ABNORMALITIES IN ACID-BASE BALANCE

Objectives

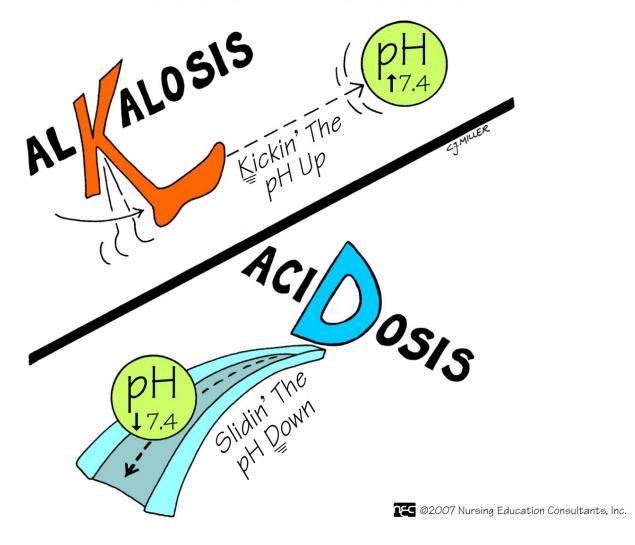
- To explain the principles of blood gas and acid-base analysis
- To interpret blood gas analysis and diagnose various acid base disorders
- Describe causes of acid base disorders
- Understand use of acid base nomograms

Abnormalities in Acid-Base Balance

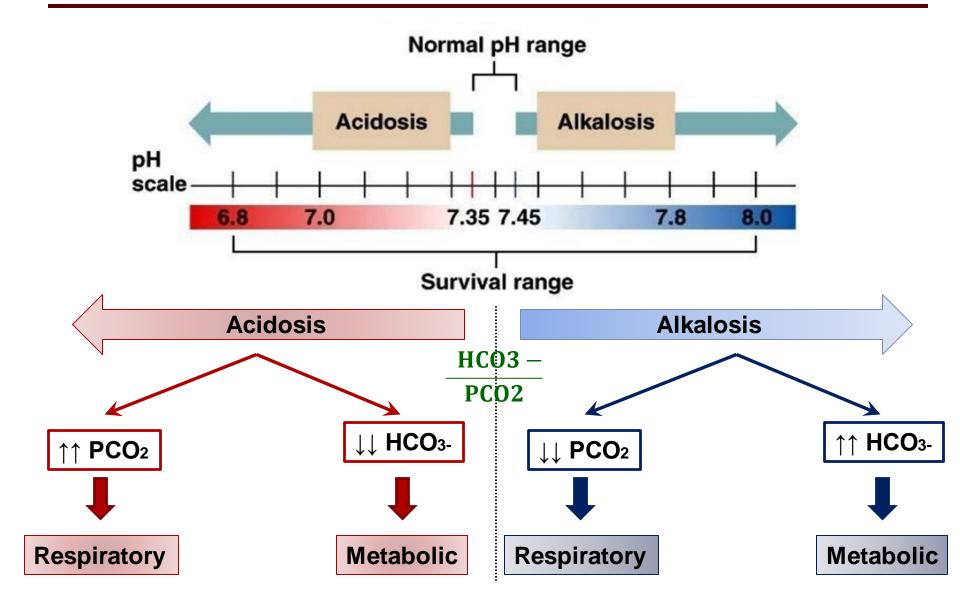
- Many critical illnesses can disturb acid-base balance.
- Acid-base disturbances may indicate an underlying disease or organ damage.
- Accurate interpretation of acid-base disturbances requires the following:
 - ✓ Arterial blood gases.
 - Plasma electrolytes.
 - ✓ Knowledge of the compensatory physiologic mechanisms.

Abnormalities in Acid-Base Balance

ACIDOSIS - ALKALOSIS



Abnormalities in Acid-Base Balance



Primary Acid-Base Disorders

There are 4 *primary* acid-base disorders

Defence mechanism of the body Compensatory/secondary response

Respiratory acidosis

Respiratory alkalosis

Metabolic acidosis

Metabolic alkalosis

The kidney compensates "Renal compensation"

The lung compensates "Respiratory compensation"

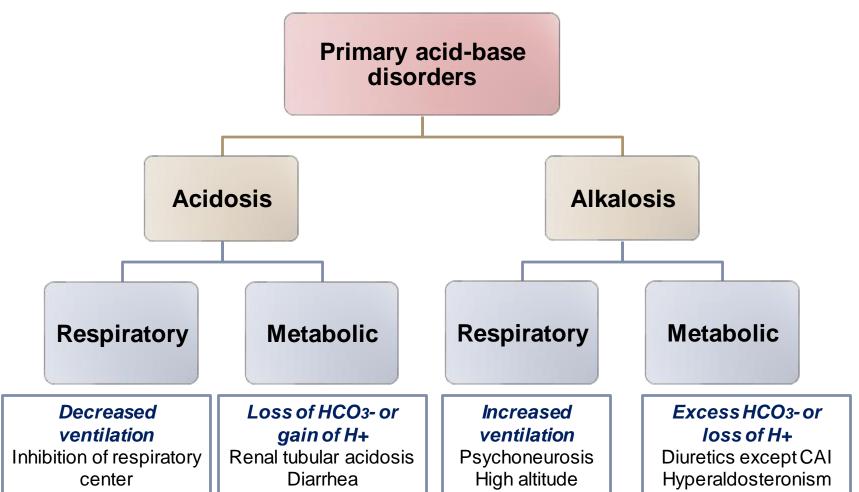
If a person develops any of these disorders,

What will the body try to do?

Fundamentals in Acid-Base Disorders

- Acid-base disorders are classified by changes in pH, PCO₂ and HCO₃⁻
- There are 4 <u>primary</u> acid-base disorders:
 - ➤ Respiratory acidosis: ↑ PCO₂
 - ➤ Respiratory alkalosis: ↓ PCO₂
 - Metabolic acidosis: ↓ [HCO₃-]
 - Metabolic alkalosis: ↑ [HCO₃-]
- The body normally attempts to correct the primary acidbase disturbances by a <u>secondary</u> or <u>compensatory</u> response trying to restore pH towards normal.
 - > The *kidneys* compensate for primary *respiratory disorders*.
 - > The *lungs* compensate for primary *metabolic disorders*.

Primary Acid-Base Disturbances



ventilation
Inhibition of respiratory
center
Lung damage
Airway obstruction
COPD

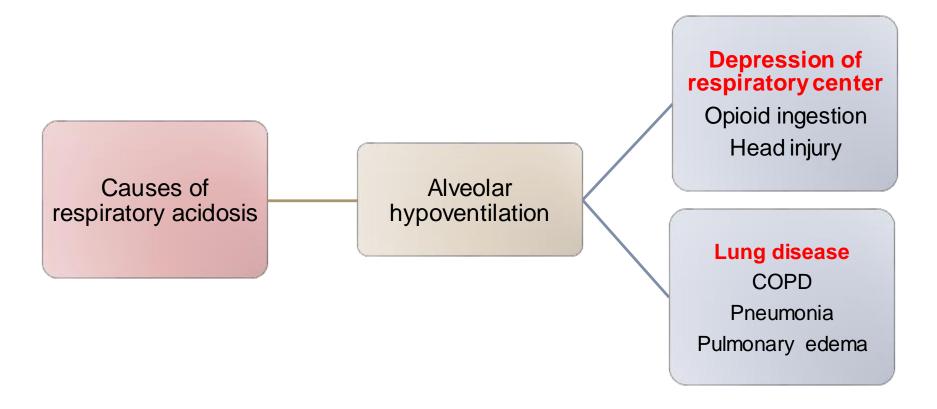
Renal tubular acidosis
Diarrhea
Diabetes
Ingestion of acids
(alcohol or aspirin)
Ch. renal failure

Psychoneurosis
High altitude
Mechanical
overventilation
Pregnancy

Diuretics except CAI
Hyperaldosteronism
Vomiting gastric
contents
Ingestion of antacids

Respiratory Acidosis

- Respiratory acidosis = ↓ pH + ↑ PCO2
 - Due to alveolar hypoventilation.



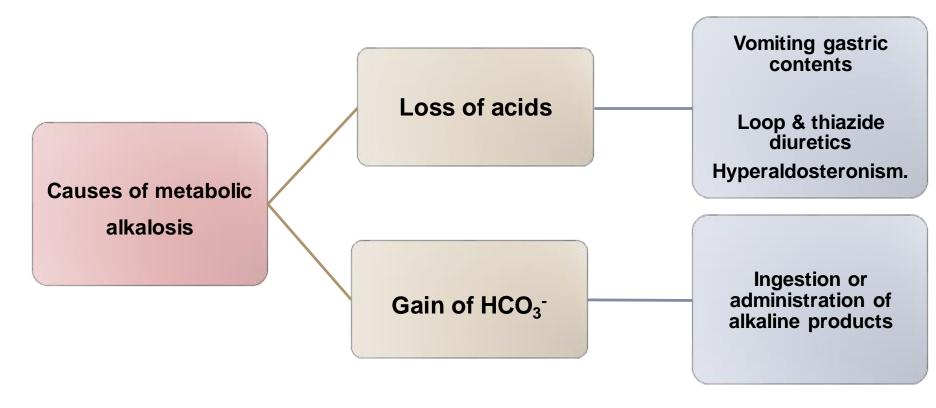
Respiratory Alkalosis

- Respiratory alkalosis = ↑ pH + ↓ PCO2
 - Due to alveolar hyperventilation.



Metabolic Alkalosis

- Metabolic alkalosis = ↑ pH +↑ [HCO₃-]
 - ➤ Due to loss of acids.
 - ➤ Due to gain of HCO₃-



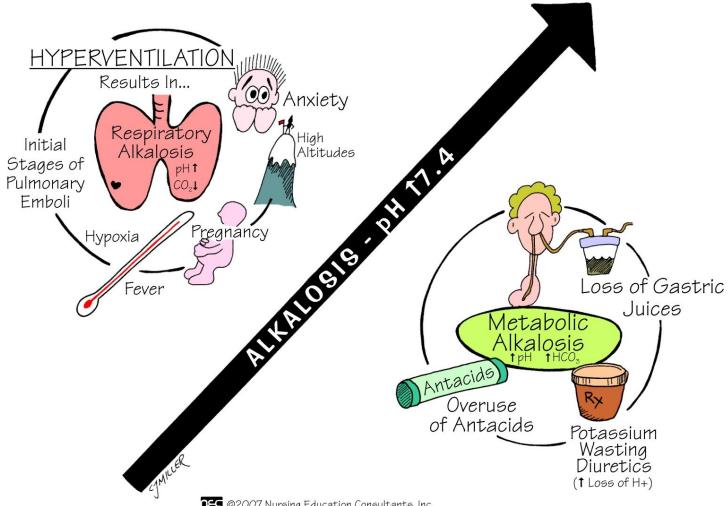
(Appel & Downs. 2008. Understanding acid-base balance; Dooley & Sisson. Acid-base disorders)

Metabolic Acidosis

Metabolic acidosis = \downarrow pH due to \downarrow [HCO₃⁻] Due to acid gain. ↑acid production Due to loss of HCO3 Lactic acidosis Diabetic ketoacidosis Salicylate poisoning Starvation Gain of acids **↓** acid elimination Causes of metabolic Renal failure acidosis Through kidneys RTA CAL Loss of HCO₃-Aldosterone deficiency **Through GIT** Diarrhea

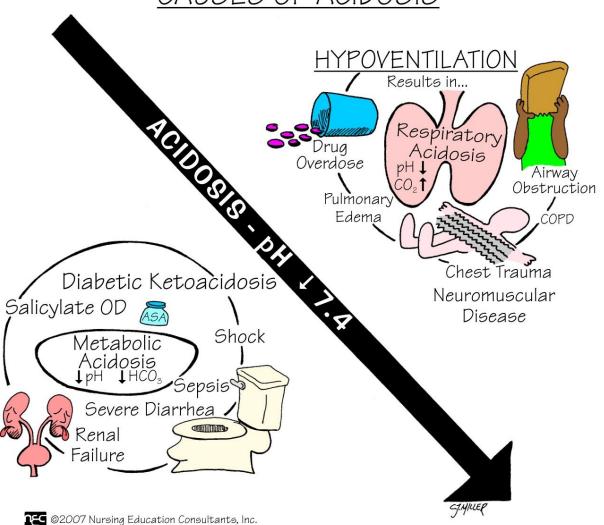
(Appel & Downs. 2008. Understanding acid-base balance; Dooley & Sisson. Acid-base disorders)

CAUSES OF ALKALOSIS



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CAUSES OF ACIDOSIS



Compensatory Mechanisms

Primary Disturbance	Compensatory Mechanism
Respiratory Acidosis	Increase HCO3
Respiratory Alkalosis	Decrease HCO3
Metabolic Acidosis	Decrease PCO2
Metabolic Alkalosis	Increase PCO2

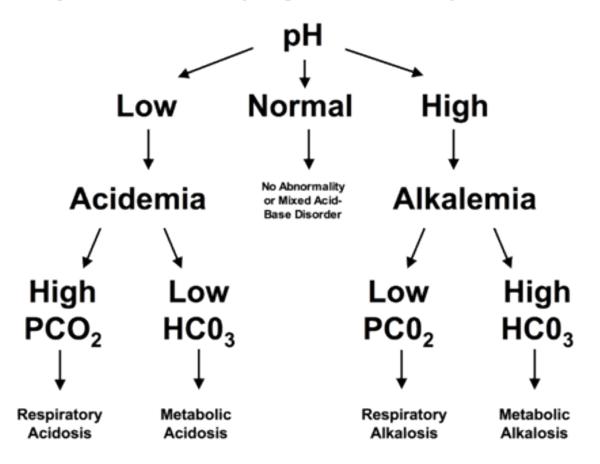
Summary of Primary Acid-Base Disorders

Acid Base Disorders

Disorder	pH	[H*]	Primary disturbance	Secondary response
Metabolic acidosis	1	1	↑ [HCO³.]	↓ pco₂
Metabolic alkalosis	1	\	↑ [HCO³.]	↑ pco₂
Respiratory acidosis	1	1	↑ pCO ₂	↑ [HCO³-]
Respiratory alkalosis	1	1	↑ pco²	↑ [HCO³-]

Interpretation of Acid-Base Disturbances

Figure 1: Identifying the Primary Process



Normal values,

pH =7.35-7.45 PCO2 =35-45 mmHg HCO3-= 22-28 mmol/L

Simple Acid-Base Disturbances

	рН	PCO2 (mmHg)	HCO3 (mEq/L)
Normal	7.35-7.45	35-45	22-28
Respiratory acidosis	Decrease	Increase	Increase
Respiratory alkalosis	Increase	Decrease	Decrease
Metabolic acidosis	Decrease	Decrease	Decrease
Metabolic alkalosis	Increase	Increase	Increase

- A patient known to have COPD presented with 3-day history of fever, SOB, and cough productive of yellowish sputum. His ABGs showed:
 - pH = 7.25
 - PCO₂ = 80 mmHg.
 - $[HCO_3^-] = 34 \text{ mEq/L}.$

 A 21 year old man with IDDM presents to ER with mental status changes, nausea, vomiting, abdominal pain and rapid respirations. His ABGs showed:

- pH = 7.2
- PCO₂ = 20 mmHg
- $[HCO_3^-] = 8 \text{ mEq/l}$

Check this video:

https://www.youtube.com/watch?v=raEKXVfuWTo

 A 2-year old child who is lethargic and dehydrated has a 3-day history of vomiting. His ABGs showed:

- pH = 7.56
- PCO₂ = 44 mmHg
- $[HCO_3^-] = 37 \text{ mEq/l}$

 A 20-year old student suffered a panic attack while awaiting an exam. Her ABGs showed:

- pH = 7.6
- PCO₂ = 24 mmHg.
- $[HCO_3^-] = 23 \text{ mEq/L}.$

Other Acid-Base Disorders

Simple acid-base disorders

Result from a single
 primary abnormality with appropriate physiologic compensation.

Mixed acid-base disorders

 Result from multiple primary processes.

Mixed Acid-Base Disturbances

 Occurs when a patient has more than one primary acid base disorder that occur at the same time.

Examples:

- Respiratory alkalosis/acidosis along with a metabolic acidosis/alkalosis.
- ✓ Two metabolic acid-base disorders occurring simultaneously.

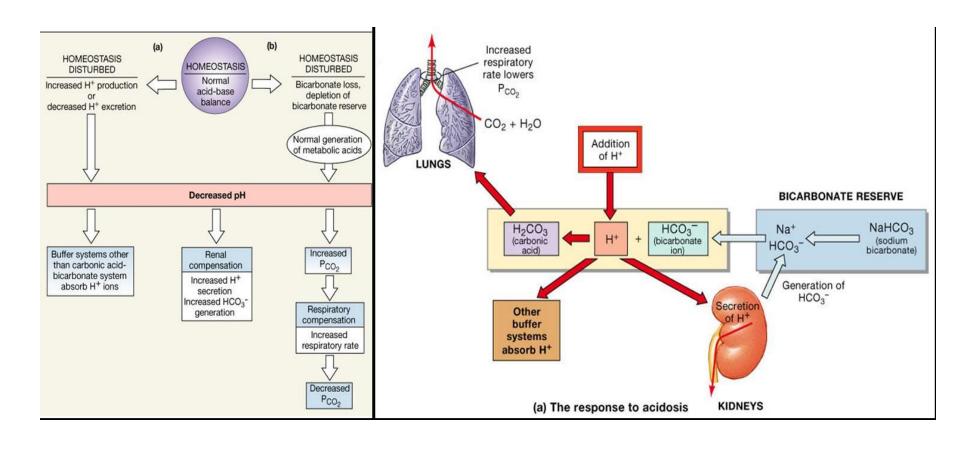
 A 69 year old patient known to have COPD presented with a 3-day history of abdominal pain and diarrhea. His ABGs showed;

•
$$pH = 6.96$$

- PCO₂ = 55mmHg
- $[HCO_3^-] = 12 \text{ mmol/L}$

THANK YOU

Body's Response to Acidosis



Body's Response to Alkalosis

