Structure and function of hemoglobin

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Hemoglobin (Hb)

- A hemeprotein found only in red blood cells
- Oxygen transport function
- Contains heme as a prosthetic group
- Heme reversibly binds to oxygen

The heme group

- A complex of protoporphyrin IX and ferrous iron (Fe²⁺)
- Fe²⁺ is present in the center of heme
- Binds to four nitrogens of the porphyrin ring
- Plus two additional bonds with:
 - Histidine residue of globin chain
 - Oxygen

The heme group: Fe²⁺– porphyrin complex with bound O₂

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Types of Hb

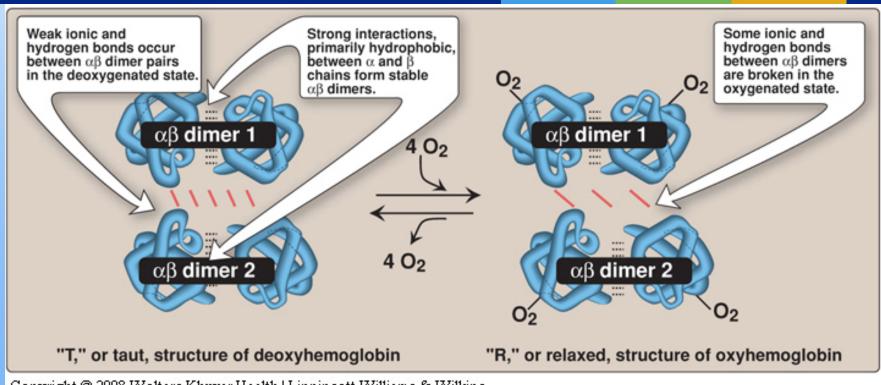
Normal:	HbA (97%)
	HbA ₂ (2%)
	HbF (1%)
	HbA _{1c}
Abnormal:	Carboxy Hb
	Met Hb
	Sulf Hb

Hemoglobin A (HbA)

- Major Hb in adults
- Composed of four polypetide chains:
 - **Two** α and two β chains
- Contains two dimers of αβ subunits
- Held together by noncovalent interactions
- Each chain is a subunit with a heme group in the center that carries oxygen
- A Hb molecule contains 4 heme groups and carries 4 moelcules of O₂

Polypeptide chains β chains (146 a.a.) Fe²⁺ α chains (141 a.a.) Heme (protoporphyrin + iron)

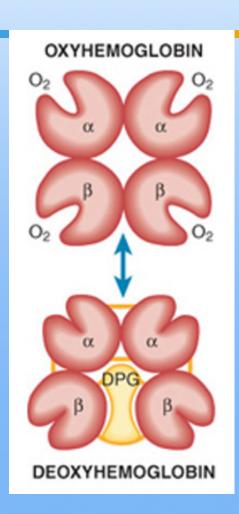
HbA structure



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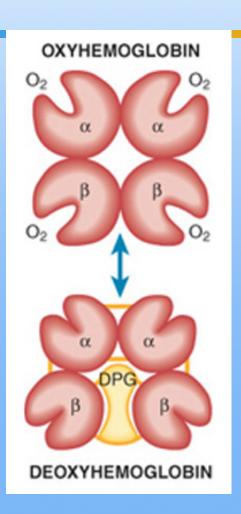
T-form of Hb

- The deoxy form of Hb
- Taut form
- The movement of dimers is constrained
- Low oxygen affinity form



R-form of Hb

- The oxygenated form of Hb
- Relaxed form
- The dimers have more freedom of movement
- High-oxygen-affinity form



Hemoglobin function

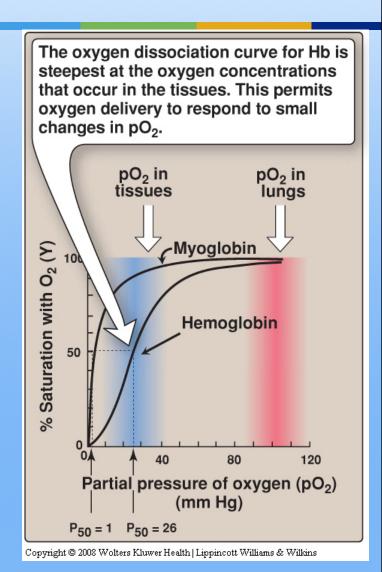
- Carries oxygen from the lungs to tissues
- Carries carbon dioxide from tissues back to the lungs
- Normal level:
 - Males: 14-16 g/dL
 - Females: I3-I5 g/dL

Factors affecting oxygen binding

- Three allosteric effectors:
 - pO₂ (partial oxygen pressure)
 - pH of the environment and pCO₂
 (partial carbon dioxide pressure)
 - Availability of 2,3-bisphosphoglycerate

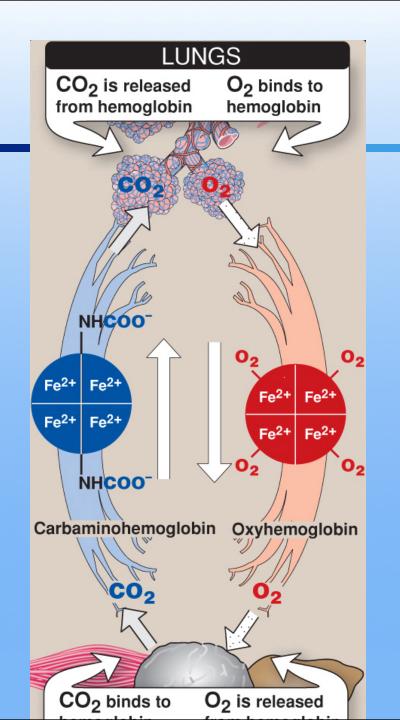
Oxygen Dissociation Curve (ODC)

- The curve is sigmoidal
- Indicates cooperation of subunits in O₂ binding
- Binding of O₂ to one heme group increases O₂ affinity of others
- Heme-heme interaction



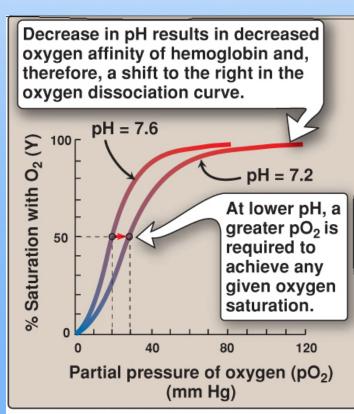
pO2 (Partial oxygen pressure)

- p50 (mm Hg): the pressure at which Hb is 50% saturated with O₂
- Indicates affinity of Hb to O₂
- High affinity \rightarrow slow unloading of O_2
- Low affinity \rightarrow fast unloading of O_2
- Lung pO_2 is 100 mm \rightarrow Hb saturation 100%
- Tissue pO₂ is 40 mm → Hb saturation reduces
- Hence, oxygen is delivered to tissues



The Bohr effect

- Effect of pH and pCO₂ on:
 - Oxygenation of Hb in the lungs
 - Deoxygenation at the tissues
- Tissues have lower pH (acidic) than lungs
- Due to proton generatation: $CO_2 + H_20 ----> HCO_3^- + H^+$
- Protons reduce O₂ affinity of Hb
- Causing easier O₂ release into the tissues
- The free Hb binds to two protons



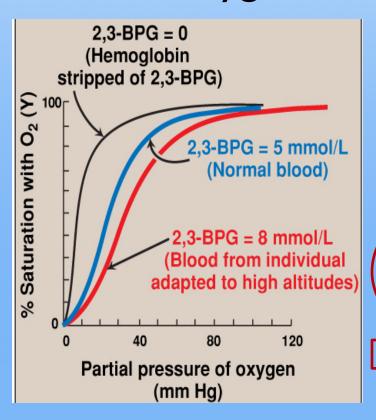
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The Bohr Effect

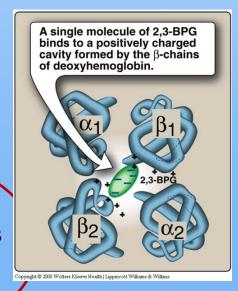
- Protons are released and react with HCO³ to form CO₂ gas
- The proton-poor Hb now has greater affinity for O₂
- The Bohr effect removes insoluble CO2 from blood stream
- Produces soluble bicarbonate

Availability of 2,3-bisphosphoglycerate

- Binds to deoxy-Hb and stabilizes the T-form
- When oxygen binds to Hb, BPG is released



At high altitudes there is -increase in no. of RBCs -Increase in conc. Of Hb -Increase in BPG



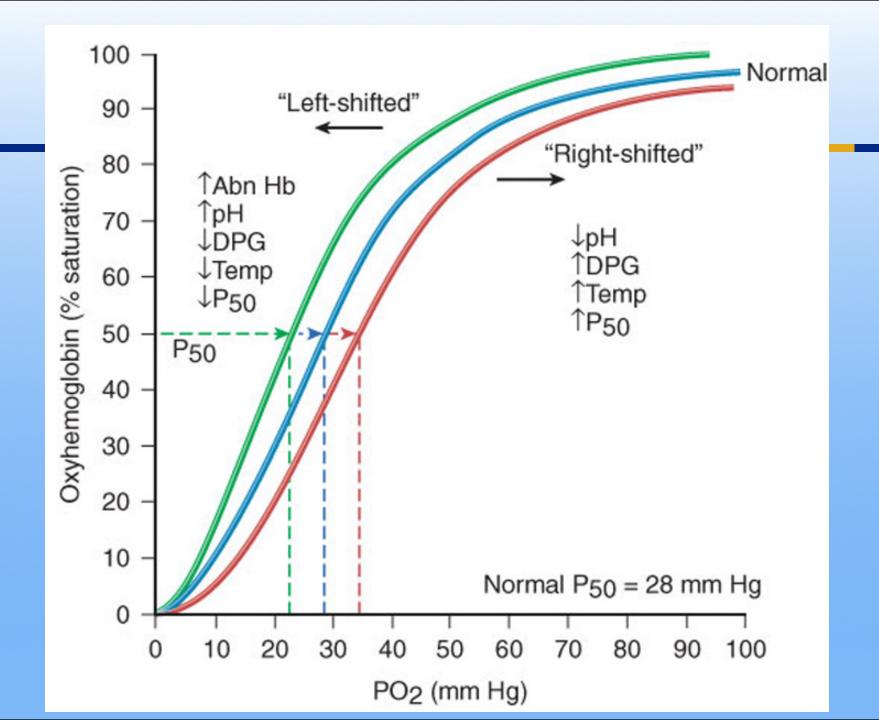
High altitude and O_2 affinity

- High altitude decreases Hb O2 affinity
- Hypoxia
 - Increases 2,3 DPG levels
 - Decreases O2 affinity
 - Increases O₂ delivery to tissues

High O₂ affinity

High O2 affinity occurs due to:

- Alkalosis
- High levels of Hb F
- Multiple transfusion of 2,3 DPG-depleted blood



Fetal Hemoglobin (HbF)

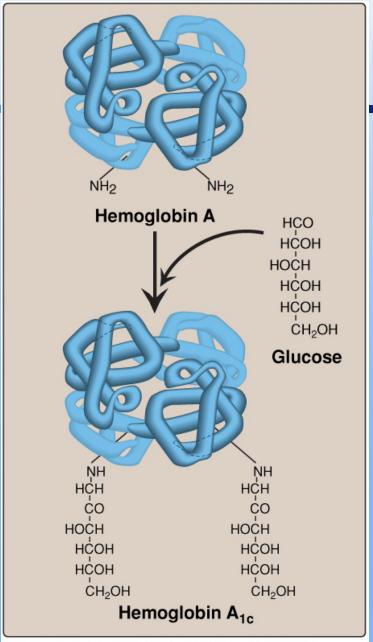
- Major hemoglobin found in the fetus and newborn
- **Tetramer** with two α and two γ chains
- Higher affinity for O₂ than HBA
- Transfers O₂ from maternal to fetal circulation across placenta



- Appears ~12 weeks after birth
- Constitutes ~2% of total Hb
- \blacksquare Composed of two α and two δ globin chains



- HbA is slowly and nonenzymatically glycosylated
- Glycosylation depends on plasma glucose levels
- HbAIc levels are high in patients with diabetes mellitus



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Abnormal Hbs

- Unable to transport O₂ due to abnormal structure
- □ Carboxy-Hb: CO replaces O₂ and binds 200X tighter than O₂(in smokers)
- Met-Hb: Contains oxidized Fe3+ (~2%) that cannot carry O₂
- Sulf-Hb: Forms due to high sulfur levels in blood (irreversible reaction)

References

■ Lippincott's Illustrated Reviews- Biochemistry (pp 25-34)