

**UJIL** Biochemistry

The instructions for the OSPE -Most likely it will be one station in the exam for biochemistry -you have about 3 -4 mins per station

-they have their own calculator



## LAB ORIENTATION

### Color Index :

### • important

• Examples of question may come



## OBJECTIVES

You should be familiar with :



General safety rules followed in biochemistry laboratory.

Safety with laboratory equipment.

Basic emergency procedures.





The basics of spectrophotometer and general equipment to be used In the lab during Biochemistry Practical sessions

# GENERAL SAFETY RULES

Always wear appropriate clothes and personal protective tools. -lab coat

- -safety goggles
- -masks
- gloves
- -<u>No</u> opens shoes
- -<u>No</u> eye lenses.



- -Always wash your hands with soap and water after handling chemicals.
- -during lab work, keep your hands away from your face.
- -tie back long hair



Roll up long sleeves.
Know the location of the fire extinguisher, fire blanket, eyewash station and first aid kit.
Keep your work area uncluttered.

-Take to the lab station only what's necessary .



-It is suggested that you wear glasses rather than contact lenses.

-Never eat or drink during a lab work.





-Lab safety is everyone's responsibility -Lab safety policy and procedures must be strictly followed.

### SAFETY WITH LABORATORY EQUIPMENT:

## ELECTRICAL SAFETY

Lay electrical cords where no one can trip on them.



Unplug cords by pulling the plug and not the cord.

Be sure your hands and your lab area are dry before using electrical equipment.



Never use any laboratory equipment unless you are trained & have been authorized to do so.

> As well as injuring yourself you may cause very costly damage



\*Note: example for questions What does RACE&PASS stand for?

## FIRE SAFETY - R.A.C.E:

<u>Procedures to follow in the event</u> of a fire emergency R - Remove or secure individuals in immediate danger. <u>A</u> - <u>A</u>ctivate the alarm by pulling a fire pull station located in the corridors and calling 953. <u>C</u> - <u>C</u>onfine the fire by closing windows, vents, and doors. E - Evacuate to a safe area

## FIRE SAFETY - EXTINGUISHER P.A.S.S:



## **BIOLOGICAL SAFETY** :

### WASTE DISPOSAL :

All biological samples are considered potentially infectious Should be handled and processed using strict precautions -For disposal of contaminated waste, use containers with yellow plastic bags

-Regular waste like papers etc go into containers with black/white plastic bags

-All sharp objects such as needles, scalpels and even broken glassware go into yellow-red sharps container

### CLINICAL BIOCHEMISTRY LABORATORIES:



### BIOCHEMICAL TEST PROFILES



### CLINICAL BIOCHEMISTRY FOR DIAGNOSIS OF DISEASES

Biochemical laboratory tests are crucial tools for diagnosis of many human diseases:



# LAP EQUIPMENTS





Light <u>source</u> which works with visible wavelengths (400-700 nm)

### Spectrophotometer

Most of visible spectrophotometer are composed of:

Detector

<u>Monochromator</u> filter for choosing desired wavelength.

<u>Sample</u> holder (cuvette).



Figure I

# DNA EXTRACTION

### Color Index :

#### important ۲

Examples of question may come



## OBJECTIVES

Understand the principle behind DNA extraction and purification.

Perform DNA extraction, purification and measurement according to the provided protocol (spin protocol).

> Interpret the results in terms of quantity, purity and yield.

Have a knowledge about some molecular techniques and applications.

### PRINCIPLES

Genomic DNA is extracted from peripheral blood samples preserved in EDTA using QIAamp DNA Blood Mini Kit, spin protocol.

The principle of the test includes lysis of the nucleated cells using lysis buffer, which has high salt concentration that breaks the cellular membrane; after the lysing step, DNA is allowed to bind to the spin column membrane for separating the DNA from the cell debris; removal of the contaminants with wash buffers; and elution of pure DNA. The measurement of the purified DNA is performed by UV absorbance at 260nm and 280nm. <u>DNA concentration is determined by measuring at 260nm</u>, and the purify of the purified DNA is determined on the bases of 260nm/280nm ratio. <u>A pure DNA falls in the accepted ratio, which ranges from 1.7 up to 1.9</u>.

Lysis of the nucleated cells Contaminants (any other substance other than DNA)

Thanks to Med438



## SPIN PROTOCOL OF DNA PURIFICATION FROM BLOOD



mixture in cuvettes tube by pipettes then into UV-spectrophotometer then do step 33

### QUANTIFICATION OF THE PURIFIED DNA

### measurements:

- Measure the Absorbance at 260nm and at 280nm.
- Assess the DNA purity= 260/280 ratio. (Accepted ratio: 1.7 - 1.9)
- <u>Calculate DNA Conc</u>.:Provided A260 = 1.0, DNA is 50µg/ml, unknown DNA Conc. can be calculated by cross multiplication

A260= 1.0 A260= 0.5 DNA conc. = 50 µg/ml DNA conc. ?

القاعده باختصار و ثابته: x50µg/m

Note: In case of diluting the eluted sample, multiplies the final concentration by the dilution factor. This can be adjusted by the spectrophotometer.

اذا قالوا 3 times dilution یعنی اضرب ب ۳ بس



- Volume of DNA solution: 0.2 ML (200 microliter) - Final DNA Conc.: 30µg/ml

Then, the yield ( $\mu$ g) = 0.2ml × 30  $\mu$ g/ml = 6.0  $\mu$ g

# MOLECULAR TECHNIQUE AND APPLICATIONS

Note:- Almost all molecular biology techniques can be utilized for diagnosis and research

### a. Amplification techniques:

e.g. Polymerase Chain Reaction (PCR)

Applications of PCR:

- Comparison of a normal gene with a mutant form of the gene.

- Detection of low-abundance nucleic acid sequences.

- Forensic analysis of DNA samples.
- Prenatal diagnosis.



# SUMMARY

### 🖈 How to calculate the purity ?

260nm/280nm ratio

<u>Example:</u> If you have - A260 = 0.50 - A280 = 0.35 Then Answer: 0.50/0.35 = 1.428 ( it's not pure because DNA ratio should be between 1.7-1.9 )

2 How to calculate DNA conc ?

260nm x 50

<u>Example:</u>

If you have - A260 = 0.50 Then Answer: 0.50 × 50 = 25 µg/ml



#### 0.2 x final DNA conc

Example:

We will use the conc for the previous question - conc = 25 µg/ml Then Answer: 0.2ml x 25 µg/ml = 5 µg





#### Explain what does PASS represent ?



Calculate the DNA conc & the purity ?

If A260 = 0.5 , A280 = 0.29

Calculate the DNA yield

-If the volume of DNA solution 200µl (0.2 ml) -final DNA conc 20µg/ml





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 Biochemistry team wishes
 you all the best

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