

#### Renal Block





#### By the end of this lecture, the students will be able to:

Discuss the general physiological and pathological factors that favor kidney stones formation

- List the types of kidney stones, their chemical constituents and characteristics
- Identify the etiological causes of each type of kidney stone

Discuss the diagnosis, treatment and prevention of kidney stones



# Overview

■ Introduction Conditions causing kidney stone formation Types of kidney stones - Calcium salts - Uric acid -Mg ammonium  $PO_4$ - Cystine - Other (xanthine, etc.)

Laboratory investigations



# What are kidney stones?

Renal calculi (kidney stones) are formed in renal tubules, ureter or bladder

Composed of metabolic products present in glomerular filtrate

These products are in high conc.
 Near or above maximum solubility



High conc. of metabolic products in glomerular filtrate

- Changes in urine pH
- Urinary stagnation

Deficiency of stone-forming inhibitors in urine



High conc. of metabolic products in glomerular filtrate is due to:

- Low urinary volume (with normal renal function) due to restricted fluid intake
- Increased fluid loss from the body
- Increased excretion of metabolic products forming stones
- High plasma volume (high filtrate level)
- Low tubular reabsorption from filtrate



Changes in urine pH due to:
 Bacterial infection

- Precipitation of salts at different pH

- A persistently acidic urine → promotes uric acid precipitation
- A persistently alkaline urine (due to upper urinary tract infection) -> promotes Mg Ammonium
   Phosphate crystals (Struvite stones)

Urinary stagnation is due to:
 Obstruction of urinary flow



Deficiency of stone-forming inhibitors:

- Citrate, pyrophosphate, glycoproteins inhibit growth of calcium phosphate and calcium oxalate crystals
- In type I renal tubular acidosis, hypocitraturia leads to renal stones



# Types of kidney stones

Calcium salts
Uric acid
Mg ammonium PO<sub>4</sub>
Cystine
Other (xanthine, etc.)



**80% of kidney stones contain calcium:** Mostly Ca-Oxalate and less often Ca-Phosphate

The type of salt depends on
 Urine pH
 Availability of oxalate

General appearance:

- White, hard, radio-opaque
- Calcium oxalate: present in ureter (small)
- Calcium PO<sub>4</sub>: staghorn in renal pelvis (large)



### **Causes of calcium salt stones:**

### Hypercalciuria:

- Increased urinary calcium excretion
- Men: > 7.5 mmols/day
- Women > 6.2 mmols/day
- Due to hypercalcemia (most often due to 1<sup>ary</sup> hyperparathyroism)
- sometimes, Ca<sup>++</sup> salts stones are found with no hypercalcemia



### Hyperoxaluria:

- Causes the formation of calcium oxalates without hypercalciuria
- Diet rich in oxalates
- Increased oxalate absorption in fat malabsorption

#### Primary hyperoxaluria:

- Due to inborn errors
- Urinary oxalate excretion: > 400 μmol/24 Hours



**Calcium oxalate stones** 



#### Treatment:

- Treatment of primary causes such as infection, hypercalcemia, hyperoxaluria
- Oxalate-restricted diet
- Increased fluid intake (if no glomerular failure)
- Acidification of urine (by dietary changes)
   Calcium salt stones are formed in alkaline urine



# Uric acid stones

About 8% of renal stones contain uric acid
 May be associated with hyperuricemia (with or without gout)
 Form in acidic urine

General appearance:

- Small, friable, yellowish
- May form staghorn (if big)
- Radiolucent (plain x-rays cannot detect)
- Visualized by ultrasound or i.v. pyelogram



#### Uric acid stones



# Uric acid stones

#### **Treatment:**

- Treatment of cause of hyperuricemia.
- Purine-restricted diet
- Alkalinization of urine (by dietary changes)
- Increased fluid intake



# Mg ammonium PO<sub>4</sub> stones

About 10% of all renal stones contain Mg amm. PO<sub>4</sub>

- Also called struvite kidney stones
- Associated with chronic urinary tract infection
  - Microorganisms (such as from *Proteus* genus) that metabolize urea into ammonia
  - Causing urine pH to become alkaline leading to stone formation

Commonly associated with staghorn calculi
 75% of staghorn stones are of struvite type



Mg ammonium phosphate (struvite) stone



# Mg ammonium PO<sub>4</sub> stones

#### **Treatment:**

- Treatment of infection
- Urine acidification
- Increased fluid intake
- In some cases, it may require complete stone removal (percutaneous nephrolithotomy)



# **Cystine stones**

A rare type of kidney stone
Due to homozygous cystinuria
Form in acidic urine
Soluble in alkaline urine
Faint radio-opaque



Cystine stone





#### **Treatment:**

- Increased fluid intake
- Alkalinization of urine (by dietary changes)
- Penicillamine (binds to cysteine to form a compound more soluble than cystine)



Laboratory investigations of kidney stones

# If stone has formed and removed:

## Chemical analysis of stone helps to:

- Identify the cause
- Advise patient on prevention and future recurrence



Laboratory investigations of kidney stones

# If stone has not formed:

- This type of investigation identifies causes that may contribute to stone formation:
  - Serum calcium, uric acid and PTH analysis
  - Urinalysis: volume, calcium, oxalates and cystine levels
  - Urine pH > 8 suggests urinary tract infection (Mg amm. PO<sub>4</sub>)
- Urinary tract imaging:
   CT, ultrasound and i.v. pyelogram





# Clinical Chemistry and metabolic Medicine 7<sup>th</sup> Edition, pp. 36.

### The National Kidney Foundation, USA (www.kidney.org)