



Pharmacokinetics II: Bioavailability and Distribution

Pharmacology Department

By the end of the lectures, students should be able to define the following:

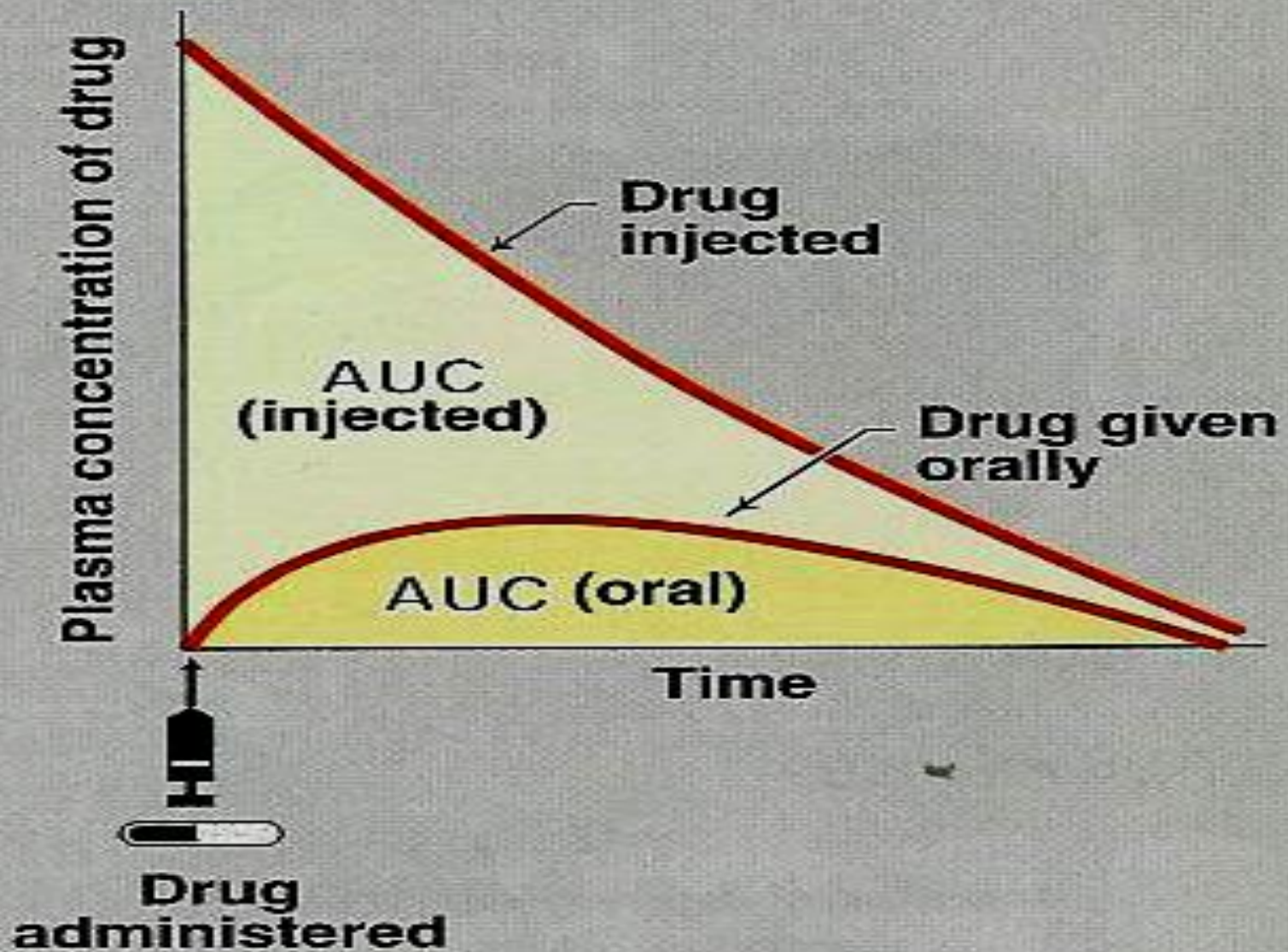
- ▶ **Major body fluid compartments**
- ▶ **Concept of compartments.**
- ▶ **Apparent volume of distribution (vd).**
- ▶ **Plasma protein binding.**
- ▶ **Tissue binding.**

Bioavailability

▶ Is the amount of unchanged drug that enters systemic circulation after administration and becomes available to produce pharmacological actions

▶ **Bioavailability (F) = $\frac{\text{AUC (oral)}}{\text{AUC (I.V.)}} \times 100$**

$$\text{Bioavailability} = \frac{\text{AUC oral}}{\text{AUC injected}} \times 100$$



Bioavailability

- ▶ **I.V. provides 100% bioavailability i.e. $F=1$.**
- ▶ Subcutaneous, intramuscular, oral, rectal, and other extra vascular routes of administration require that the drug be absorbed first, which can reduce bioavailability.

Absolute bioavailability

- ▶ The bioavailability of a drug after administration by any route is compared to its intravenous standard formulation.

Relative bioavailability

- ▶ is determined when two products are compared to each other, not to an intravenous standard.
- ▶ This is commonly calculated in the drug industry to determine that the generic formulation is bioequivalent to another formulation.
- ▶ e.g. **Tylenol** (paracetamol 500 mg) compared to **Panadol** (paracetamol 500 mg).

Relative bioavailability

- ▶ is important to get an idea of how **different formulations** or **routes of administration** differ in their bioavailability.
- ▶ dosage adjustment is required when changing formulations or routes of administration.

Bioequivalence

- ▶ Two pharmaceutical products are **bioequivalent** when the rate and extent of bioavailability of active ingredients in two products are the same.

Factors affecting bioavailability:

- Factors controlling drug absorption
- First pass effect

Factors affecting absorption :

- **Route of administration.**
- **Dosage forms** (depending on particle size and disintegration, ease of dissolution).

(suspension > capsule > tablet)

- **Molecular weight of drug.**
- **Lipid solubility**
- **Degree of ionization**
- **Drug solubility (aqueous preparation better than oily, suspension preparations)**
- **Chemical instability in gastric pH**
(Penicillin & insulin)

Factors affecting absorption :

- **Surface area available for absorption.**
 - small intestine has large surface area than stomach due to intestinal microvilli.
- **Blood flow to absorptive site**
 - **greater** blood flow increases bioavailability
 - Intestine has greater blood flow than stomach
- **Intestinal motility (transit time)**
 - Diarrhea reduce absorption

➤ Gastric emptying

- drugs that increase gastric emptying enhances absorption (metoclopramide).

➤ Drug interactions

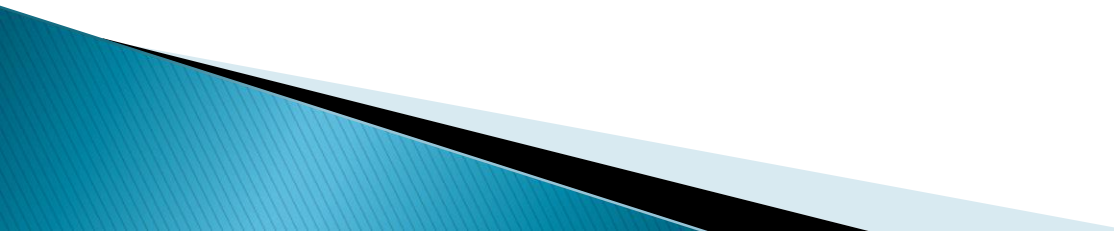
➤ Food

- **Slow** gastric emptying
- generally slow absorption
- Tetracycline, aspirin, penicillin V
- **A fatty meal** increase the absorption of fat soluble antifungal drug (e.g. griseofulvin)

Distribution

Distribution

Is the process by which drugs leave blood circulation and enters the interstitium and/or the cells of the tissues.



**Drug
Administration**



Absorption

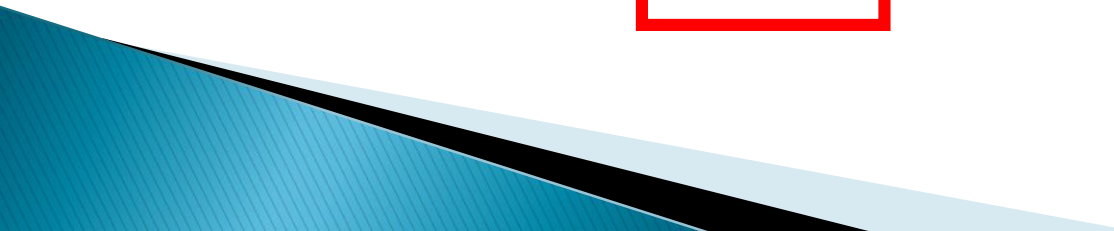


Blood



Extracellular

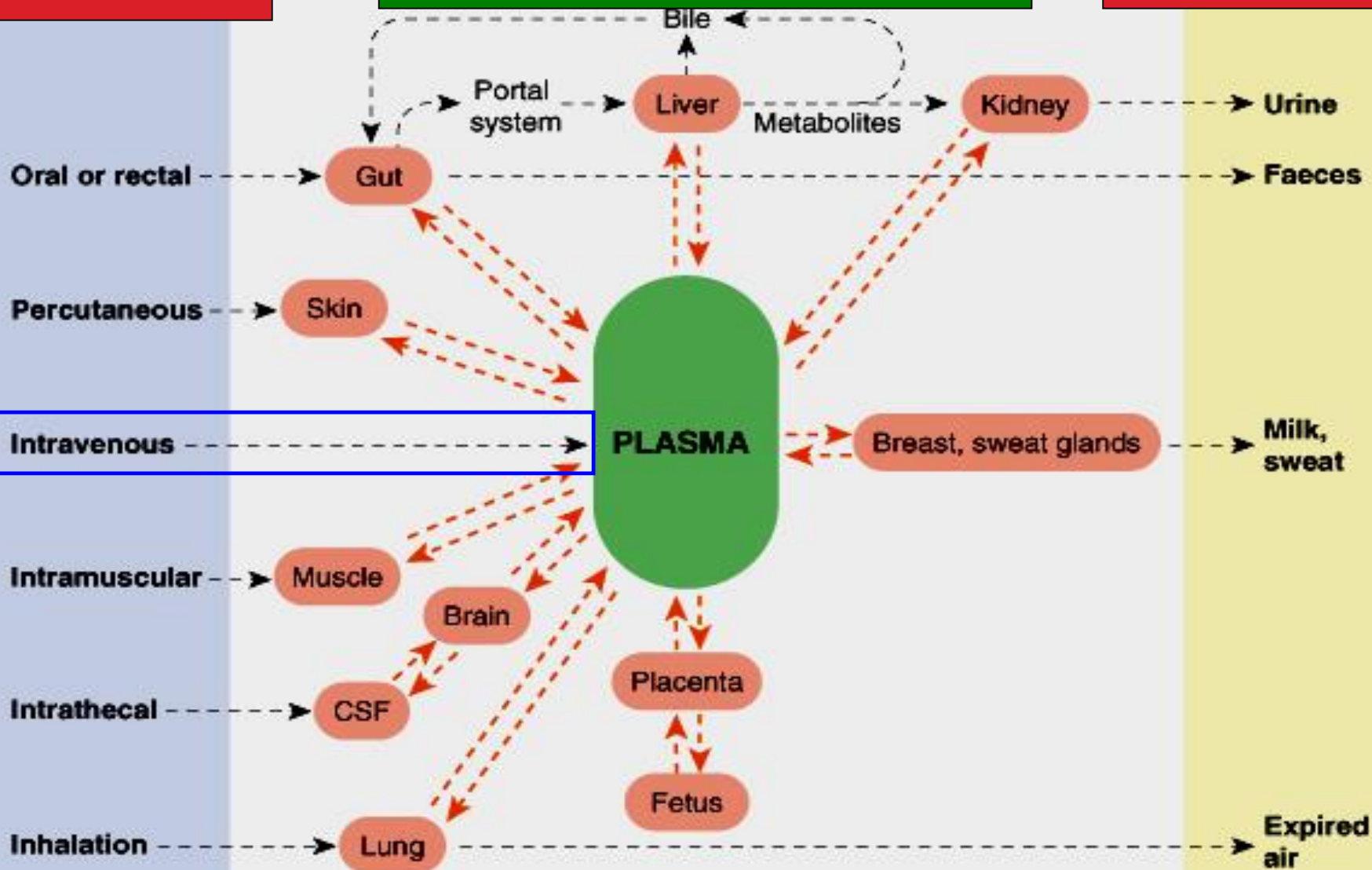
Intracellular

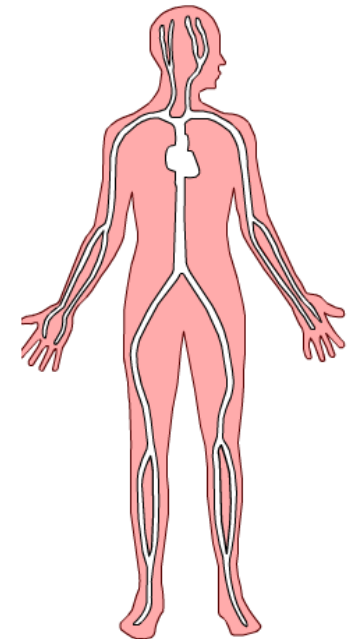
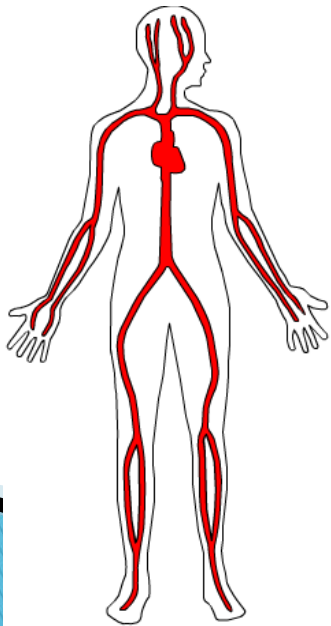
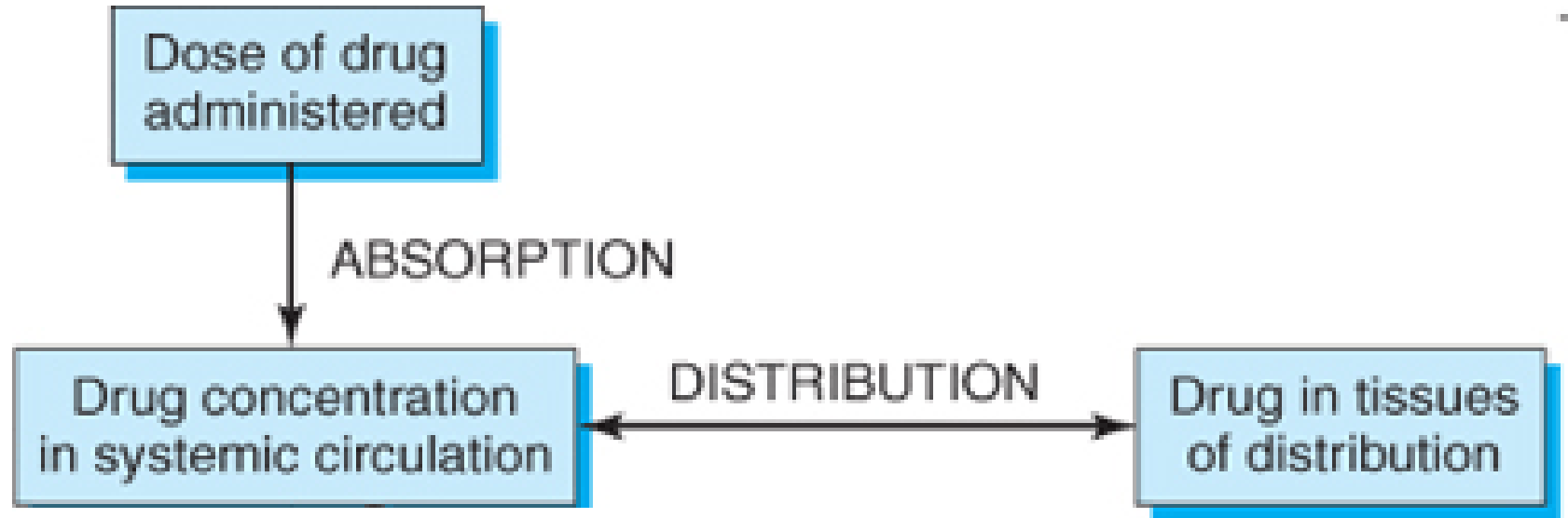


Sites of Administration

Absorption & distribution

Elimination





Apparent Volume of Distribution (Vd)

is the ratio of drug amount in the body (dose) to the concentration of drug in blood.

$$Vd \text{ (L)} = \frac{\text{Dose (mg)}}{\text{plasma concentration (mg/L)}}$$

Why is Vd important?

- To calculate loading dose
- Large Vd = means **long duration of action**

The major body fluid compartments are

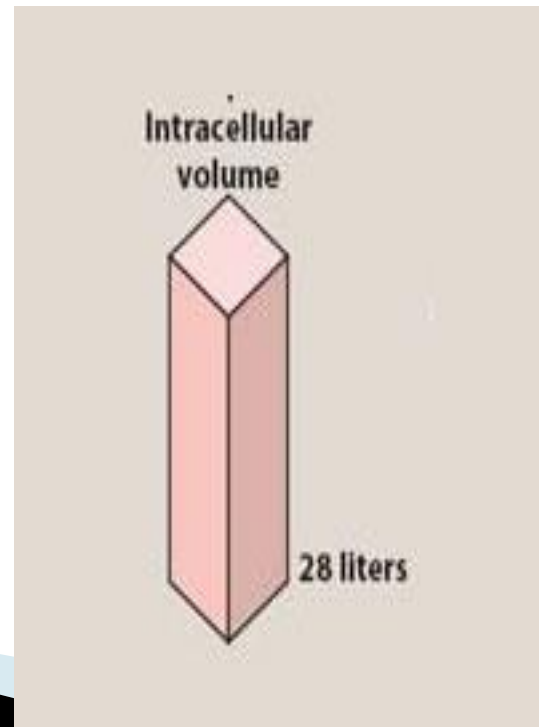
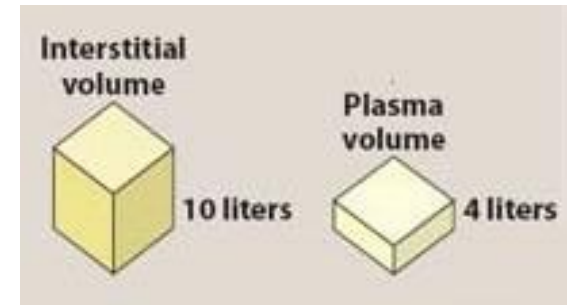
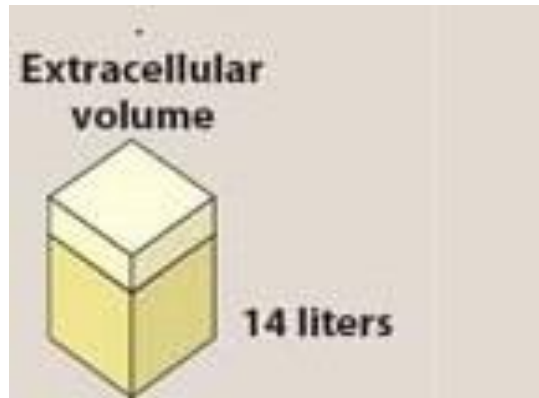
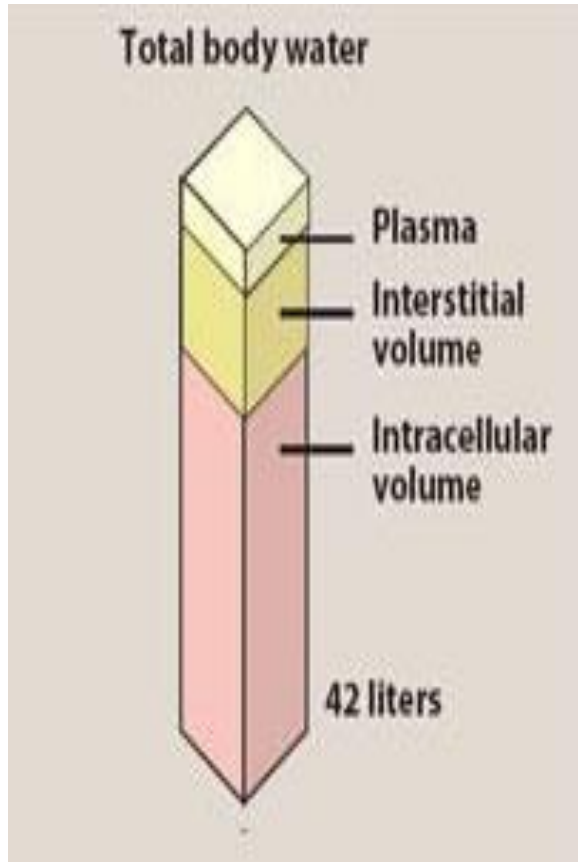
Extracellular fluid (22%)

- Plasma (5 % of body weight = 4 liters).
- Interstitial fluid (16 % = 10 liters).

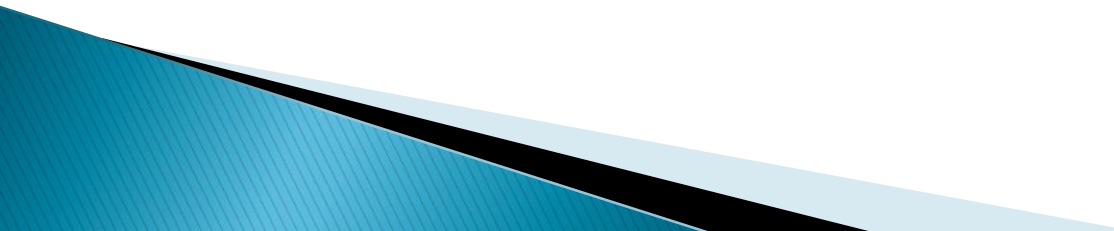
Intracellular fluid (35 %)

fluid present inside all cells in the body (28 L).

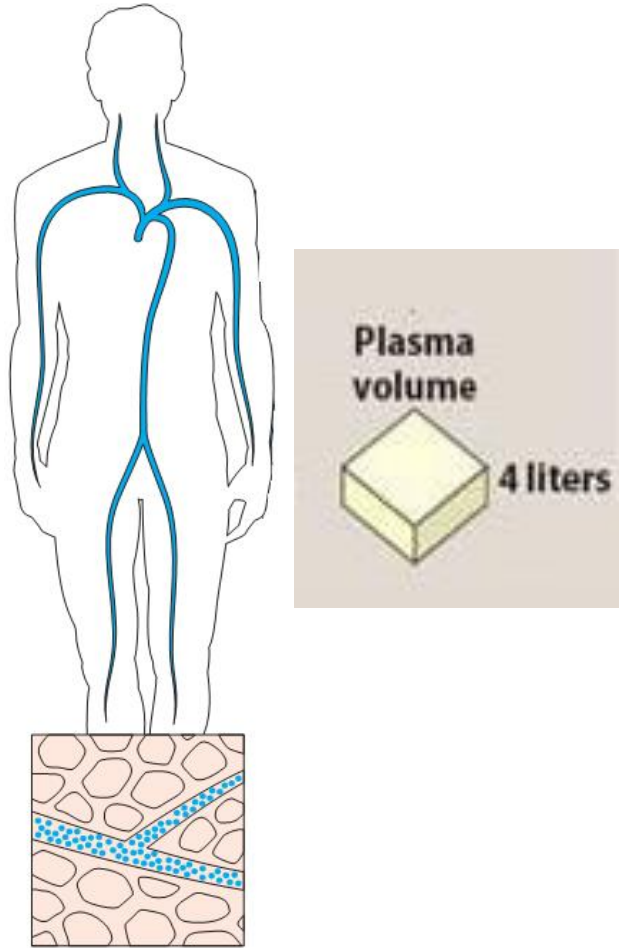
Volume of distribution



Drugs may distribute through:

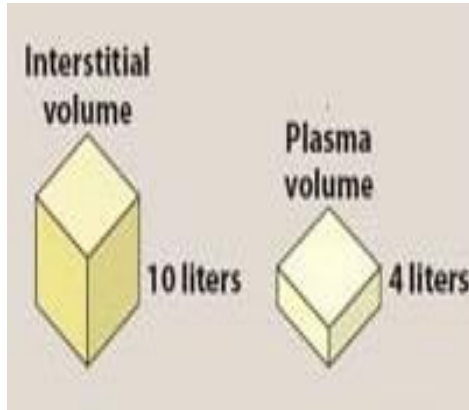
- ▶ One compartment
 - ▶ Two compartments
 - ▶ Multi-compartments
- 

Plasma compartment



- ▶ V_d : around 4 L.
- ▶ Very high molecular weight drugs, or drugs that bind to plasma proteins
- ▶ Can not moves across endothelial cells of capillaries
- ▶ Drugs are trapped in blood
- ▶ Example: heparin 4L

Extracellular fluid



- ▶ Distribute through extracellular fluids.
- ▶ Pass endothelium into interstitial fluids BUT can not cross cell membranes to intracellular fluids.
- ▶ Drugs that have a low molecular weight but are hydrophilic.
- ▶ Vd: between 4 and 14 L.
 - E.g. atracuronium 11 L

Total body water (extracellular and intracellular)



- ▶ Diffusion to intracellular fluid
- ▶ For lipid soluble drugs
- ▶ Vd equal to total body water.
 - Ethanol 38 L (34–41)
- ▶ Drug that binds strongly to tissues. Vd higher than total body water.
 - Digoxin: 385 L

Volume of Distribution (V_d)

Drugs with low V_d

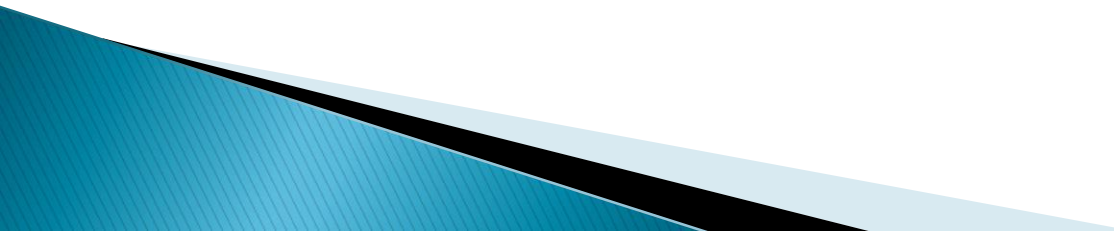
- ▶ **distributed in extracellular compartments (plasma & interstitial fluid).**
- ▶ **Polar comp or lipid insoluble drugs. e.g. gentamycin, atracurium**
- ▶ **High MW e.g. heparin – insulin.**
- ▶ **High plasma protein binding e.g. warfarin (anticoagulant).**
- ▶ **Do not cross BBB or placental barriers.**

Volume of Distribution (V_d)

Drugs with high V_d

- ▶ Have higher concentrations in tissues than in plasma.
- ▶ Lipid soluble.
- ▶ Distributed intracellularly
- ▶ e.g. digoxin, phenytoin, morphine

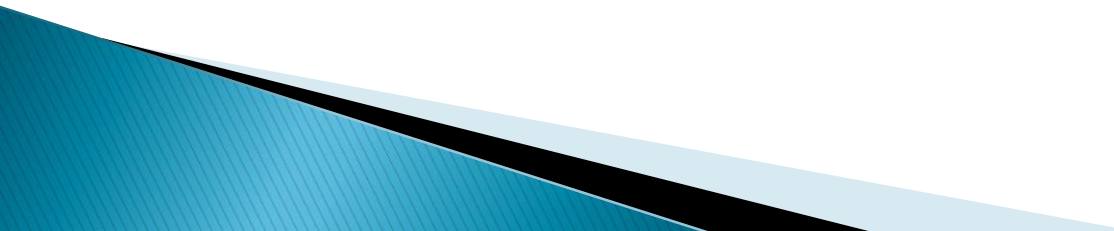
FACTORS AFFECTING DISTRIBUTION

1. Cardiac output and blood flow.
 2. **Physical and chemical properties of the drug.**
 - Molecular weight
 - Pka.
 - Lipid solubility.
 3. Capillary Permeability
 4. Plasma protein binding
 5. Tissue binding.
- 

Blood flow to organs

- ▶ The greater the blood flow to tissues, the more distribution that occurs from plasma to interstitial fluids.
- ▶ Drugs distribute more rapidly to brain, liver and kidney > more than skeletal muscles & fat.

Physical and chemical properties of drug

- ▶ Most lipid soluble drugs (unionized, uncharged, nonpolar) cross biological membranes
 - ▶ Hydrophilic drugs (ionized, charged, polar) do not readily cross membranes but go through slit junctions in endothelial cells of capillaries.
- 

Capillary permeability

- ▶ Endothelial cells of capillaries in tissues other than brain have wide slit junctions allowing easy movement, permeation and distribution.
- ▶ Brain has tight junction **Blood Brain Barrier (BBB)**.

Blood brain barrier (BBB):

- ▶ Only lipid soluble drugs or actively transported drugs can cross BBB.
- ▶ Hydrophilic drugs (ionized or polar drugs) can not cross BBB.
- ▶ Inflammation as in meningitis increase permeability to hydrophilic drugs
- ▶ e.g. penicillin & gentamycin

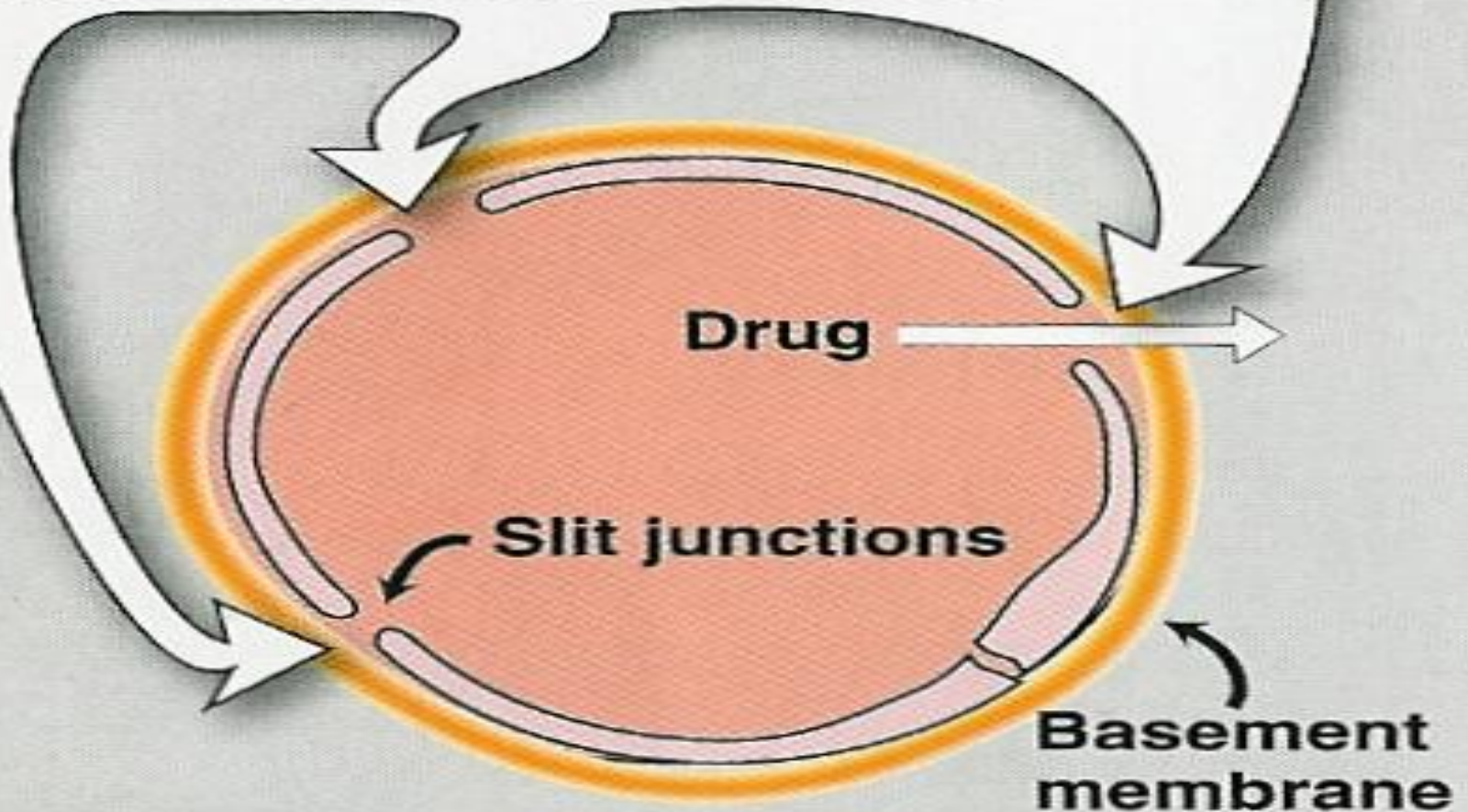
Placental barrier

- ▶ Lipid soluble drugs can cross placental barrier and enter the fetal blood.

A

Structure of endothelial cells in the liver

Large fenestrations allow drugs to exchange freely between blood and interstitium in the liver.



B

Structure of a brain capillary

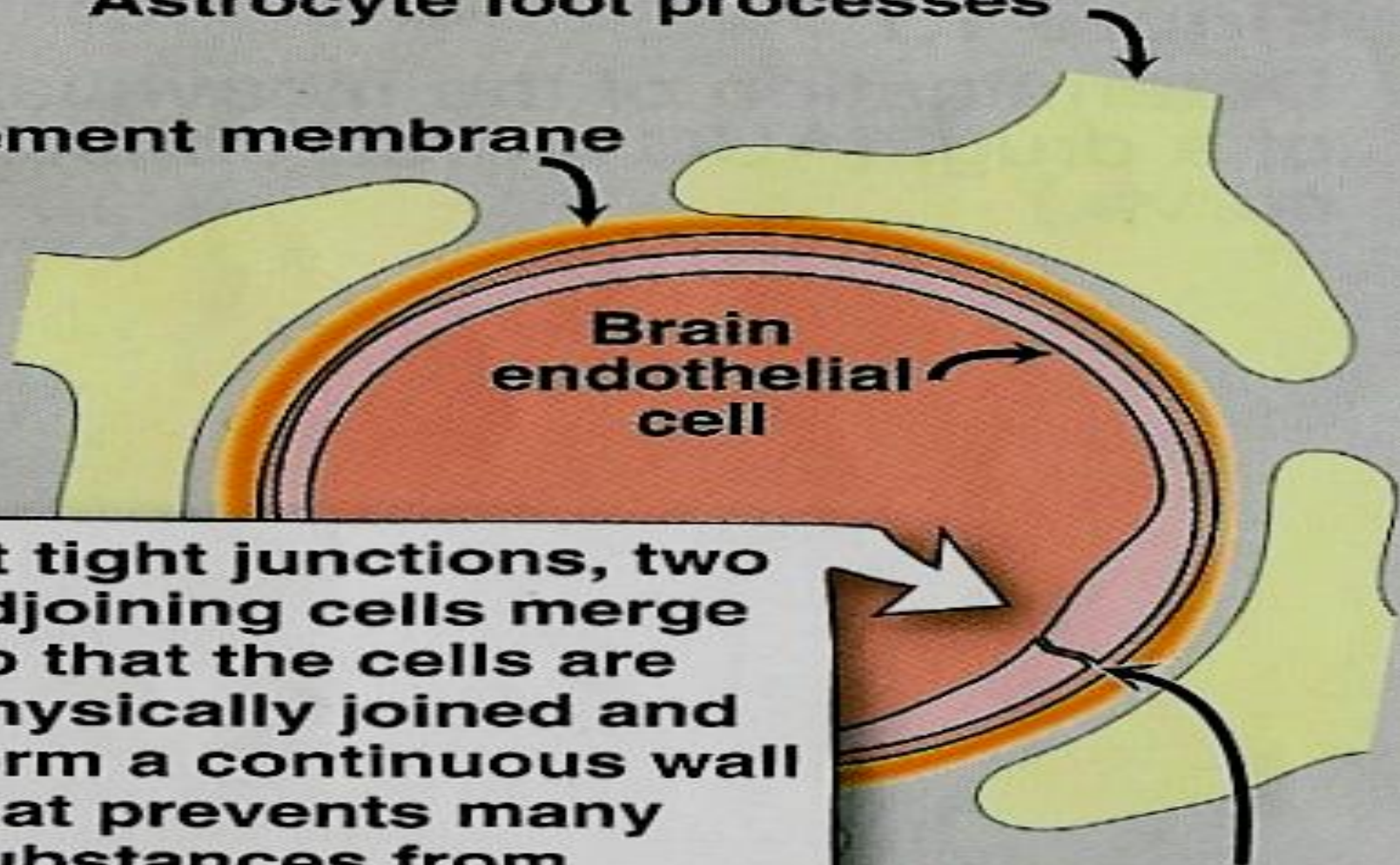
Astrocyte foot processes

Basement membrane

Brain endothelial cell

At tight junctions, two adjoining cells merge so that the cells are physically joined and form a continuous wall that prevents many substances from entering the brain.

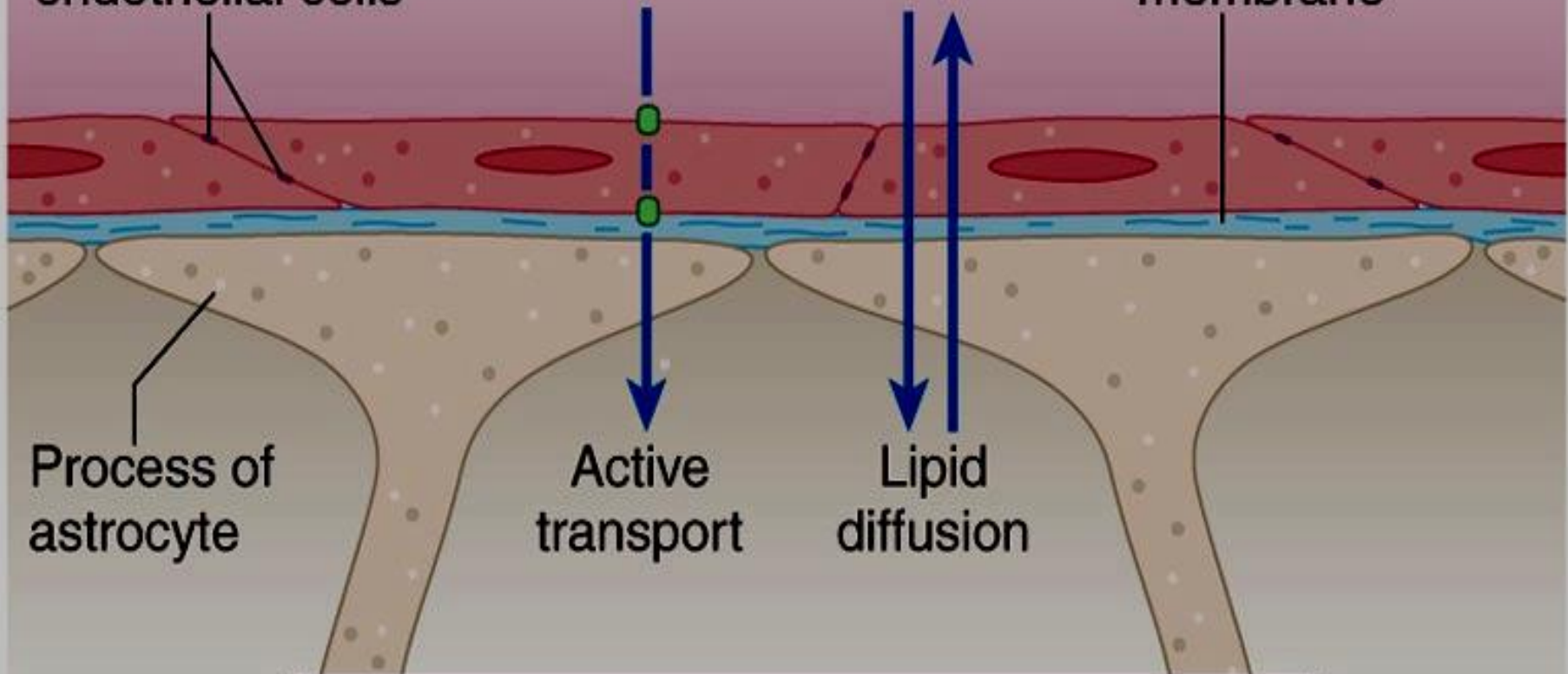
Tight junction



Blood capillary in central nervous system

Tight junctions between endothelial cells

Basal membrane



Process of astrocyte

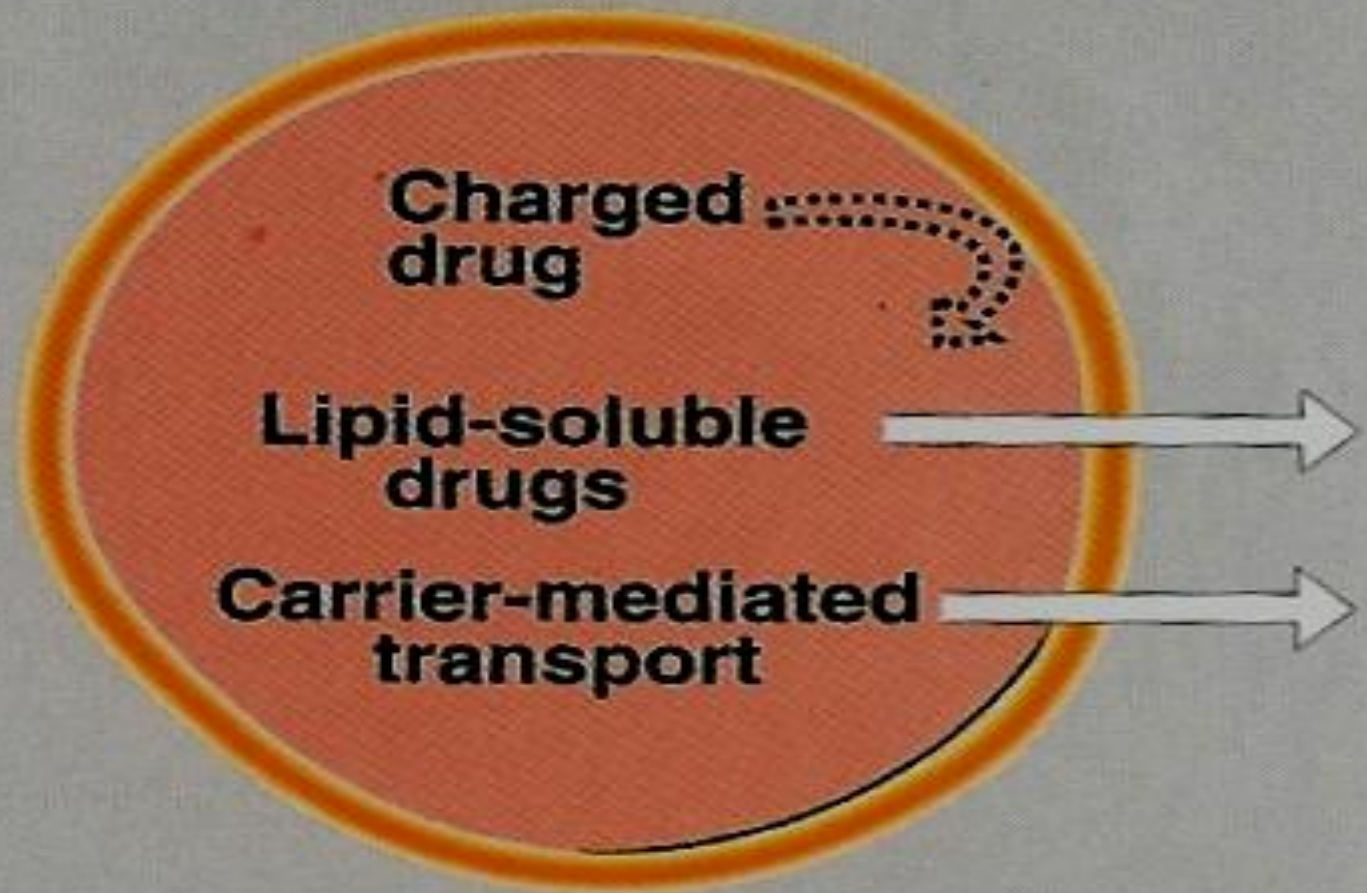
Active transport

Lipid diffusion

Glial cells, neurons, and extracellular fluid

C

Permeability of a brain capillary



Binding of Drugs

- **Plasma proteins binding.**
- **Tissue proteins binding.**

Plasma protein binding:

- Extensive plasma protein binding will cause more drug to stay in the blood compartment .
- Therefore, drugs which bind strongly to plasma protein tend to have lower distribution (V_d).

Plasma Proteins

Albumin

Has affinity for **acidic drugs** as warfarin, phenytoin, aspirin

alpha 1 -acid glycoproteins

Has affinity for **basic drugs** (cationic) as diazepam, quinidine.

Plasma protein binding

- ▶ drugs which bind strongly to plasma protein tend to have lower distribution (V_d).
- ▶ In blood, drugs exist in two forms bound and unbound forms in equilibrium



Tissues Binding

Drugs can bind to specific tissues and will have high volume of distribution (V_d)

Tetracycline bind to bone

Bound form of drug

- **non diffusible form**
- **can not cross endothelial barrier**
- **can not combine with receptors**
- **inactive**
- **not available for metabolism & excretion**
- **has long duration of action ($t_{1/2}$).**

Unbound form of drug

- **diffusible form**
- **cross endothelial barrier**
- **combine with receptors**
- **active**
- **available for metabolism & excretion**
- **has short duration of action ($t_{1/2}$).**

Characters & consequences of Binding

- ▶ **Usually reversible.**
- ▶ **determines volume of distribution (vd)**
- ▶ **Slows drug metabolism & excretion.**
- ▶ **Prolongs duration of drug action ($t_{1/2}$).**
- ▶ **Result in clinically important drug interactions**

Displacement

- ▶ Competition for the same binding site on the plasma proteins may occur between two drugs
→ displacement of one drug & increasing its concentrations & effects.
- ▶ Aspirin + Albumin-warfarin →
Albumin-aspirin + free warfarin → bleeding.