INTRODUCTION TO METABOLISM

OBJECTIVES

By the end of this lecture the First Year students will be able to:

- Understand the concept of metabolic pathways
- Identify types and characteristics of metabolic pathways (anabolic and catabolic)
- Identify ATP as the energy source for cells

METABOLISM

- All the chemical reactions taking place inside a cell are collectively known as METABOLISM
- Metabolism consists of:
 - Energy consuming (anabolic) pathways
 - Energy producing (catabolic) pathways

PATHWAY VS CHEMICAL REACTION

Metabolic Pathway

- A multi-step sequence of chemical reactions
- Product of first reaction becomes a substrate for second reaction
- Integrated pathways: Metabolism

GLYCOLYSIS IS A METABOLIC PATHWAY

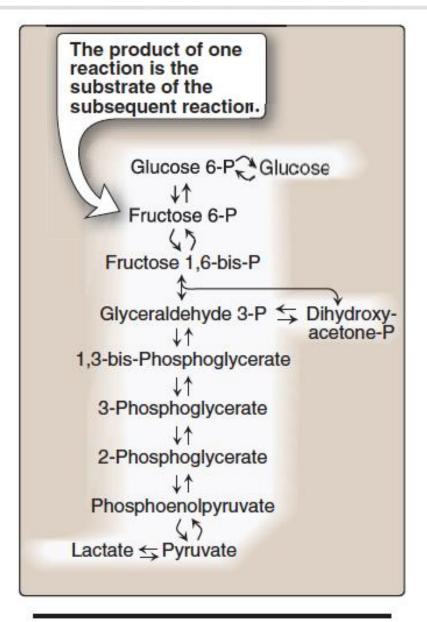
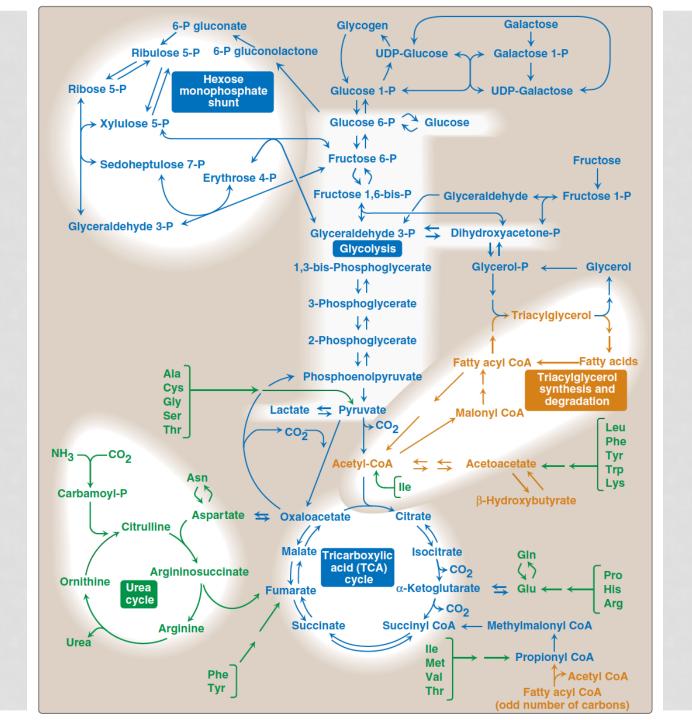


Figure 8.1

Glycolysis, an example of a metabolic pathway.

METABOLIC MAP

 Different pathways can intersect to form an integrated and purposeful network of chemical reactions called "The Metabolic Map"



CLASSIFICATION

- Most pathways can be classified as:
 - Catabolic
 - Anabolic

 Pathways that regenerate a component are called cycles

CATABOLIC PATHWAYS

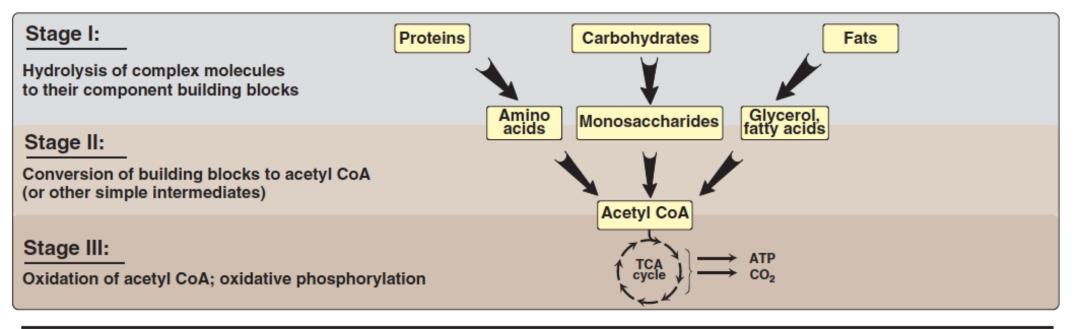


Figure 8.3

Three stages of catabolism.

ANABOLIC PATHWAYS

- Formation of precursor molecules into complex molecules
- Endergonic reactions (require ATP)
- A divergent process (few precursors form more complex products)

CATABOLISM VS ANABOLISM

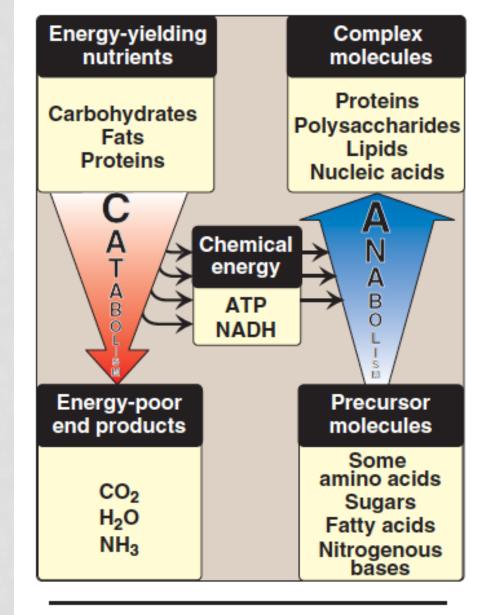


Figure 8.4

Comparison of catabolic and anabolic pathways.

COMPARISON OF CATABOLIC AND ANABOLIC PATHWAYS

Anabolic

- Simple to complex molecules
- Endergonic
- Involves reduction
- ▶ Require NADPH
- Divergent process

Catabolic

- Complex to simple molecules
- Exergonic
- Involves oxidation
- ▶ Require NAD+
- Convergent process

AMPHIBOLIC PATHWAYS

- Amphi means dual
- Amphibolic: Both catabolic and anabolic

Example:

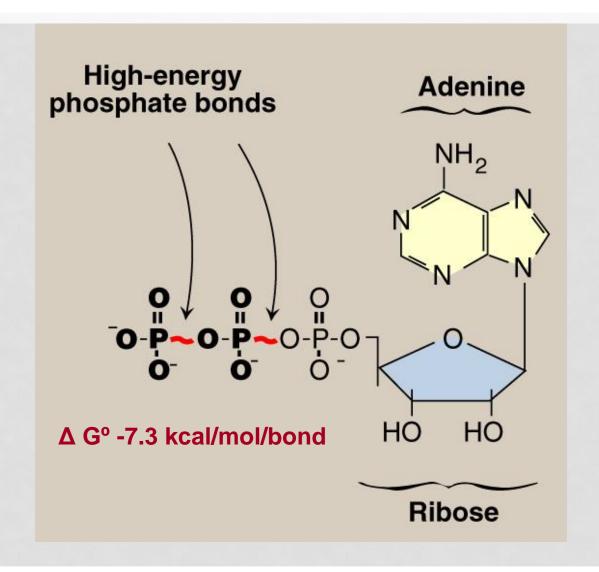
- Krebs cycle is mainly a catabolic cycle but with some anabolic features
- Krebs cycle is used for the synthesis of glucose from amino acids
- It is amphibolic

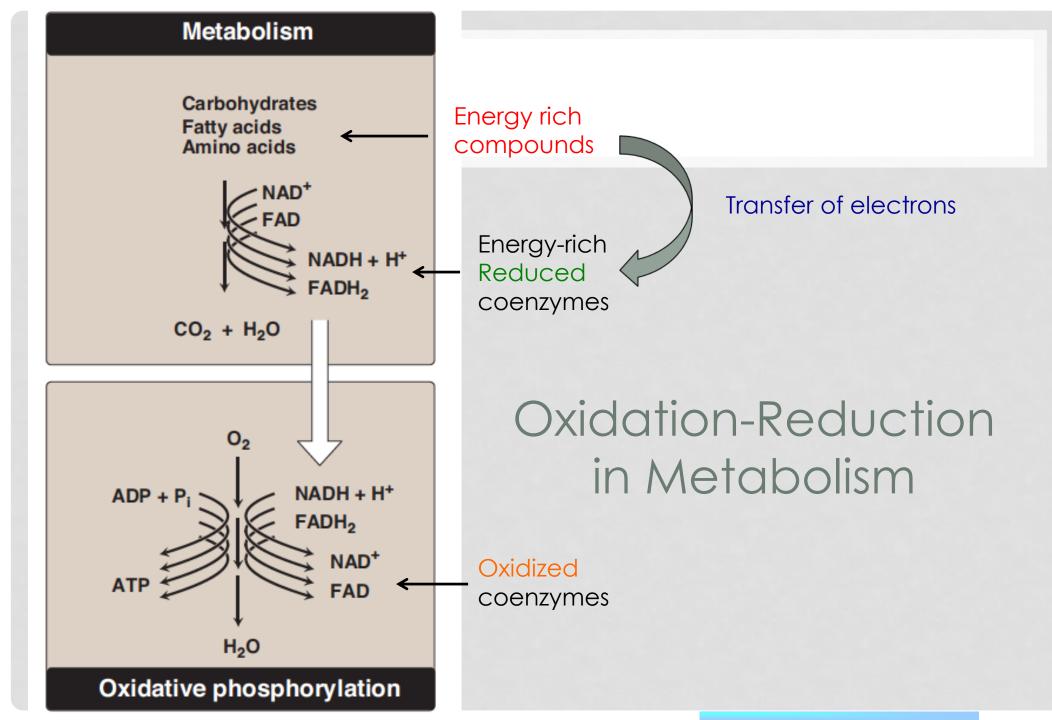
ENERGY CURRENCY: ATP

$$ATP + H_2O \longrightarrow ADP + P_i$$

- The free energy liberated by the hydrolysis of ATP is used to drive the endergonic reactions
- ATP is formed from ADP and P_i when fuel molecules are oxidized
- This ATP-ADP cycle is the fundamental mode of energy exchange in biological systems

ADENOSINE TRIPHOSPHATE (ATP)





OXIDATION / REDUCTION

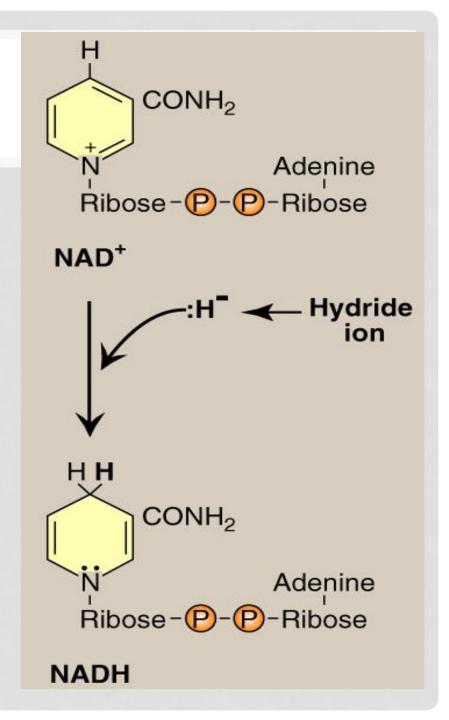
Oxidation:

Loss of hydrogen Loss of electrons

Reduction:

Gain of hydrogen Gain of electrons

NAD+ / NADH



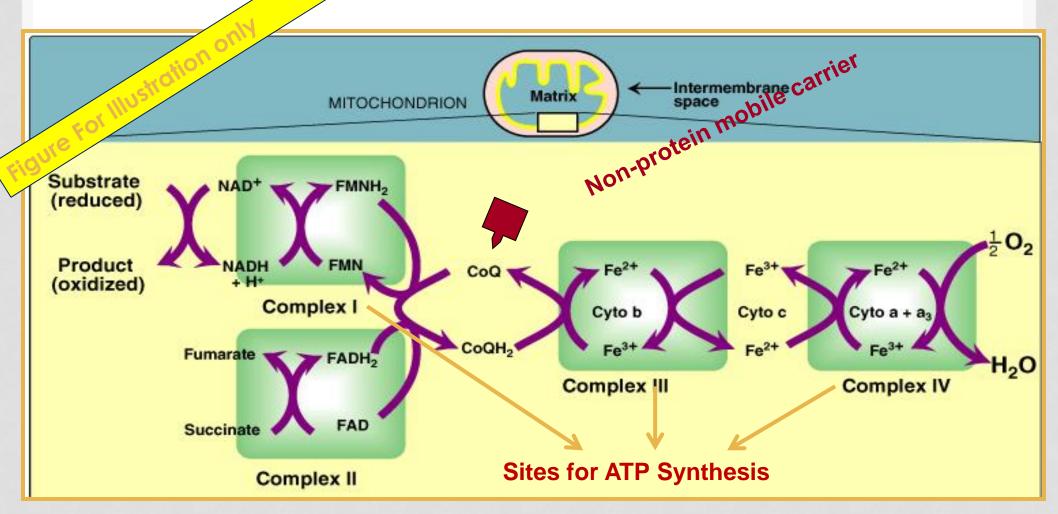
REGULATION OF METABOLISM

- Intracellular signals:
 - Substrate availability
 - Product inhibition
 - Allosteric activators or inhibitors
- Intercellular communications:
 - Chemical signaling (hormones):
 Second messenger
 - cAMP, cGMP
 - Ca⁺⁺/phosphatidylinositol

METABOLIC FUEL

- Carbohydrates and lipids (mainly) and proteins (little extent) are used for energy production
- Glucose and fatty acids are a major source of energy
- Amino acids are a minor source of energy
- Glucose is the major metabolic fuel of most tissues

ELECTRON TRANSPORT CHAIN (ETC)



Electron transport and ATP synthesis are tightly coupled processes

TAKE HOME MESSAGE

- Metabolism is the sum of all biochemical pathways that occur inside the cells
- A metabolic pathway is a multistep sequences of enzyme-catalyzed reactions
- Catabolism is a convergent process that provides energy to cells in the form of ATP
- Anabolism is a divergent process that consumes energy for the synthesis of complex molecules
- Metabolic pathways are tightly regulated and highly integrated
- ATP is the energy currency of the cells

REFERENCES

• Lippincott's Illustrated Reviews, Biochemistry, 5th edition, Denise R. Ferrier, Lippincott Williams & Wilkins, USA, pp 91-94.