

بِسْمِ اللَّهِ الرَّحْمَنِ الرَّحِيمِ



# **BODY FLUIDS**

## *DISTRIBUTION*

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*Dr. Taj*

# FLUID COMPARTMENTS

**EXTRA CELLUAR  
FLUID**  
33 %

**INTRA CELLULAR  
FLUID**  
67 %

**PLASMA**  
20 %

**INTERSTITIAL  
FLUID**  
80 %

**TRANSCELLULAR  
FLUID**

**CSF**  
Intra ocular  
Pleural  
Peritoneal  
Synovial  
Digestive Secretions

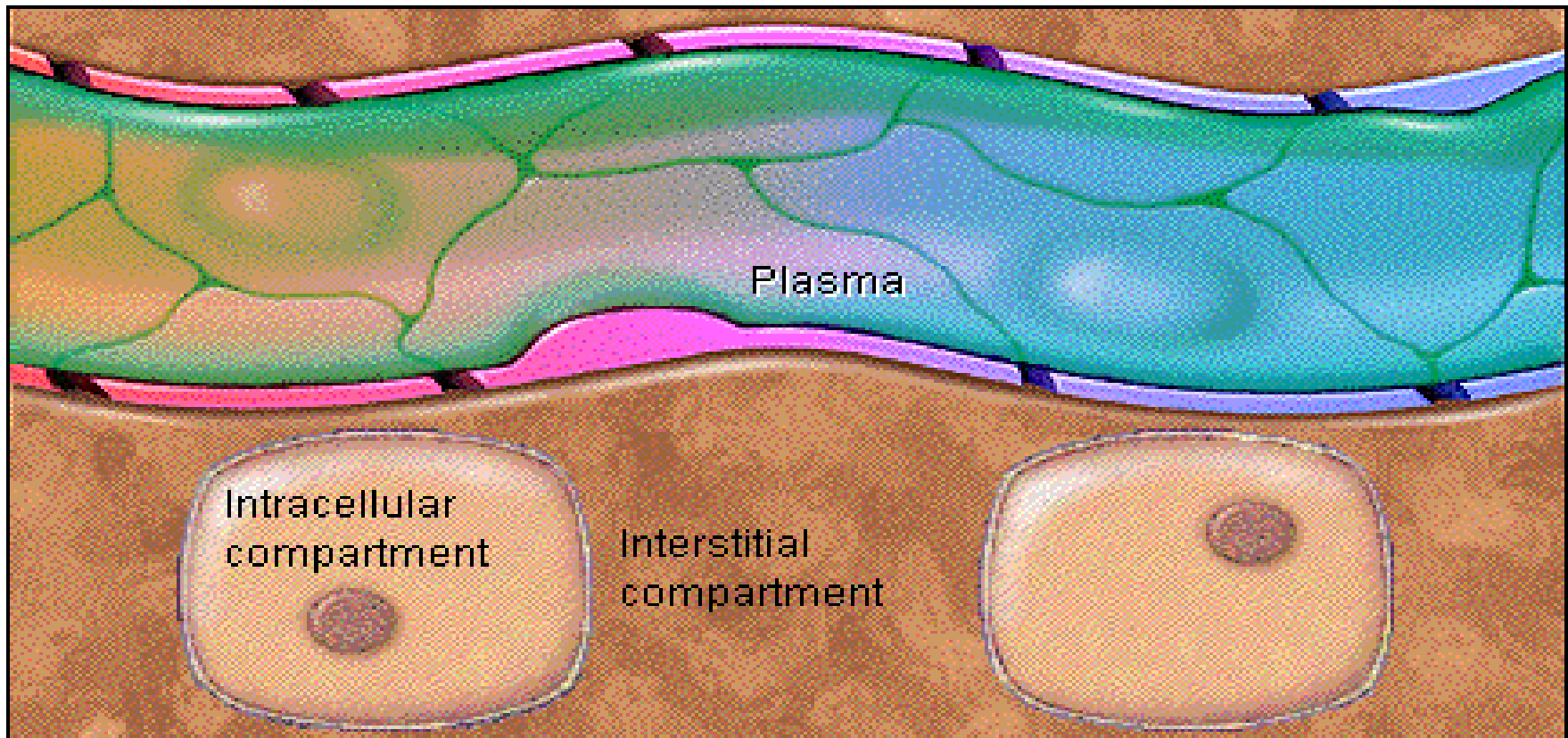
# FLUID COMPARTMENTS

The three major fluid compartments:

- **Intracellular fluid (ICF)** is the fluid within cells, also known as cytosol.
- **Extracellular fluid (ECF)** is the fluid found outside of cells.

There are two major kinds of extracellular fluid:

- **Interstitial fluid** is the fluid surrounding the cells.
- **Plasma** is the fluid component of blood.



|                                     | EXTRACELLULAR FLUID | INTRACELLULAR FLUID |
|-------------------------------------|---------------------|---------------------|
| Na <sup>+</sup> -----               | 142 mEq/L           | 10 mEq/L            |
| K <sup>+</sup> -----                | 4 mEq/L             | 140 mEq/L           |
| Ca <sup>++</sup> -----              | 2.4 mEq/L           | 0.0001 mEq/L        |
| Mg <sup>++</sup> -----              | 1.2 mEq/L           | 58 mEq/L            |
| Cl <sup>-</sup> -----               | 103 mEq/L           | 4 mEq/L             |
| HCO <sub>3</sub> <sup>-</sup> ----- | 28 mEq/L            | 10 mEq/L            |
| Phosphates -----                    | 4 mEq/L             | 75 mEq/L            |
| SO <sub>4</sub> <sup>-</sup> -----  | 1 mEq/L             | 2 mEq/L             |
| Glucose -----                       | 90 mg/dl            | 0 to 20 mg/dl       |
| Amino acids ----                    | 30 mg/dl            | 200 mg/dl ?         |
| Cholesterol } -----                 | 0.5 g/dl            | 2 to 95 g/dl        |
| Phospholipids } -----               |                     |                     |
| Neutral fat } -----                 |                     |                     |
| PO <sub>2</sub> -----               | 35 mm Hg            | 20 mm Hg ?          |
| PCO <sub>2</sub> -----              | 46 mm Hg            | 50 mm Hg ?          |
| pH -----                            | 7.4                 | 7.0                 |
| Proteins -----                      | 2 g/dl              | 16 g/dl             |
|                                     | (5 mEq/L)           | (40 mEq/L)          |

# TRANSPORT MECHANISMS

- **Cell membrane is selectively permeable (semi-permeable) to some molecules and ions.**
- **Not permeable (impermeable) to proteins, nucleic acids, and other molecules.**

# TRANSPORT MECHANISMS

## Passive Transport

Requires  
Kinetic Motion

Diffusion

- Simple
- Facilitated

## Active Transport

Requires Energy of  
ATP

Primary Active

Secondary Active

- Cotransport
- Counter Transport

**Bulk Transport**

- Endocytosis
- Exocytosis
- Phagocytosis

# CATEGORIES OF TRANSPORT ACROSS THE PLASMA MEMBRANE

- **Categorized by their energy requirements:**
    - **Passive transport:**
      - Net movement down a concentration gradient.
      - Does not require metabolic energy (ATP).
    - **Active transport:**
      - Net movement against a concentration gradient.
      - Requires ATP.
-



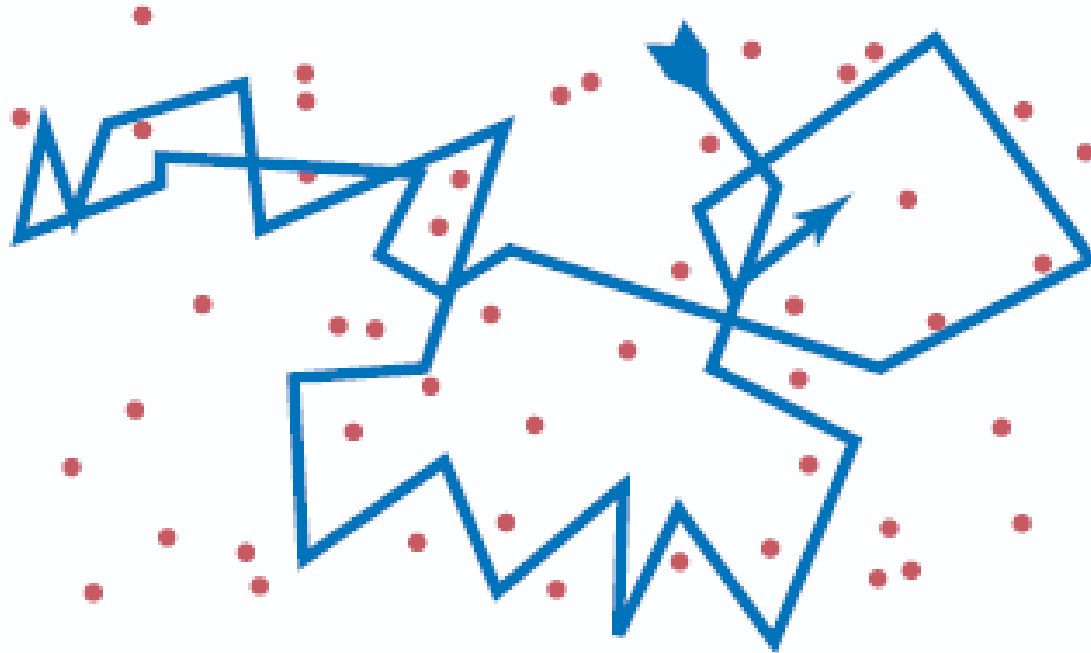
# CATEGORIES OF TRANSPORT ACROSS THE PLASMA MEMBRANE

- **Mechanisms to transport molecules and ions through the cell membrane:**
  - **Carrier mediated transport:**
    - **Facilitated diffusion and active transport.**
  - **Non-carrier mediated transport.**
    - **Simple Diffusion and osmosis.**

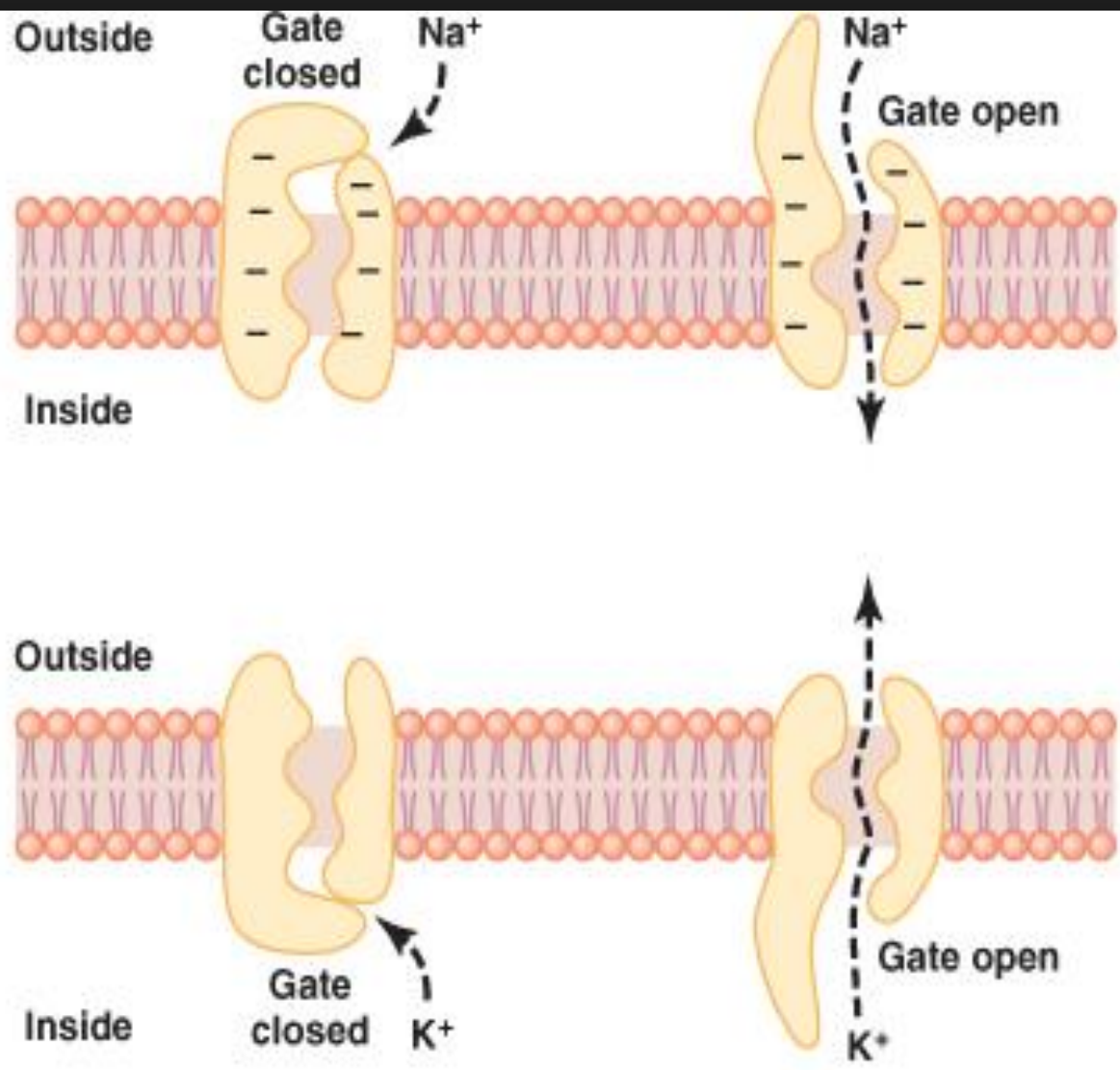
# DIFFUSION

- Molecules/ions are in constant state of random motion due to their thermal energy.
- Eliminates a concentration gradient and distributes the molecules uniformly.
- Physical process that occurs whenever there is a concentration difference across the membrane and the membrane is permeable to the diffusing substance.

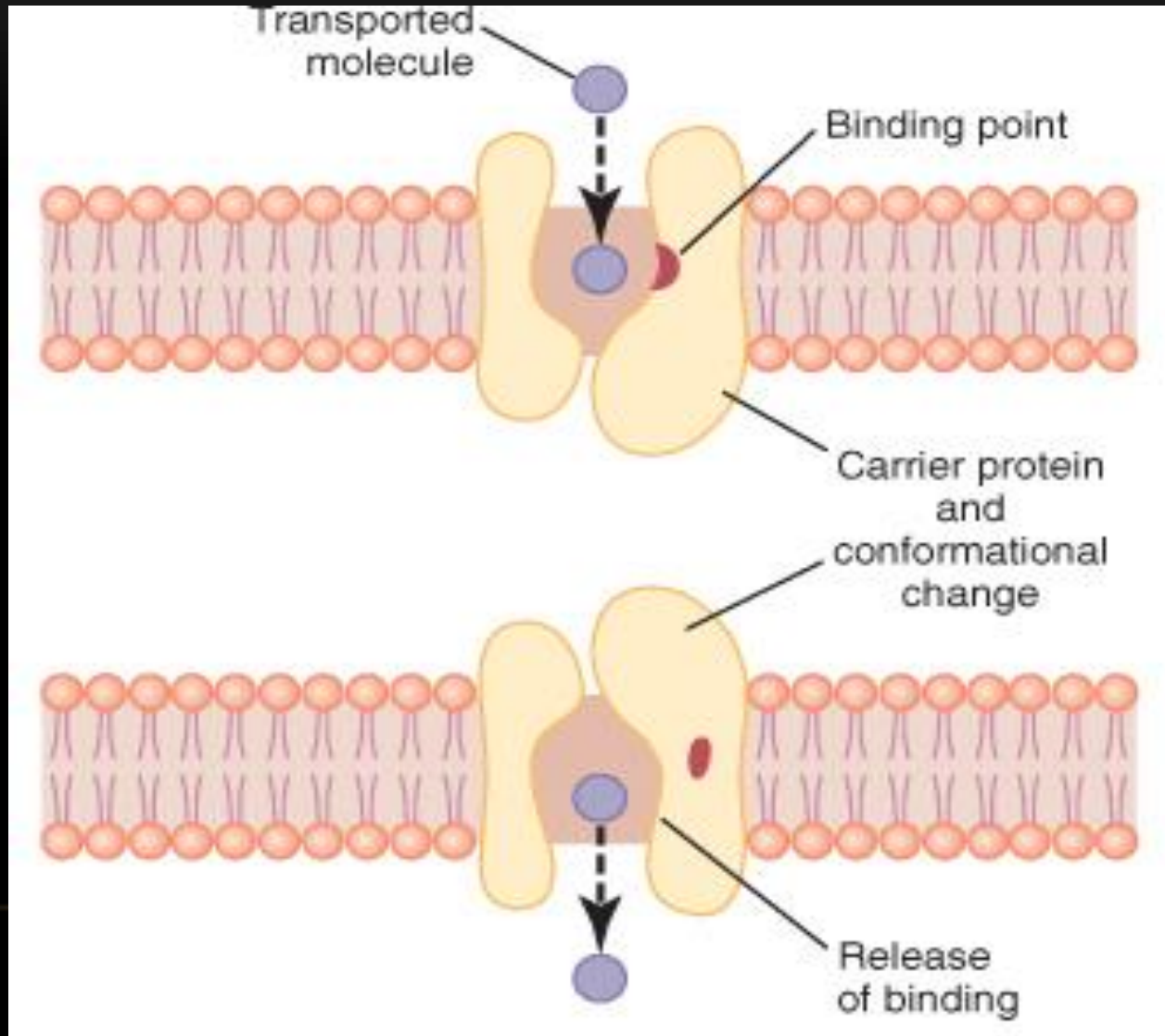
# Simple Diffusion



# Simple Diffusion



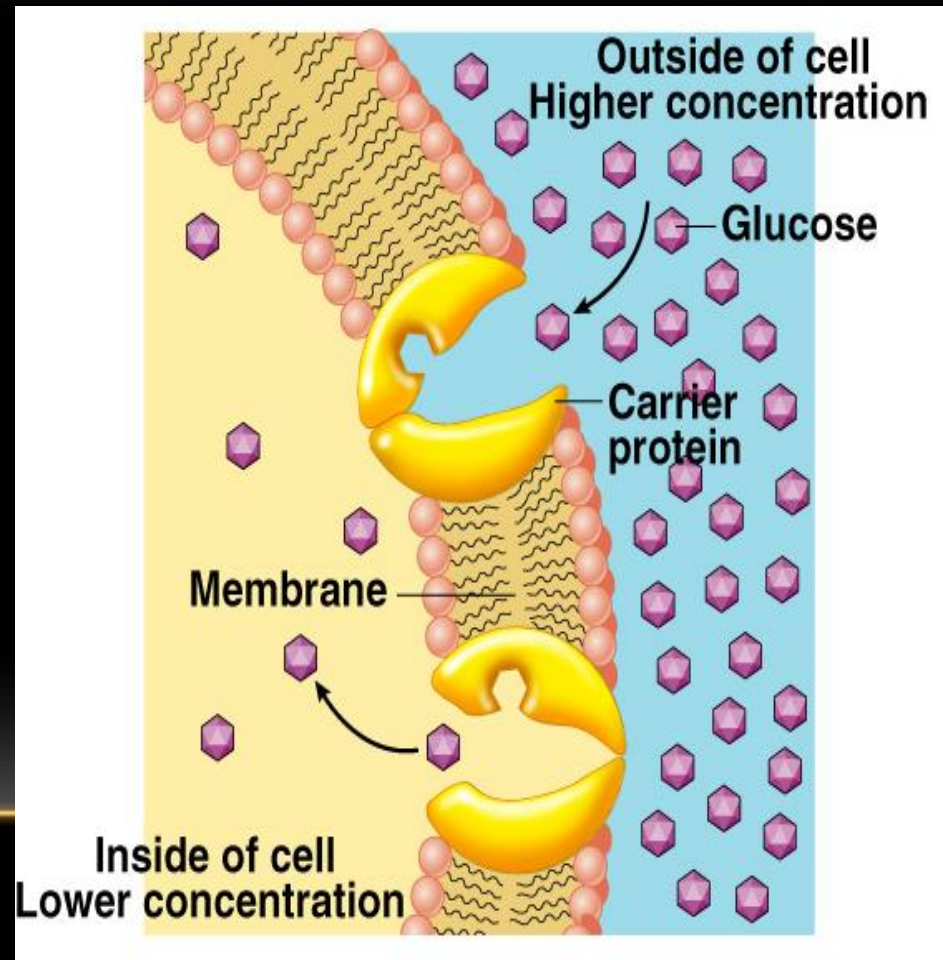
# Facilitated Diffusion



# FACILITATED DIFFUSION

It is Passive

- **ATP not needed.**
  - Powered by thermal energy of diffusing molecules.
- **Involves transport of substance through plasma membrane down concentration gradient by carrier proteins.**
  - Transport carriers for glucose designated as GLUT.



# DIFFUSION THROUGH PLASMA MEMBRANE

- Cell membrane is **PERMEABLE** to:
  - **Non-polar molecules ( $O_2$ ).**
  - **Lipid soluble molecules (steroids).**
  - **Small polar covalent bonds ( $CO_2$ ).**
  - **$H_2O$  (small size, lack charge).**
- Cell membrane **IMPERMEABLE** to:
  - **Large polar molecules (glucose).**
  - **Charged inorganic ions ( $Na^+$ ).**

# FACTORS AFFECTING NET RATE OF DIFFUSION

## **Permeability depends on:**

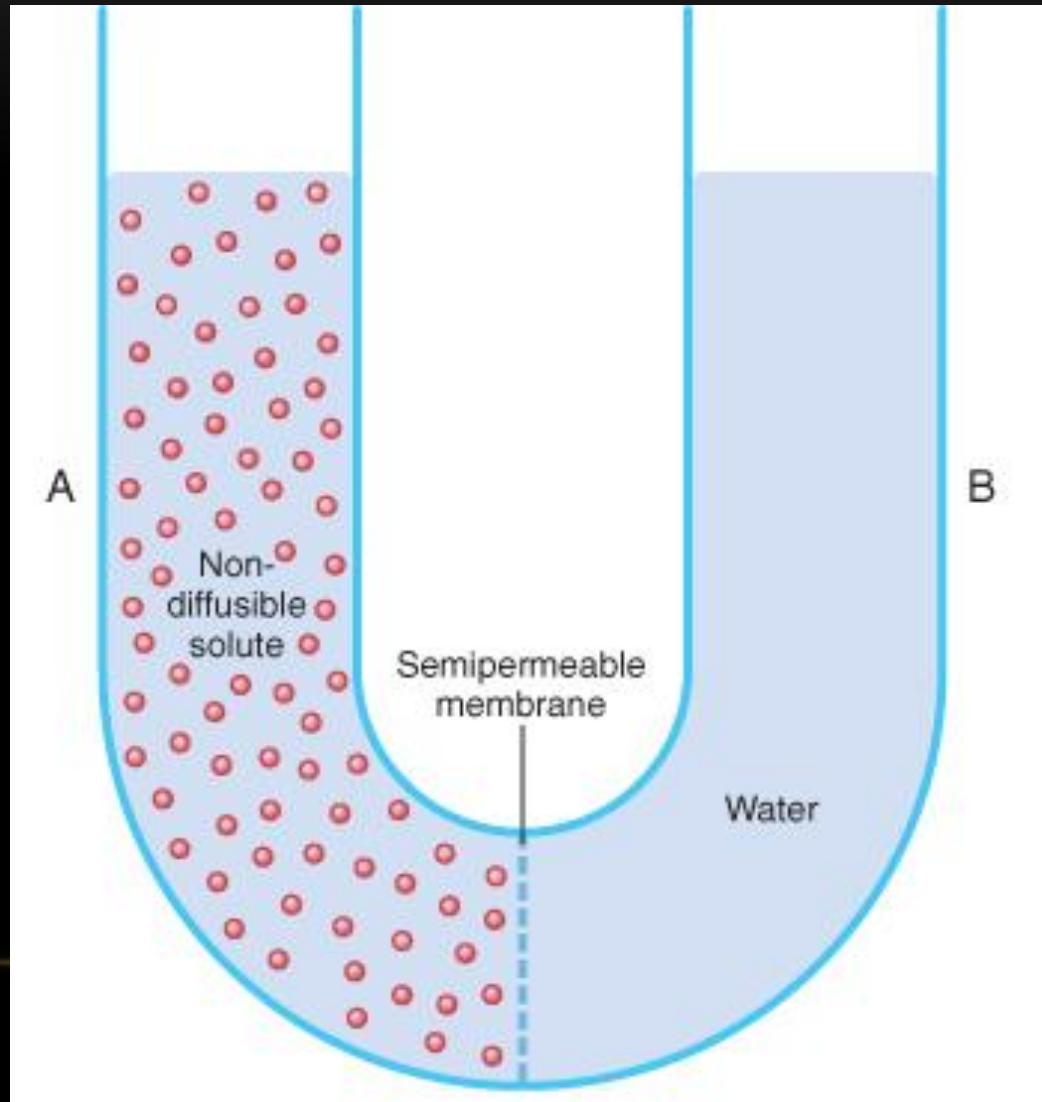
- **Thickness of the membrane.**
- **Lipid Solubility of the substance.**
- **Presence of Protein Channels.**
- **Temperature of the compartments.**
- **Molecular weight of the substance.**
- **Surface Area of the membrane.**
- **Concentration Difference of the substances.**
- **Electrical Potential difference of the substances.**
- **Pressure Difference between the sides.**

## **NOTE:**

**Neuronal plasma membrane is 20 x more permeable to  $K^+$  than  $Na^+$**

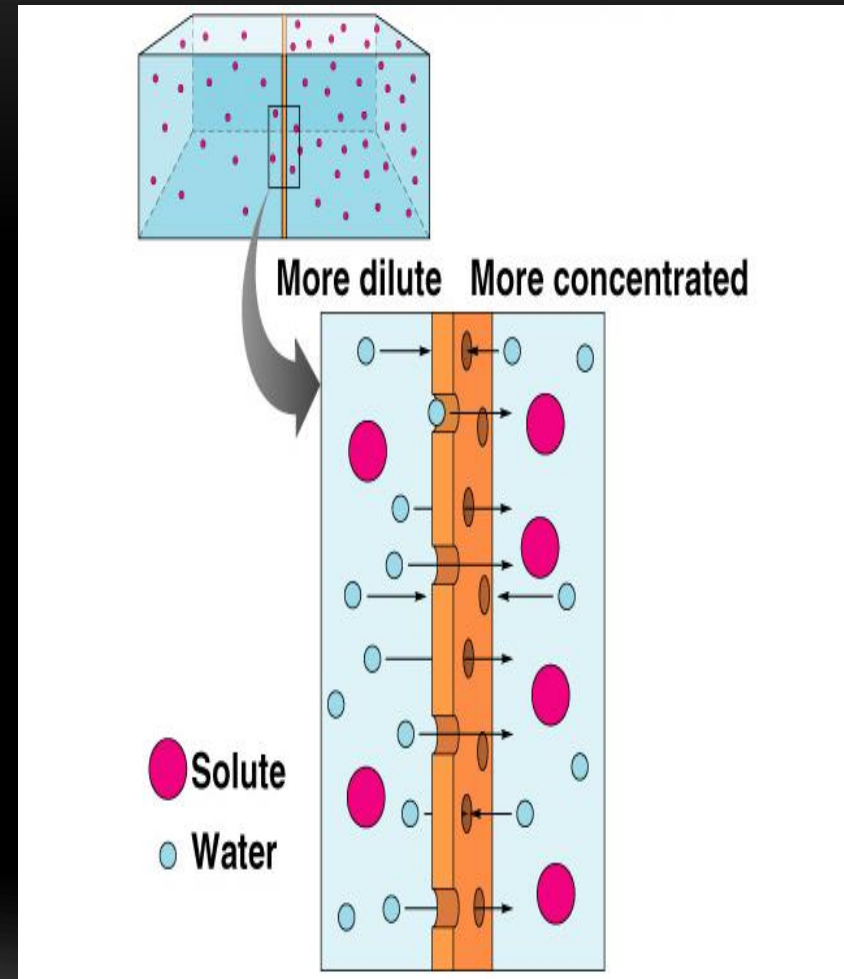


# Osmosis



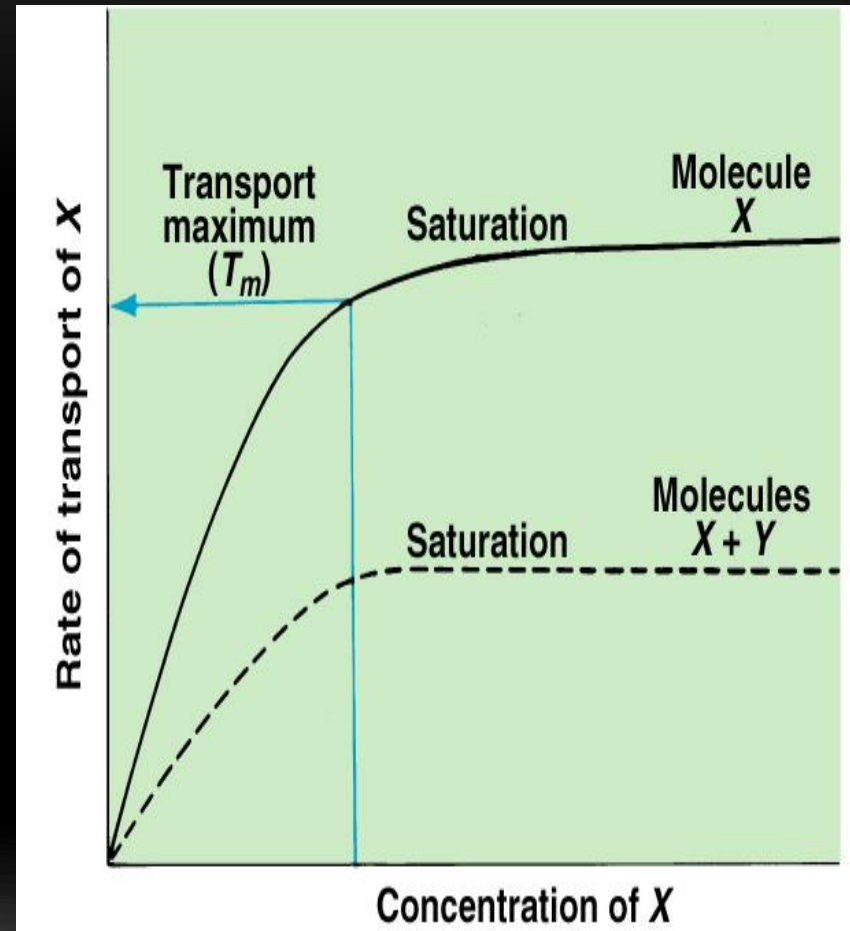
# OSMOSIS

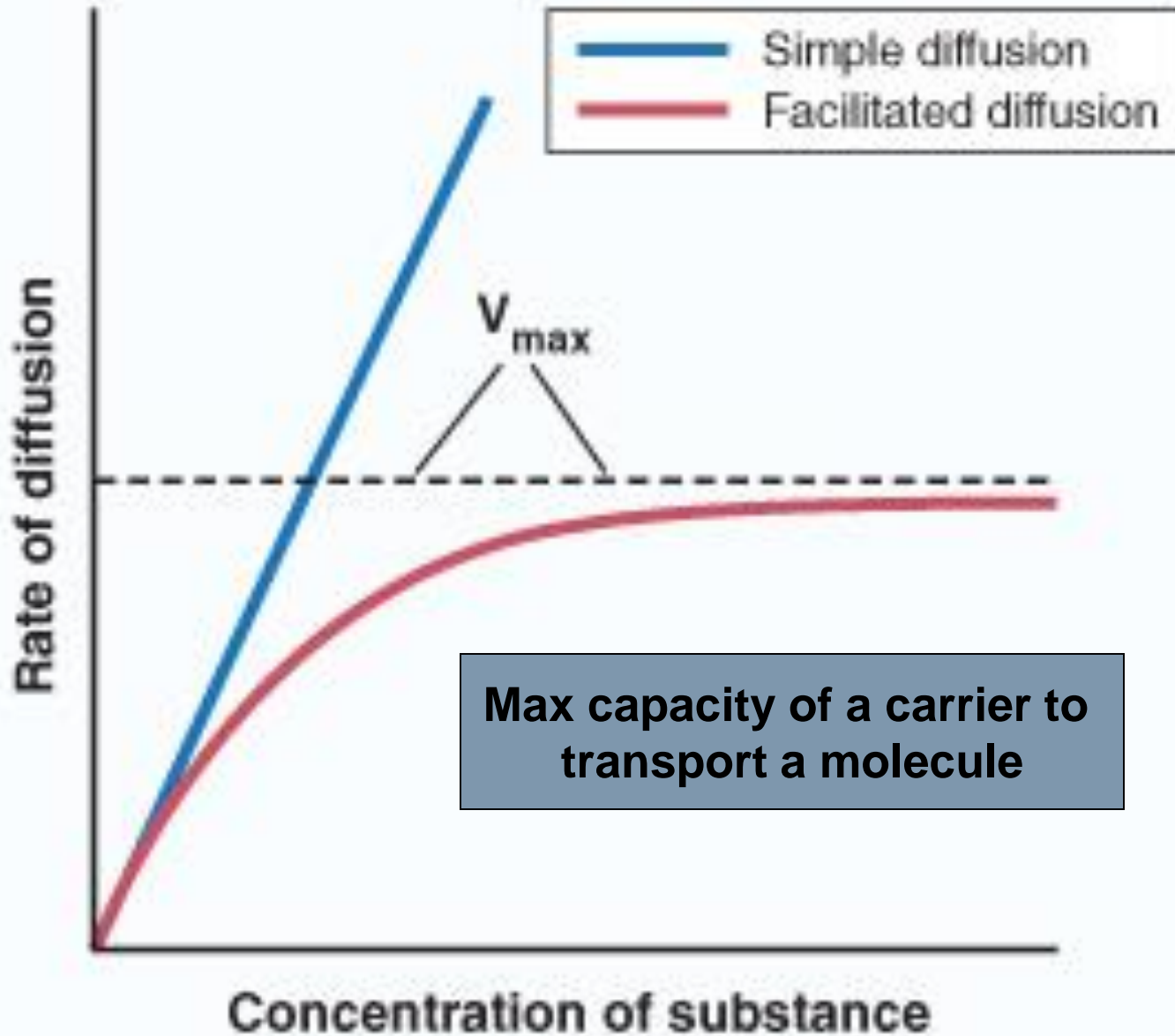
- **Net diffusion of  $H_2O$  across a selectively permeable membrane.**
- **Movement of  $H_2O$  from a high  $[H_2O]$  to lower  $[H_2O]$  area until equilibrium is reached.**
- **2 requirements for osmosis:**
  - **Must be difference in [solute] on the 2 sides of the membrane.**
  - **Membrane must be impermeable to the solute.**
- **Osmotically active solutes:**
  - **Solutes that cannot pass freely through the membrane.**



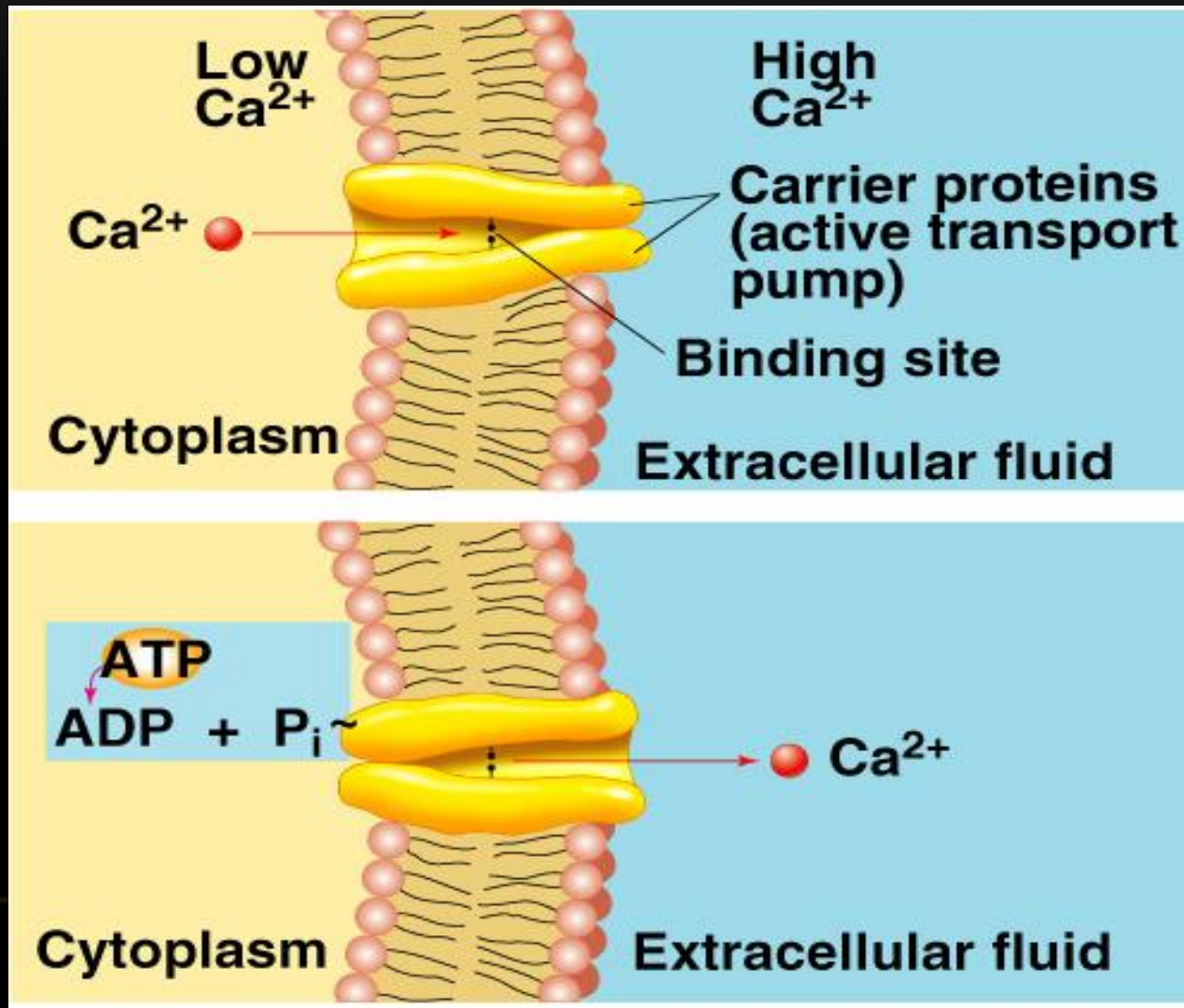
# CARRIER-MEDIATED TRANSPORT

- Molecules that are too large and polar to diffuse are transported across plasma membrane by protein carriers.
- **Characteristics of protein carriers:**
  - **Specificity:**
    - Interact with specific molecule only.
  - **Competition:**
    - Molecules with similar chemical structures compete for carrier site.
  - **Saturation:**
    - $T_m$  (transport maximum):
    - Carrier sites have become saturated.



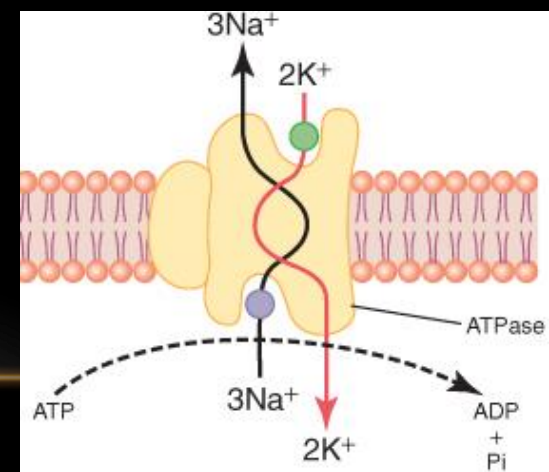
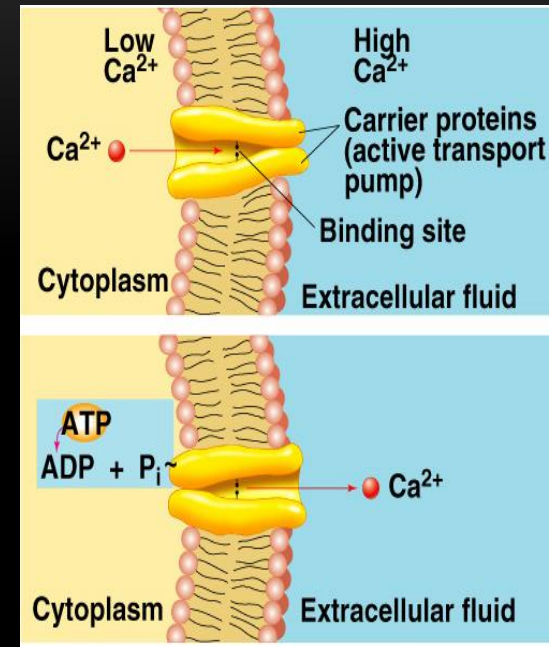


# Primary Active Transport

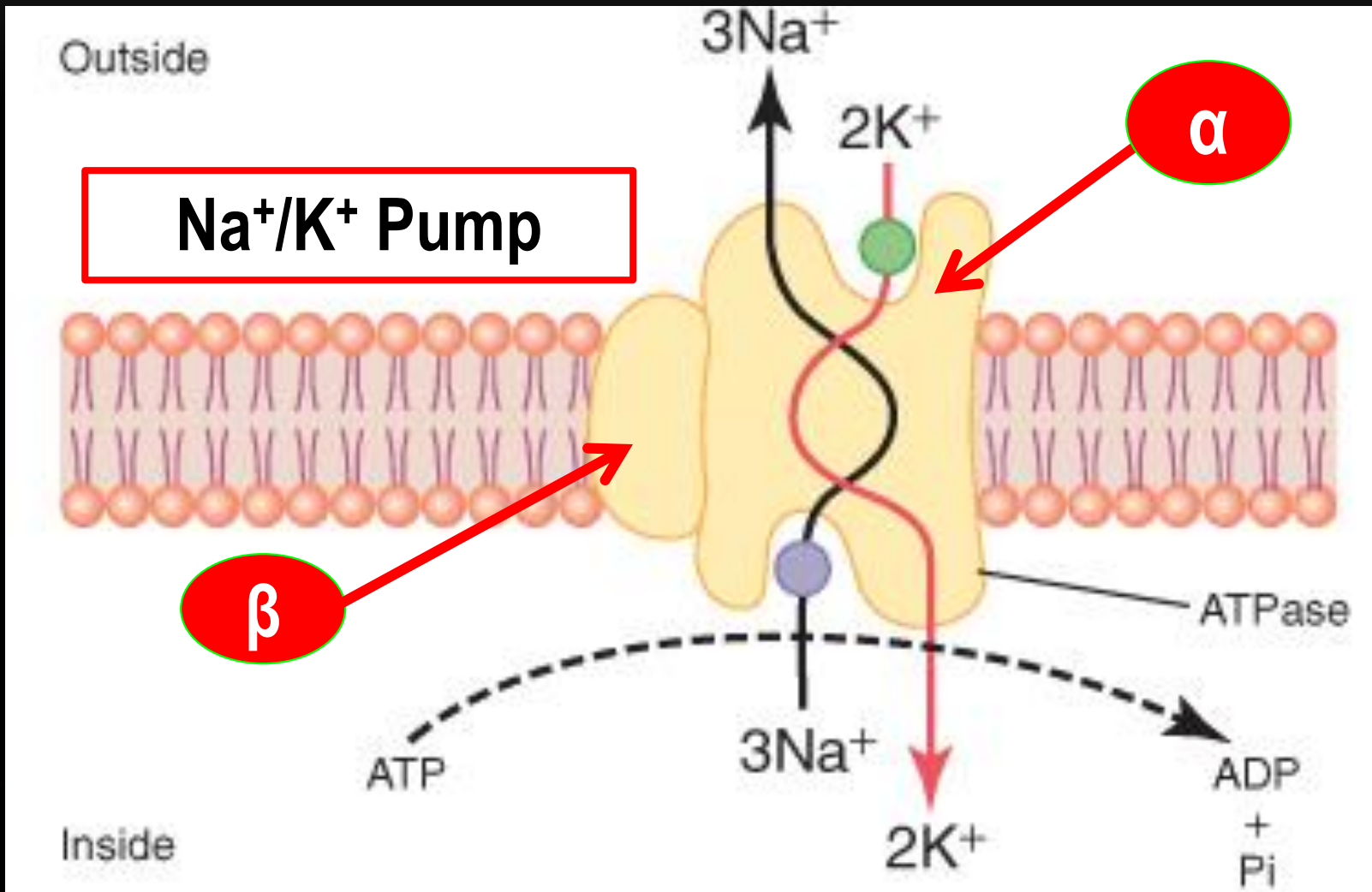


# PRIMARY ACTIVE TRANSPORT

- **Hydrolysis of ATP directly required for the function of the pump.**
- **Molecule or ion binds to “recognition site” on one side of pump.**
- **Binding stimulates phosphorylation (breakdown of ATP) of carrier protein.**
- **Carrier protein undergoes conformational change.**
- **Hinge-like motion releases transported molecules to opposite side of membrane.**

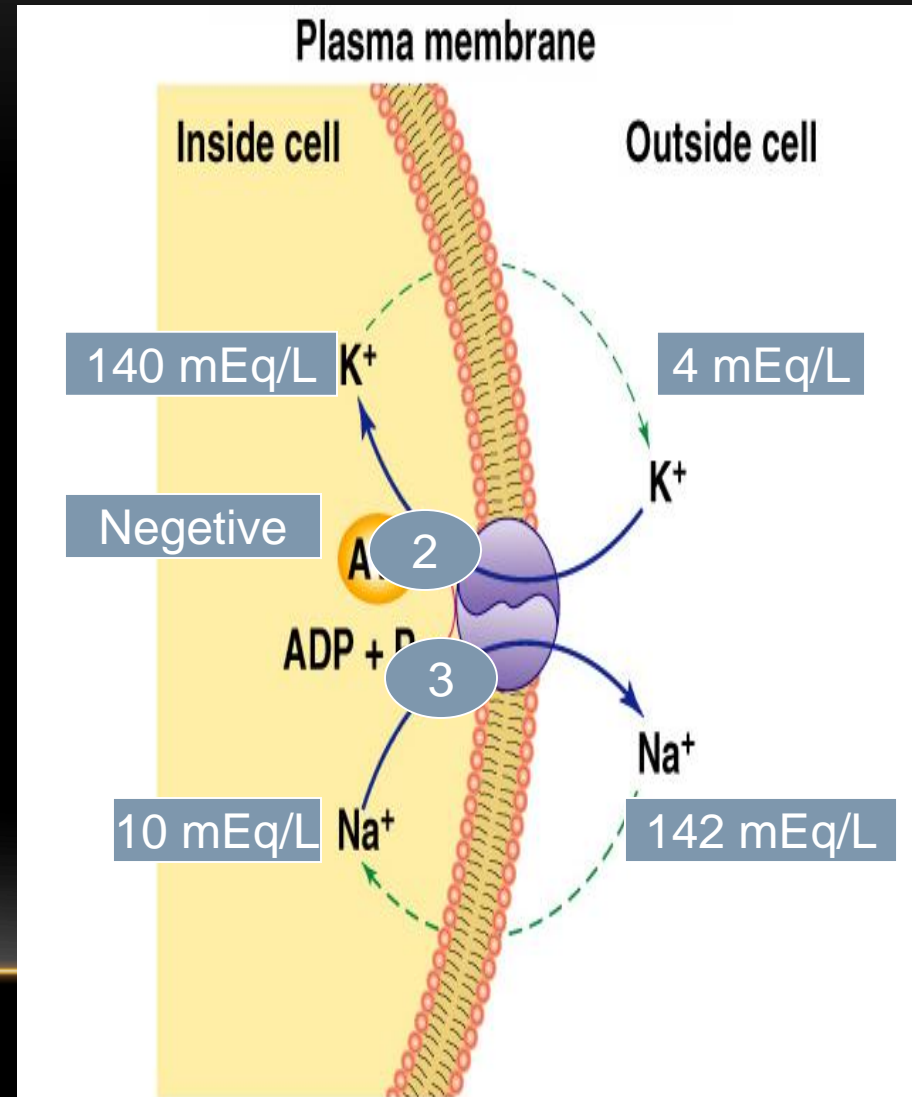


# Primary Active Transport



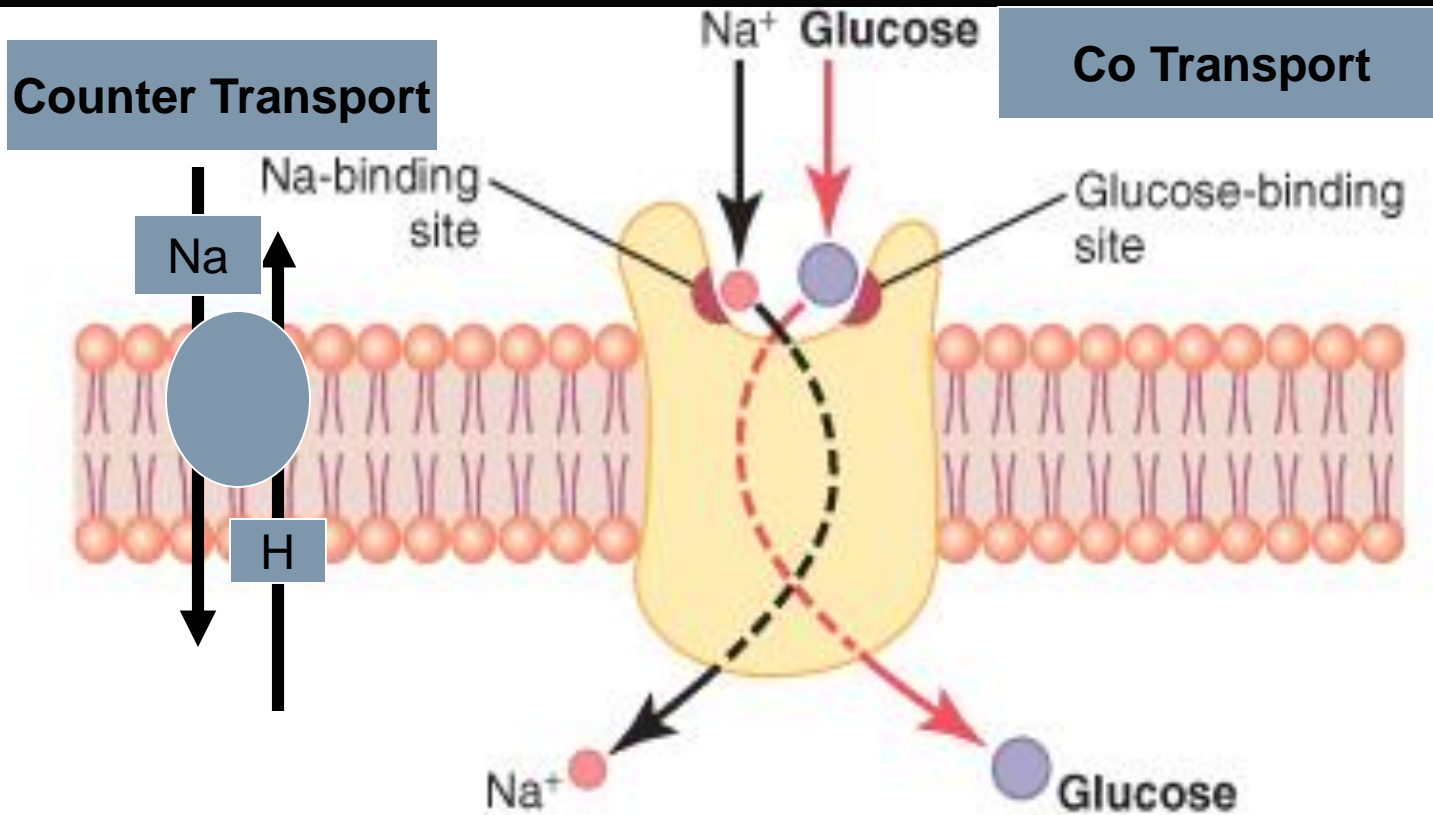
# Na<sup>+</sup>/K<sup>+</sup> PUMP

- Carrier protein is also an ATP enzyme that converts ATP to ADP and Pi.
  - Actively extrudes **3 Na<sup>+</sup>** and transports
  - **2 K<sup>+</sup>** inward against concentration gradient.
- Steep gradient serves 4 functions:
  - Provides energy for “coupled transport” of other molecules.
  - Regulates resting calorie expenditure and BMR.
  - Involvement in electrochemical impulses.
  - Promotes osmotic flow.





# Secondary Active Transport

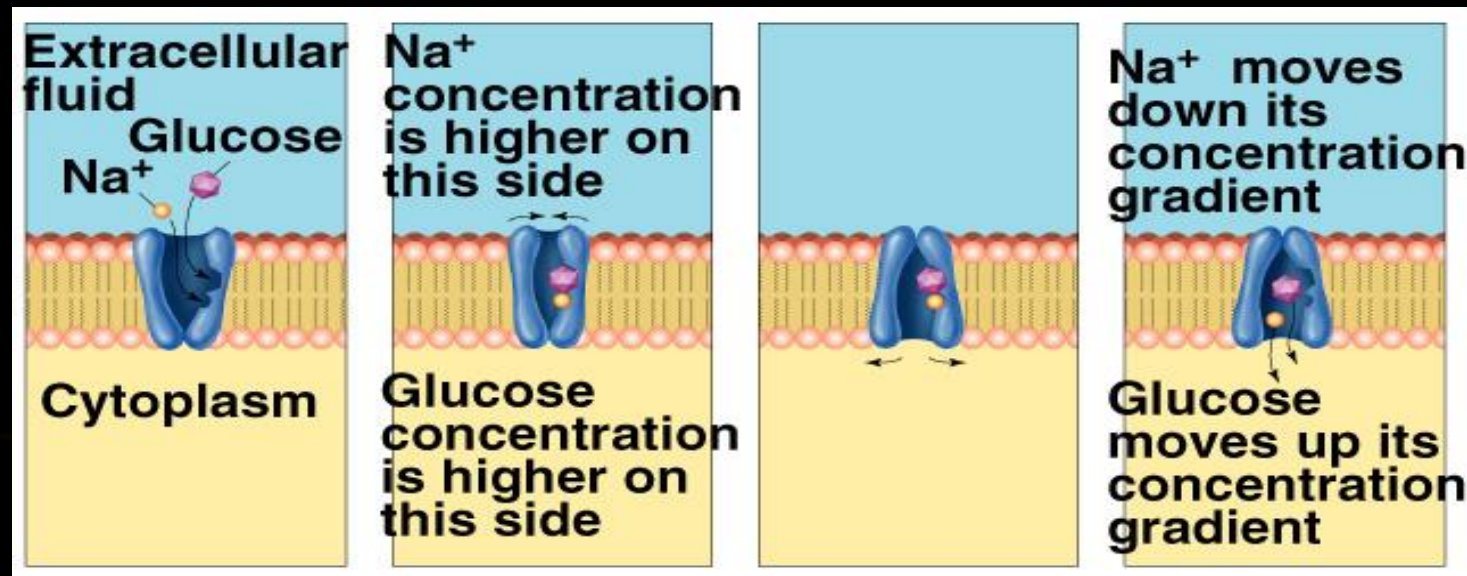


# SECONDARY ACTIVE TRANSPORT

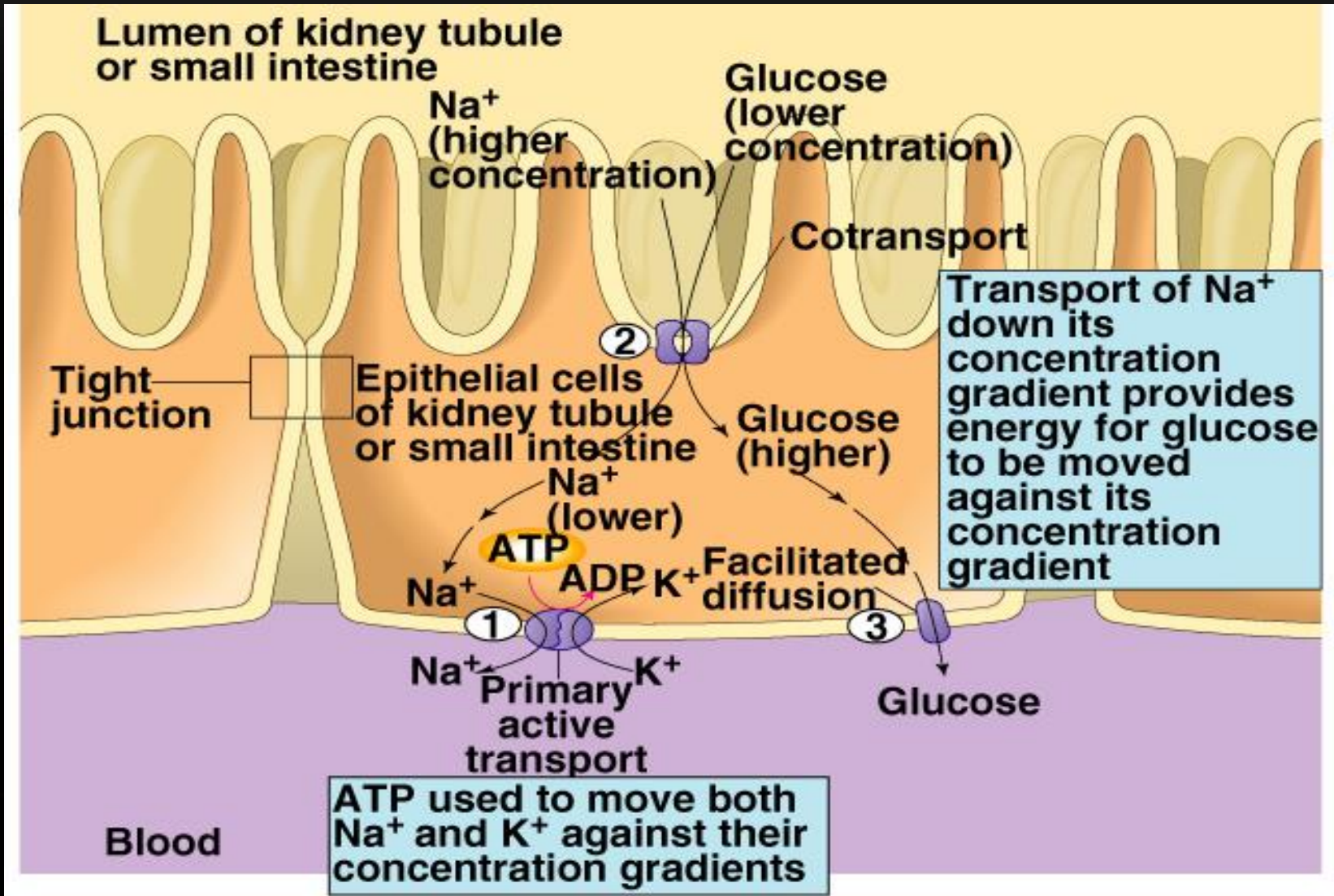
- **Co-transport (symport):**
    - Molecule or ion moving in the same direction as  $\text{Na}^+$ .
  - **Counter-transport (antiport):**
    - Molecule or ion moving in the opposite direction of  $\text{Na}^+$ .
  - **Glucose transport is an example of:**
    - Co-transport.
    - Facilitated diffusion.
-

# Secondary Active Transport

- **Coupled transport.**
- **Energy needed for “uphill” movement obtained from “downhill” transport of  $\text{Na}^+$ .**
- **Hydrolysis of ATP by  $\text{Na}^+/\text{K}^+$  pump required indirectly to maintain  $[\text{Na}^+]$  gradient.**



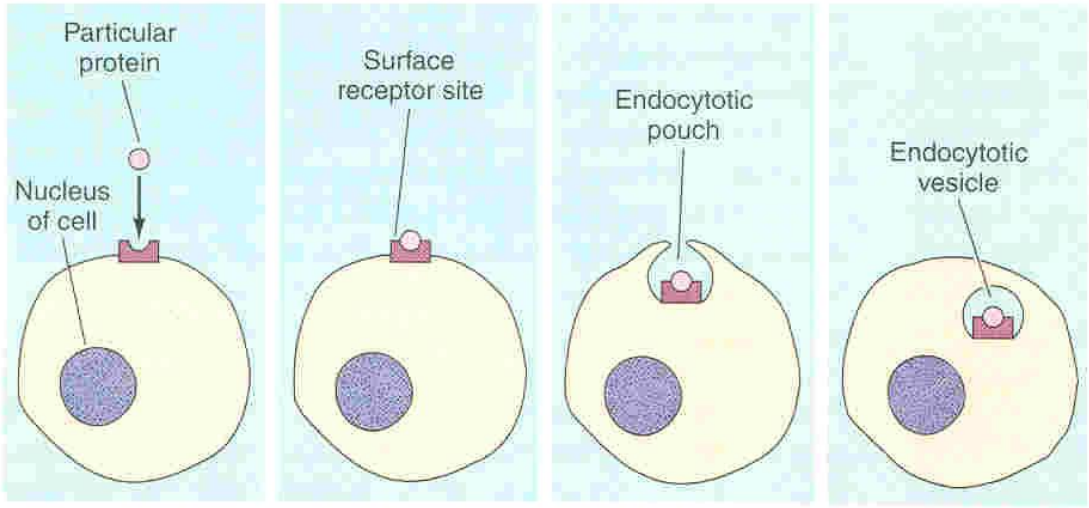
# SECONDARY ACTIVE TRANSPORT (CONT...)



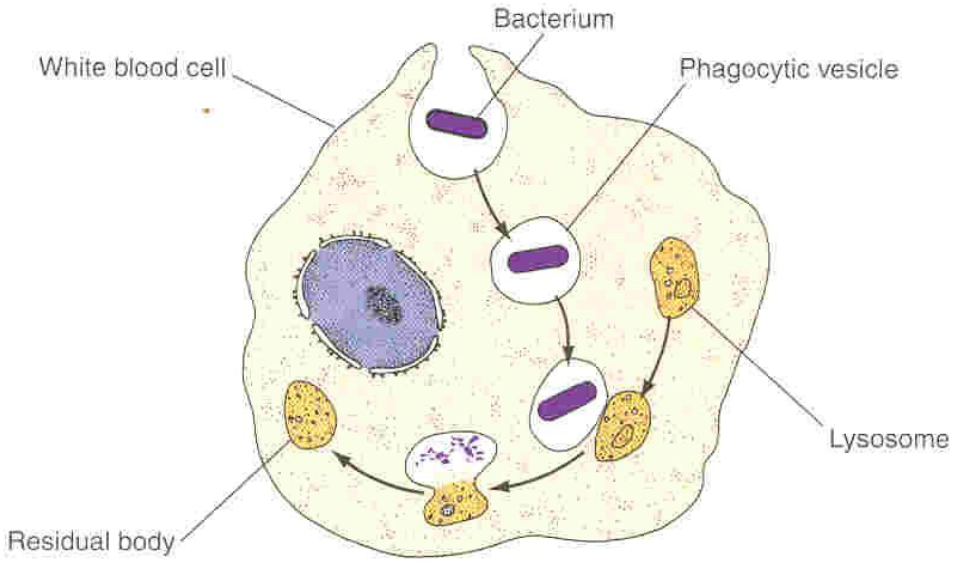
# BULK TRANSPORT

- Movement of many large molecules, that cannot be transported by carriers, at the same time.
- **Exocytosis:**
  - Fusion of the membrane-bound vesicles that contains cellular products with the plasma membrane.
- **Endocytosis:**
  - Exocytosis in reverse.
  - Specific molecules can be taken into the cell because of the interaction of the molecule and protein receptor.

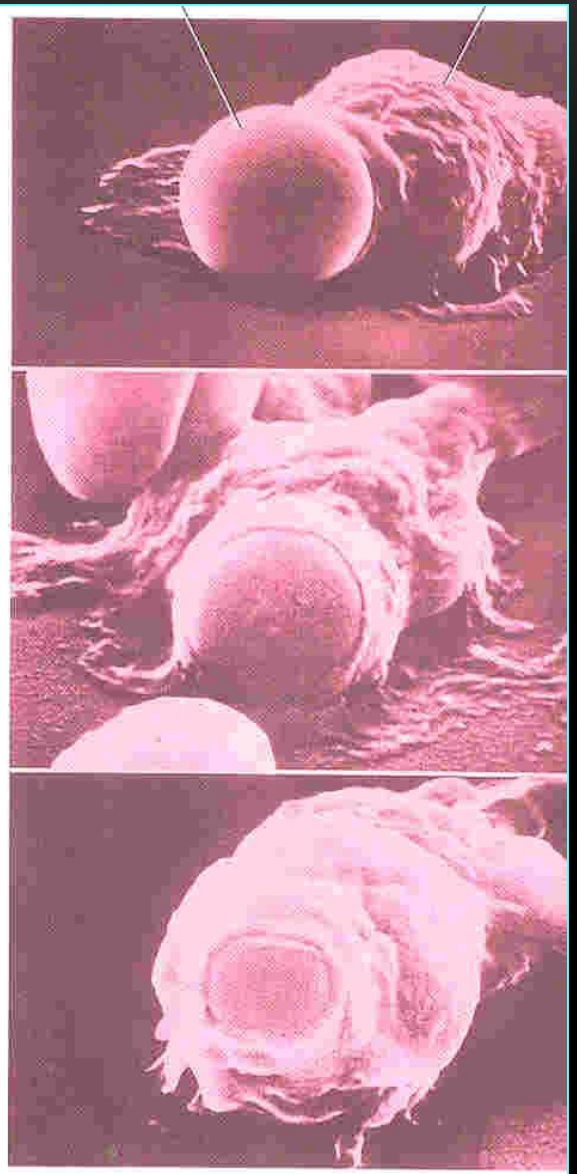




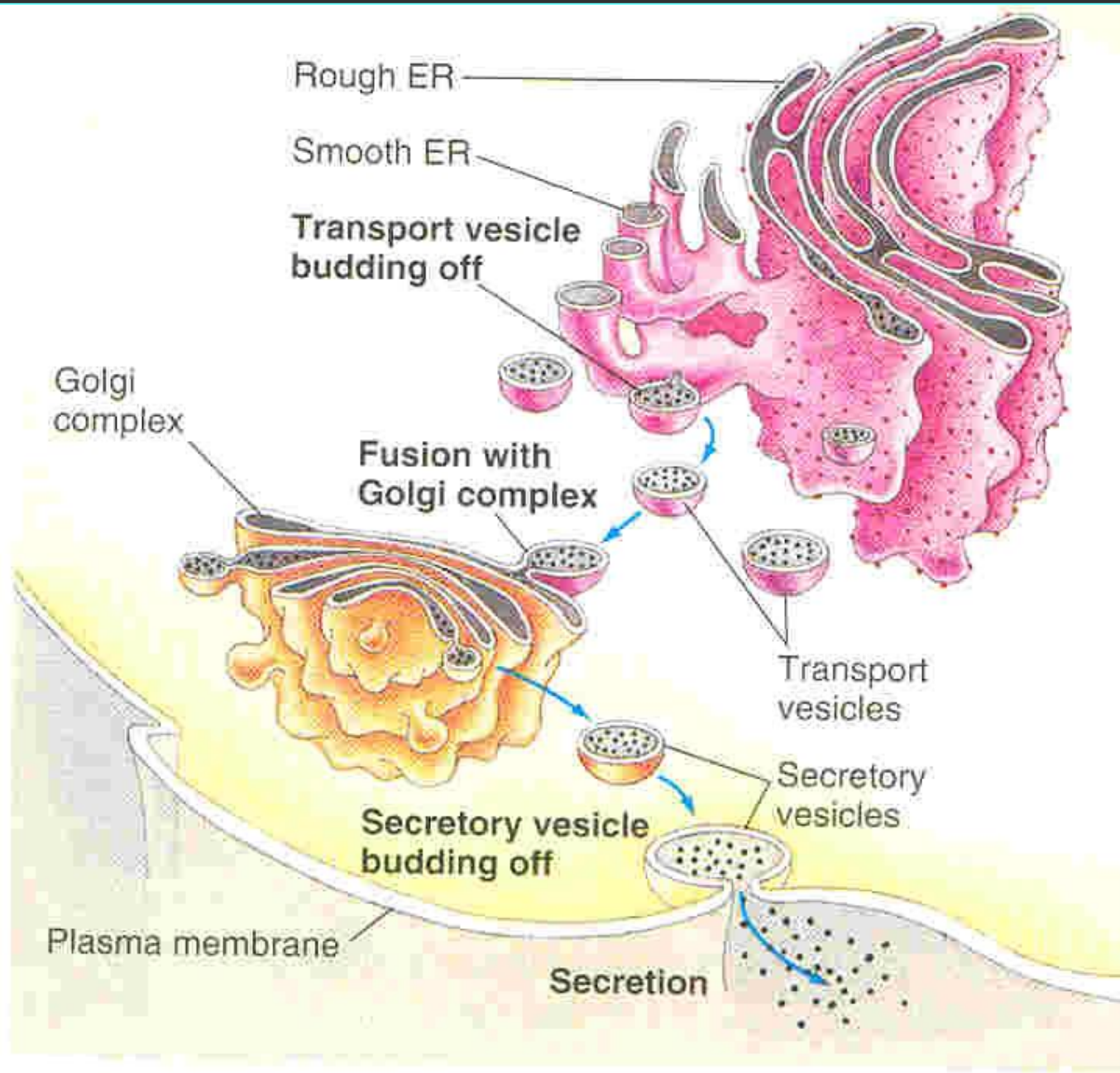
(a) Receptor-mediated endocytosis

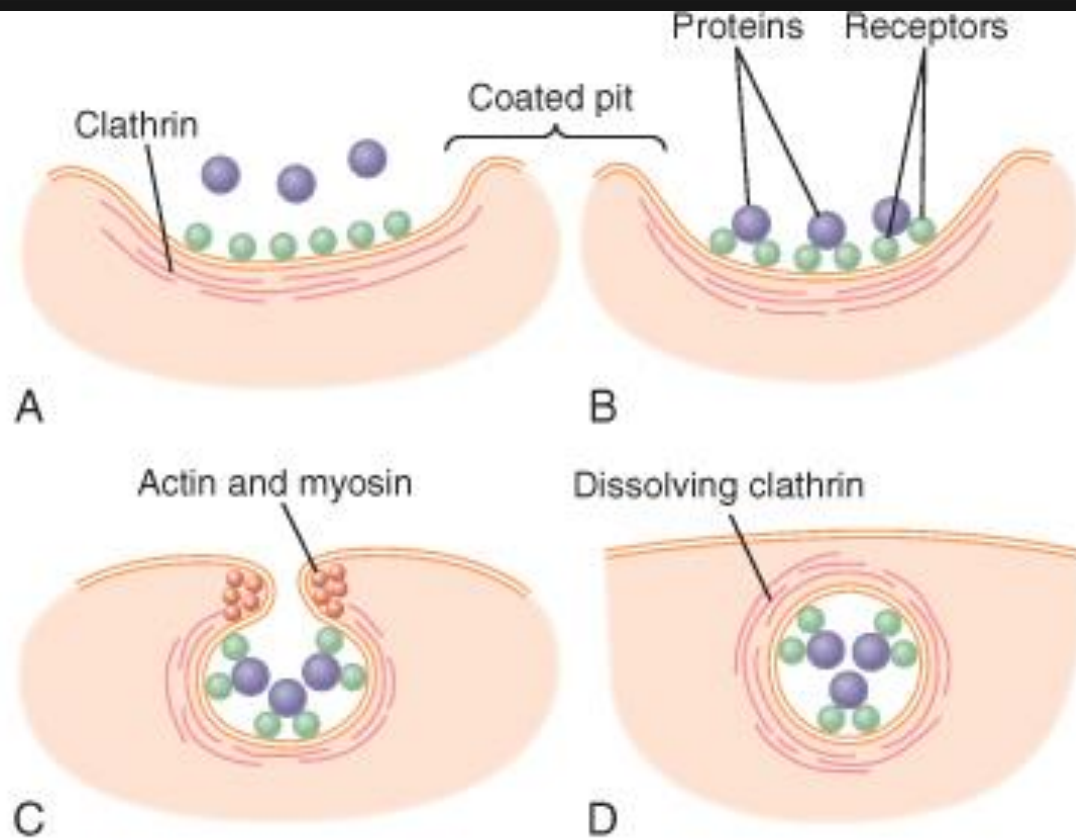


(b) Phagocytosis



(c)







# TONICITY AND ITS EFFECT ON MOVEMENT OF H<sub>2</sub>O

- **Isotonic:**

- **Equal tonicity osmolality (300 mosm/l) to plasma.**
  - **RBCs will not gain or lose H<sub>2</sub>O.**

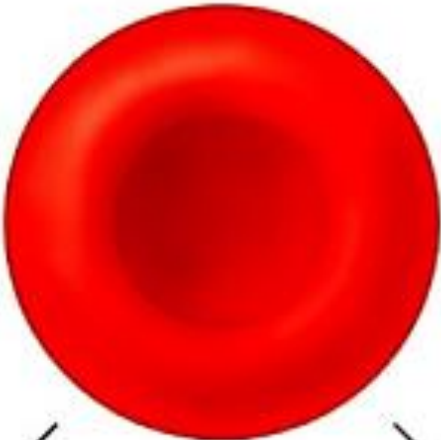
- **Hypotonic:**

- **Osmotically active solutes in a lower osmolality and osmotic pressure than plasma.**
  - **RBC will hemolyse.**

- **Hypertonic:**

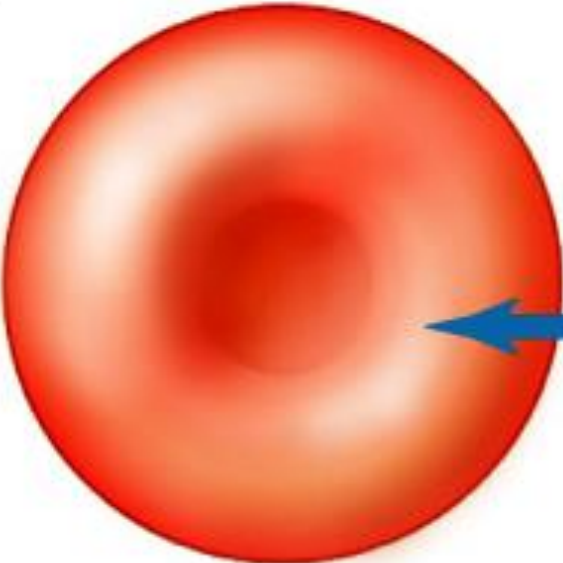
- **Osmotically active solutes in a higher osmolality and osmotic pressure than plasma.**
  - **RBC will crenate.**

# Isotonic solution

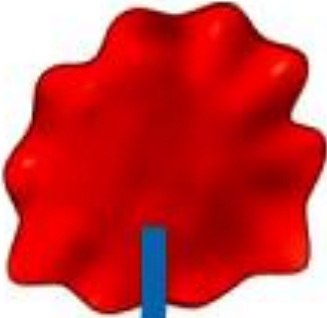


Hypotonic solution

Hypertonic solution



H<sub>2</sub>O



H<sub>2</sub>O



# TRANSPORT ACROSS EPITHELIAL MEMBRANES

- **In order for a molecule or ion to move from the external environment into the blood, it must first pass through an epithelial membrane.**

- **Absorption:**

- Transport of digestion products across the intestinal epithelium into the blood.

- **Reabsorption:**

- Transport of molecules out of the urinary filtrate back into the blood.

- **Transcellular transport:**

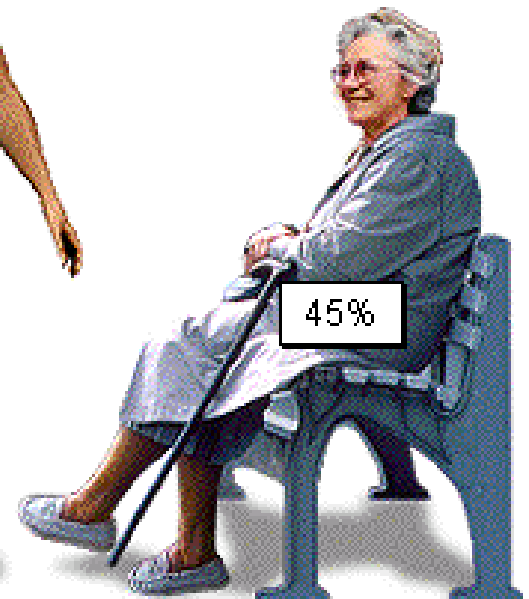
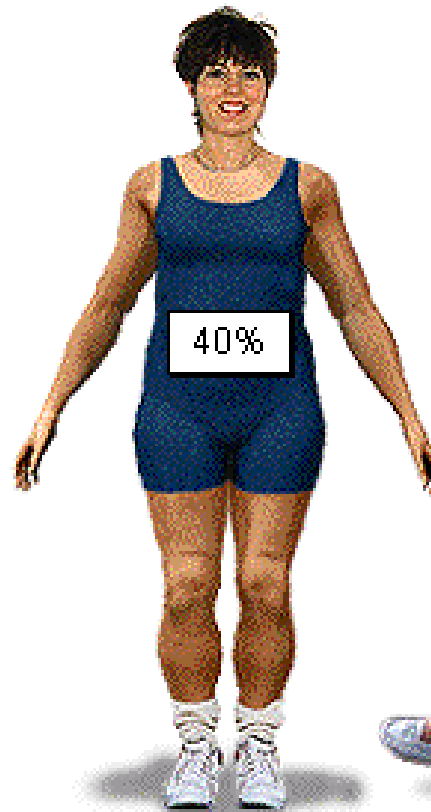
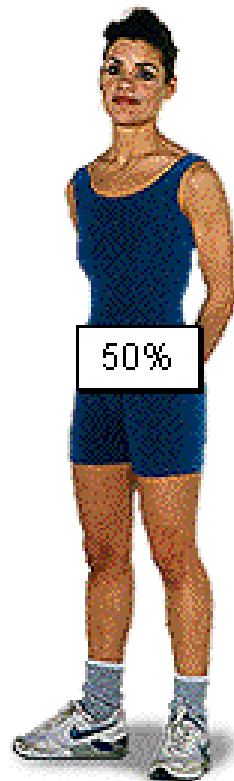
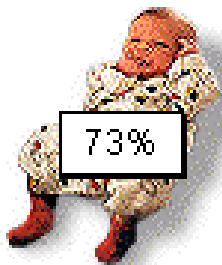
- Moves material through the cytoplasm of the epithelial cells.

- **Paracellular transport:**

- Diffusion and osmosis through the tiny spaces between epithelial cells.

# PERCENTAGE OF WATER IN THE BODY

Click each of the people below to determine the approximate percentage of water in their bodies.



# FACTORS AFFECTING

Total Body Water varies depending on body fat:

- infant: 73%
- male adult: 60%
- female adult: 40-50%
- effects of obesity
- Old age 45%

# DEHYDRATION

- What are the common causes of dehydration?
  - What are the common clinical features of dehydration?
  - How is dehydration classified?
-

# Are you dehydrated?

Check your urine



Yes



No

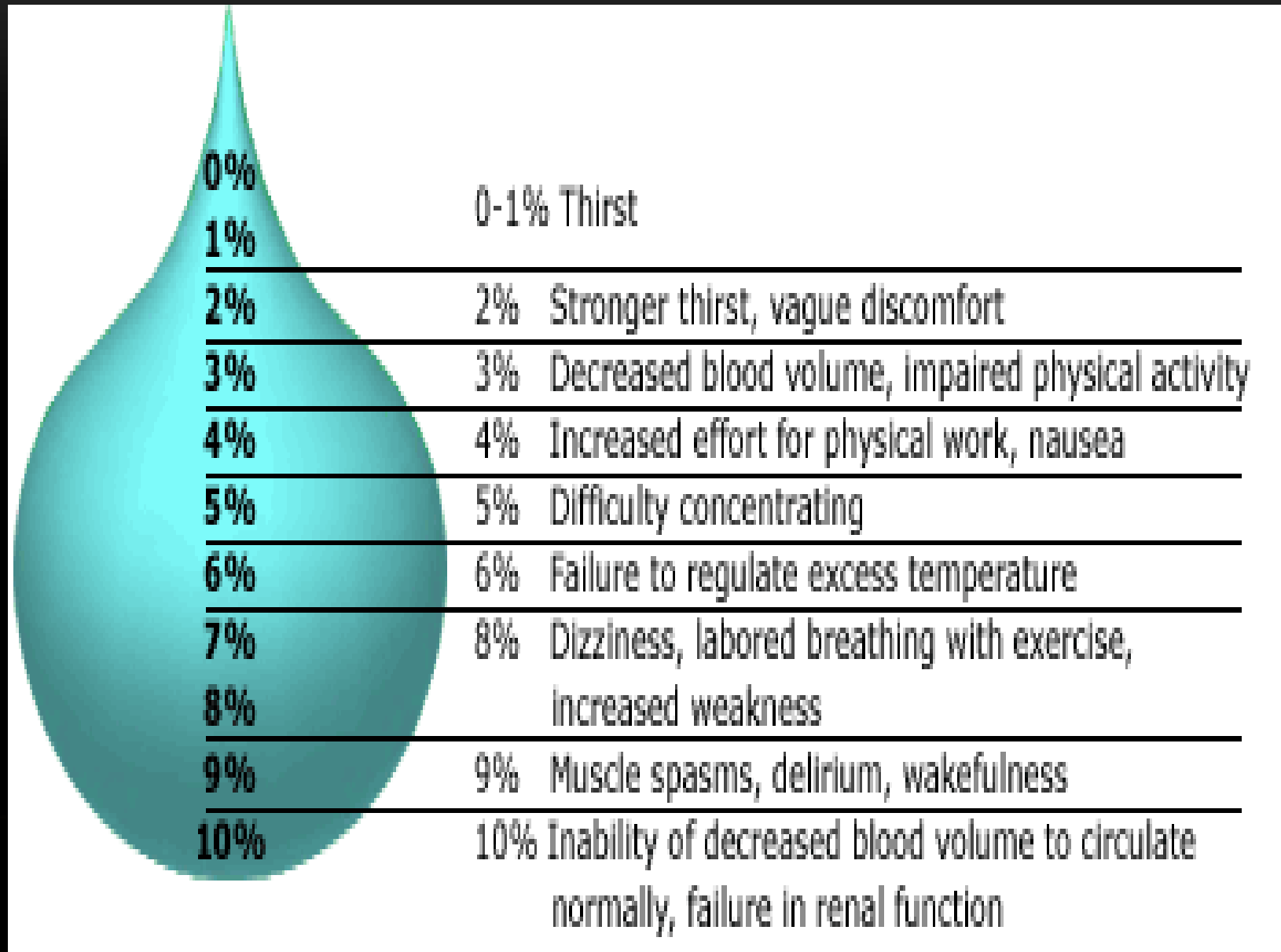
# Sign and Symptoms of Dehydration



- Dry or sticky mouth
- Lethargy
- Sunken eyes
- Weight loss
- Low or no urine input
- Dark yellow urine
- Poor skin turgor
- Delayed capillary refill
- Dizziness
- Confusion/changes in mental status
- Lack of tears/sweat
- Falls/difficulty walking
- Low blood pressure
- Rapid heart rate
- Abnormal labs/electrolytes

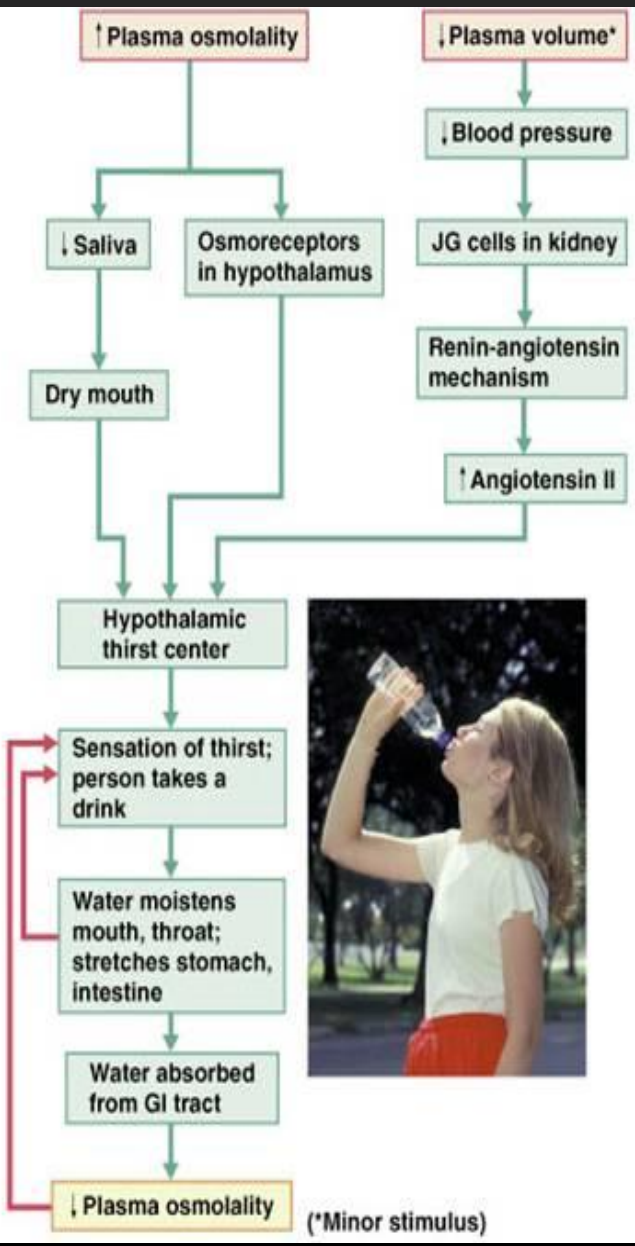


# SYMPTOMOLOGY OF DEHYDRATION



# MANAGEMENT OF DEHYDRATION

- What are the different methods used for rehydration?
    - **Volume replacement**
    - **Electrolyte replacement**
  - What are the substances used for rehydration?
    - **Uses of Isotonic, Hypotonic and Hypertonic solution**
-



- Key:**
- ← Increases, stimulates
  - ← Reduces, inhibits
  - Initial stimulus
  - Physiological response
  - Result

# Home Remedies for Dehydration



Increase Your Water Intake



Homemade ORS



Watery Fruits and Vegetables



Yogurt



Banana



Coconut Water



## Additional Tips

- Rest in a cool place and avoid any further exertion.
- During dehydration, stick to bland foods for a couple of days.
- To prevent dehydration, take a refillable water bottle with you when going out.

To explore more, visit

**Top10**  
Home Remedies

[www.Top10HomeRemedies.com](http://www.Top10HomeRemedies.com)

THANK YOU

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