

Structure And Function Of Hemoglobin



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A hemeprotein found only in **RBCs**

Heme (prosthetic group)
+
Globin protein chains

Consist of

Definition

Function

- O_2 transport from the lungs to the tissues
- and then transport CO_2 back from the tissues to the lungs.

HEMOGLOBIN (Hb)

Normal level

Normal level (g/dL):
Males: 14-16
Females: 13-15

Types

HEME group

Abnormal

Normal

Caroxy Hb

Met Hb

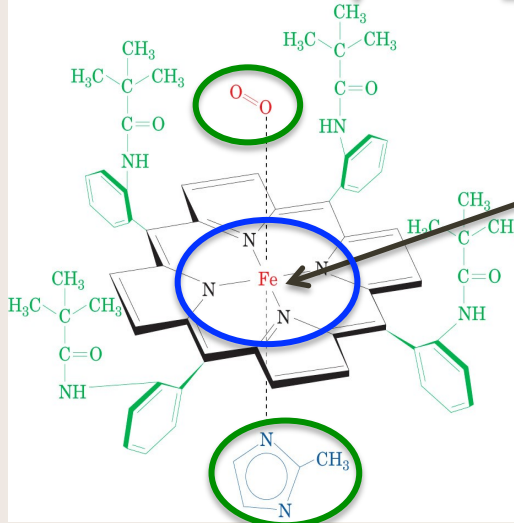
Sulf Hb

Hb A (97%)

HbA₂ (2%)

HbF (1%)

HbA_{1c}



protoporphyrin IX
+
Ferrous iron (Fe²⁺)

= Heme group

- ☐ Fe²⁺ is present in the center of the Heme
- ☐ Fe²⁺ binds to four nitrogen atoms of the porphyrin ring
- ☐ Forms two additional bonds with histidine residue of globin chain and Oxygen

Forms of Hemoglobin

R-form (Relaxed)	T-form (Taut or tense)
The oxygenated form of Hb	The deoxy (without O ₂) form of Hb
High-oxygen-affinity	Low-oxygen-affinity
The dimers have more freedom of movement	The movement of dimers is constrained

Types of Hemoglobin

Hemoglobin A (HbA)

Major Hb in adults

Composed of four polypeptide chains:

Two α and two β chains

Contains two dimers of $\alpha\beta$ subunits which are held together by non-covalent interactions

Each chain is a subunit with a heme group in the center that carries oxygen

A Hb molecule contains 4 heme groups and carries 4 molecules of O₂

Fetal Hemoglobin (HbF)

Major hemoglobin found in the fetus and newborn

Tetramer with two α and two γ chains

Higher affinity for O₂ than HbA

Transfers O₂ from maternal to fetal circulation across placenta

HbA₂

Appears ~12 weeks after birth

Constitutes ~2% of total Hb

Composed of two α and two δ globin chains

HbA_{1c}

HbA undergoes non-enzymatic glycosylation

Glycosylation depends on plasma glucose levels

HbA_{1c} levels are high in patients with diabetes mellitus

Factors affecting oxygen binding* (allosteric effectors)

pO_2
(partial pressure of O_2)

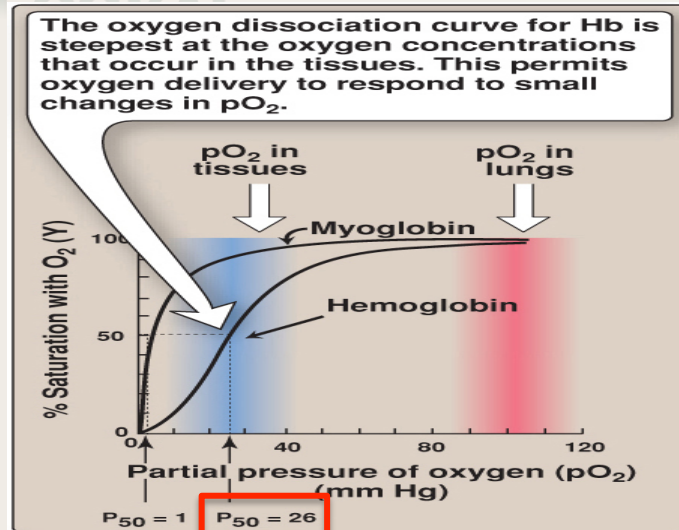
pH of the
environment

pCO_2
(partial pressure of
carbon dioxide)

Availability of
2,3 biphosphoglycerate

OXYGEN DISSOCIATION CURVE

- ❑ The curve is sigmoidal¹ (s-shaped)
- ❑ Indicates cooperation of subunits in O_2 binding
- ❑ Binding of O_2 to one heme group increases O_2 affinity of others
- ❑ Heme-Heme interaction²



*: About 97% of oxygen transported in the blood is bound to hemoglobin.

1- The sigmoidal curve reflect structural changes that are initiated at one heme group and transmitted to other heme groups in the Hb tetramer

2- The affinity of Hb for the last O_2 bound is about 300 times greater than its affinity for the first O_2 bound

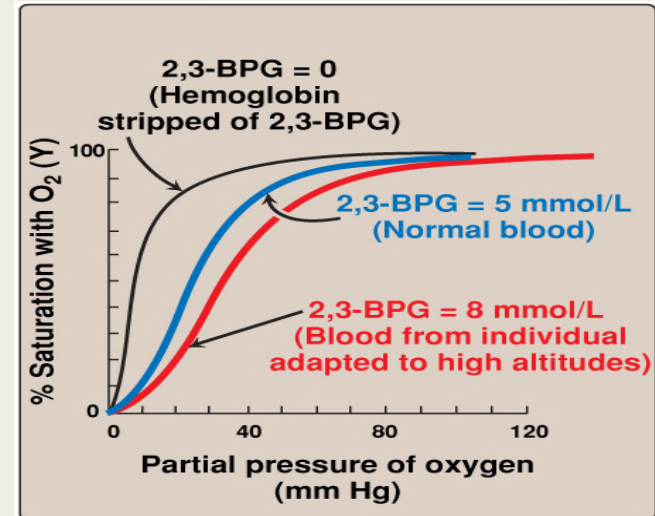
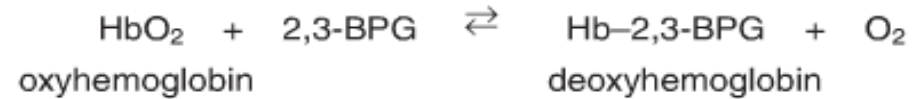
P_{50} (mm Hg)

- ✓ It is the pressure at which Hb is 50% saturated with O_2
 - ✓ Indicates affinity of Hb to O_2
 - ❑ High affinity \rightarrow slow unloading of O_2 .
 - ❑ Low affinity \rightarrow fast unloading of O_2 .
(inverse proportion)
 - ❑ Lung pO_2 is 100 mm \rightarrow Hb saturation 100%
 - ❑ Tissue pO_2 is 40 mm \rightarrow Hb saturation reduces.
- \Rightarrow Hence O_2 is delivered to tissues.
- ❑ P_{50} of hemoglobin = 26-28 mm Hg

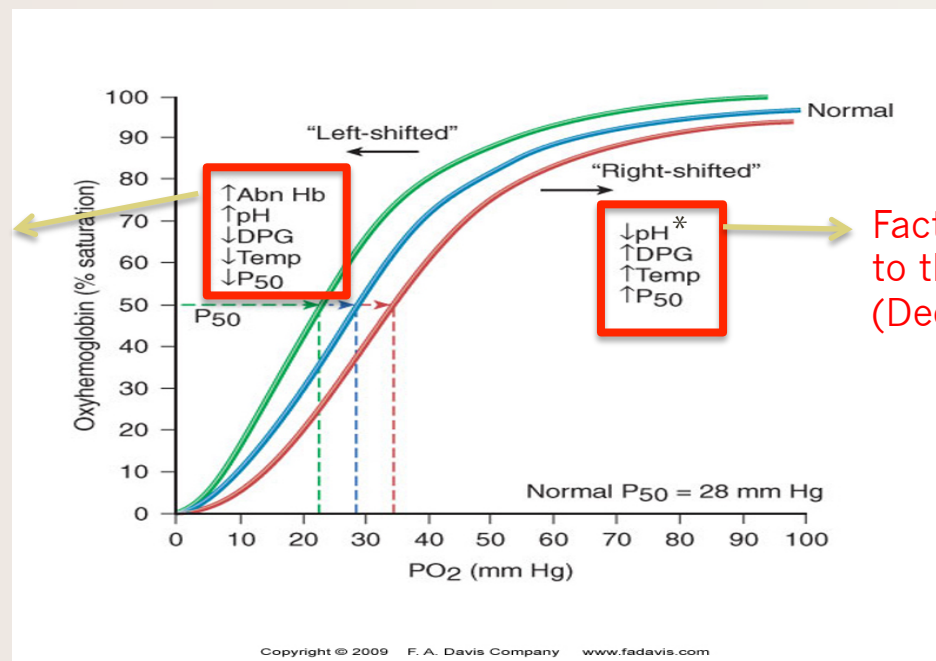
Availability of 2,3 bisphosphoglycerate

Binds to deoxy-hb and stabilizes the T-form

When oxygen binds to Hb, BPG is released



Factors that shift the curve to the LEFT
(Increase Hb affinity to O₂)



Factors that shift the curve to the RIGHT
(Decrease Hb affinity to O₂)

High altitude & O₂ affinity

In hypoxia and high altitude:

- 2,3 BPG levels rise
- This decreases O₂ affinity of Hb
- Thus increases O₂ delivery to tissues

Causes of high O₂ affinity

- Alkalosis
- High levels of Hb F
- Multiple transfusion of 2,3 DPG-depleted blood

*: Decreased PH means increased release of H⁺

The Bohr effect

Effect:

- ❑ removes insoluble CO₂ from blood stream
- ❑ Produces soluble bicarbonate

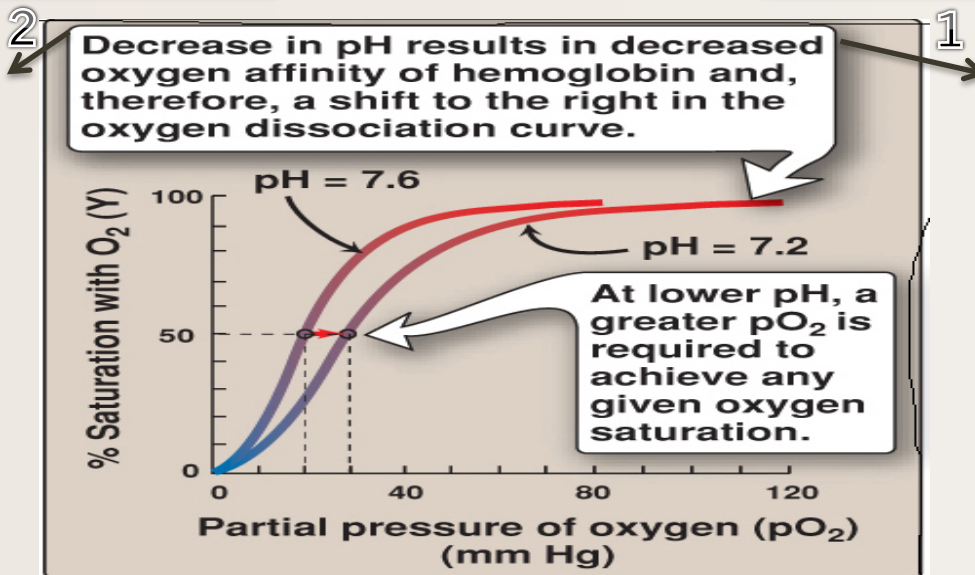
Definition:

- Effect of pH and pCO₂ on:
- ❑ Oxygenation of Hb in the lungs
 - ❑ Deoxygenation of Hb in tissues

Tissues have lower pH (acidic) than lungs due to proton generation:

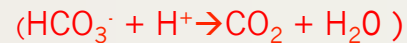


Protons reduce O₂ affinity of Hb
=> Shift to the right



The free Hb binds to two protons.

Protons are released and react with HCO₃⁻ to form CO₂ gas



The proton-poor Hb now has greater affinity for O₂ (in lungs)

Abnormal hemoglobin:

Unable to transport O₂ due to abnormal structure

Carboxy-Hb:

CO replaces O₂ and binds 200X tighter than O₂
(in smokers)

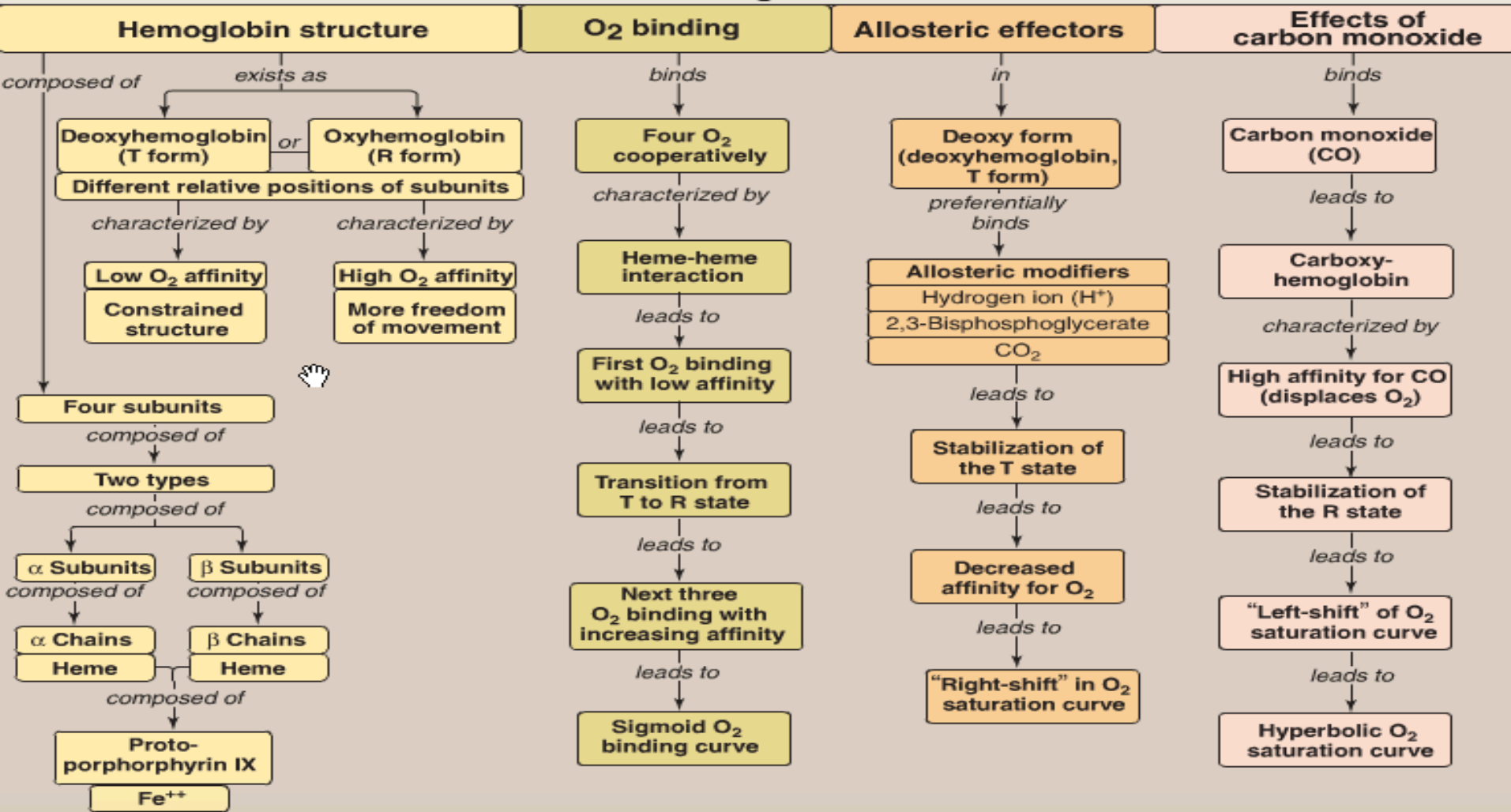
Met-Hb:

Contains oxidized Fe³⁺ (~2%)
that cannot carry O₂

Sulf-HB:

Forms due to high sulfur levels in blood
(irreversible reaction)

Hemoglobin



1) The form of Iron that is bound to hemoglobin is:

- A- Ferric iron**
- B- Ferrous iron**
- C- Ferritin**
- D- All of them**

2) Which ONE of the following statements is true about the T-form of hemoglobin :

- A- Relaxed-form**
- B- Oxygenated-form of Hb**
- C- Have freedom of movement.**
- D- Low oxygen affinity**

3) HbA2 is composed of :

- A- two α and two γ chains**
- B- two α and two δ globin chains**
- C- four α globin chains**
- D- Two α and two β chains**

4) Which ONE of the following cause shift of O_2 dissociation curve to the left

- A- Increased H^+**
- B- High PH**
- C- Increased 2,3 BPG**
- D- Increased P_{50}**

5) 2,3 bisphosphoglycerate binds to Hb in the form of:

- A- Oxyhemoglobin**
- B- Deoxy-hemoglobin**
- C-Fetal Hb**
- D- None of them**

6) What do we expect to see in the CBC of a patient from ABHA :

- A- High levels of hemoglobin**
- B-Low O_2 affinity**
- C- Decreased RBCs**
- D- Low levels of 2,3 bisphosphoglycerate**

7) Which ONE of the following replaces O_2 and binds 200X tighter than O_2

- A-Met hemoglobin**
- B- Carboxy-hemoglobin**
- C- Sulf-hemoglobin**
- D- Fetal hemoglobin**

8) Which of the following result from Bohr effect:

- A- Removes insoluble CO_2 from blood stream**
- B- Produces soluble bicarbonate**
- C- Produce CO_2 to the blood stream**
- D- A&B**

Answers: 1) B 2) D 3) B 4) B 5) B 6) A 7) B 8) D



Thank You!

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