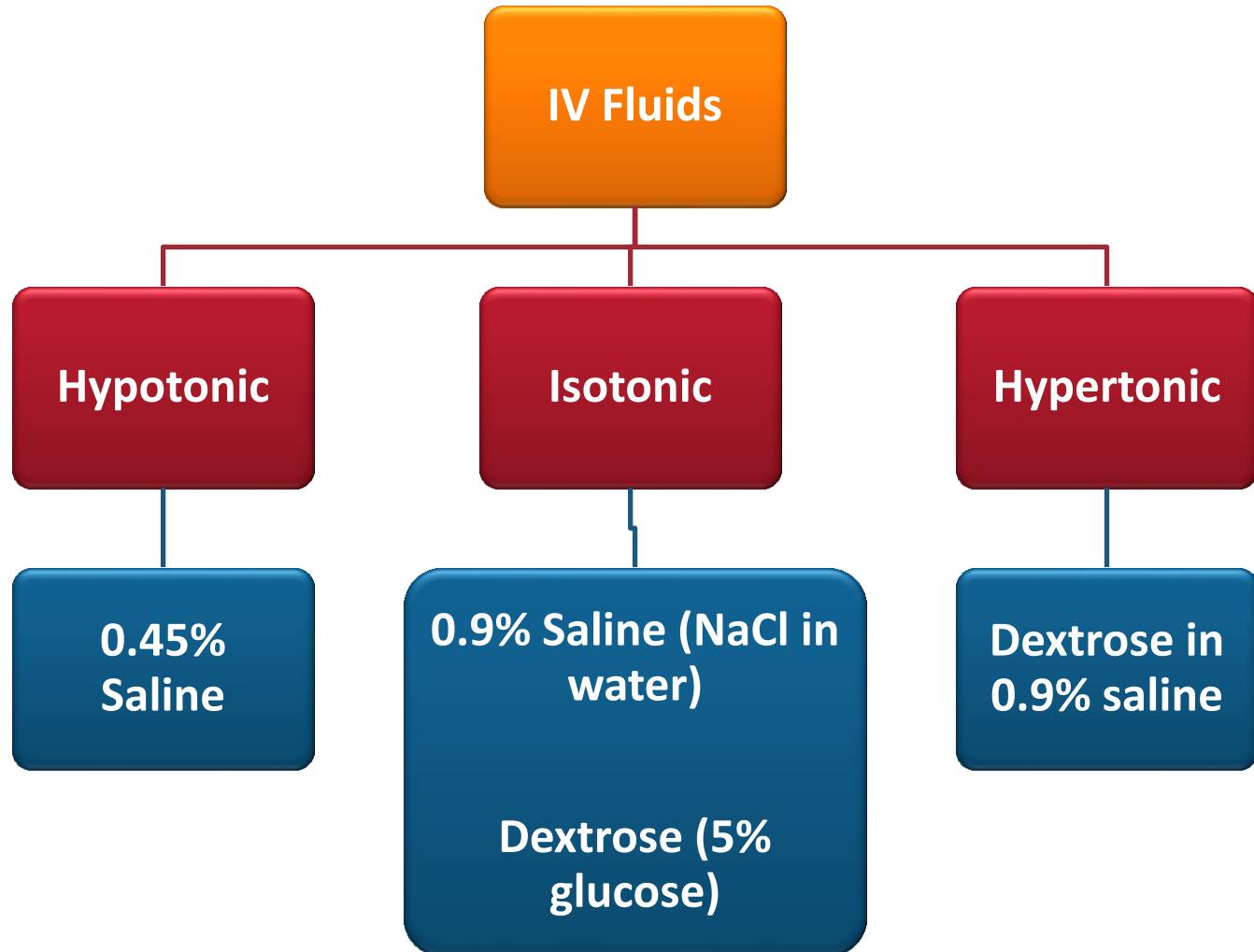


How is it related to patient care?

1. Physicians administer fluids to patients for various reasons that may change the osmolarity of ECF.
 2. Abnormalities in electrolytes can change the osmolarity of ECF and can affect the cells
 - a. **Hypernatremia** (\uparrow Na conc. In ECF)
 - b. **Hyponatremia** (\downarrow Na conc. In ECF)
- There are many types of IV fluids with different tonicities.

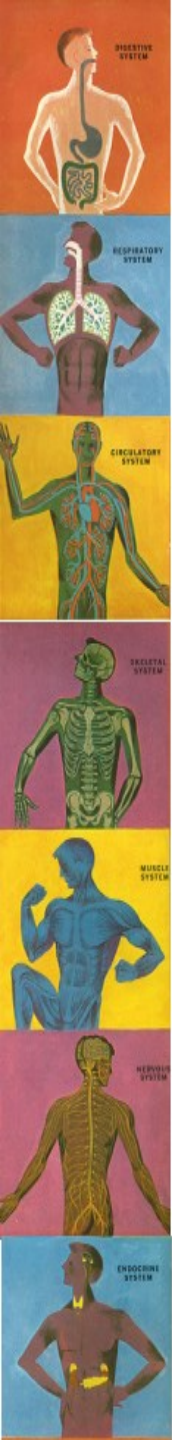
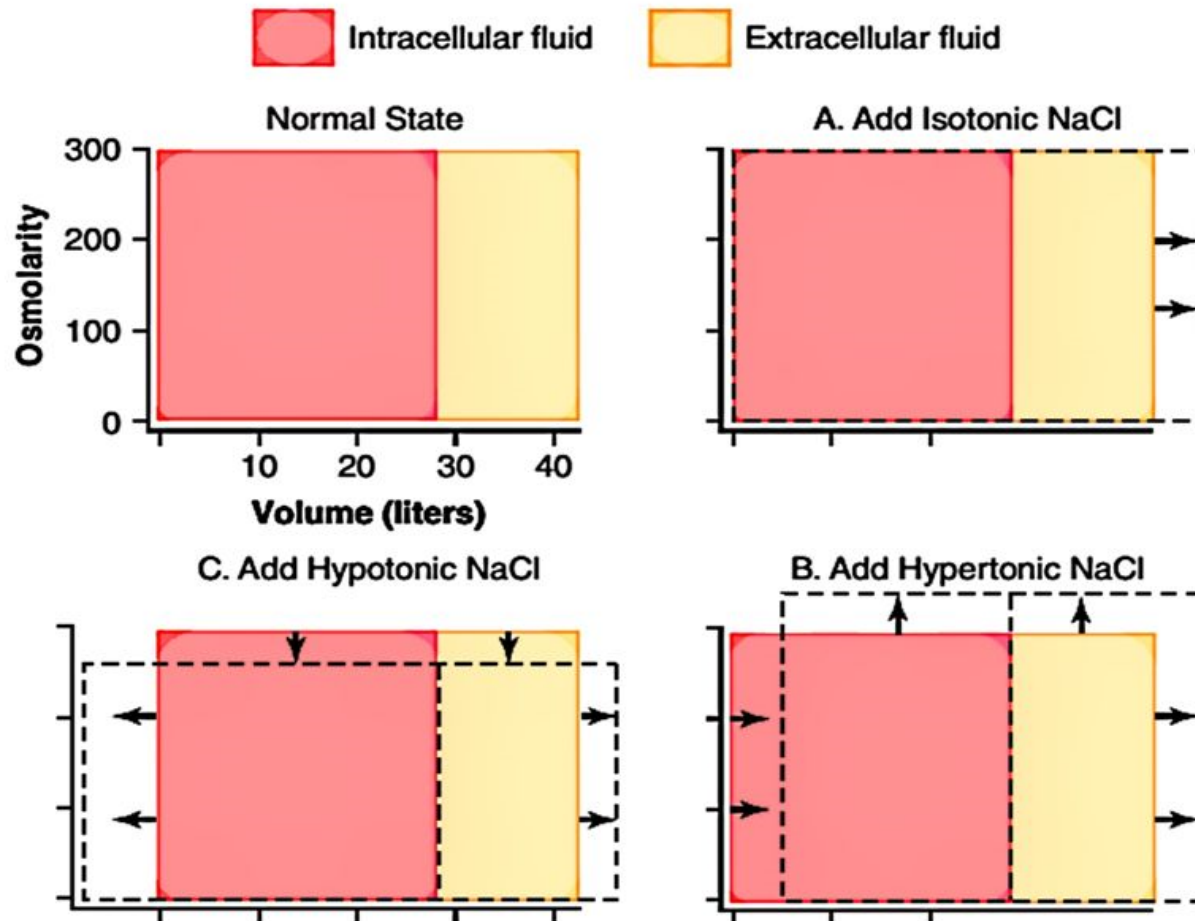


Types of IV Fluids

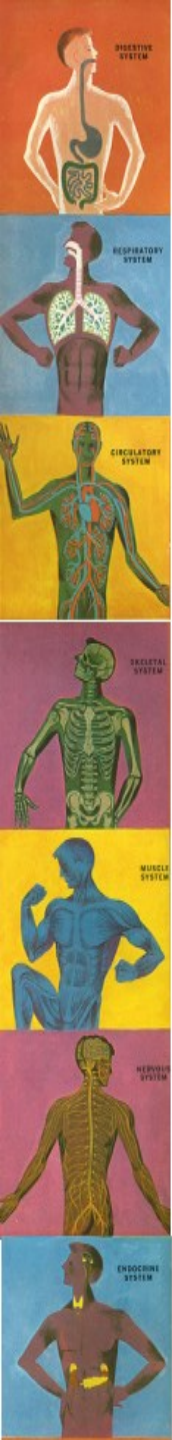


What happens to the different compartments with the administration of IV fluids?

Effect of adding isotonic, hypertonic, and hypotonic solutions to the extracellular fluid after osmotic equilibrium. The normal state is indicated by the solid lines, and the shifts from normal are shown by the shaded areas.



What happens to the different compartments with the administration of IV fluids?



Adding *hypertonic* solution

- \uparrow ECF osmolarity
- Water moves out of cells into ECF.
- ICF volume decreases.
- ECF volume increases.
- \uparrow ICF osmolarity

Adding *isotonic* solution

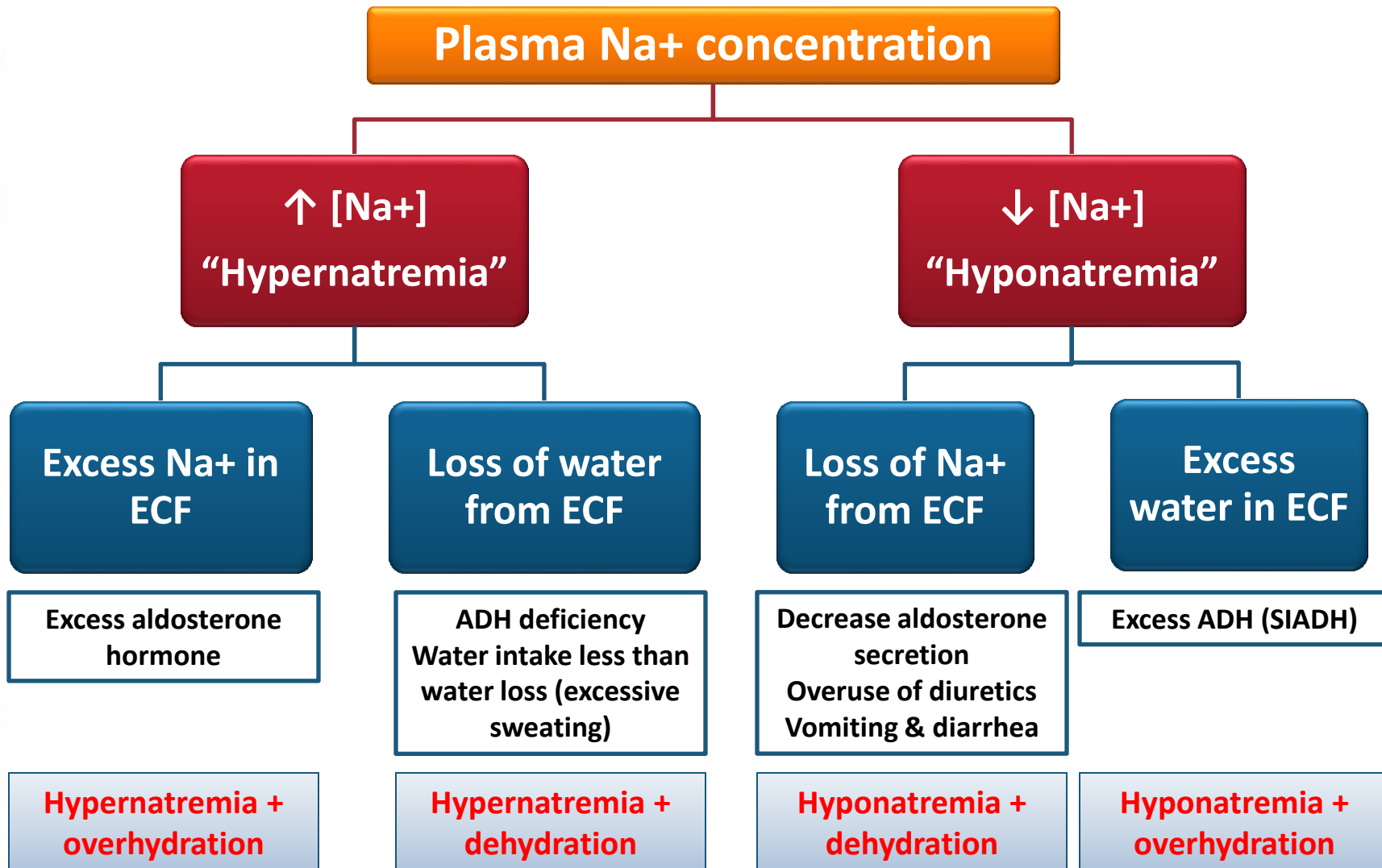
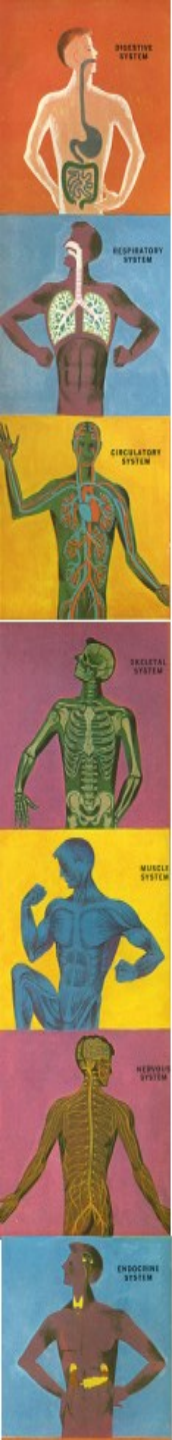
- No change in ECF osmolarity.
- No osmosis occurs through cell membrane.
- ECF volume increases.

Adding *hypotonic* solution



Activity

Clinical Abnormalities of Fluid Volume Regulation (Hypo- & Hyponatremia)



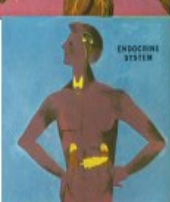
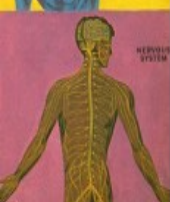
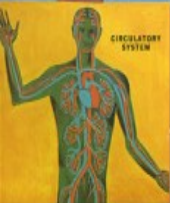
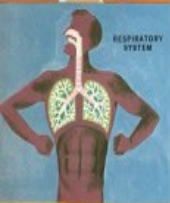
Clinical Manifestations of Hypo- & Hypernatremia

Hyponatremia

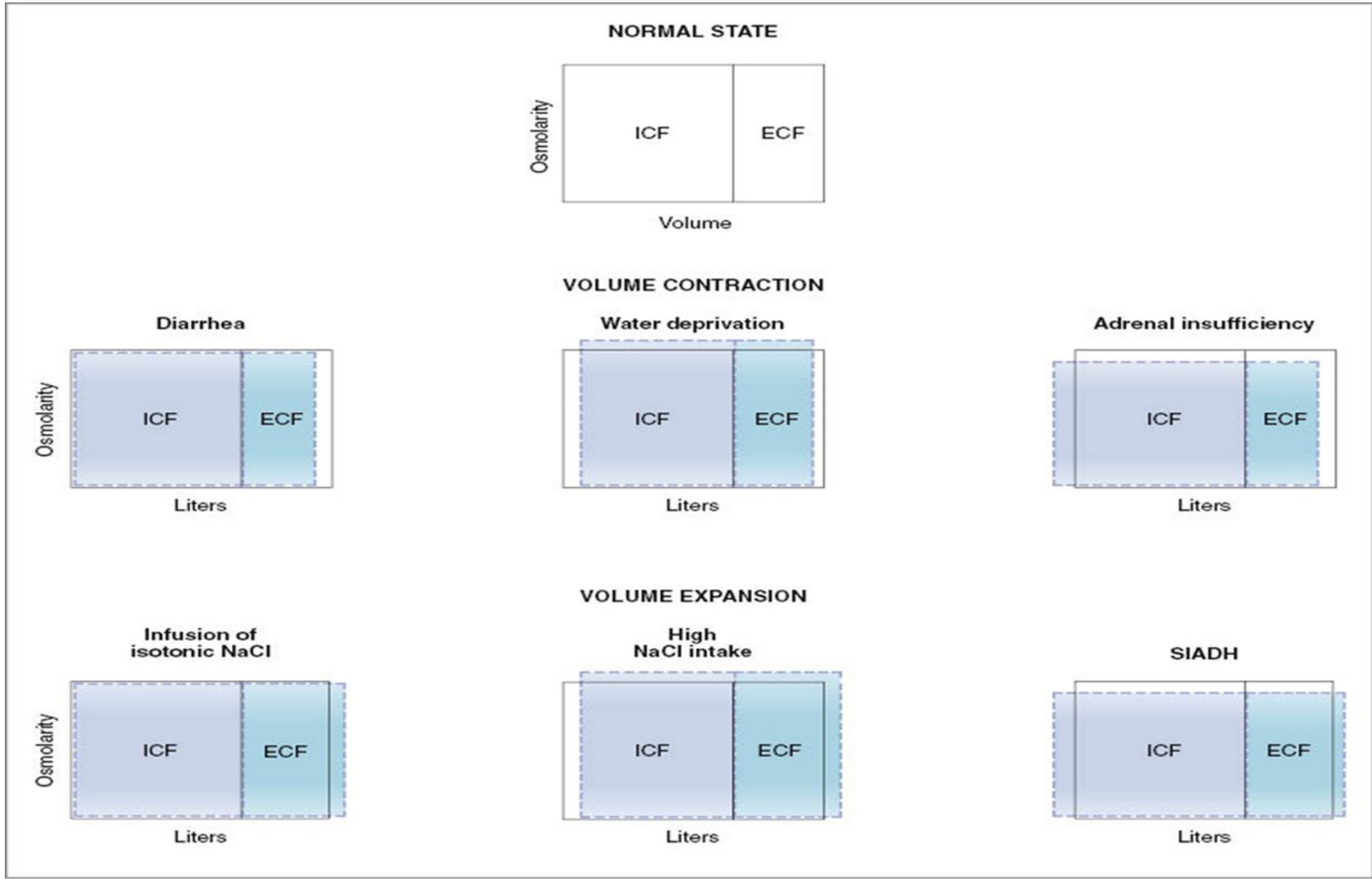
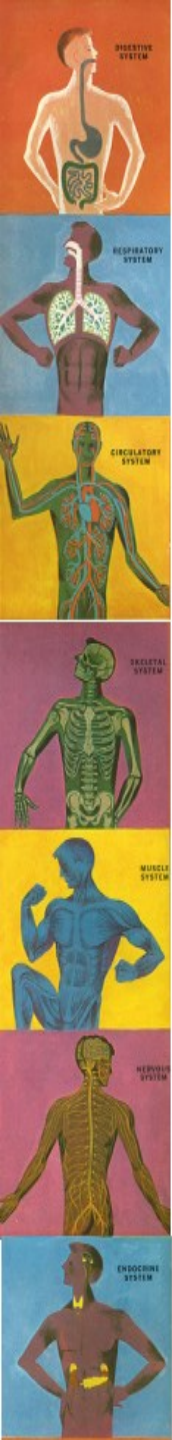
- $\downarrow\downarrow$ Plasma $[Na^+]$ \rightarrow
 $\downarrow\downarrow$ ECF osmolarity
causing cells to swell.
- Brain cell edema leads to neurologic symptoms;
 - Headache.
 - Nausea.
 - Lethargy & disorientation.
- $[Na^+] < 115-120\text{mmol/L}$
 \rightarrow seizures, coma,
permanent brain damage
& death.

Hypernatremia

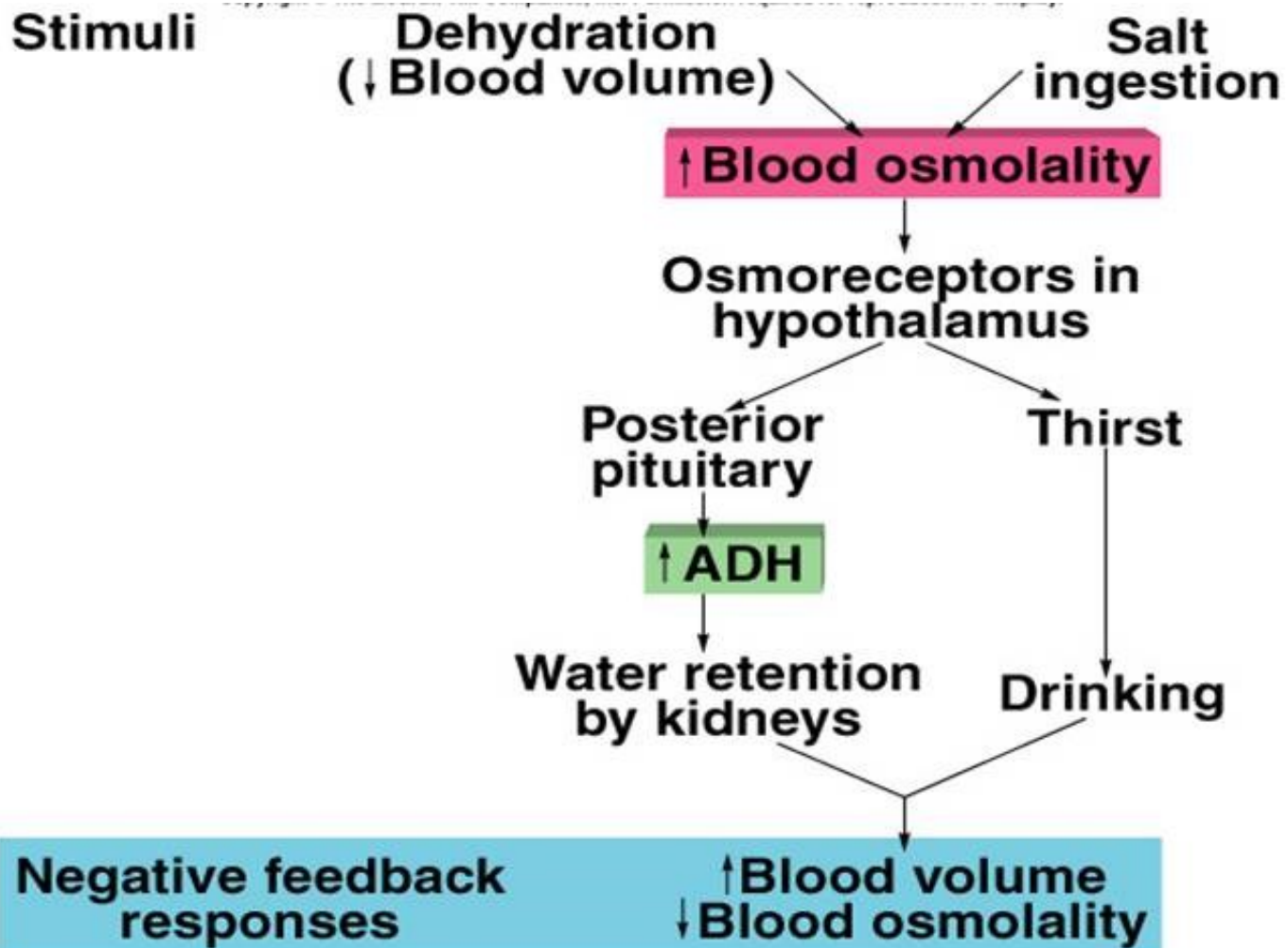
- $\uparrow\uparrow$ Plasma $[Na^+]$ \rightarrow $\uparrow\uparrow$
ECF osmolarity causing
cells to shrink.



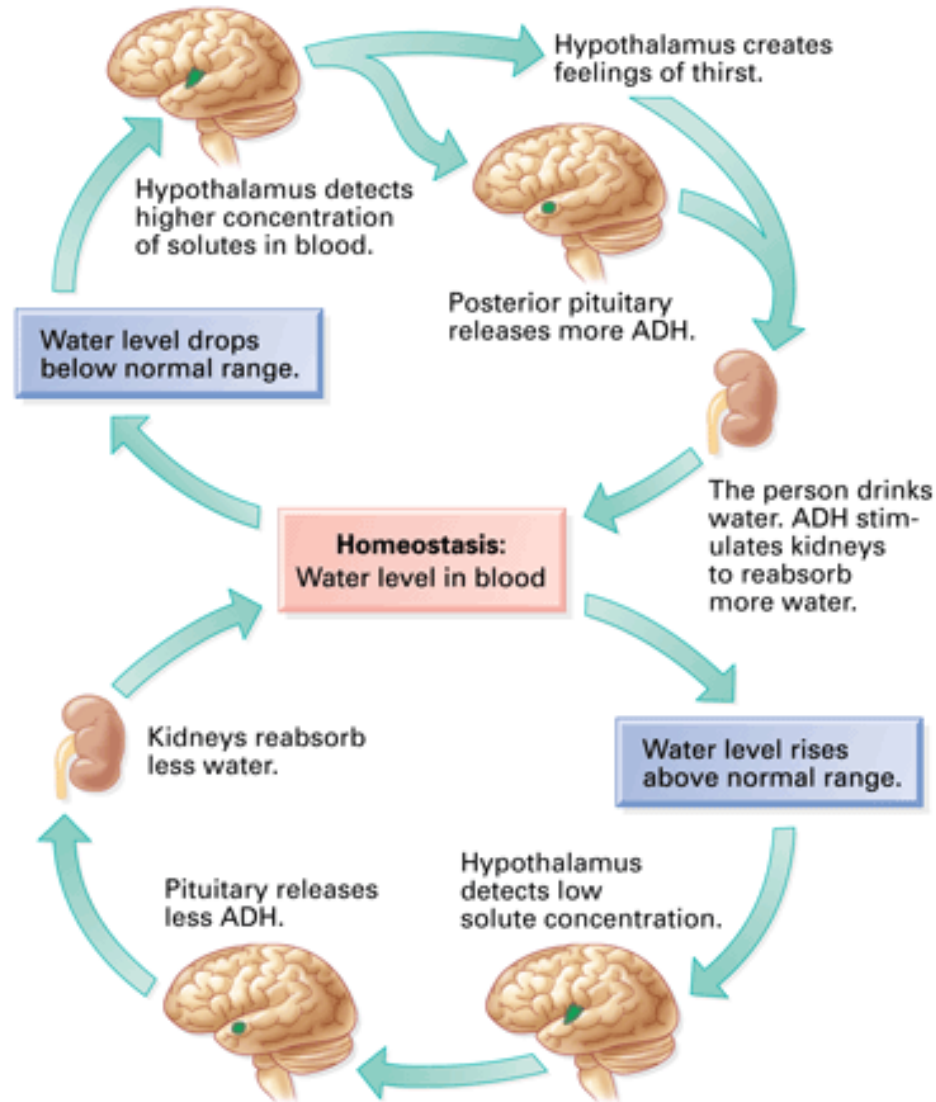
What happens to body fluid compartments in different clinical situations?



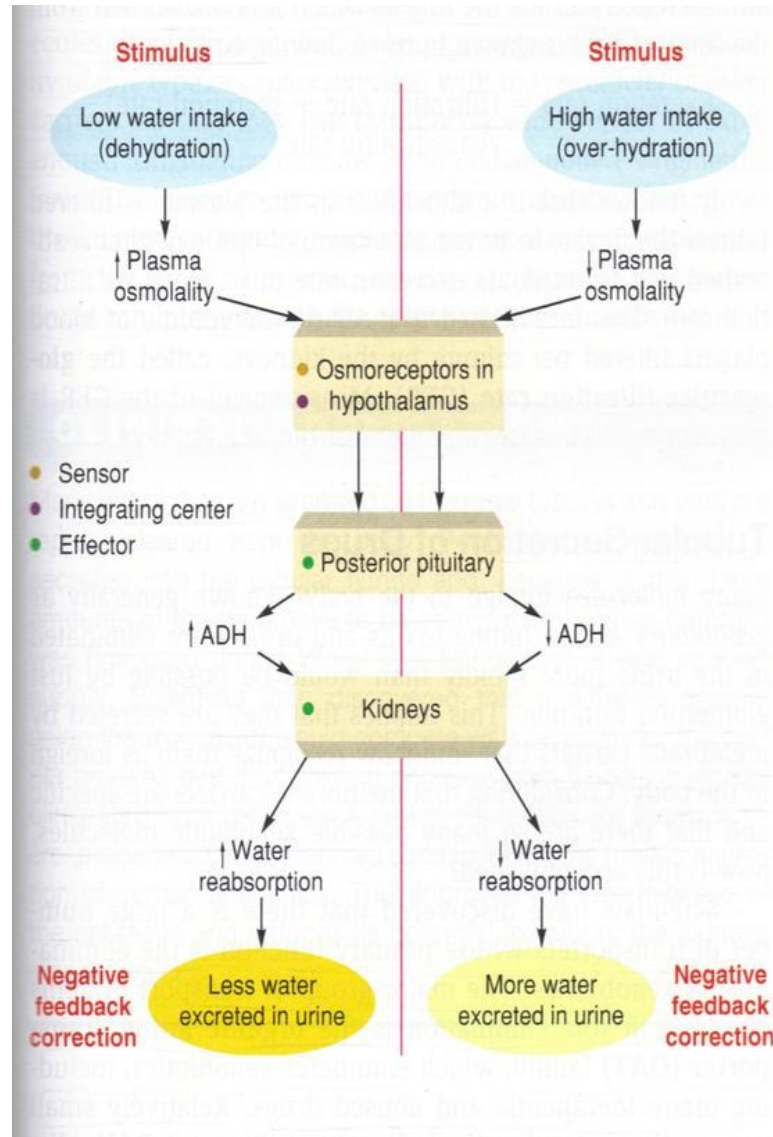
What are the feedback mechanisms operating in fluid balance control?



Feedback Mechanisms Controlling Fluid Balance



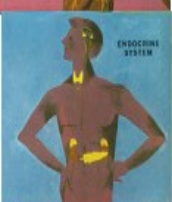
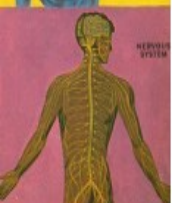
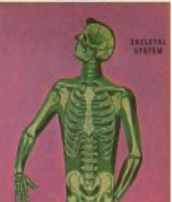
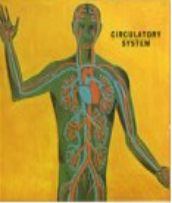
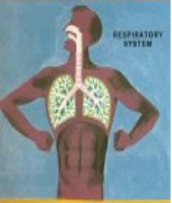
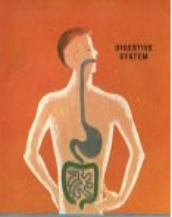
Feedback Mechanisms Controlling Fluid Balance



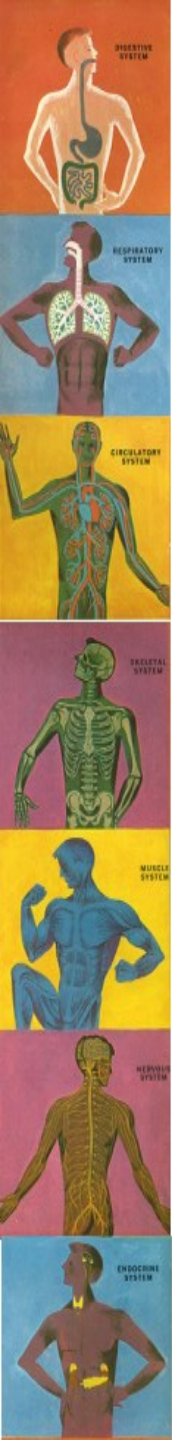
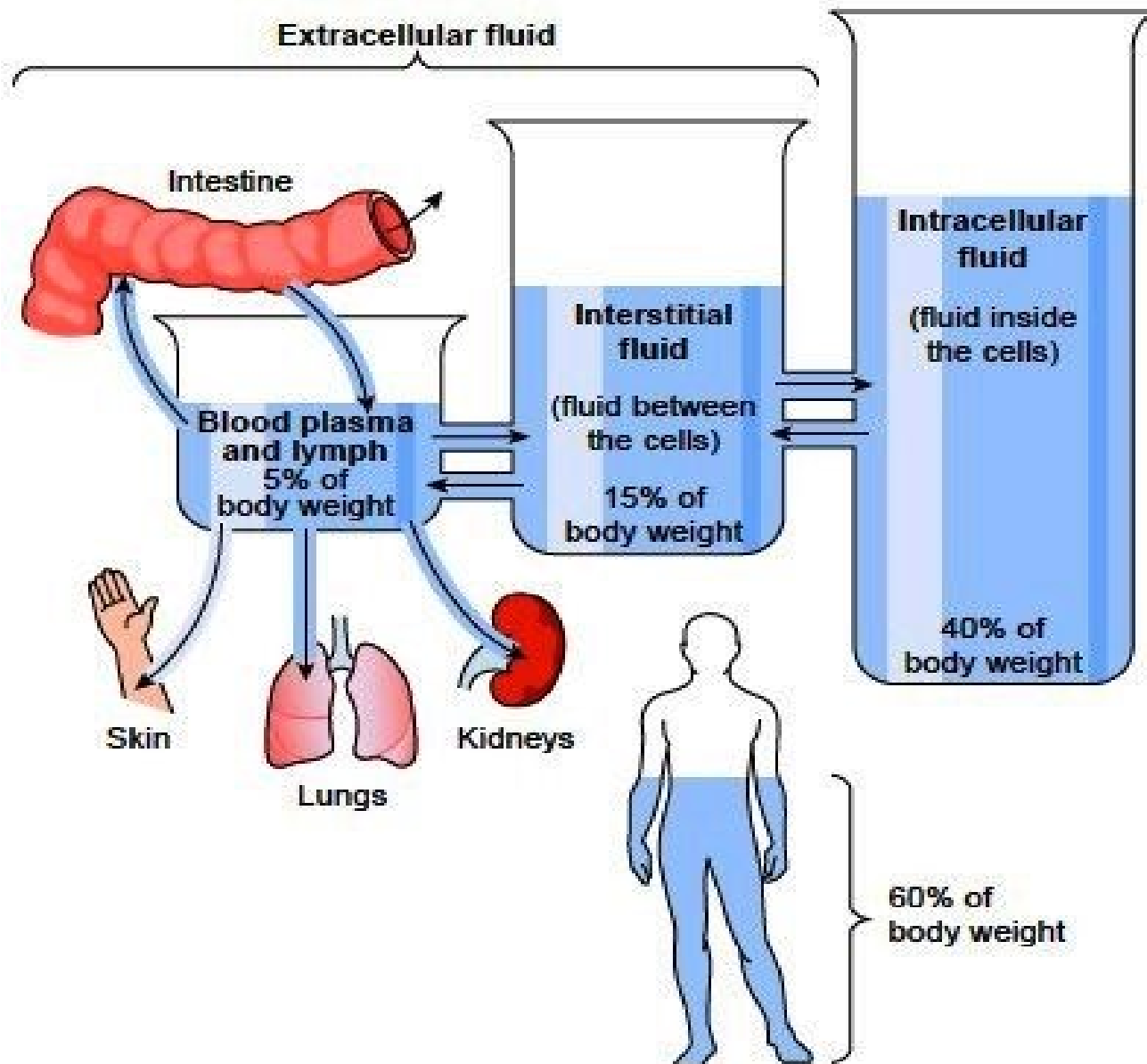
Water distribution in the body

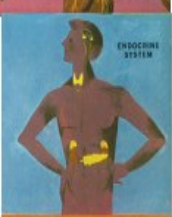
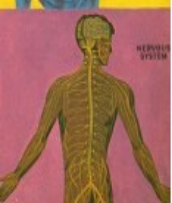
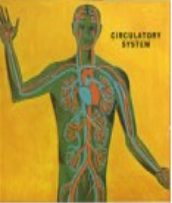
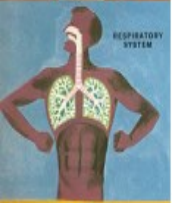
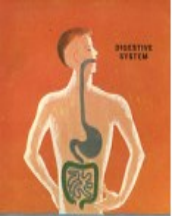
An exercise

- After learning about water content of the human body, a **60kg** medical student was curious to calculate her own body water content. What would you expect the **total volume of her body water** to be?
- And what is the volume of the following compartments?
 - ICF
 - ECF
 - IF
 - Plasma



In summary





Thank you