## Kidney Stones

Renal Block

1 Lecture



### **Objectives**

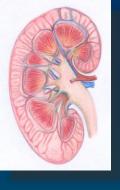
By the end of this lecture, the students will be able to:

- Discuss the general physiological and pathological factors that favor kidney stones formation
- List the types of kidney stones, their chemical constituents and characteristics
- ■Identify the etiological causes of each type of kidney stone
- ■Discuss the diagnosis, treatment and prevention of kidney stones



#### Overview

- Introduction
- Conditions causing kidney stone formation
- Types of kidney stones
  - Calcium salts
  - Uric acid
  - $\overline{\text{-}}$  Mg ammonium PO $_4$
  - Cystine
  - Other (xanthine, etc.)
- Laboratory investigations



### What are kidney stones?

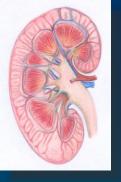
Renal calculi (kidney stones) are formed in renal tubules, ureter or bladder

Composed of metabolic products present in glomerular filtrate

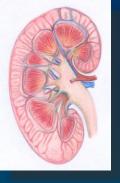
- These products are in high conc.
  - Near or above maximum solubility



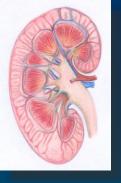
- High conc. of metabolic products in glomerular filtrate
- Changes in urine pH
- Urinary stagnation
- Deficiency of stone-forming inhibitors in urine



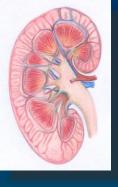
- High conc. of metabolic products in glomerular filtrate is due to:
  - Low urinary volume (with normal renal function) due to restricted fluid intake
  - Increased fluid loss from the body
  - Increased excretion of metabolic products forming stones
  - High plasma volume (high filtrate level)
  - Low tubular reabsorption from filtrate



- Changes in urine pH due to:
  - Bacterial infection
  - Precipitation of salts at different pH
- A persistently acidic urine → promotes uric acid precipitation
- A persistently alkaline urine (due to upper urinary tract infection)→ promotes Mg Ammonium Phosphate crystals (Struvite stones)
- Urinary stagnation is due to:
  - Obstruction of urinary flow

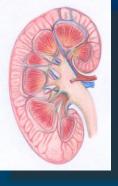


- Deficiency of stone-forming inhibitors:
  - Citrate, pyrophosphate, glycoproteins inhibit growth of calcium phosphate and calcium oxalate crystals
  - In type I renal tubular acidosis, hypocitraturia leads to renal stones



### Types of kidney stones

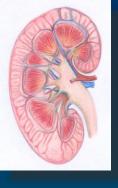
- Calcium salts
- Uric acid
- Mg ammonium PO<sub>4</sub>
- Cystine
- Other (xanthine, etc.)



#### 80% of kidney stones contain calcium:

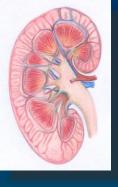
Mostly Ca-Oxalate and less often Ca-Phosphate

- ■The type of salt depends on
  - -Urine pH
  - -Availability of oxalate
- ■General appearance:
  - -White, hard, radio-opaque
  - -Calcium oxalate: present in ureter (small)
  - -Calcium  $PO_4$ : staghorn in renal pelvis (large)



#### Causes of calcium salt stones:

- Hypercalciuria:
  - Increased urinary calcium excretion
  - Men: > 7.5 mmols/day
  - Women > 6.2 mmols/day
  - Due to hypercalcemia (most often due to 1<sup>ary</sup> hyperparathyroism)
  - sometimes, Ca<sup>++</sup> salts stones are found with no hypercalcemia



#### ■ Hyperoxaluria:

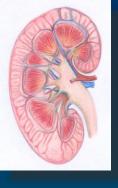
- Causes the formation of calcium oxalates without hypercalciuria
- Diet rich in oxalates
- Increased oxalate absorption in fat malabsorption

#### ■ Primary hyperoxaluria:

- Due to inborn errors
- Urinary oxalate excretion: > 400 μmol/24 Hours

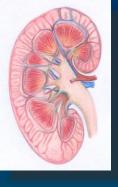


Calcium oxalate stones



#### **■** Treatment:

- Treatment of primary causes such as infection, hypercalcemia, hyperoxaluria
- Oxalate-restricted diet
- Increased fluid intake (if no glomerular failure)
- Acidification of urine (by dietary changes)
  - Calcium salt stones are formed in alkaline urine



#### Uric acid stones

- About 8% of renal stones contain uric acid
- May be associated with hyperuricemia (with or without gout)
- Form in acidic urine

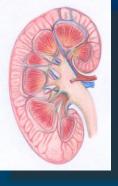
#### General appearance:

- Small, friable, yellowish
- May form staghorn (if big)
- Radiolucent (plain x-rays cannot detect)
- Visualized by ultrasound or i.v. pyelogram





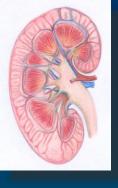
Uric acid stones



#### Uric acid stones

#### **Treatment:**

- Treatment of cause of hyperuricemia.
- Purine-restricted diet
- Alkalinization of urine (by dietary changes)
- Increased fluid intake

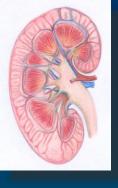


## Mg ammonium PO<sub>4</sub> stones

- About 10% of all renal stones contain Mg amm. PO<sub>4</sub>
- Also called struvite kidney stones
- Associated with chronic urinary tract infection
  - Microorganisms (such as from *Proteus* genus) that metabolize urea into ammonia
  - Causing urine pH to become alkaline leading to stone formation
- Commonly associated with staghorn calculi
- 75% of staghorn stones are of struvite type



Mg ammonium phosphate (struvite) stone



### Mg ammonium PO<sub>4</sub> stones

#### **Treatment:**

- Treatment of infection
- Urine acidification
- Increased fluid intake
- In some cases, it may require complete stone removal (percutaneous nephrolithotomy)

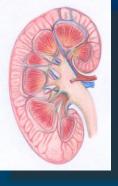


### Cystine stones

- A rare type of kidney stone
- Due to homozygous cystinuria
- Form in acidic urine
- Soluble in alkaline urine
- Faint radio-opaque



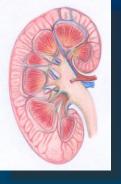
Cystine stone



### Cystine stones

#### **Treatment:**

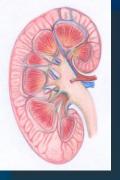
- Increased fluid intake
- Alkalinization of urine (by dietary changes)
- Penicillamine (binds to cysteine to form a compound more soluble than cystine)



# Laboratory investigations of kidney stones

#### If stone has formed and removed:

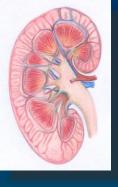
- **■** Chemical analysis of stone helps to:
  - Identify the cause
  - Advise patient on prevention and future recurrence



# Laboratory investigations of kidney stones

#### If stone has not formed:

- This type of investigation identifies causes that may contribute to stone formation:
  - Serum calcium, uric acid and PTH analysis
  - Urinalysis: volume, calcium, oxalates and cystine levels
  - Urine pH > 8 suggests urinary tract infection (Mg amm. PO<sub>4</sub>)
- Urinary tract imaging:
  - CT, ultrasound and i.v. pyelogram



#### References

- Clinical Chemistry and metabolic Medicine 7<sup>th</sup> Edition, pp. 36.
- The National Kidney Foundation, USA (www.kidney.org)