






## *Biochemistry orientation lab L 1*

-  **DOCTOR'S NOTE**
-  **VERY IMPORTANT!**
-  **EXTRA INFORMATION**



# OBJECTIVES

**The students should be able to understand & become familiar with:**

- 1) General safety rules followed in Biochemistry laboratory
- 2) Safety with laboratory equipments
- 3) Basic emergency procedures
- 4) Biological safety and waste disposal
- 5) The basics of spectrophotometer and general equipments to be used in the lab during Biochemistry practical sessions

# GENERAL SAFETY RULES

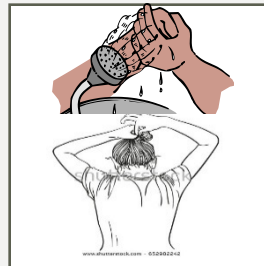
## VERY IMPORTANT

- Lab safety is everyone's responsibility
- Lab safety policy and procedures must be strictly followed



Always use appropriate clothes and personal protective tools (Lab coat, safety goggles, masks, gloves, no open shoes, no eye lenses)

- After handling chemicals, always wash your hands with soap and water.
- During lab work, keep your hands away from your face.  
Tie back long hair.



- Roll up loose sleeves.
- Know the location of the fire extinguisher, fire blanket, eyewash station, and first aid kit.
- Keep your work area uncluttered. Take to the lab station only what is necessary.

- It is suggested that you wear glasses rather than contact lenses.
- Never eat or drink during a lab work.



# CLINICAL BIOCHEMISTRY FOR DIAGNOSIS OF DISEASES

- Biochemical laboratory tests are crucial tools for diagnosis of many human diseases:

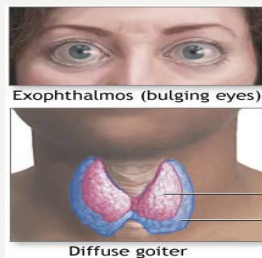
Kidney diseases e.g., nephrotic syndrome



Liver diseases e.g., hepatitis and jaundice



Endocrine diseases e.g., Thyrotoxicosis



Graves' disease is a common cause of hyperthyroidism, an over-production of thyroid hormone, which causes enlargement of the thyroid and other symptoms such as exophthalmos, heat intolerance and anxiety

Normal thyroid  
Enlarged thyroid

ADAM.



\*Just read it

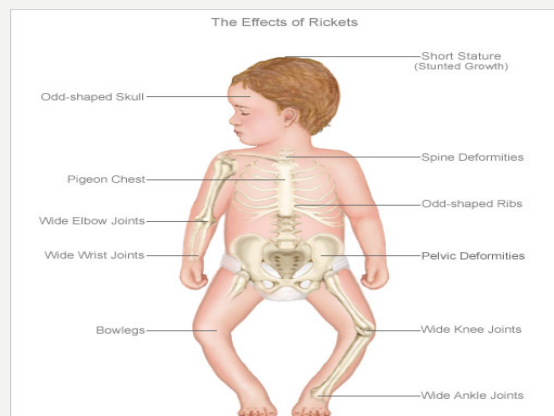
Cancers & malignancy e.g., prostate cancer.

Inherited diseases e.g., PKU



Metabolic diseases e.g., diabetes mellitus

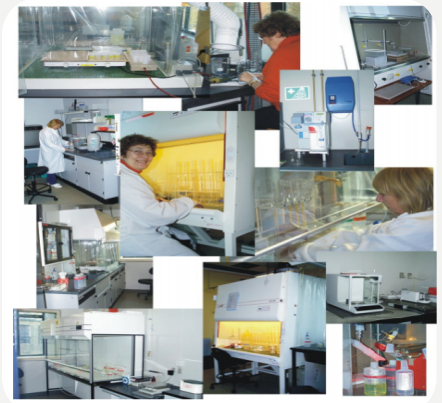
Skeletal diseases e.g., rickets



# Very important!!

## Safety with laboratory equipments:

Never use any laboratory equipment unless you are trained & have been authorized to do so As well as injuring yourself you may cause very costly damage



## Electrical safety:

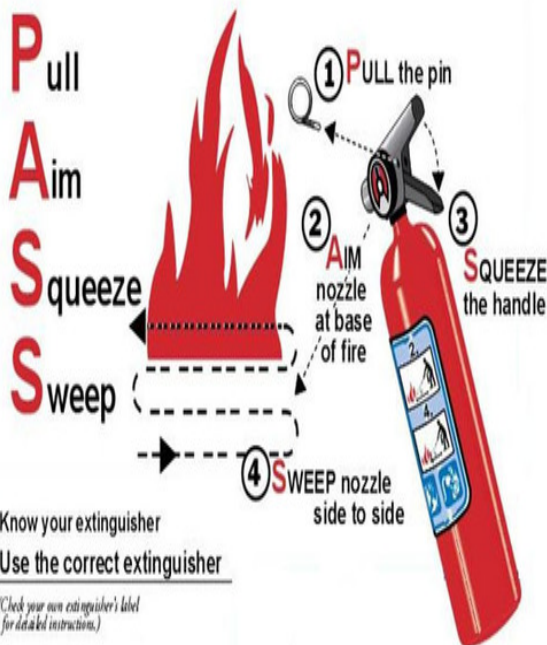
- \*Lay electrical cords where no one can trip on them.
- \* Be sure your hands and your lab area are dry before using electrical equipment.
- \* Unplug cords by pulling the plug and not the cord.



## Fire extinguisher

## Emergency procedures

To operate an extinguisher:



## Evacuation Instructions

- R** - Rescue and relocate anyone in immediate danger
- A** - Alert others by activating the building fire alarm
- C** - Confine the emergency by closing the doors
- E** - Evacuate immediately. Do not use elevators. Use stairs.

**Very important!!**

## **Biological safety:**

All biological samples are considered potentially infectious  
Should be handled and processed using strict precautions.

## **Waste disposal :**



All sharp objects such as needles, scalpels and even broken glassware go into yellow-red sharps container



For disposal of contaminated waste, use containers with yellow plastic bags



Regular waste like papers etc go into containers with black/white plastic bags

\*Just read it

# LAP EQUIPMENT



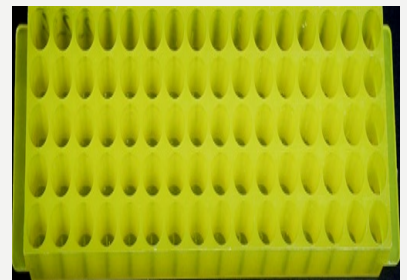
Microcentrifuge UV-spectrophotometer Water bath

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Vortex Automatic pipettes Eppendorf tube Tips

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Cuvettes

Rack- test tube

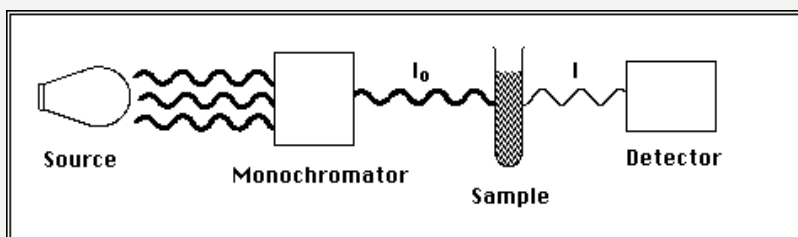
Rack- eppendorf tube



# SPECTROPHOTOMETER

Most of visible spectrophotometers are composed of:

- Light source which works with visible wavelengths (400-700 nm)
- Monochromator filter for choosing desired wavelength
- Sample holder (cuvette)
- Detector
- Meter or recorder
- UV light



- **Sample : Nucleated cells**

- **Principle:**

1. **Lysis of nucleated cells.**

2. **Removal of contaminants:** Any substance other than DNA, e.g., proteins

3. **Measurements:** UV absorbance at 260nm and 280nm

**Purity of DNA solution:** 260/280 ratio

**DNA concentration:** Absorbance at 260nm

- **Steps:** **memorize them in order**
  - **Lysis** of nucleated cells using lysis buffer.
  - **Binding** of DNA to the membrane of spin column.
  - **Wash:** using wash buffer.
  - **Elution** of pure DNA

# MEASUREMENTS NOTICE THAT WE HAVE 3 MEASUREMENTS ALL OF THEM ARE IMPORTANT!

- Measure the Absorbance at **260nm** ( is for DNA)
- Measure the Absorbance at **280nm**. (is for protein)

- Assess the DNA purity: 260/280 ratio  
(Accepted ratio: **1.7 - 1.9**) No unit



- Calculate DNA Conc.: Provided A<sub>260</sub> = 1.0, DNA is 50 µg/ml, unknown DNA Conc. can be calculated by cross multiplication

$$A_{260} = 1.0 \quad \text{DNA conc.} = 50 \mu\text{g/ml}$$

$$A_{260} = 0.5 \quad \text{DNA conc. ?}$$

Note:- In case of diluting the eluted sample, multiplies the final concentration by the dilution factor.

This can be adjusted by the spectrophotometer.

- pure sample of DNA has a range of 1.7 to 1.9 at 260/280 and is relatively free from protein contamination .
- A DNA preparation that is contaminated will have a 260/280 ratio lower than 1.7

## 1. DNA purity:

What do we mean by A 260/280 ratio? when we did the practical, the spectrophotometer sends UV light and the DNA absorbs it at 260 nm and the proteins absorb the UV light at 280 nm so let's say that the UV light absorbed at 260 nm is 2.33 and at 280 nm is 1.3 so the ratio is: the absorbance at 260 nm divided by the absorbance at 280 nm

$$\text{So } 2.33/1.3 = 1.79$$

\* Team 36

المعطيات التي راح تكون في السؤال:  
A<sub>260</sub> قيم  
الثابت يحفظ  
A<sub>260</sub>=1.0 يقابله DNA concentrations =50 ug/ml  
المطلوب:  
Concentration of A<sub>260</sub> التي أعطيت بالسؤال  
µg/ml \* الوحدة  
المقصود ب Dilution  
It's diluted twice مثلاً يقولك  
Concentration التي طلعتها تروح تأخذ قيمة ال  
وتكتب الوحدة 2 تضربها في  
Diluted 3 times ذلك قال  
تضرب في 3 وتطلع الجواب والوحدة

# DNA YIELD

\*Very important

**DNA Yield** = DNA Volume x final DNA Conc.

**Example:**

If you have

**Volume of DNA solution:** 200 $\mu$ l (0.2 ml)

The volume given in  $\mu$ l so you will divide it by 1000

هنا ممكن تتسألون هل قيمة التركيز التي تضرب في الحجم احط القيمة اللي قبل ما تضرب في عامل التخفيف

ولا بعد الضرب ؟ ع حسب اذا أصلاً جاء لك بالاختبار انه تخفف حط القيمة بعد التخفيف وتضرب بالحجم في حال لم يذكر أنه تخفف بتحط التركيز الأصلي وتضرب في الحجم

**Final DNA Conc.:** 30  $\mu$ g/ml

**Then, the yield ( $\mu$ g) = 0.2 ml x 30  $\mu$ g/ml**

**= 6.0  $\mu$ g**

$\mu$ g هنا كيف صارت

ml مع ml مشطبت

ضروري كتابة الوحدة

**Very important!!**

# DNA APPLICATIONS

**Purified DNA can be used for:**

## **1. Molecular diagnosis of diseases.**

- (e.g., sickle cell anemia)

## **2. Forensic applications.**

- (e.g., paternity testing)

## **3. Molecular biology research.**

**Molecular techniques using purified DNA:**

### **a. Amplification techniques:**

**Polymerase Chain •  
Reaction (PCR)**

### **b.**

**Southern blotting.(DNA)  
Northern Blotting(ss mRNA)  
Western Blotting (Protein)**

### **c. Restriction Fragment length polymorphism (RFLP).**

# **DONE BY :**

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**WISHING YOU THE BEST OF  
LUCK ...**

