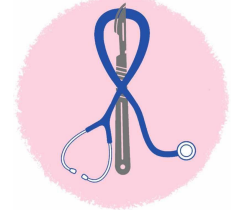


Protein Structure



MED441
KING SAUD UNIVERSITY

Revised & Reviewed
by:
Abdulaziz & Bahammam
Faye Wael Sondi



2

V1

Foundation
Block - KSU

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- Main text
- Important
- Notes
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- Girls slides'
- Extra

[Editing File](#)



Objectives

1

Understand the peptide bonding between amino acids.

2

Explain the different levels of protein structure and the forces stabilizing these structures and what happens when the protein is denatured.

3

Define the α -helix and β -sheet as the most commonly encountered secondary structures in a protein molecule.

4

Correlate the protein structure with function with hemoglobin as an example.

5

Understand how the misfolding of proteins may lead to diseases like Alzheimer's or prion disease.

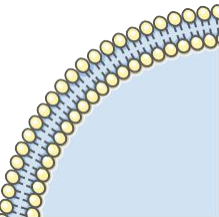
What are Proteins?

- **Proteins:** are large, complex molecules that play many critical roles in the body.
- Made up of: hundreds or thousands of **amino acids** (the smaller units - protein building blocks).
-
- Amino acids are attached to one another in long chains.
- There are mainly **20 different types** of amino acids that can be combined to make a protein.

The **sequence** of **amino acids** determines

Protein's unique three
dimensional (**3D**) **structure**

Protein's specific **function**.



What are Proteins?

The importance of proteins

They do **most** of the **work** in cells .

Required for the **structure, function, and regulation** of the body's tissues and organs.

Protein functions

Antibody

Enzyme

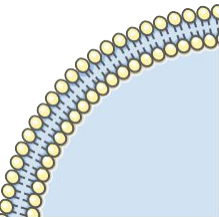
Messenger

Structural component

Transport

Storage

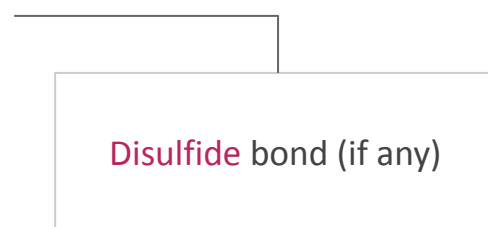
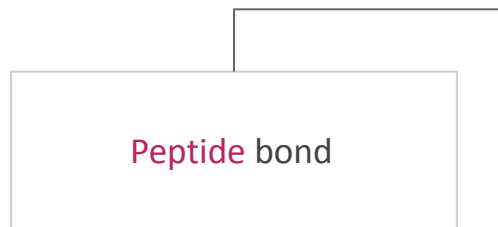
Proteins can be described according to their large range of functions in the body.



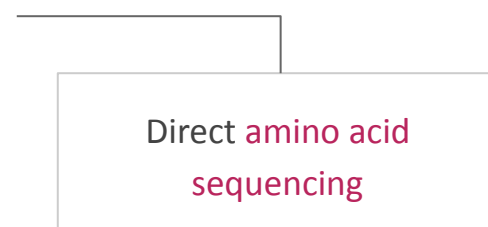
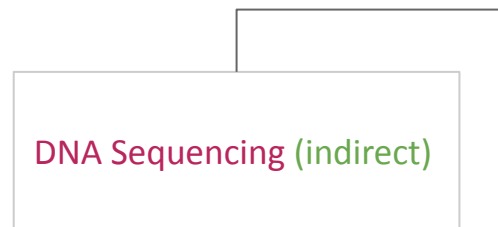
Primary Structure

- **Primary structure:** the **linear** sequence of amino acids.
- **Not functional.**

Covalent Bonds in the Primary Structure



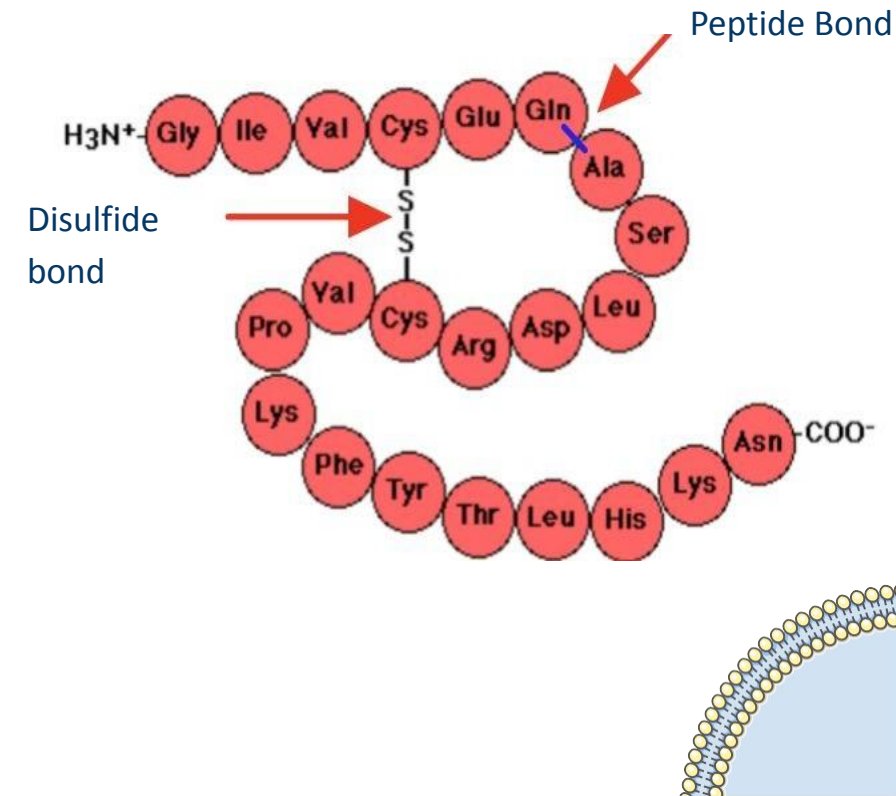
To Determine the Primary Structure Sequence



Notes 439:

- Disulfide (SS) bond links two residues of Cysteine near to each other.

- The primary structure is the simplest and the first level of structure.
- Peptide bonds are very strong and they require enzymes to be broken.



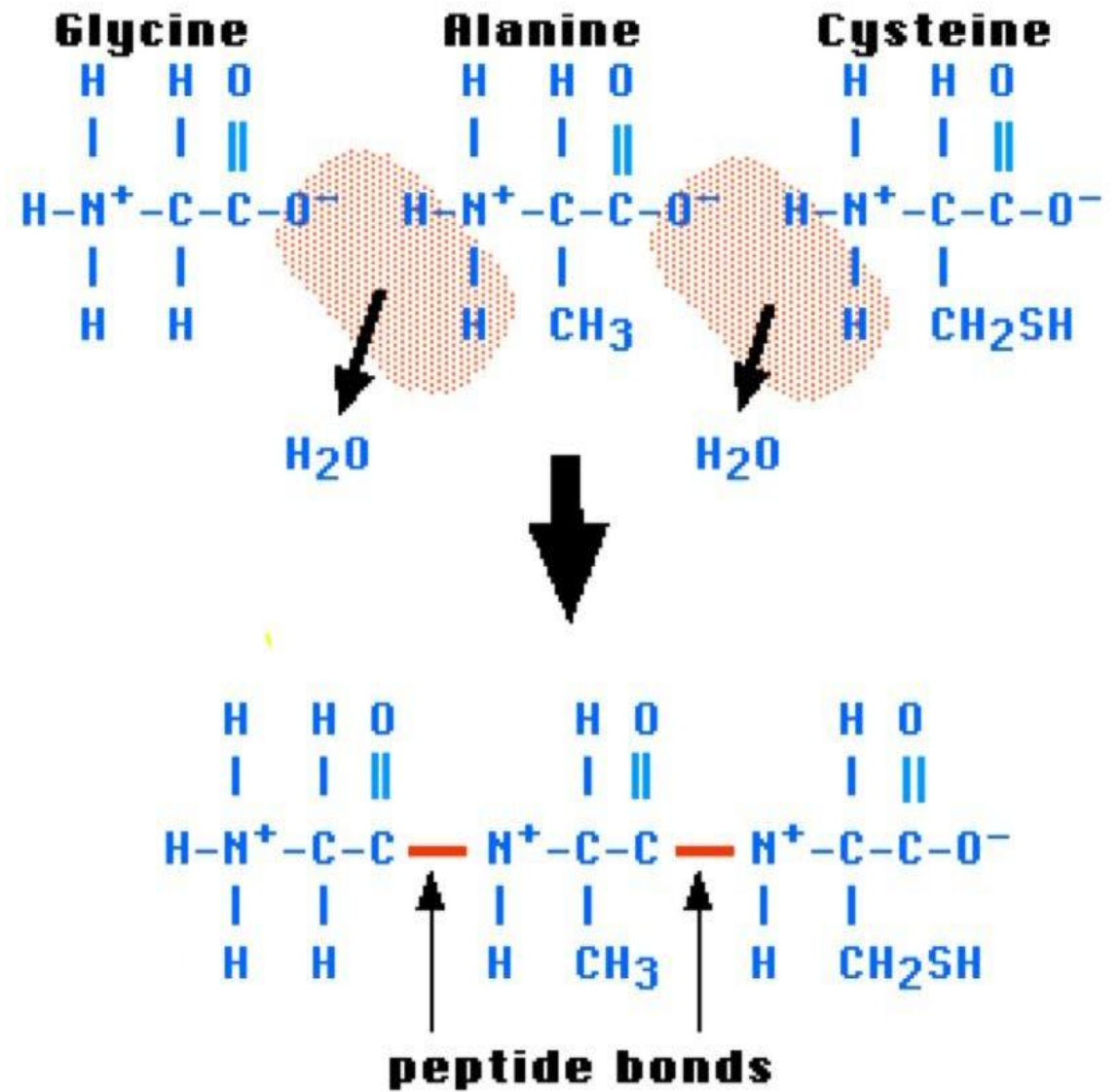
Peptide Bond (Amide Bond)

Each amino acid in a chain makes **2 peptide bonds**.

The **amino acids** at the two **ends** of a chain make only **1 peptide bond**.

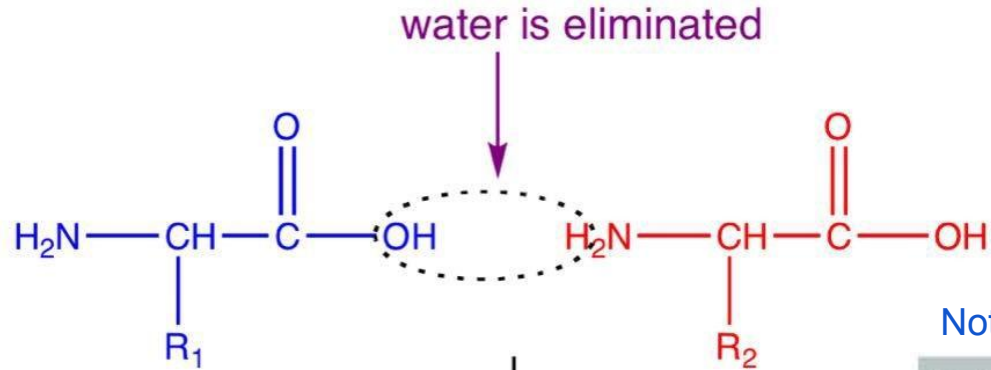
Amino (NH₂-) terminus: amino acid with a **free amino group**.

Carboxyl (COOH-) terminus: amino acid with a **free carboxylic group**.



Peptide Bond (Amide Bond)

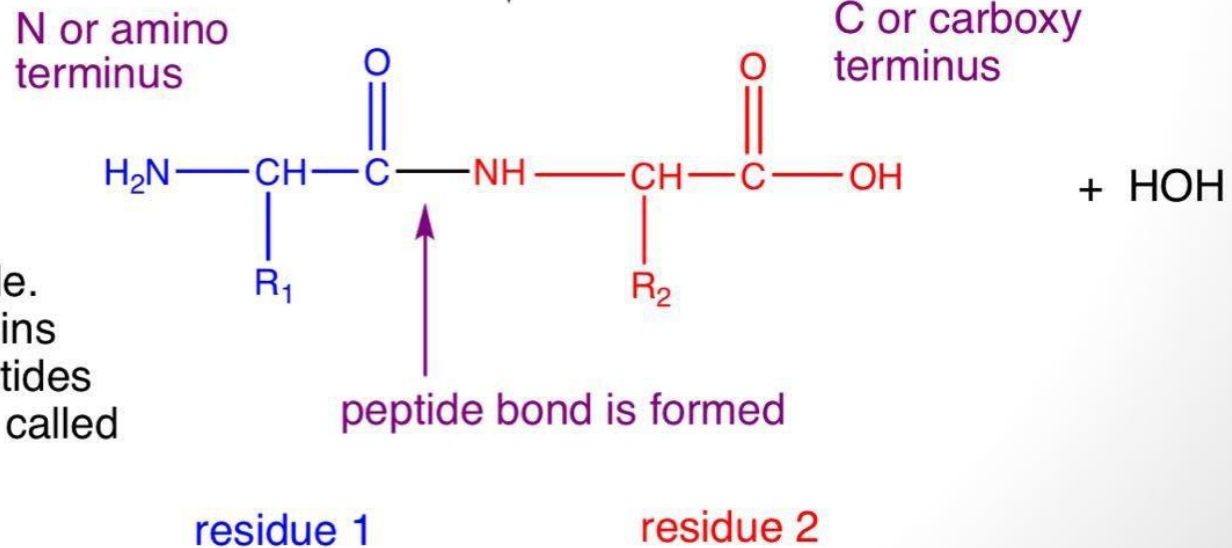
two amino acids
condense to form...



Note38:

- Residue : amino acid in a peptide chain
- We always read from N-terminus to C-terminus.

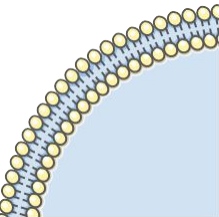
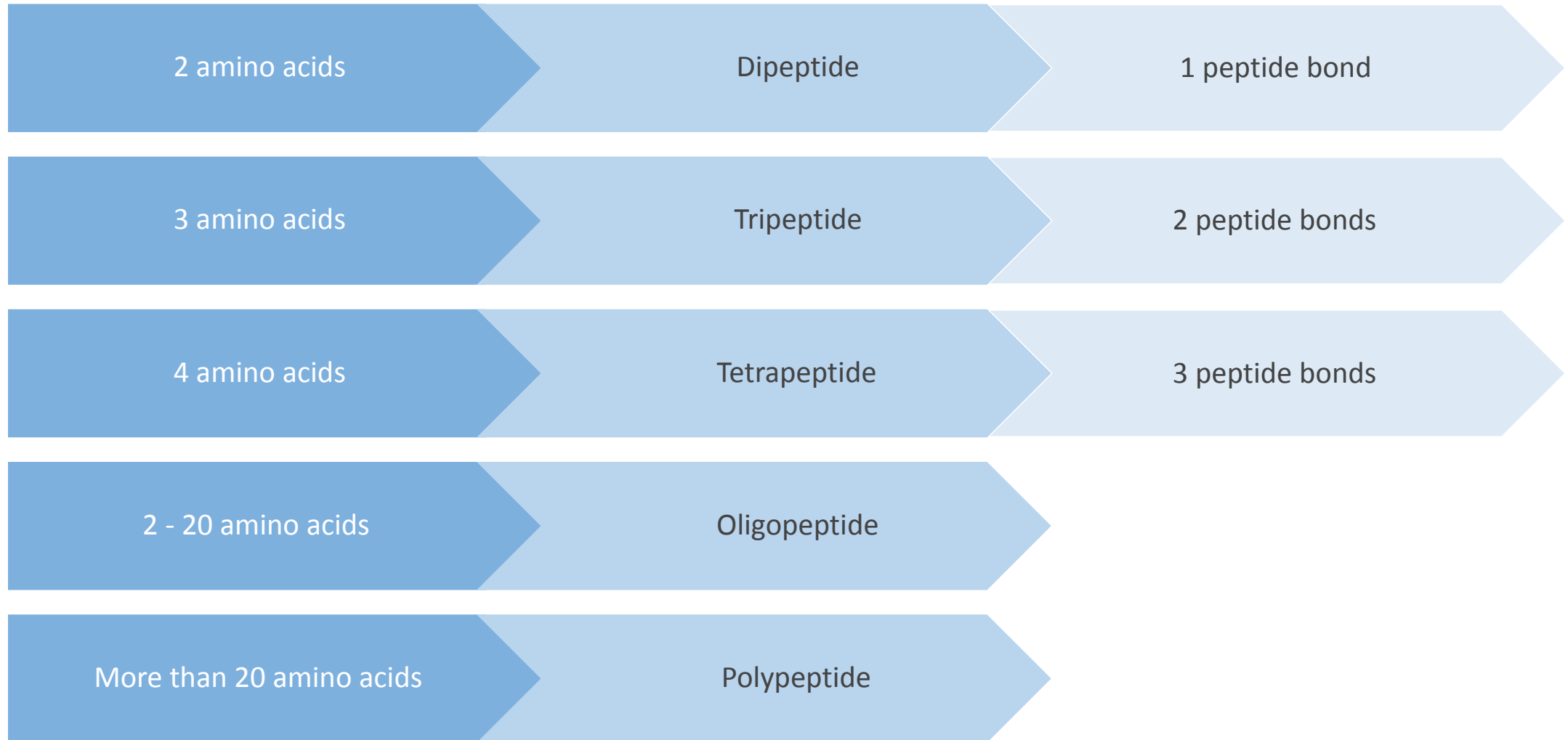
...a dipeptide. If there are more it becomes a polypeptide. Short polypeptide chains are usually called peptides while longer ones are called proteins.



Peptides

Notes 438:

- Oligo: few
- Poly: more
- Number of peptide bonds = number of amino acid - 1



Secondary Structure

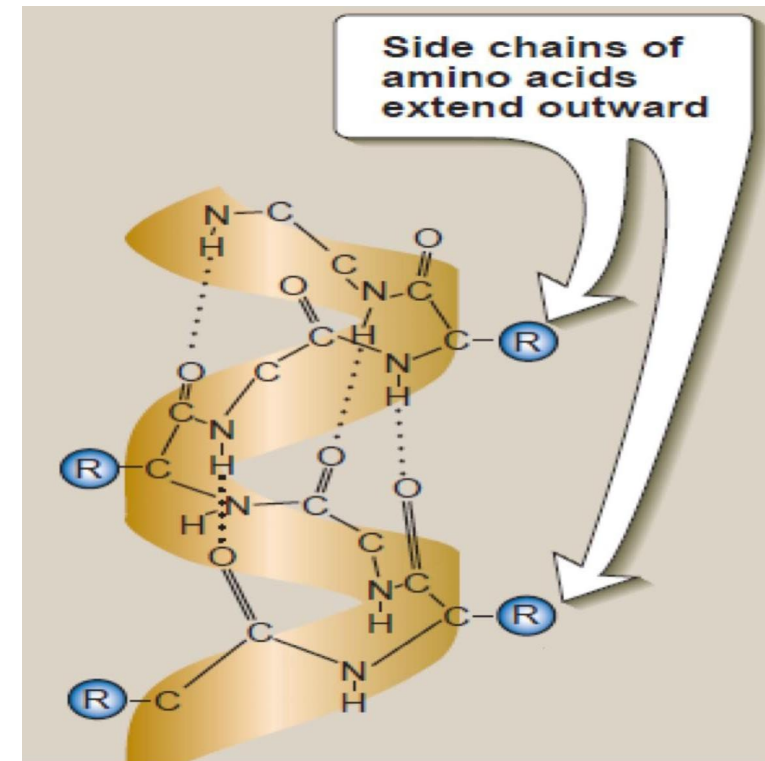
- It is regular arrangements of amino acids that are located near to each other in the linear sequence.
- Not functional.
- Excluding the conformations (3D arrangements) of its side chain.

Examples of Secondary Structure

α -helix

β -sheet

β -bends



α -Helix

α -helix is the most common and thus the most stable secondary structure.

- **α -helix:** a **right-handed** spiral, in which **side chains** of amino acids extended **outward**.
- **Hydrogen bonds stabilize the α -helix.**
 - H-bonds form between the peptide bond carbonyl oxygen and amide hydrogen.
- Amino acids per turn: each turn contains **3.6 amino acids**.
- Amino acids that disrupt an α -helix:

01

Proline:

Has imino group → interferes with the smooth helical structure.
(Proline has a ring structure that disrupts the helical structure)

02

Glutamate - Aspartate - Histidine - Lysine - Arginine:

They form ionic bonds.

03

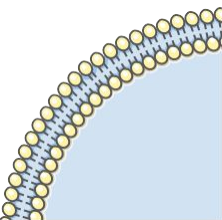
Tryptophan:

Because of the bulky side chain.
Note: there might be other amino acids but we only have Tryptophan.

04

Valine - Isoleucine:

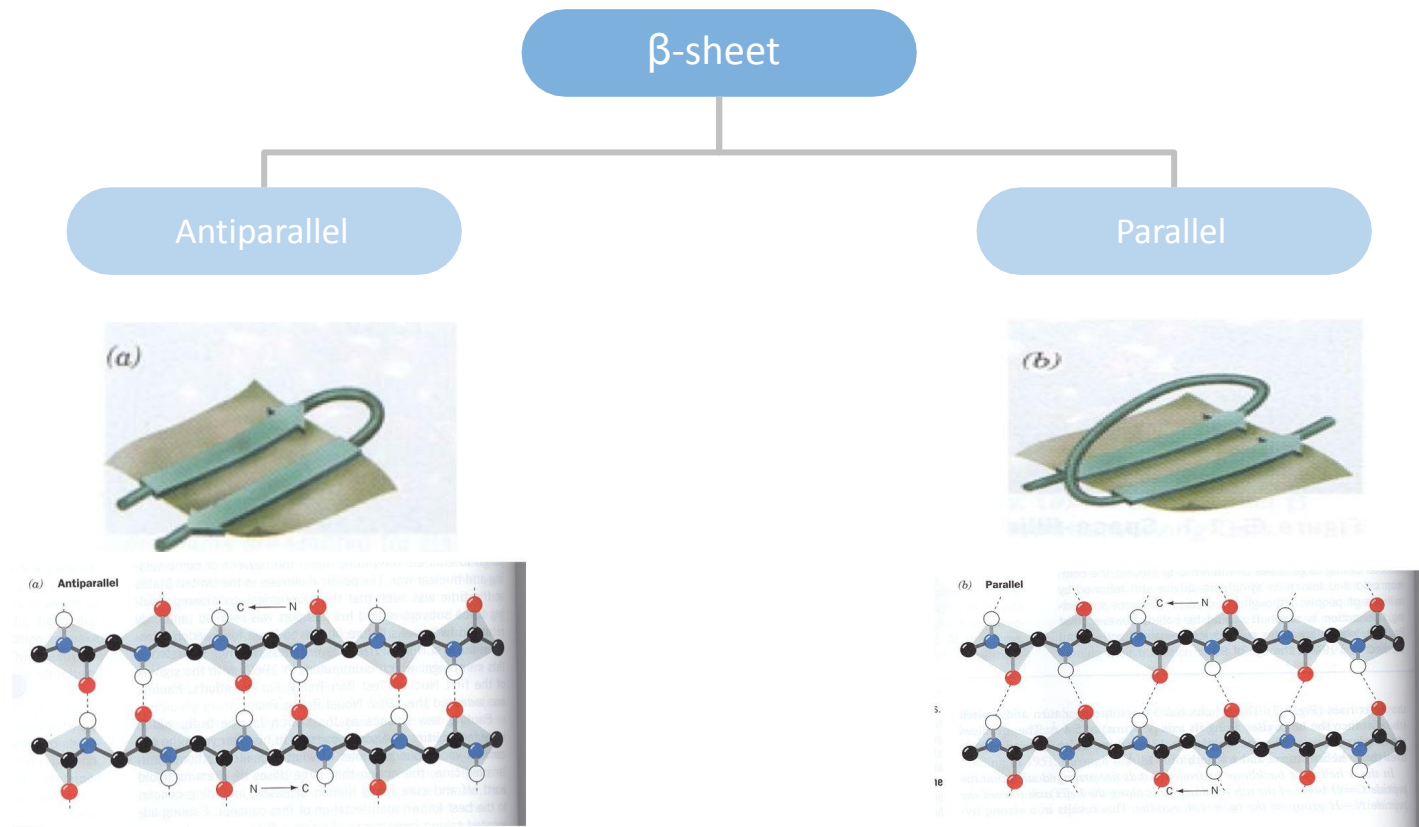
Branched amino acids at the β -carbon.



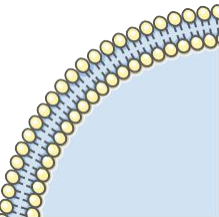
β -Sheet (Composition of a β -Sheet)

- The **tail** represents the **N-terminus**.
- The **head** represents the **C-terminus**.

- **β -sheet**: 2 or more polypeptide chains make hydrogen bonding with each other.
- Also called **pleated sheets** because they appear as folded structures with edges.

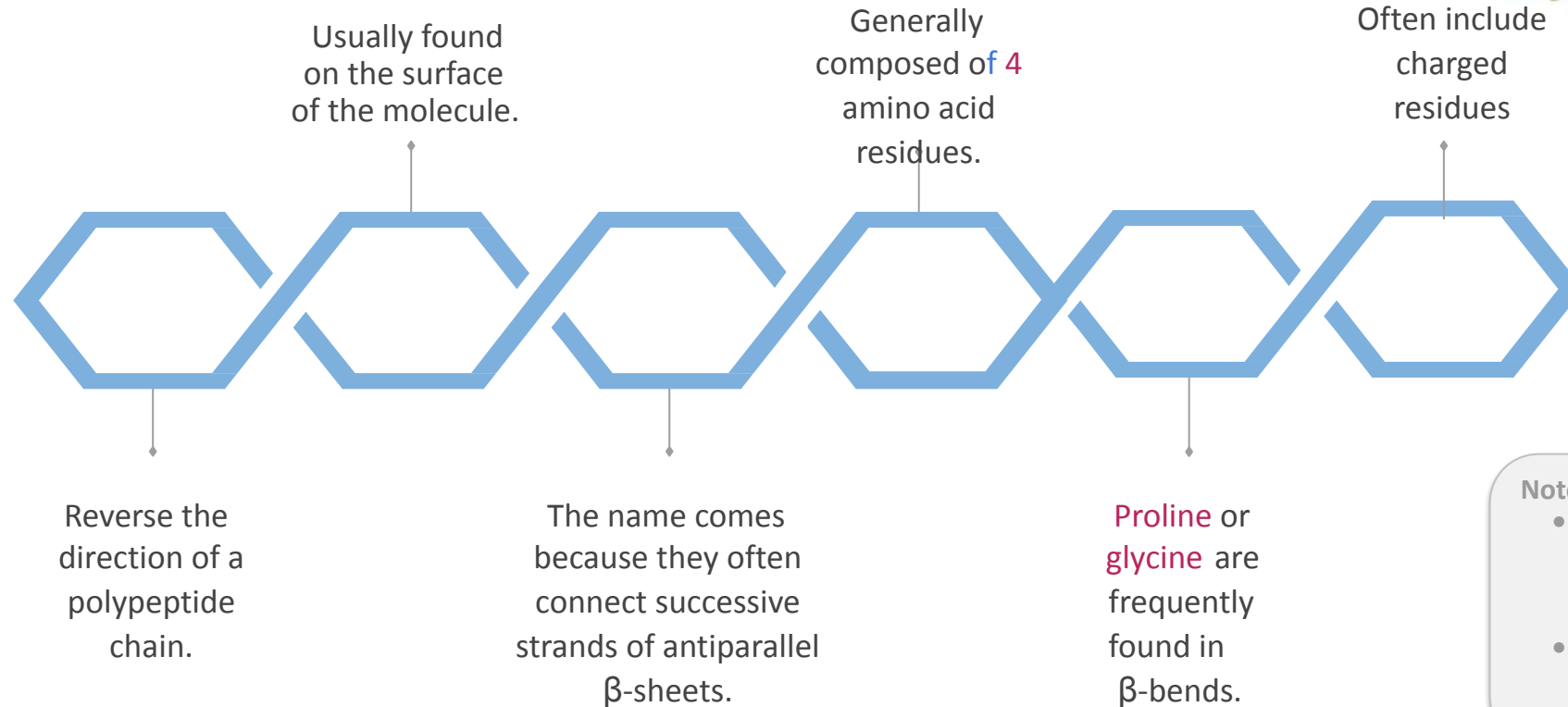
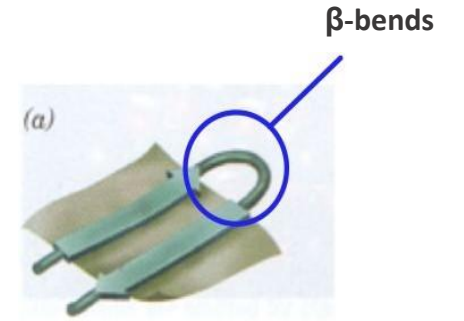


Hydrogen bonds in **parallel** direction is **less stable** than in **antiparallel** direction



Other Secondary Structure Examples?

1- β -bends (reverse turns):

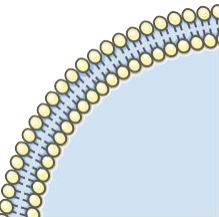


Note:

- **Glycine:** smallest amino acid which makes it easier to bend.
- **Proline:** due to bending and bending due to lack of hydrogen bonds.

2 Non Repetitive secondary structure:

- e.g. loop or coil conformation.



Supersecondary Structures

3- Supersecondary structures (motifs): a combination of secondary structural elements.

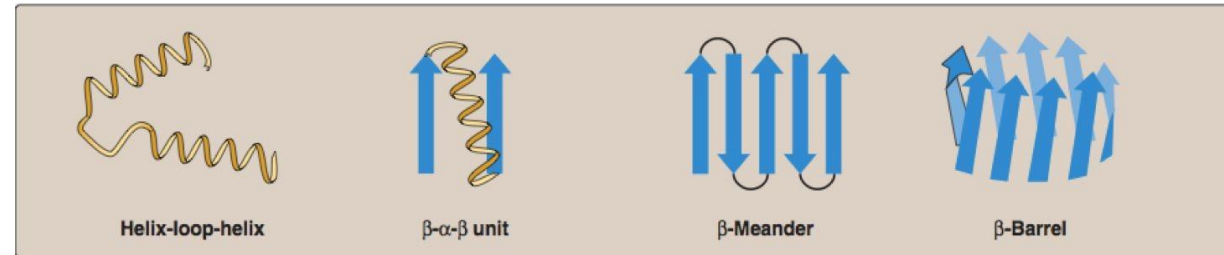
Examples of Supersecondary Structure

α α motif: 2 α helices together.

β α β motif: a helix connects 2 β sheets.

β hairpin: reverse turns connect antiparallel β sheets.

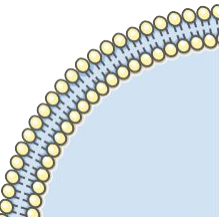
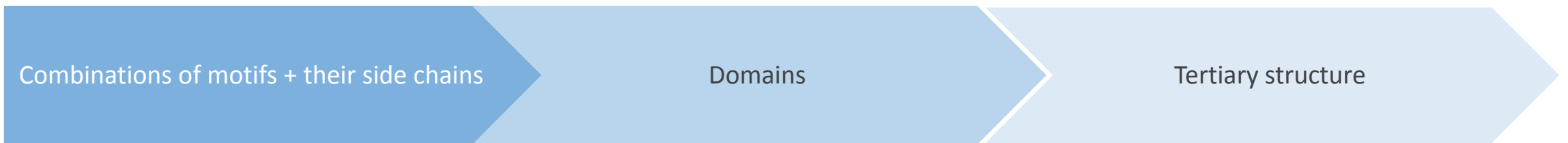
β barrels: rolls of β sheets.



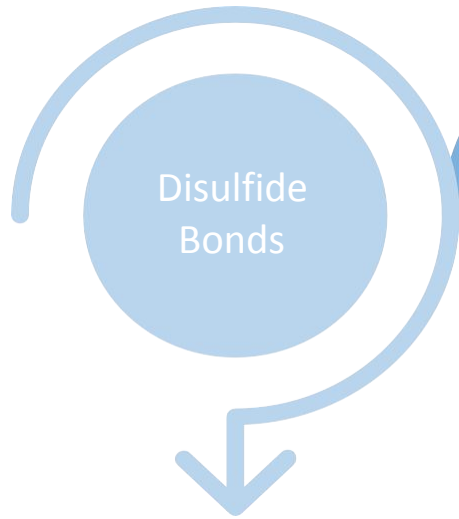
[A video to illustrate \(click\)](#)

Tertiary Structure

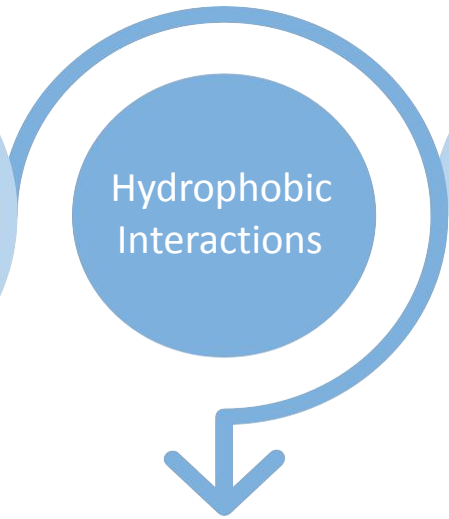
- **Tertiary structure:** the three-dimensional (3D) structure of an entire polypeptide chain **including side chains**.
- **Domains:** the fundamental **functional and 3D structural** units of a polypeptide.
 - Core of a domain: built from **combinations of** supersecondary structural elements (**motifs**) and their side chains.
- Polypeptide chain that is greater than 200 amino acids in length --> generally consist of 2 or more domains.
- Domains combined → tertiary structure.



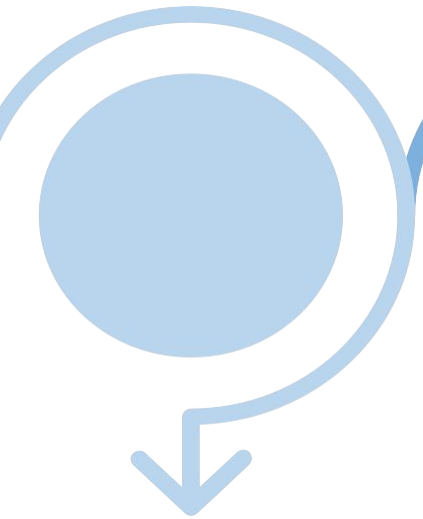
Interactions Stabilizing Tertiary Structure



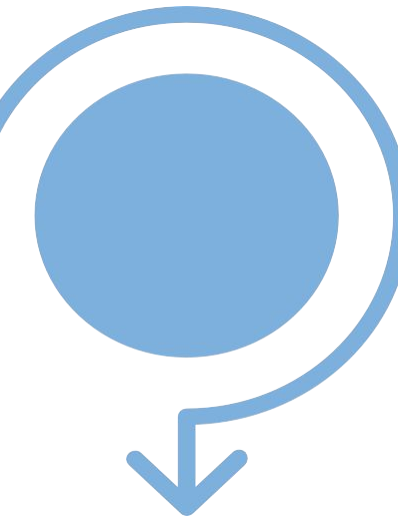
A covalent bond formed between the side chains of Cysteins.



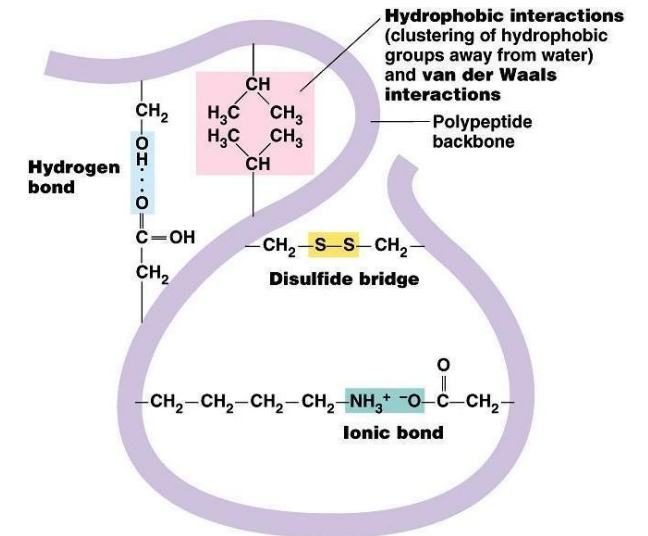
Interactions between non polar amino acids in the interior of the polypeptide or it's the most important interaction that stabilizes the tertiary structure



The hydrogen bonds between the polar and hydrophilic side chains the surface with the water molecule

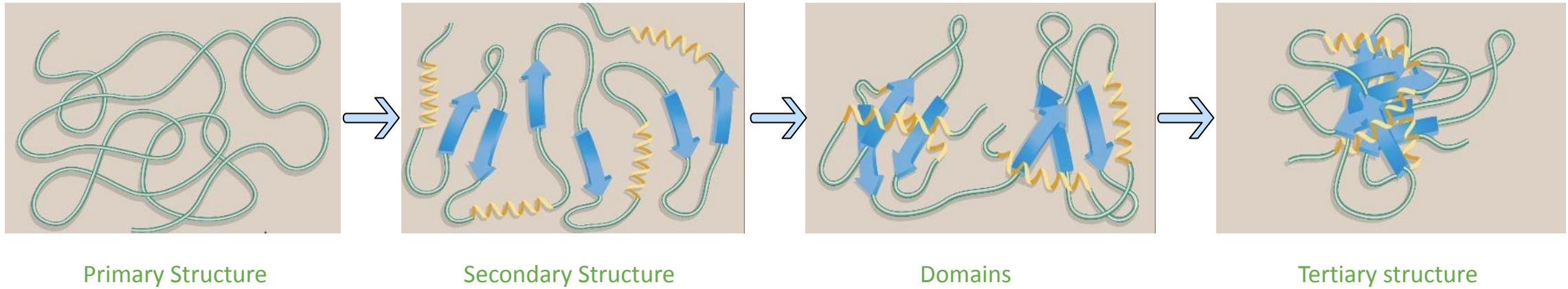


Oppositely charged side chains groups interact together.



[A video to illustrate \(click\)](#)

Levels of Folding



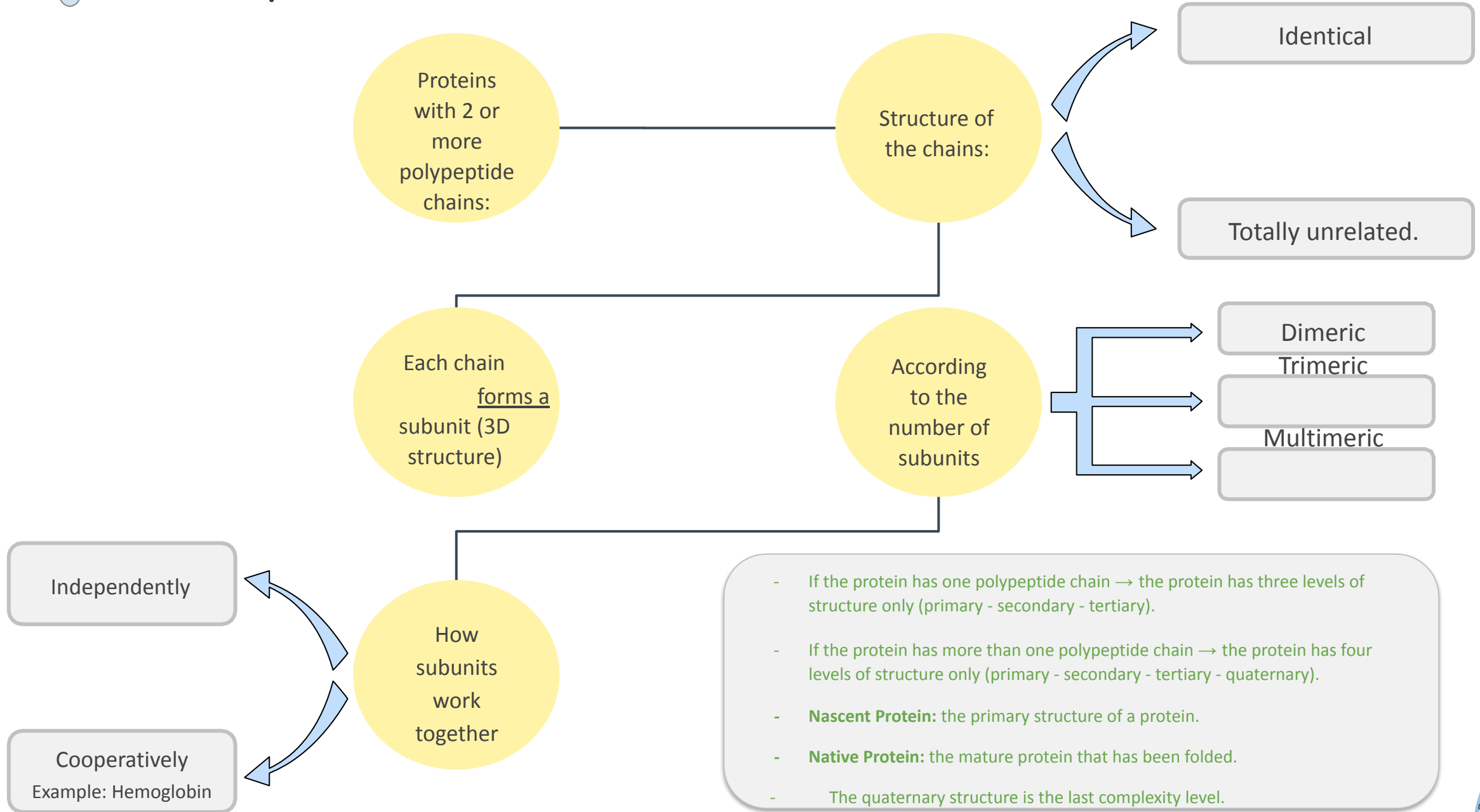
In some proteins, another level of folding (after the tertiary structure) is the quaternary structure.

Chaperons (Heat Shock Proteins)

Chaperons are folded by other chaperons.

- **Chaperons:** a specialized group of **proteins**, required for the **proper folding** of many species of proteins.
- Chaperons interact with polypeptide at various stages during the folding process.

Quaternary Structure



Hemoglobin

01

It is a **globular** protein.

02

Composed of **4** subunits: $\alpha_2\beta_2$

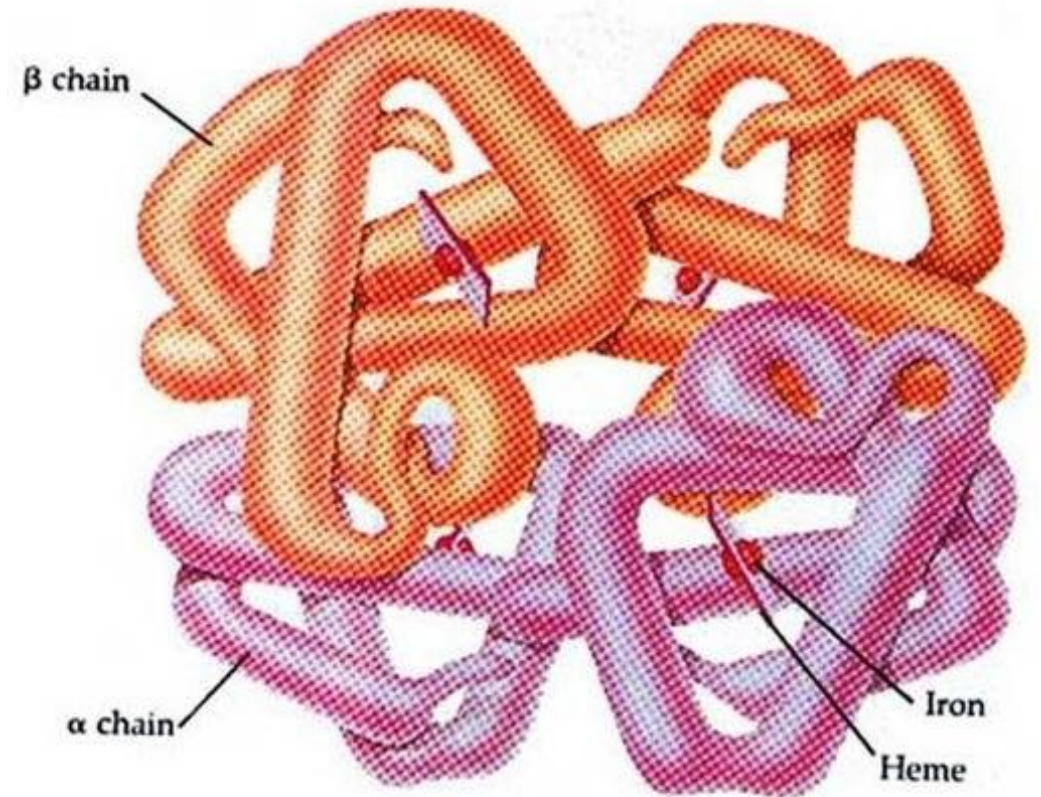
03

Multisubunit protein called **oligomer**.

04

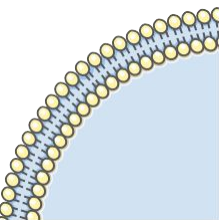
2 same subunits called **protomers**.

[A video to illustrate \(click\)](#)

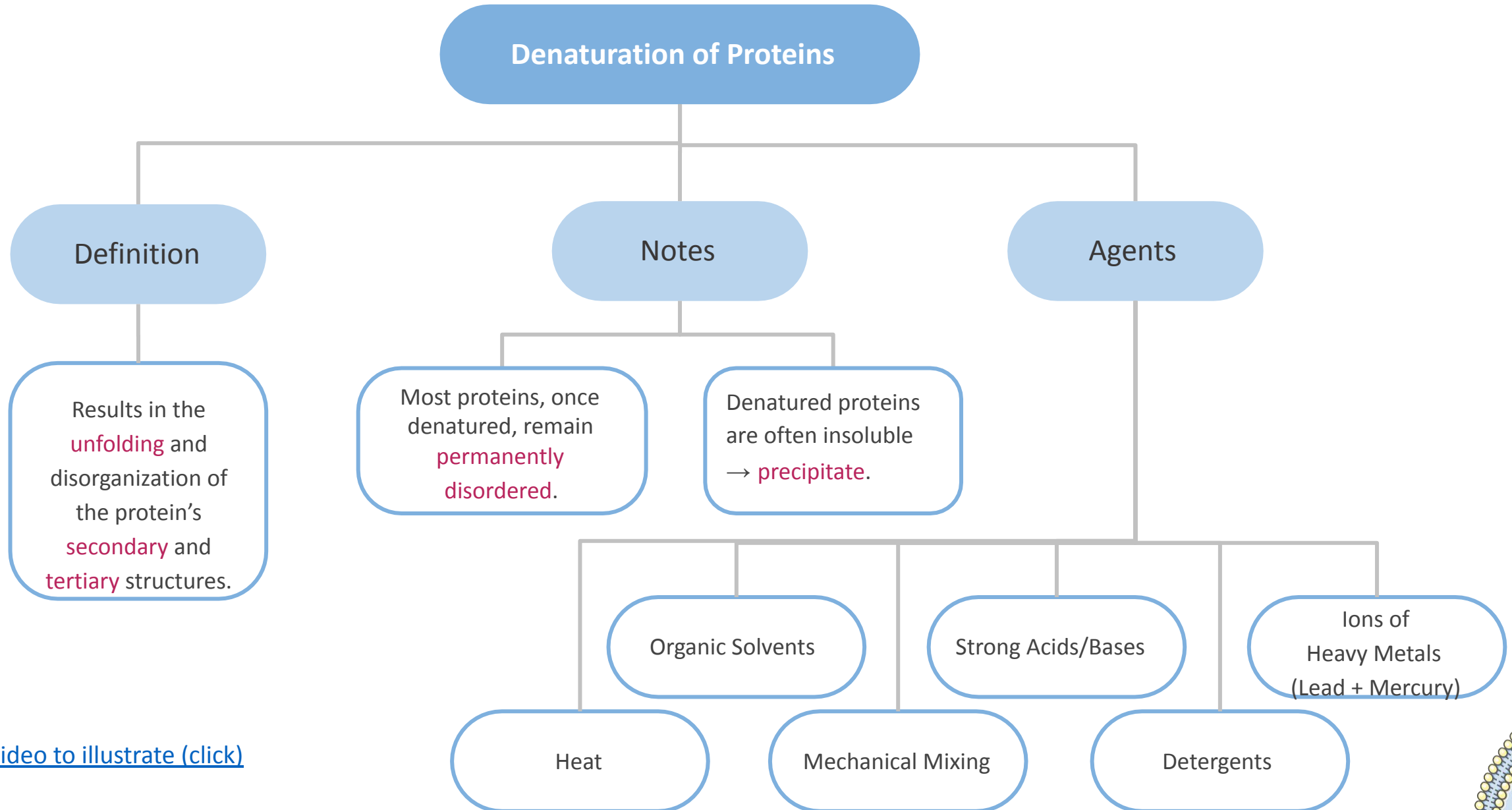


(b) Hemoglobin

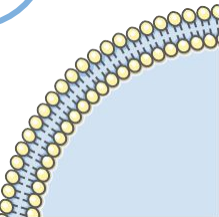
Globular is round/spherical.



Denaturation of Proteins



[A video to illustrate \(click\)](#)

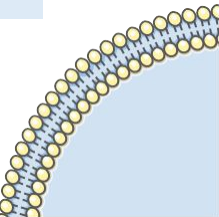


Protein Misfolding

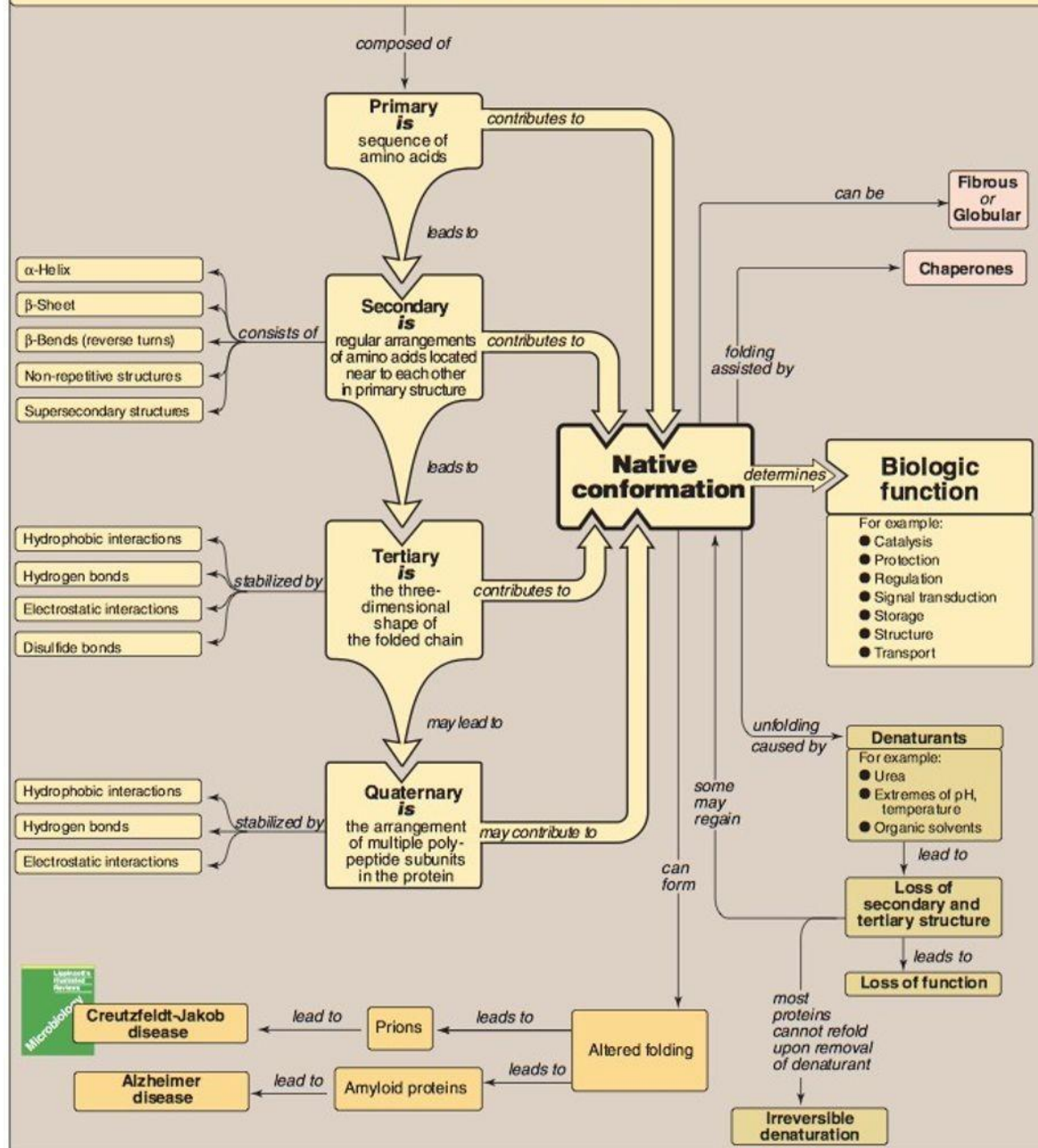
- Every **protein must fold** to achieve its normal conformation and function.
- **Abnormal folding** of proteins leads to a number of **diseases** in humans (Example: Alzheimer's - prion diseases).

Alzheimer's disease		Creutzfeldt-Jakob (Prion) Disease
β amyloid	Misfolded Protein	Prion protein
Forms fibrous deposits or plaques in the brain.	Result	Forms insoluble fibrous aggregates that damage brain cells.
	Notes	Prion protein is present in normal brain tissue but it's not misfolded.

[A video to illustrate \(click\)](#)



Hierarchy of protein structure





Take Home Messages



01

Native conformation of the protein: the functional, fully folded protein structure.



02

The unique 3D structure of the native conformation is determined by its primary structure (amino acid sequence).



03

Interactions between the amino acid side chains guide the folding of the polypeptide chain to form secondary, tertiary and sometimes quaternary structures that cooperate in stabilizing the native conformation of the protein.



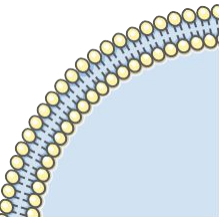
04

Protein denaturation results in unfolding and disorganization of of the protein's structure, which are not accompanied by hydrolysis of peptide bonds.



05

Disease can occur when an apparently normal protein assumes a conformation that is cytotoxic, as in the case of Alzheimer disease and Prion disease.



- 1) D 2) B 3) A 4) C 5) C

Quiz

Q1: A right-handed spiral, in which side chains of amino acids extended outward.

- A β - sheet B β - bend C α - sheet D α - helix

Q2: Reverse turns connect antiparallel β sheets.

- A $\beta \alpha \beta$ motif B β hairpin C β barrels D $\alpha \alpha$ motif

Q3: The fundamental functional and 3D structural units of a polypeptide

- A Domain B Subunit C Motif D Chaperon

Q4: Which of the following amino acids is frequently found in β - bend?

- A Isoleucine B Methionine C Proline D Lysine

Q5: The three-dimensional structure of an entire polypeptide chain including side chains.

- A Primary Structure B Secondary Structure C Tertiary Structure D Quaternary Structure

Answer Key:

Q6: List two ways used to determine the primary structure sequence.

Q7: List the forces that stabilize the tertiary structure of proteins.

Q8: What causes Creutzfeldt-Jakob disease? How does it affect the patient's brain?

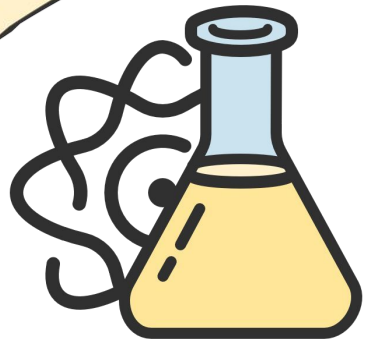
Q9: How many amino acids per turn

Q6: .
DNA Sequence - Direct Amino acid sequencing.

Q7:
Disulfide bonds - Hydrophobic interactions - Hydrogen bonds - Ionic interactions.

Q8:
Prion protein is misfolded. Thus, it forms insoluble fibrous aggregates that damage brain cells.

Q9:



Biochemistry 441

Girls



★ **Leader: Ghadah Alarify**

Members:

Yara Almufleh	Latifa Alkhdiri
Reema Alrashedi	Alanoud Alhaider
Wareef Almousa	Futoon Almotairi
Joud Alangari	Manal Aldhirgham
Fay Alluhaidan	Raaoum Jabor
Sarah Alhamlan	Norah alawlah
Arwa Almobeirek	Shahad Helmi
Jumana AL-qahtani	Rand Aldajani

Boys



★ **Leader: Khalid Alhamdi**

Members:

Ahmed Alayban	Faisal Alhmoud
Sultan Alosaimi	Abdulrahman Alnoشان
Abdullah Alomran	Ahmed Alqahtani
Bassam Alghizzi	Hamad Alshaalan
Ibrahim Aljurayyan	Anas Alharbi
Mohammed Almutairi	Mohammed Alwahibi
Turki Alkhalifa	Saad Alghadir
Malik Alshaya	Firas Alqahtani



BiochemistryTeam441@gmail.com