

Water & Sodium Disorders **(H₂O/Na⁺)**

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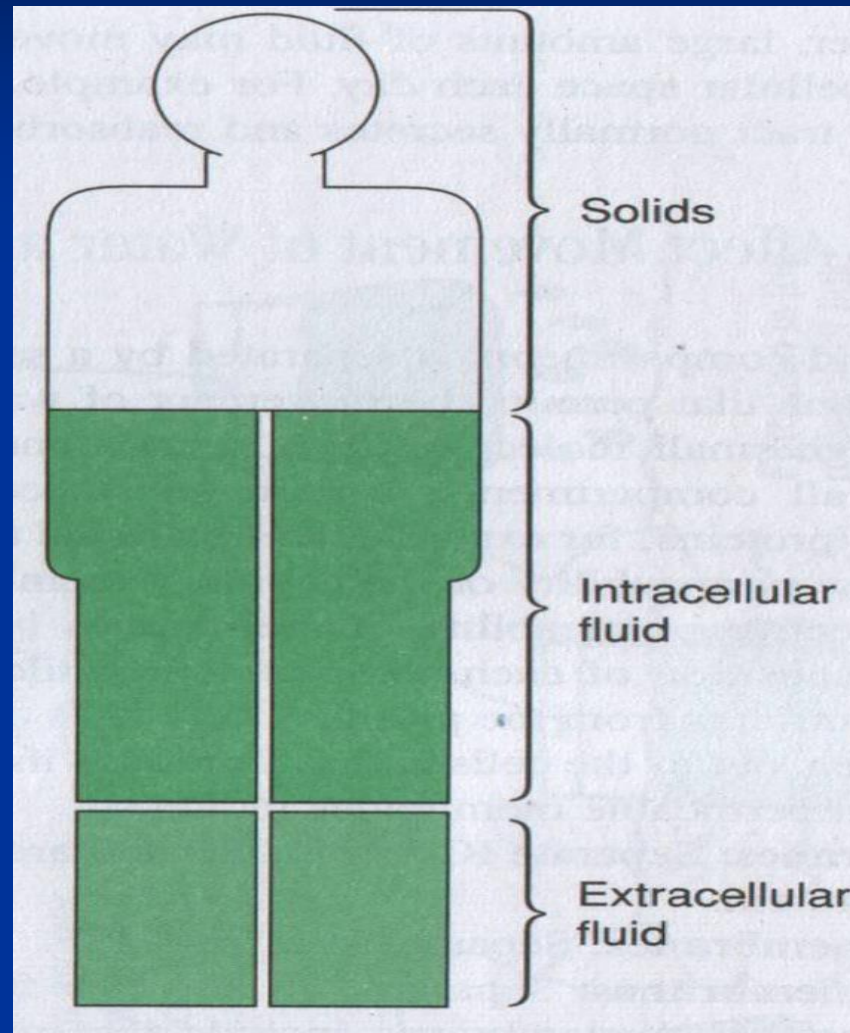
Objectives

- 1) Composition of the fluid compartments
- 2) Mechanisms which regulate fluid and Sodium balance
- 3) Disorders of water balance
- 4) Disorders of Sodium balance

Homeostasis

A relative constancy in the internal environment of the body, naturally maintained by adaptive responses that promote cell function and survival

Total Body Fluid:



Total body water ▶ TBW is affected by gender and age.



TBW = 50%



TBW = 60%

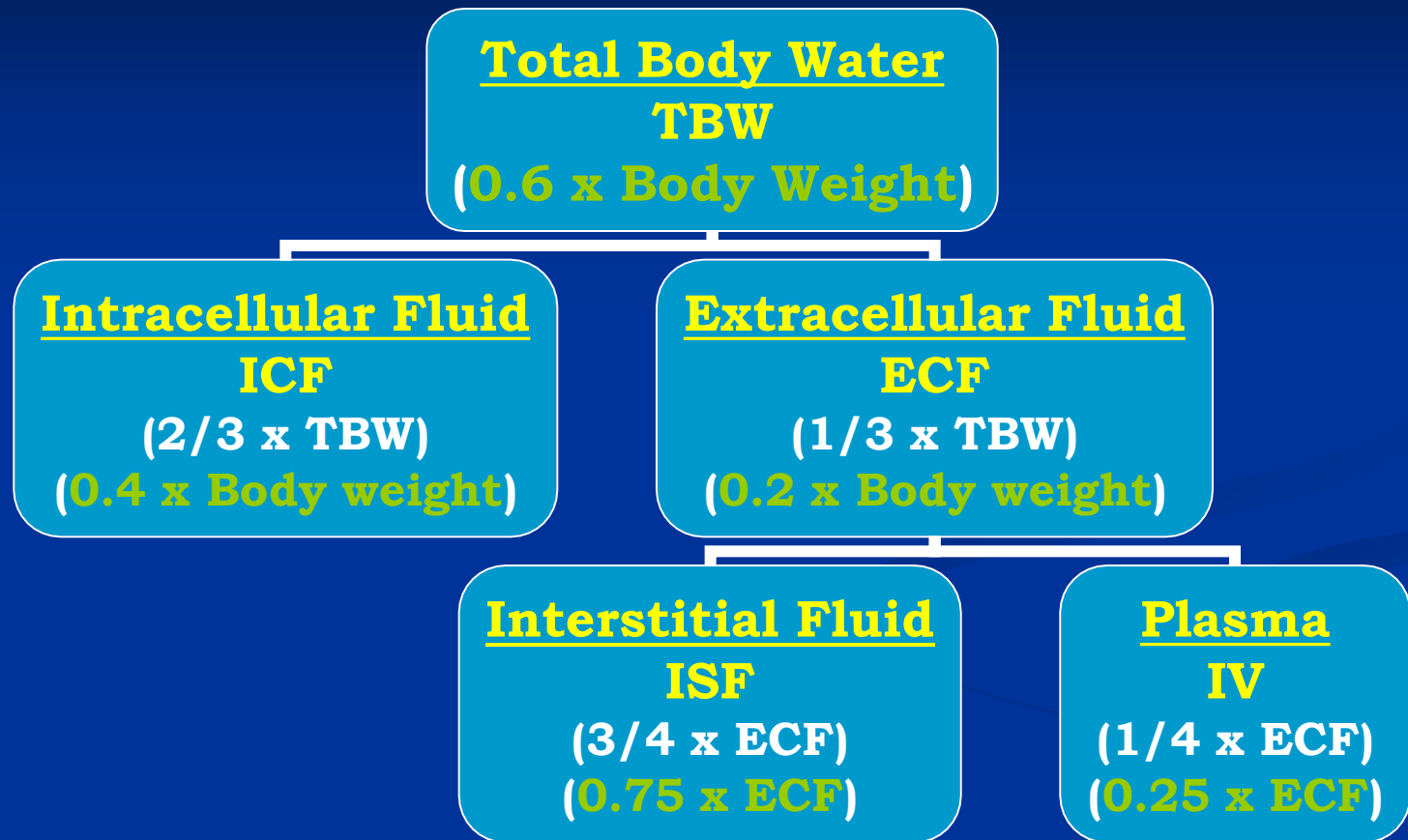


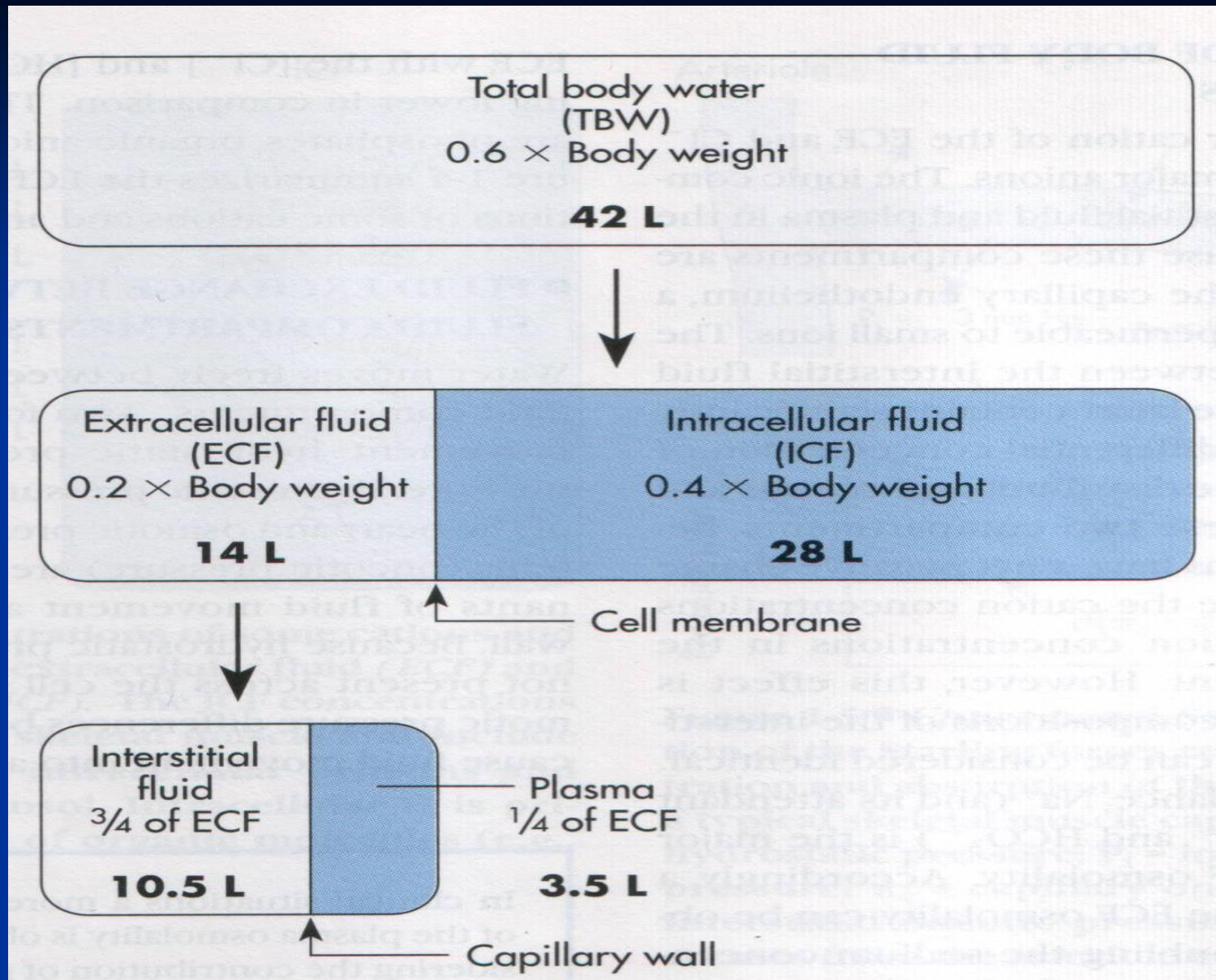
TBW = 70%



TBW = 50%

Body Fluid Compartments





Body Fluid Compartments

- Fluid compartments are separated by thin **semi-permeable membranes** with pores to allow fluid movement and molecules of a specific size to pass while preventing larger heavier molecules from passing
- The bodies fluid is composed of water and dissolved substances known as **solutes** (electrolytes or non-electrolytes)
- Electrolytes are substances dissolved in solutions and dissociated into particles called **ions**
 - **Cations**: Positively charged ions
 - **Anions**: Negatively charged ions

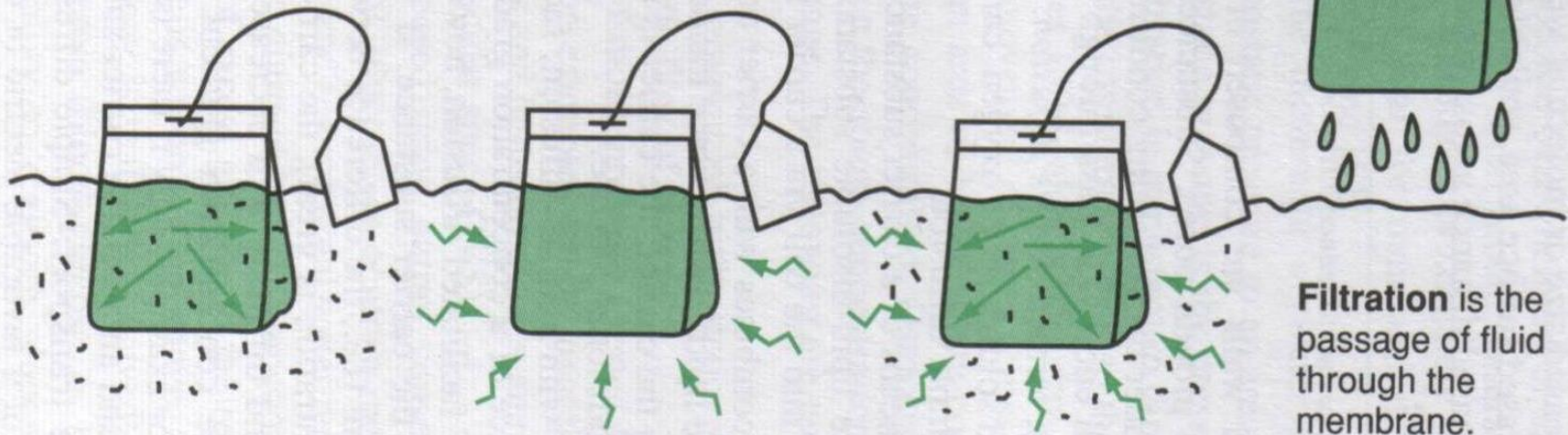
Definitions:

- ❑ Osmosis: movement of water
- ❑ Diffusion: movement of solutes
- ❑ Filtration: movement of both solutes and water
- ❑ Osmolality:
 - ❑ Osmoles in solution: mOsm/kg water
 - ❑ $\text{Calc Posm} = (2 \times \text{serum Na}^+) + \text{blood urea} + \text{glucose}$
 - ❑ For Na⁺, K⁺ and Cl⁻: 1 mEq = 1 mOsm
 - ❑ Normal osmolality of body fluids: 283-292 mOsm/kg water

Diffusion is the passage of particles through a semipermeable membrane. Tea, for example, diffuses from a tea bag into the surrounding water.

Osmosis is the movement of fluid across a semipermeable membrane from a lower concentration of solutes to a higher concentration of solutes.

Diffusion and Osmosis can occur at the same time.



Filtration is the passage of fluid through the membrane.

Fig. 1-2
Transport processes.

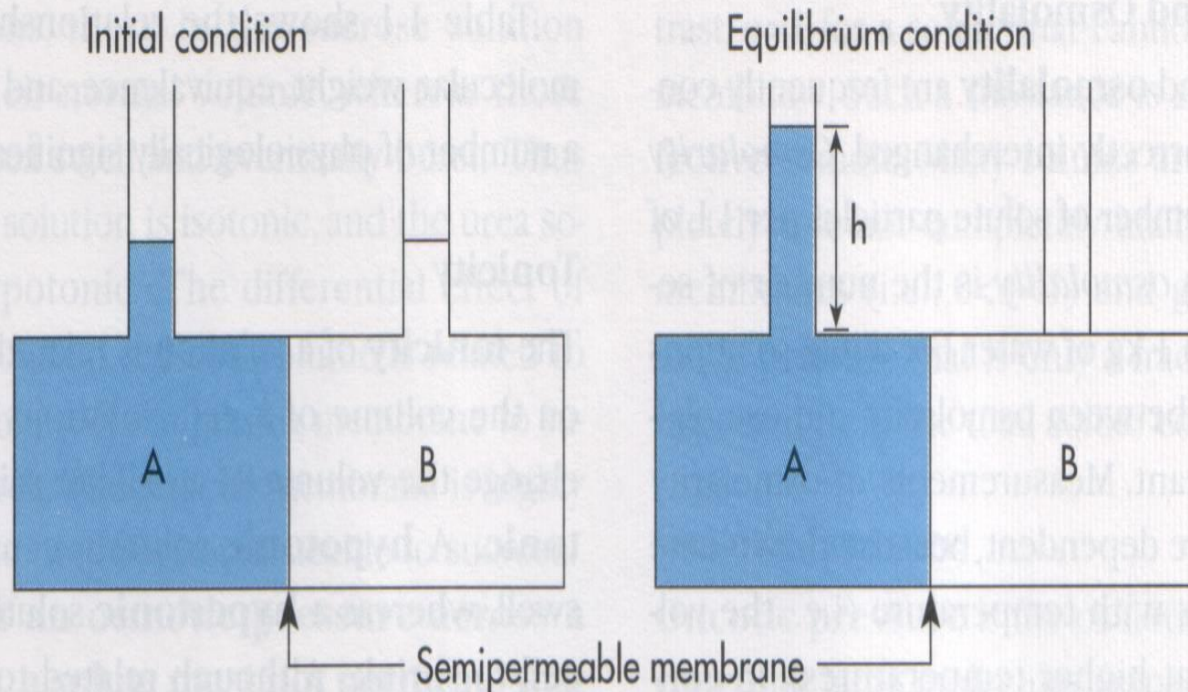
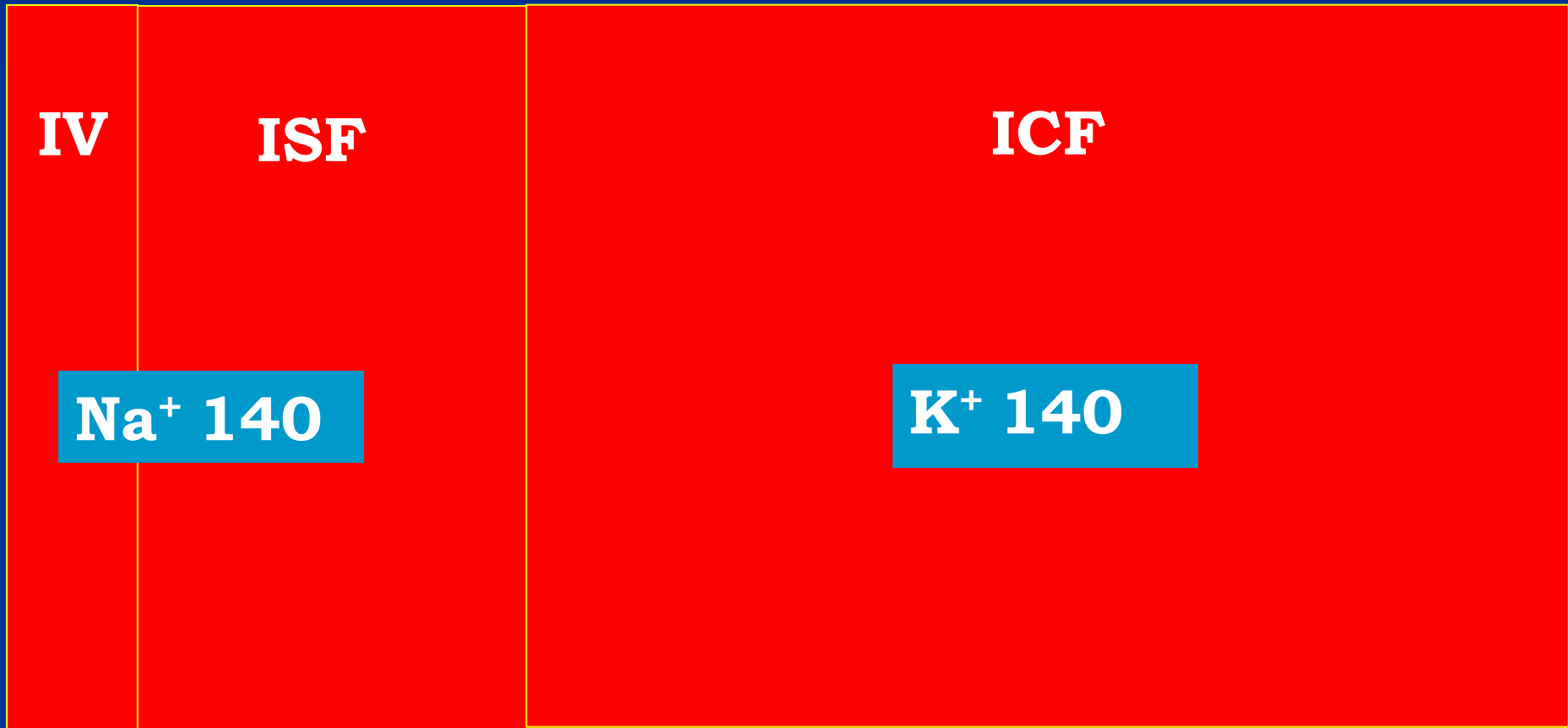


Figure 1-1 ■ Schematic representation of osmotic water movement and the generation of an osmotic pressure. The solute particles in compartment *A* cause water to move by osmosis from compartment *B* across the semipermeable membrane into compartment *A*. The water column in compartment *A* will rise until the hydrostatic pressure generated by the water column (*h*) stops the flow of water from compartment *B* into compartment *A*. This hydrostatic pressure is equal to the osmotic pressure generated by the solution in compartment *A*.

Body Fluid Compartments

- ECF and ICF are in *osmotic equilibrium*
- $ICF_{osm} = ECF_{osm} = P_{osm}$



Isotonic solution Hypertonic solution Hypotonic solution

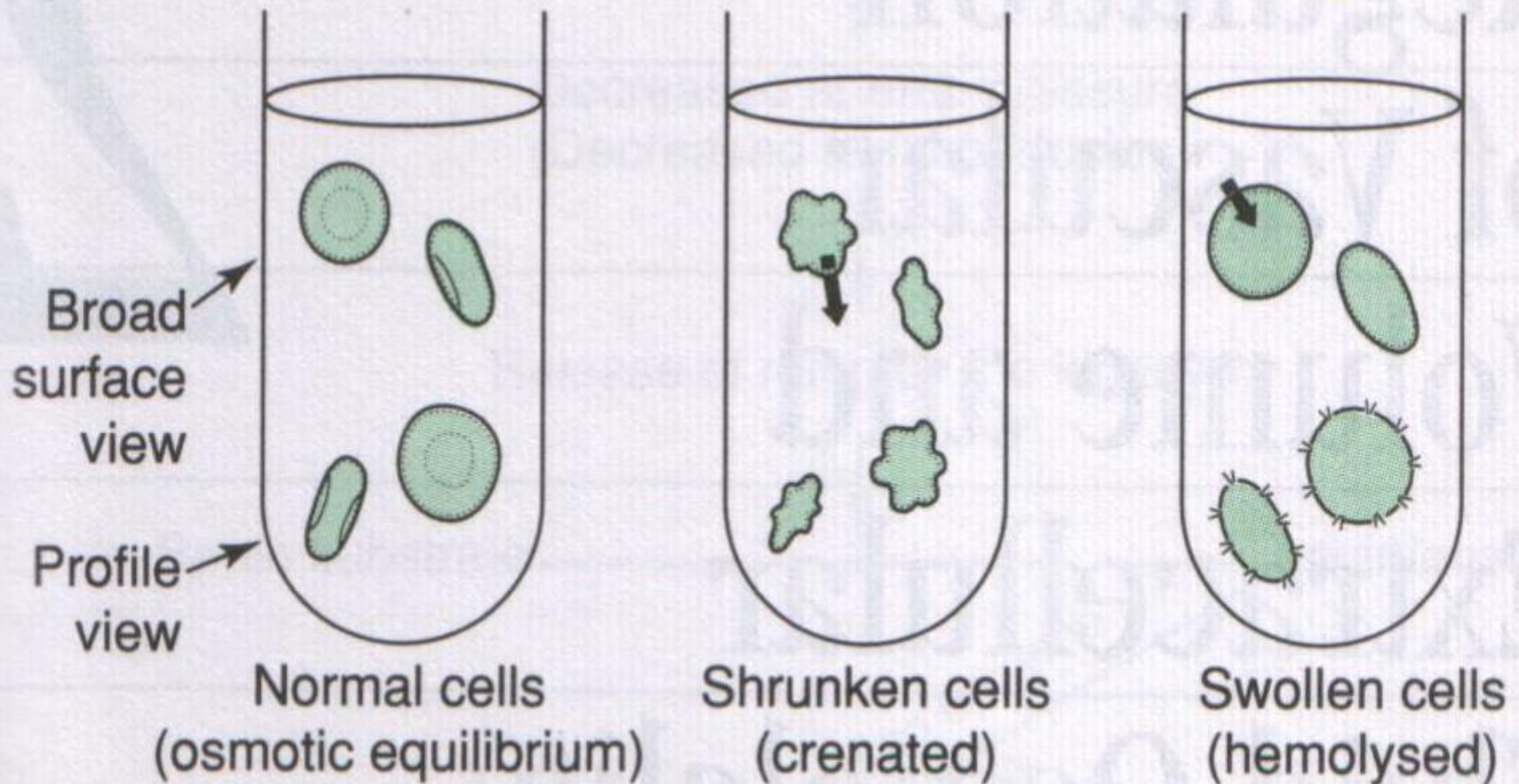
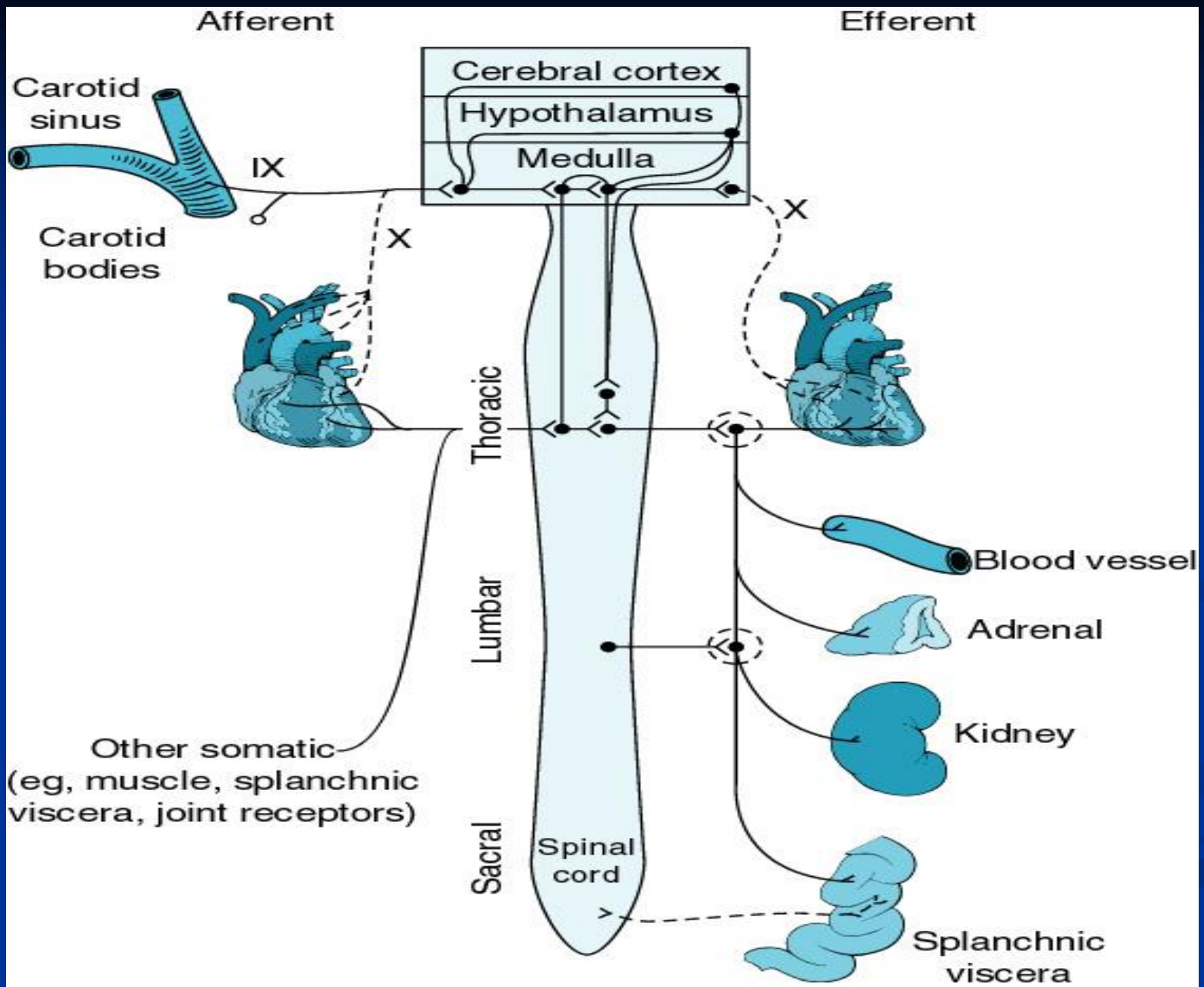


Fig. 1-3

Effect of osmotic pressure on the cells.

Regulation Mechanisms of Fluid and Electrolytes:

- Regulation of **osmolality** and **volume** is achieved through **thirst** and the **osmoreceptor-antidiuretic hormone system (vasopressin)**
 - Volume is more important than osmolality
- The regulation of **volume** also occurs through neurological and renal mechanisms
 - **The stretch receptors (baroreceptors)**
 - **The Renin-Angiotension-Aldosterone System**
 - **The Natriuretic peptides**



Afferent limb sensors of extracellular fluid volume

Cardiopulmonary (venous circulation)

Atria

Ventricular and pulmonary

Arterial

Extrarenal: aortic arch, carotid sinus,

Intrarenal: juxtaglomerular apparatus

Others

Central nervous system

Hepatic

Figure 8.4 The afferent limb (volume sensors) of the integrated homeostatic response system for extracellular volume.

Integrated homeostatic response to volume changes

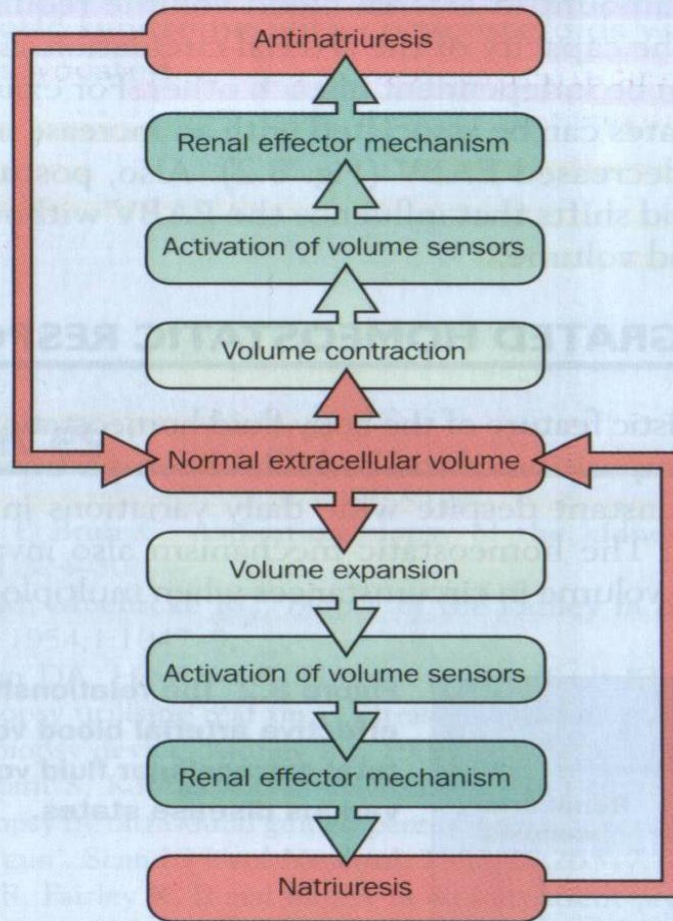
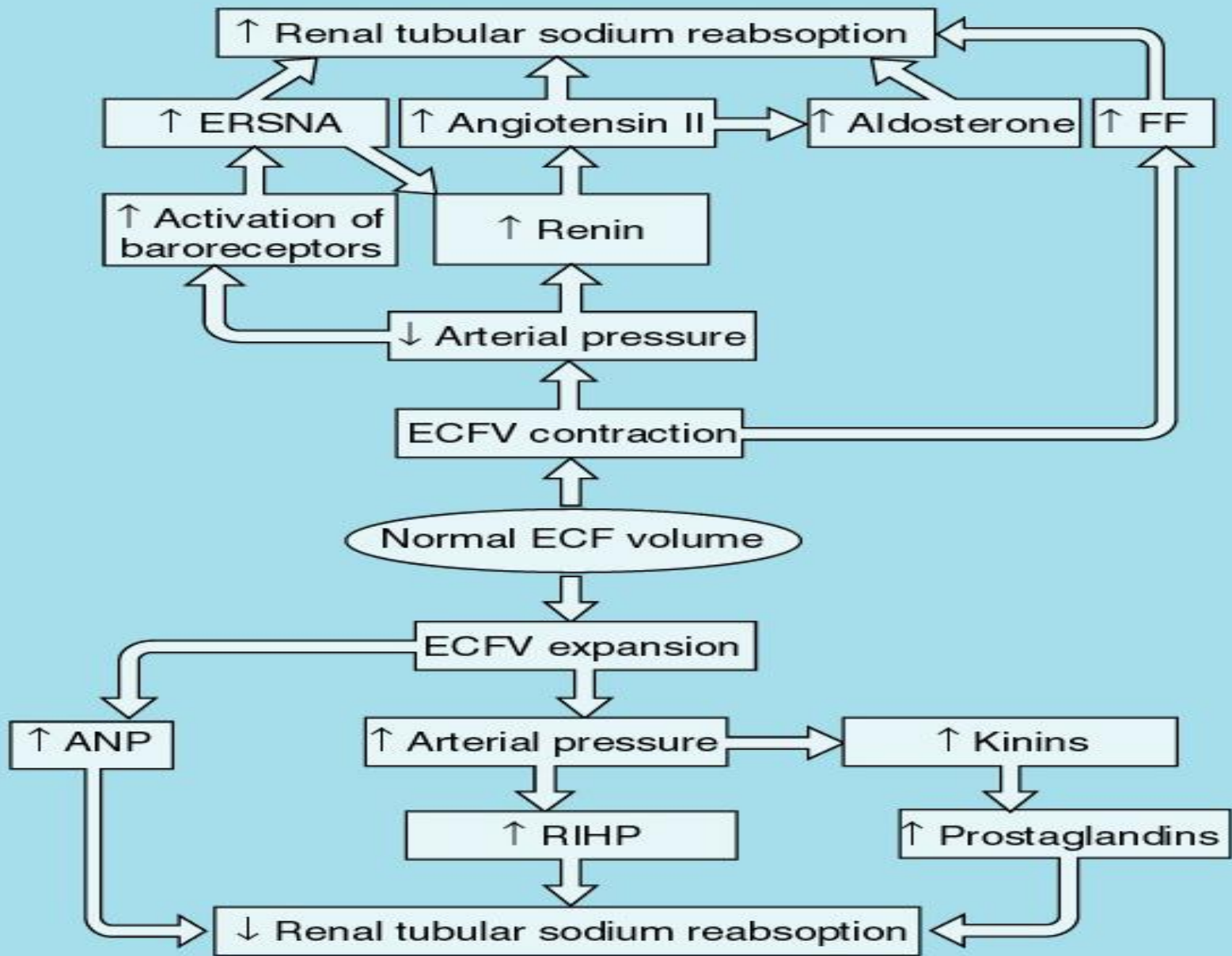
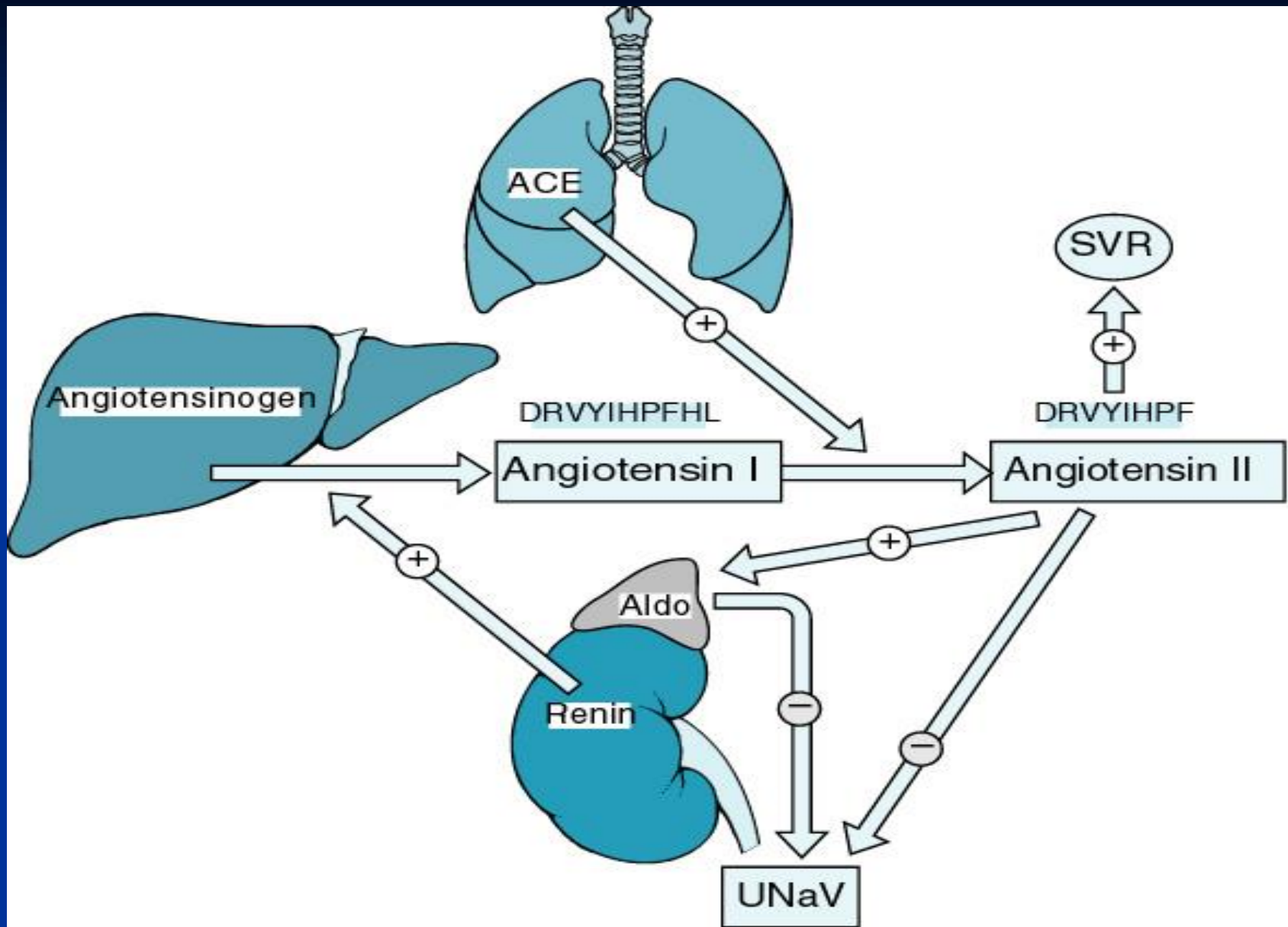
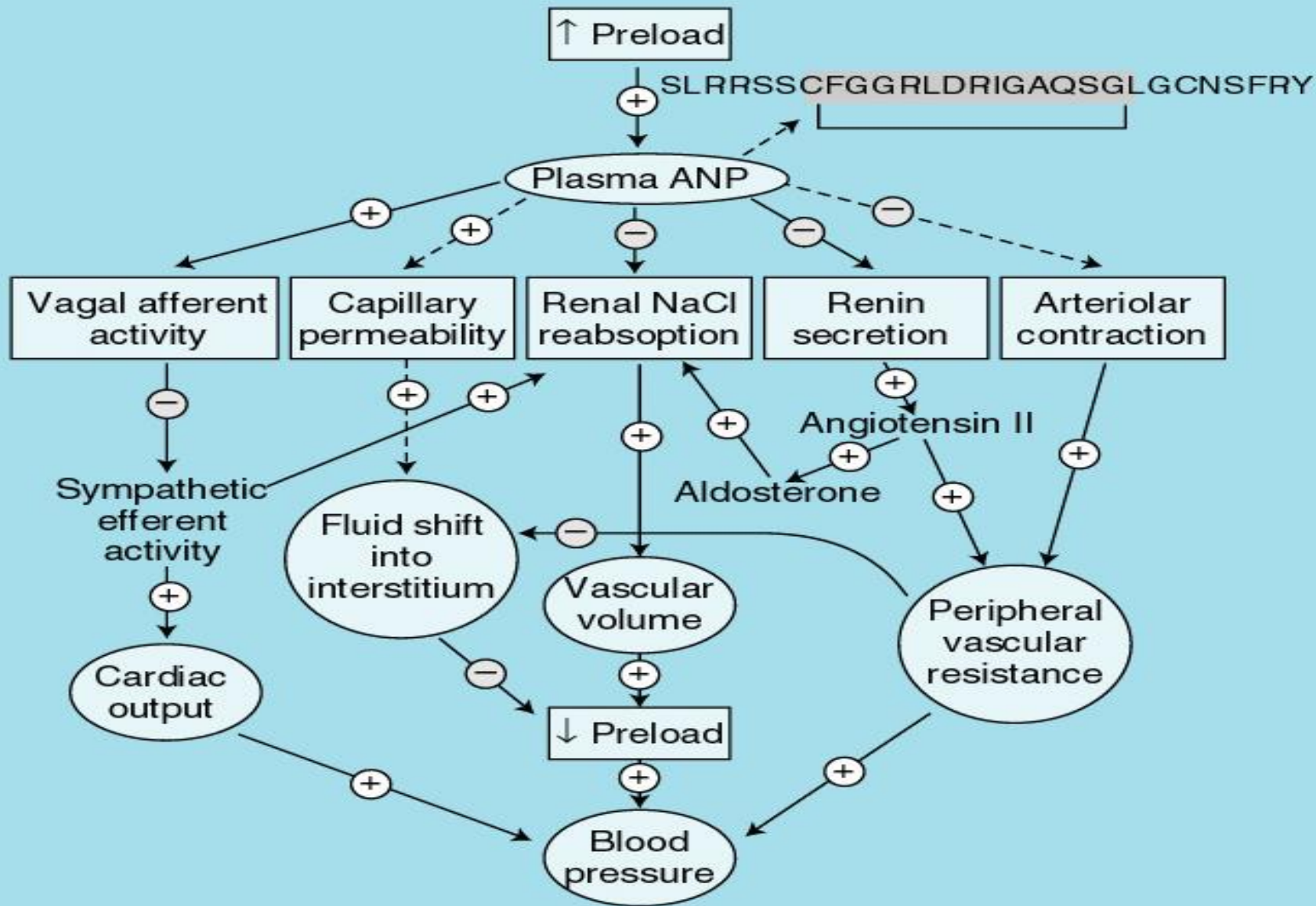
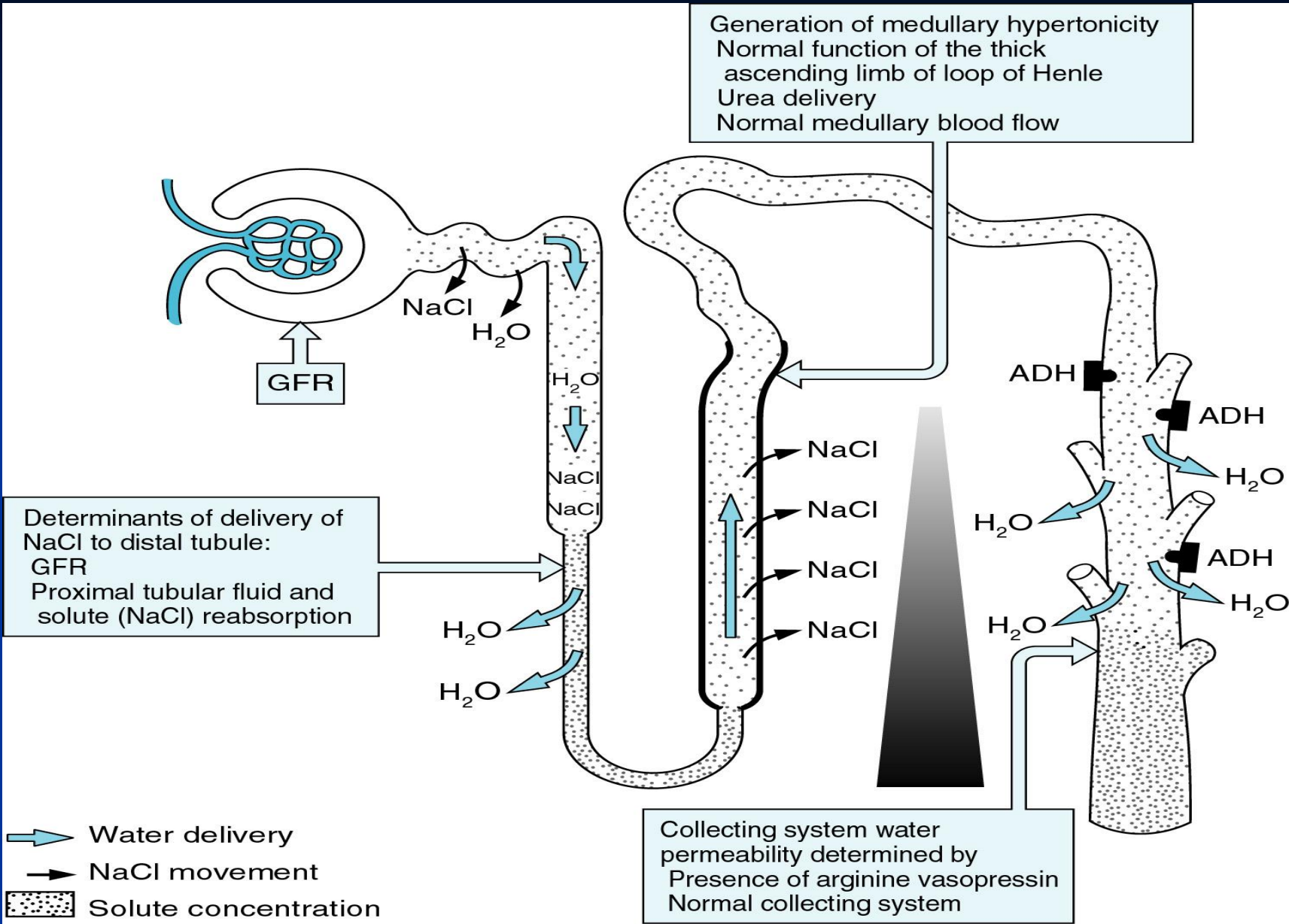


Figure 8.3 A general overview of the integrated homeostatic response system regulating extracellular fluid volume during volume contraction and expansion.

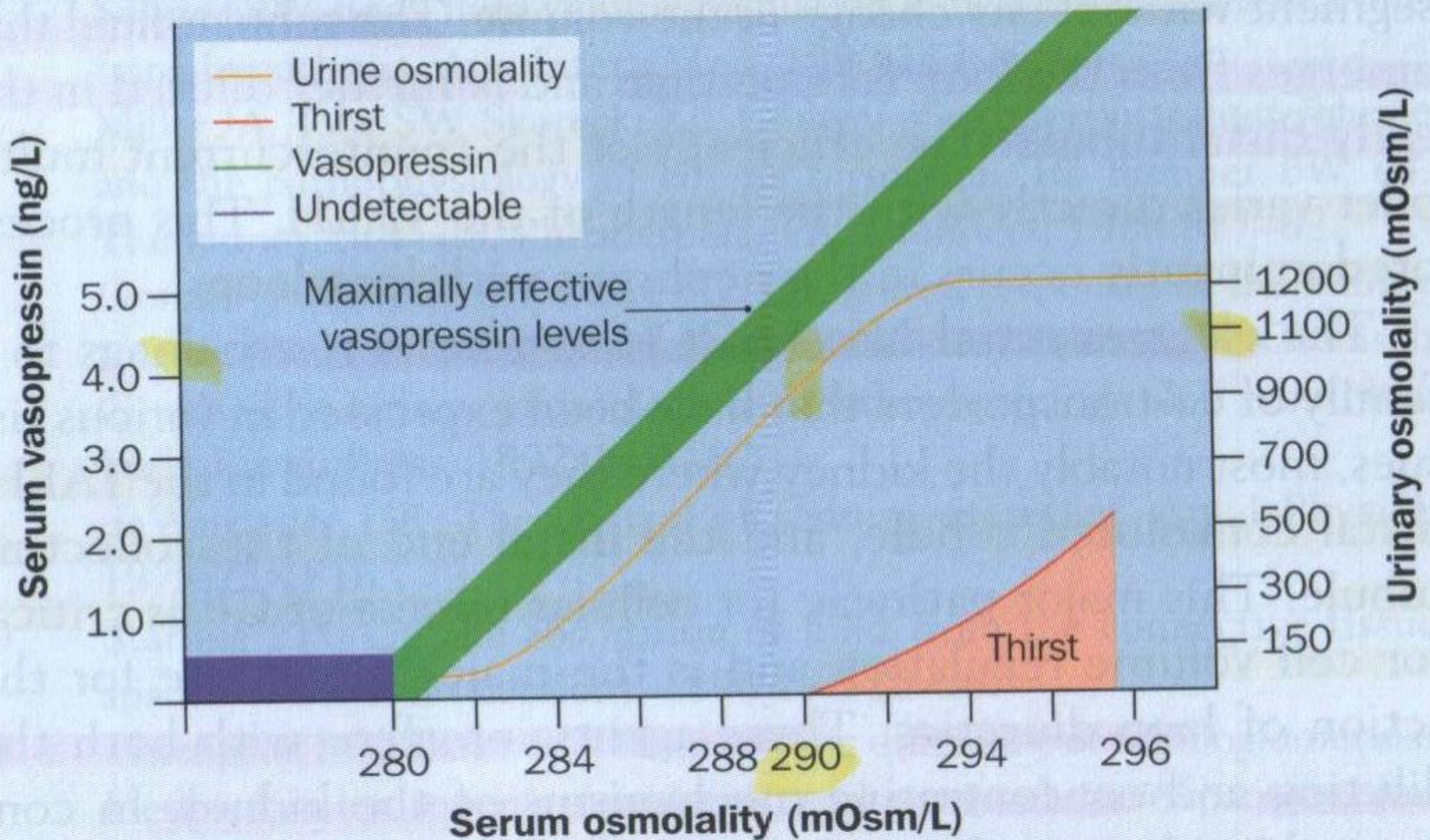




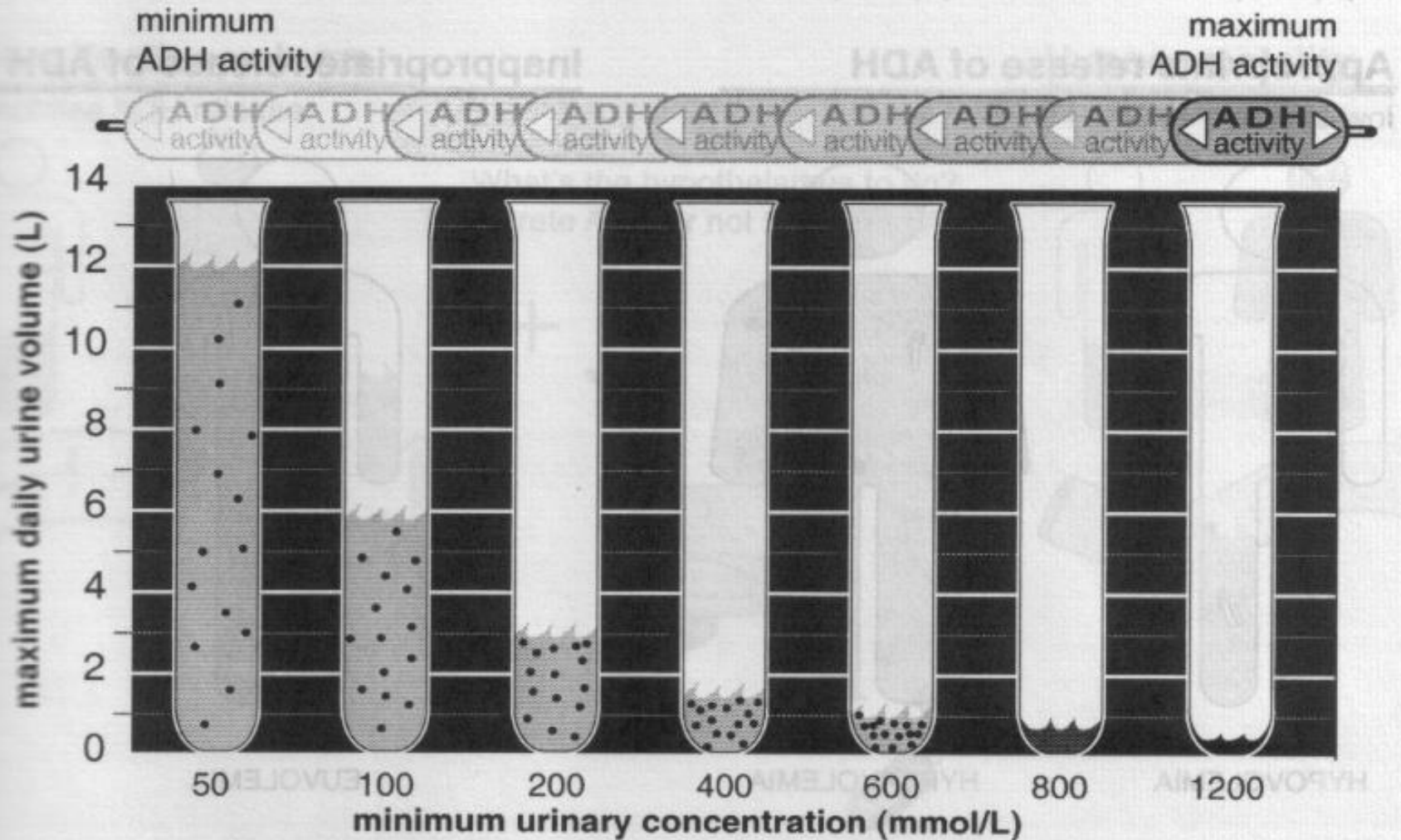




Response to changes in serum osmolality



Urine Output & Daily Solute Load



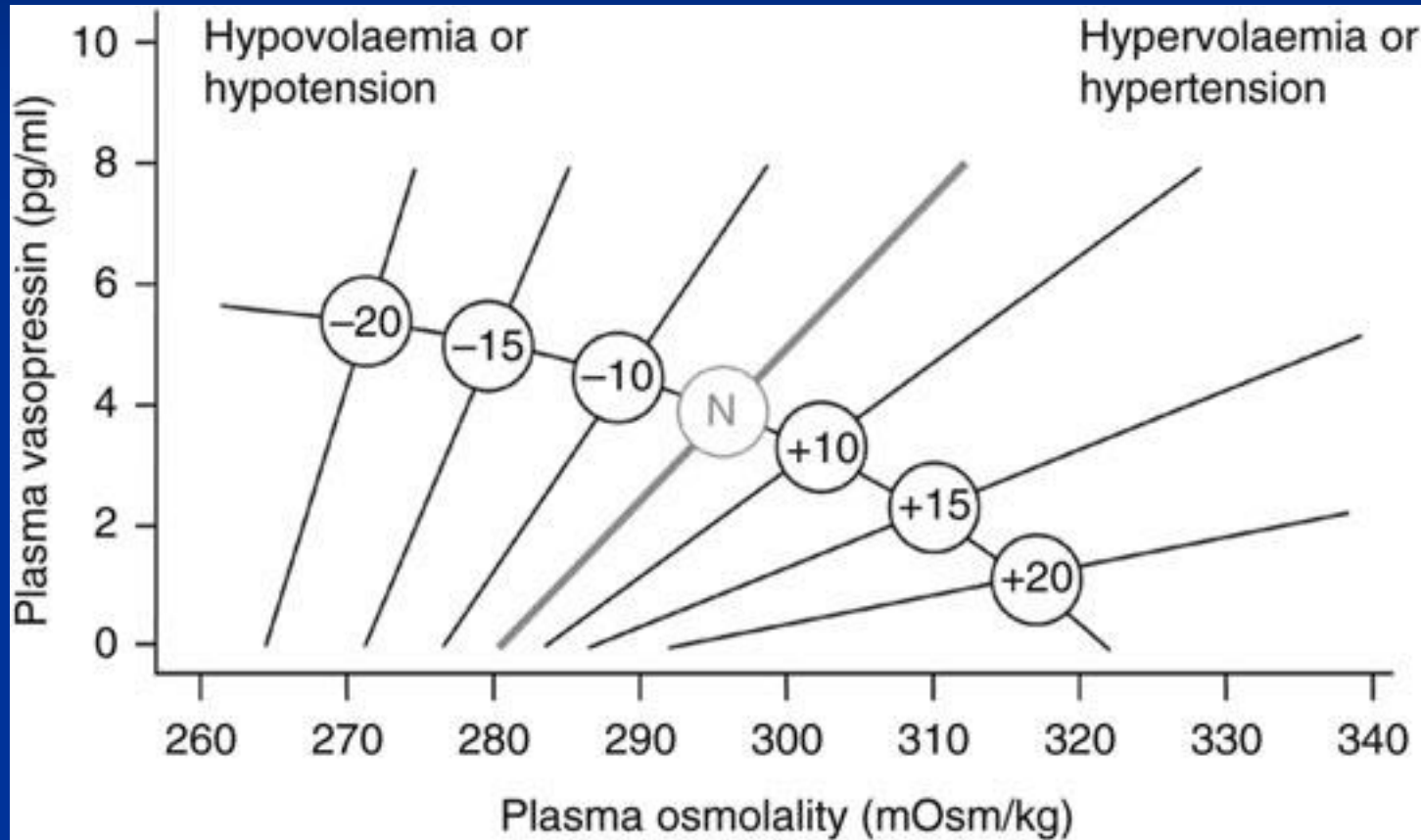
The graph shows the maximum daily urine volume for several values of the minimum urinary concentration (50 to 1200 mmol/L), assuming a daily solute load of 600 mmol/day.

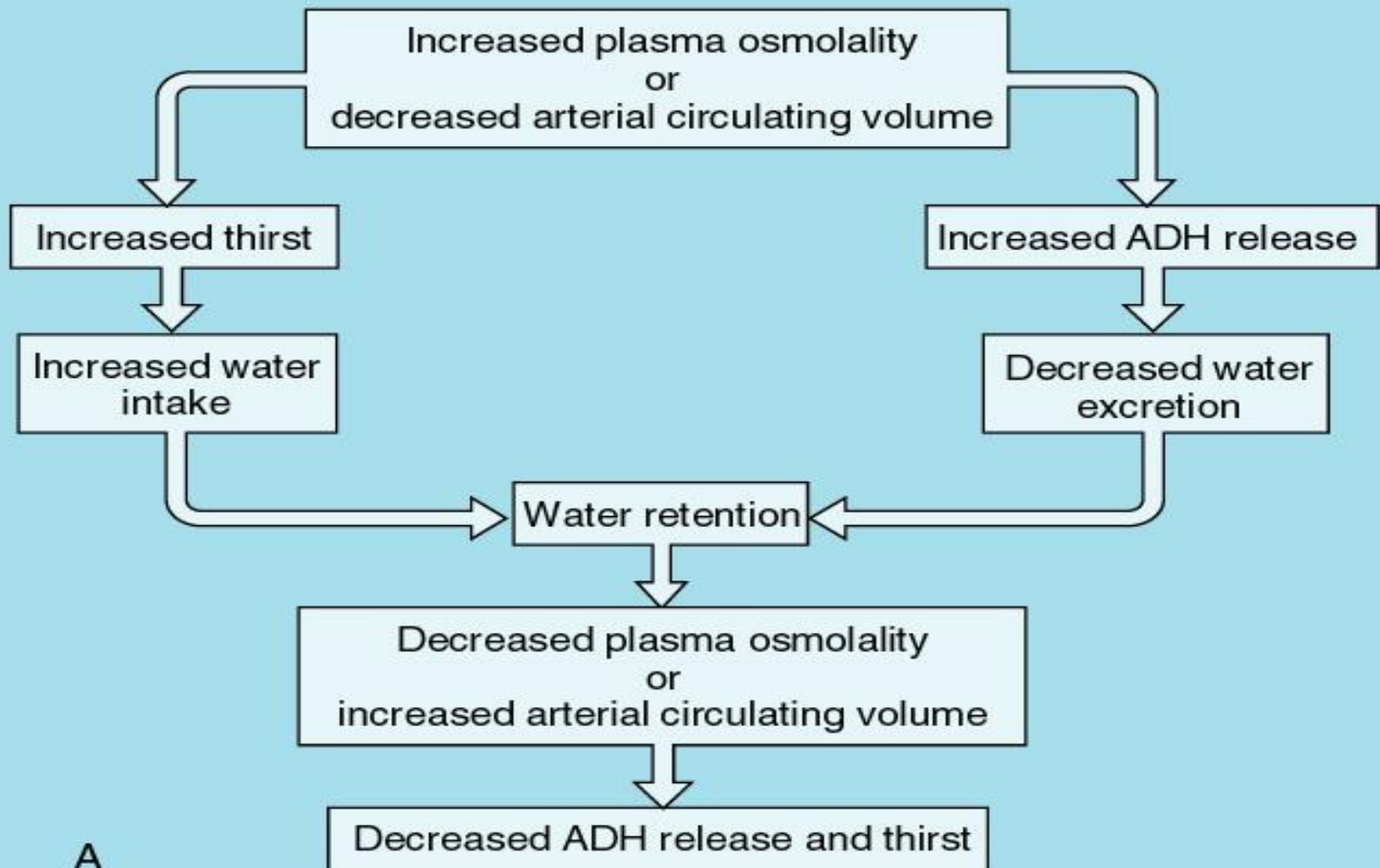
The Linear Relationship Between Urine Specific Gravity and Uosm

SG	Osmolality (mOsm/Kg H₂O)
1.010	300 – 400
1.020	700 – 800
1.030	1000 – 1200

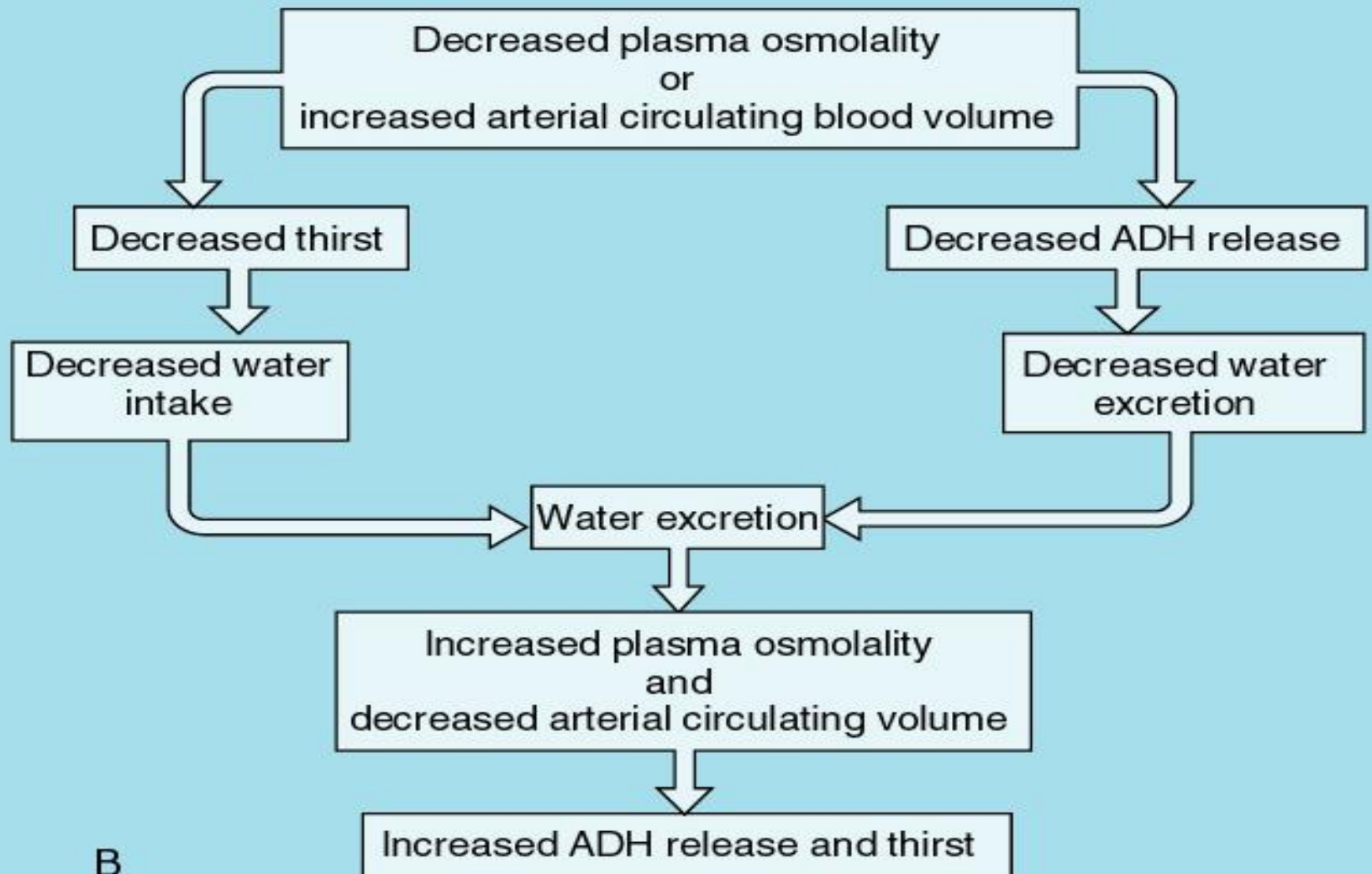
Plasma SG ~ 1.008

Effect of Hypovolemia on Osmoreceptor Gain

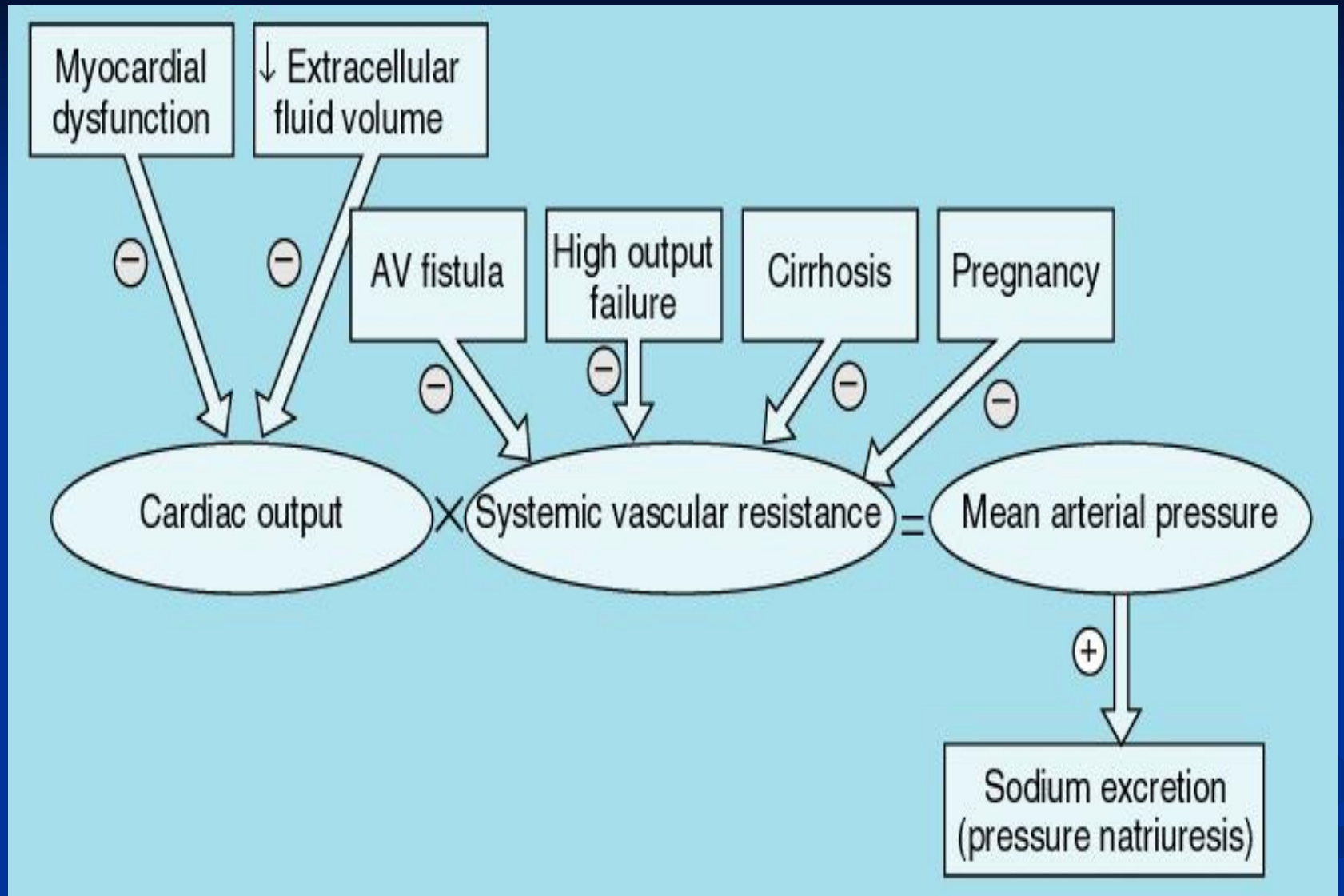




A



B



Effective Arterial Blood Volume (EABV):

- Although the absolute volume of the intravascular space is an important component of circulatory “fullness”, *the adequacy of the circulation* (more commonly called **the effective arterial blood volume or EABV**) also is determined by cardiac output and systemic vascular resistance

Effective Arterial Blood Volume (EABV):

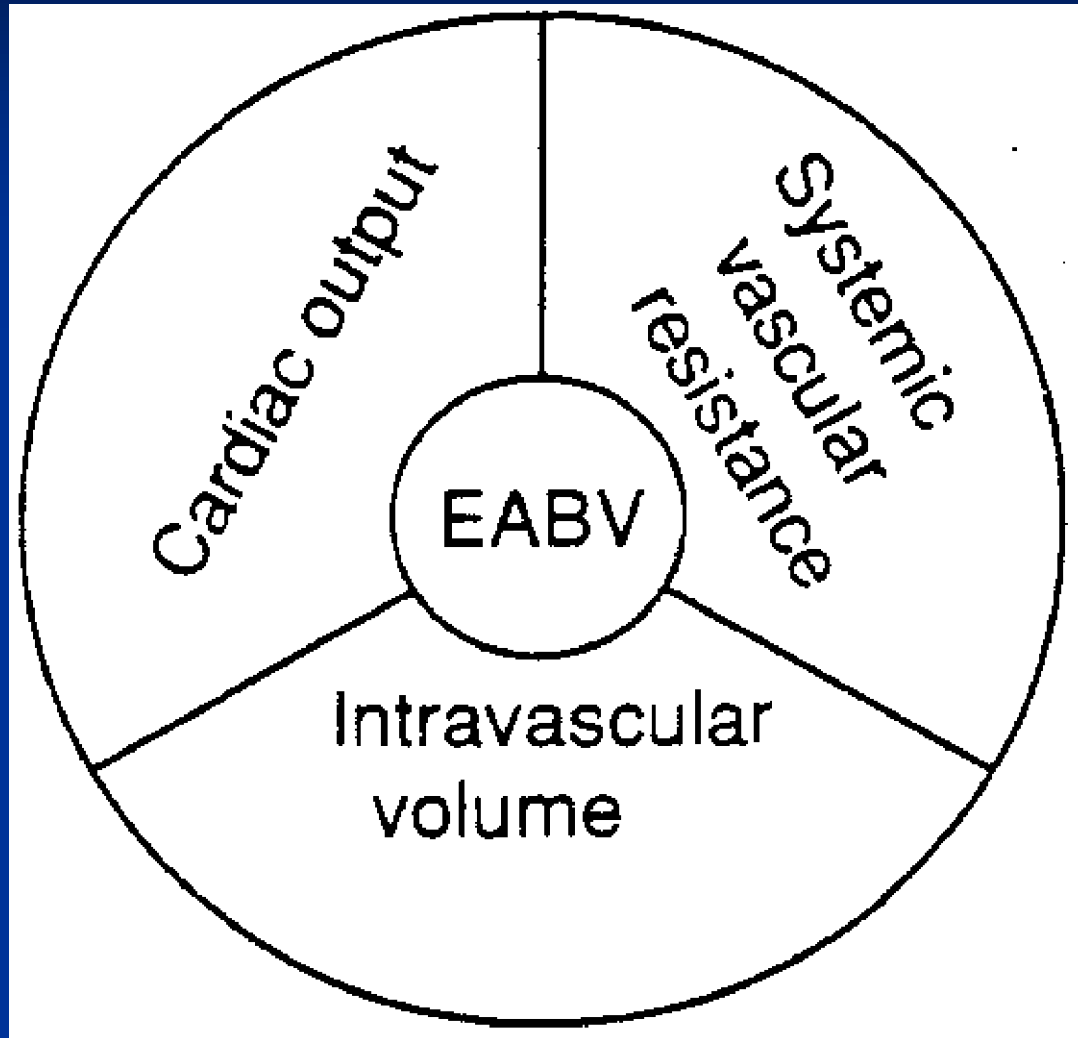
- ↓ EABV:
 - ↑ CO
 - ↑ SVR
 - ↑ Renal Na retention

- ↑ EABV:
 - ↓ CO
 - ↓ SVR
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Effective Arterial Blood Volume (EABV):

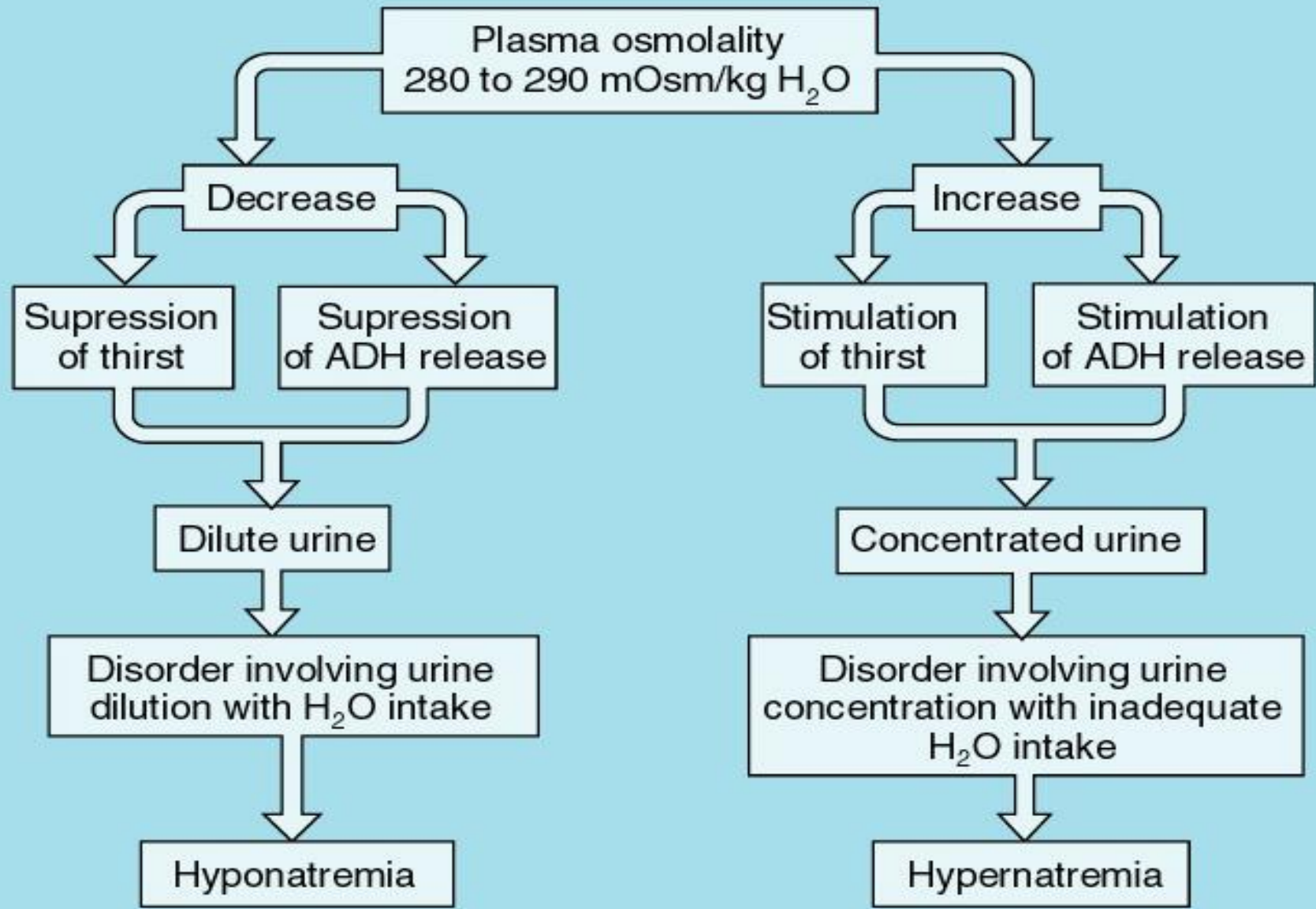
- **EABV is the amount of arterial blood volume required to adequately 'fill' the capacity of the arterial circulation**
- ECF volume and EABV can be **independent** of each other
 - **Edematous states:** increase in total ECF volume and decreased EABV
 - **Postural changes** may cause shifts that influence the EABV without affecting the total blood volume

Effective Arterial Blood Volume (EABV):



Clinical features of Hypovolaemia & Hypervolaemia

	Hypovolemia	Hypervolemia
Symptoms	Thirst	Ankle swelling
	Dizziness on standing	Abdominal swelling
	Weakness	Breathlessness
Signs	Low JVP	Raised JVP
	Postural hypotension	Peripheral oedema
	Tachycardia	Pulmonary crepitations
	Dry mouth	Pleural effusion
	Reduced skin turgor	Ascites
	Reduced urine output	Hypertension (sometimes)
	Weight loss	Weight gain
	Confusion, stupor	



Sodium and Water:

- ECF volume= **absolute** amounts of Sodium and water
- Plasma Na⁺ = **ratio** between the amounts of Sodium and water (Concentration)

- Hyponatremia = Water Excess
- Hypernatremia = Water Deficit

- Hypervolemia = Sodium Excess
 - “Edema”
- Hypovolemia = Sodium Deficit
 - “Dehydration?”

Sodium and Water:

	Hyponatremia (Water Excess)	Hypernatremia (Water Deficit)
Hypovolemia (Dehydration) (Sodium Deficit)	Hemorrhagic Shock with good oral water intake	Diarrhea in Children and Seniors
Hypervolemia (Edema) (Sodium Excess)	Advanced Congestive Heart Failure	Hemodialysis Patient after 3% Saline infusion

Tonicity

- To **compare** the osmolality of a solution to that of another solution (body fluid compartments)
- Used to compare the osmolality of intravenous solutions to that of the serum:
 - *ISOTONIC*
 - *HYPOTONIC*
 - *HYPERTONIC*

<i>Hypotonic</i>	<i>Isotonic</i>	<i>Hypertonic</i>
Solutions have more water than solutes comparing to ECF	Solutions have the same solute concentration as the ECF	Solutions have more solutes than water comparing to ECF
Water will move from ECF into ICF	It will remain in the ECF	Water will move from ICF to ECF
Distilled Water 0.45% NaCl (1/2) 0.33% NaCl (1/3)	NS (0.9% NaCl) Ringers Lactate 2/3 DW-1/3 NS 5% Dextrose in Water (D5W)	3% NaCl 10%-50% Dextrose D5W-1/2 NS D5NS Amino acid solution

Intravenous Solutions

- Crystalloids vs Colloids
- Crystalloids are intravenous solutions that contain solutes that readily cross the capillary membrane
 - Dextrose and electrolyte solutions
- Colloids are intravenous solutions that **DO NOT** readily cross the capillary membrane
 - Blood, albumin, plasma

<i>Solution</i>	<i>Gluc</i>	<i>Na⁺</i>	<i>K⁺</i>	<i>Ca⁺²</i>	<i>Cl⁻</i>	<i>Lact</i>	<i>mOsm/L</i>
D ₅ W	50	0	0	0	0	0	278
D ₁₀ W	100	0	0	0	0	0	556
NS	0	154	0	0	154	0	308
½ NS	0	77	0	0	77	0	154
D ₅ NS	50	154	0	0	154	0	293
D ₅ ½ NS	50	77	0	0	77	0	216
2/3-1/3	33	50	0	0	50	0	285
Ringer's Lactate	0	130	4	3	109	28	274

D5W: 5 g dextrose/100 mL (50 g/L)

Lytes: mEq/L

D10W: 10 g dextrose/100 mL (100 g/L)

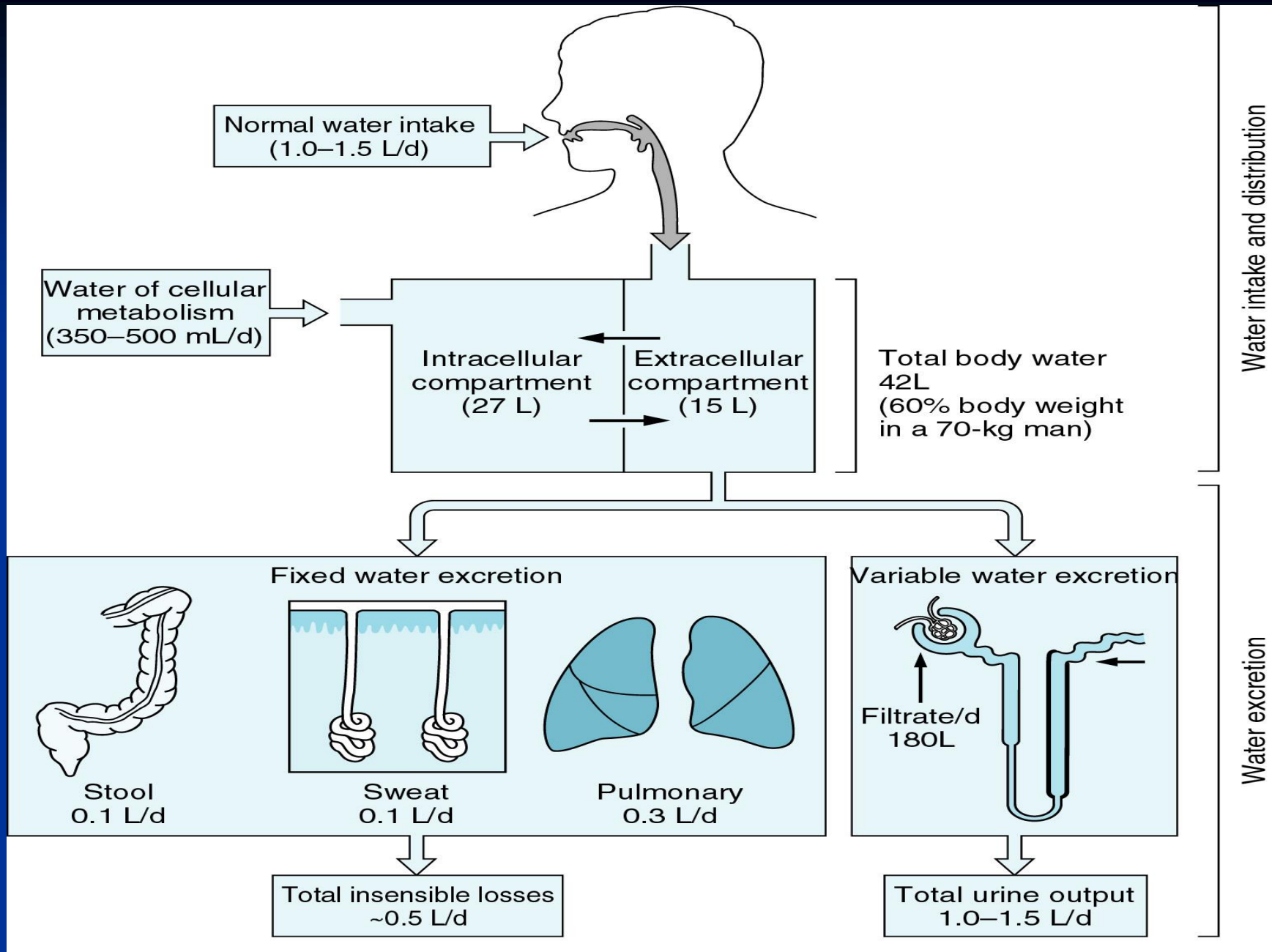
Gluc: g/L

NS (0.9% NS): 0.9 g NaCl/100 mL (9 g/L)

½ NS (0.45% NS): 0.45 g NaCl/100 mL (4.5 g/L)

2/3-1/3: 2/3 D5W (33 g/L) + 1/3 NS (0.33 g NaCl/100mL or 3.3 g NaCl/L)

Parental Fluid	ECF (1/3 TBW)		ICF (2/3 TBW)
	IV (1/4 ECF)	ISF (3/4 ECF)	
1000 ml D ₅ W	80 ml	250 ml	670 ml
1000 ml NS	250 ml	750 ml	---
Colloids (PRBC)	300 ml	---	---
1000 ml ½ NS:			
(500 ml NS)	125 ml	375 ml	---
(500ml water)	40 ml	125 ml	335 ml
Total	165 ml	500 ml	335 ml
1000 ml D ₅ ½NS	165 ml	500 ml	335 ml
1000 ml D ₁₀ W	80 ml	250 ml	670 ml
1000 ml D ₅ NS	250 ml	750 ml	---



Basal Requirements:

■ Basal Water:

- 1st 10 kg: 4 ml/kg/h +
- 2nd 10 kg: 2 ml/kg/h +
- > 20 kg: 1 ml/kg/h

■ Insensible water loss:

- Stool, breath, sweat: 800 ml/d
- Increases by 100-150 ml/d for each degree above 37 C

Basal Requirements:

■ Electrolytes:

- Na: 50-150 mmol/d (NaCl)
- Cl: 50-150 mmol/d (NaCl)
- K: 20-60 mmol/d (KCl)

■ Carbohydrates:

- Dextrose: 100-150 g/d
- IV Dextrose minimizes protein catabolism and prevents ketoacidosis

Hyponatremia

Hyponatremia

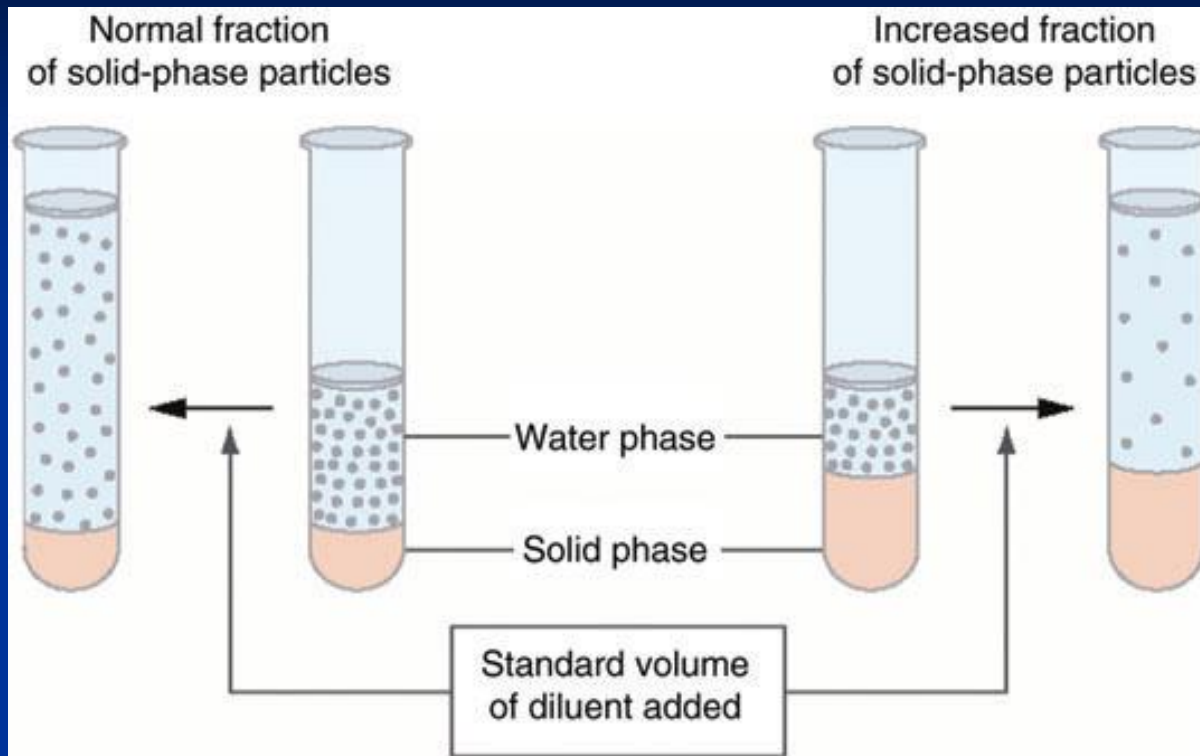
1. Normotonic or Isotonic Hyponatremia

1. *Factitious Hyponatremia*

2. *Pseudohyponatremia*

3. Results from laboratory artifact due to high concentrations of proteins or lipids

Pseudohyponatremia:



- Flame photometric or Indirect potentiometry measurement of PNa^+
- Normal Measured $\text{PNa}^+ = 153 \text{ mmol/L}$ of **Plasma Water**
- Normal Plasma Water Phase = 93% of One liter of Plasma
- Reported Plasma $\text{Na}^+ = 153 \times 0.93 = 142 \text{ mmol/L}$ of Plasma

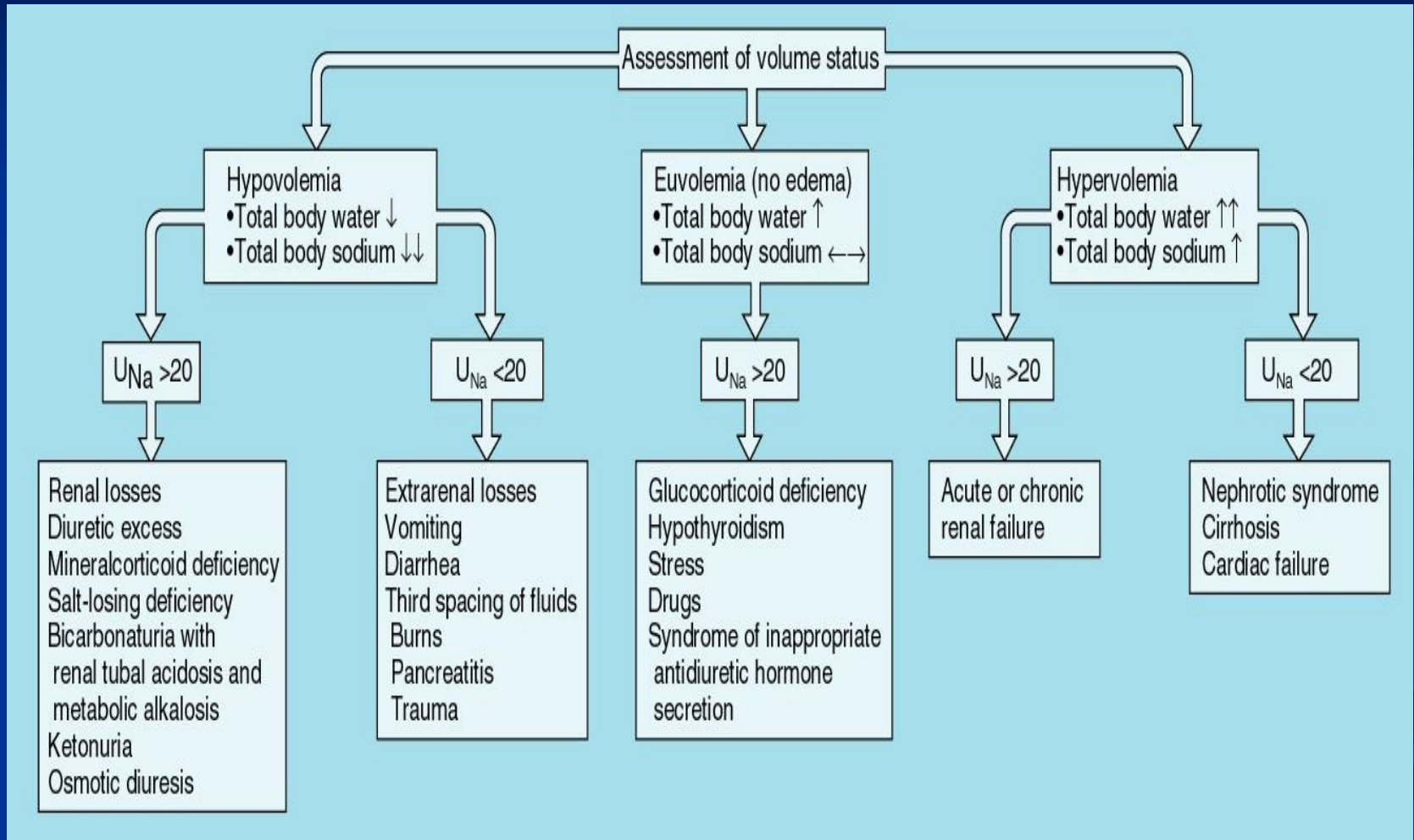
Hyponatremia

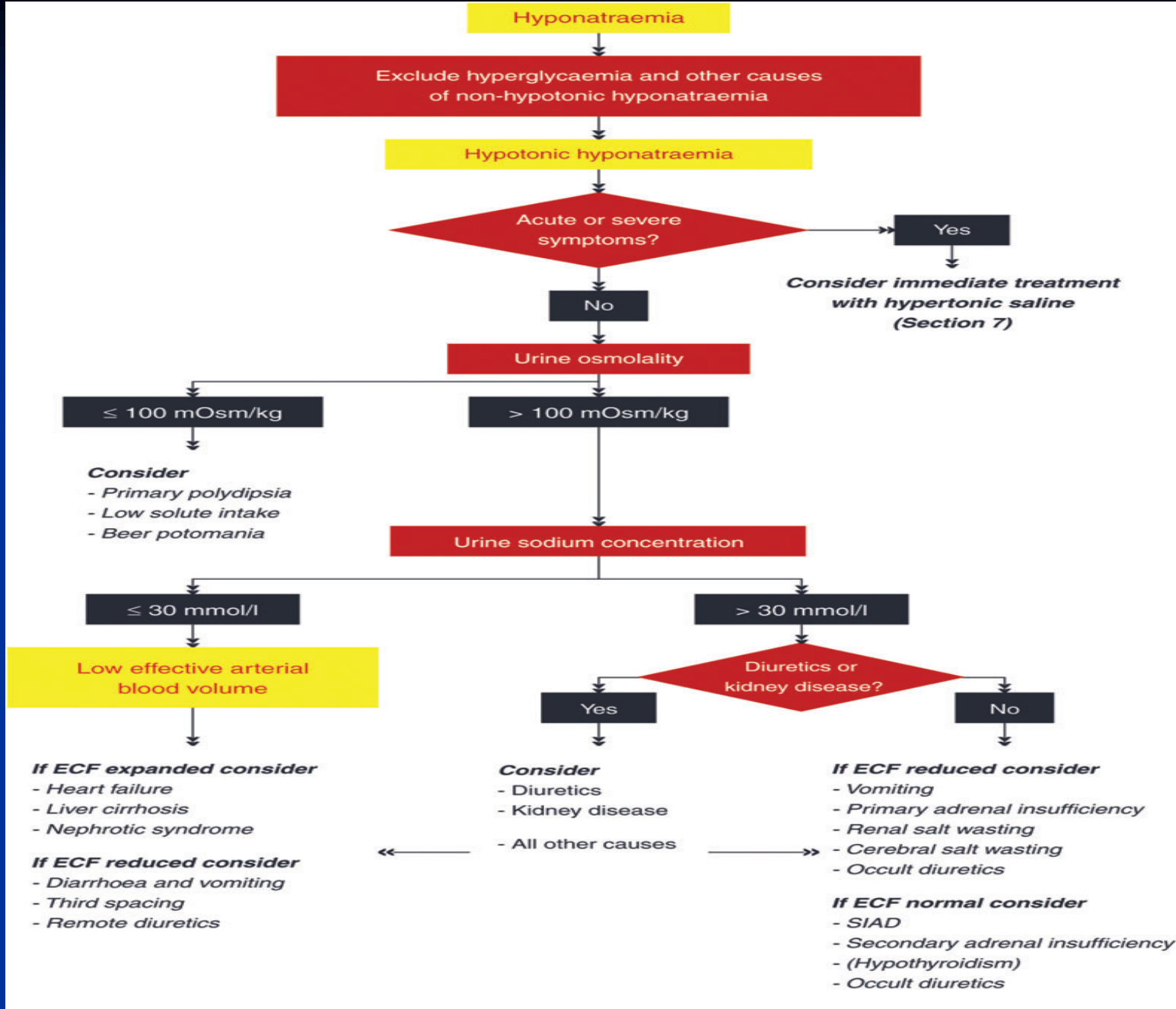
2. Hypertonic Hyponatremia

1. *Translocational Hyponatremia*

2. Results from non-Na osmoles in serum (often glucose or mannitol) drawing Na-free H₂O from cells
3. [Na⁺] declines by ~1.6 mEq/L for each 100 mg/dL [5.6 mmol/L] increase in serum glucose

Hypotonic Hyponatremia: Causes





Classification of symptoms of hyponatremia

- All symptoms that can be signs of cerebral edema should be considered as severe or moderate symptoms that can be caused by hyponatremia
- **Moderately severe**
 - Nausea without vomiting
 - Confusion
 - Headache
- **Severe**
 - Vomiting
 - Cardiorespiratory distress
 - Abnormal and deep somnolence
 - Seizures
 - Coma (Glasgow Coma Scale ≤ 8)

Management of Hyponatremia:

■ Symptoms & Signs

- Volume Status

■ Serum:

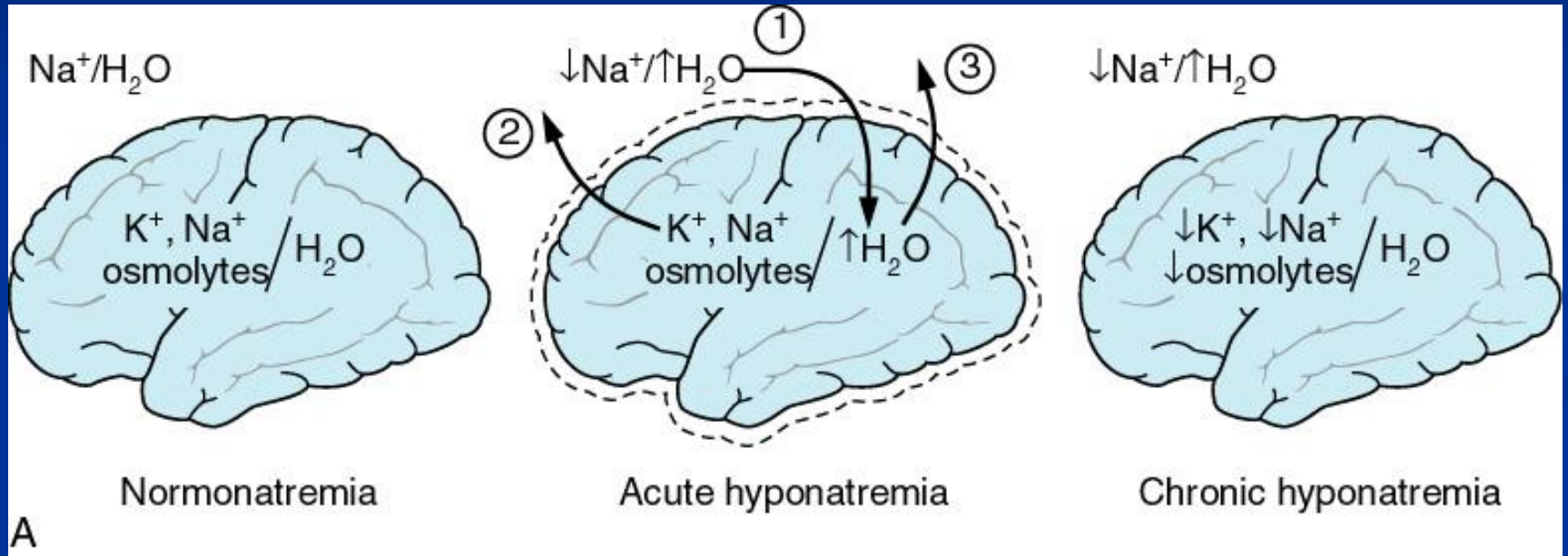
- Osmolality
- TSH, FT4, Cortisol
- Albumin, Total Proteins
- Uric Acid

■ Urine:

- Electrolytes (Na/K/Cl/Urea/Creatinine)
- Osmolality
- Urinalysis

■ Rate of correction:

- 0.5 mmol/L/h ~ 10-12 mmol/L/d



Diagnostic Criteria for SIADH

■ Essential criteria

1. Effective serum osmolality <275 mOsm/kg
2. Urine osmolality >100 mOsm/kg
3. Clinical euvolemia
4. Urine sodium concentration >30 mmol/l with normal dietary salt and water intake
5. Absence of adrenal, thyroid, pituitary or renal insufficiency
6. No recent use of diuretic agents

Diagnostic Criteria for SIADH

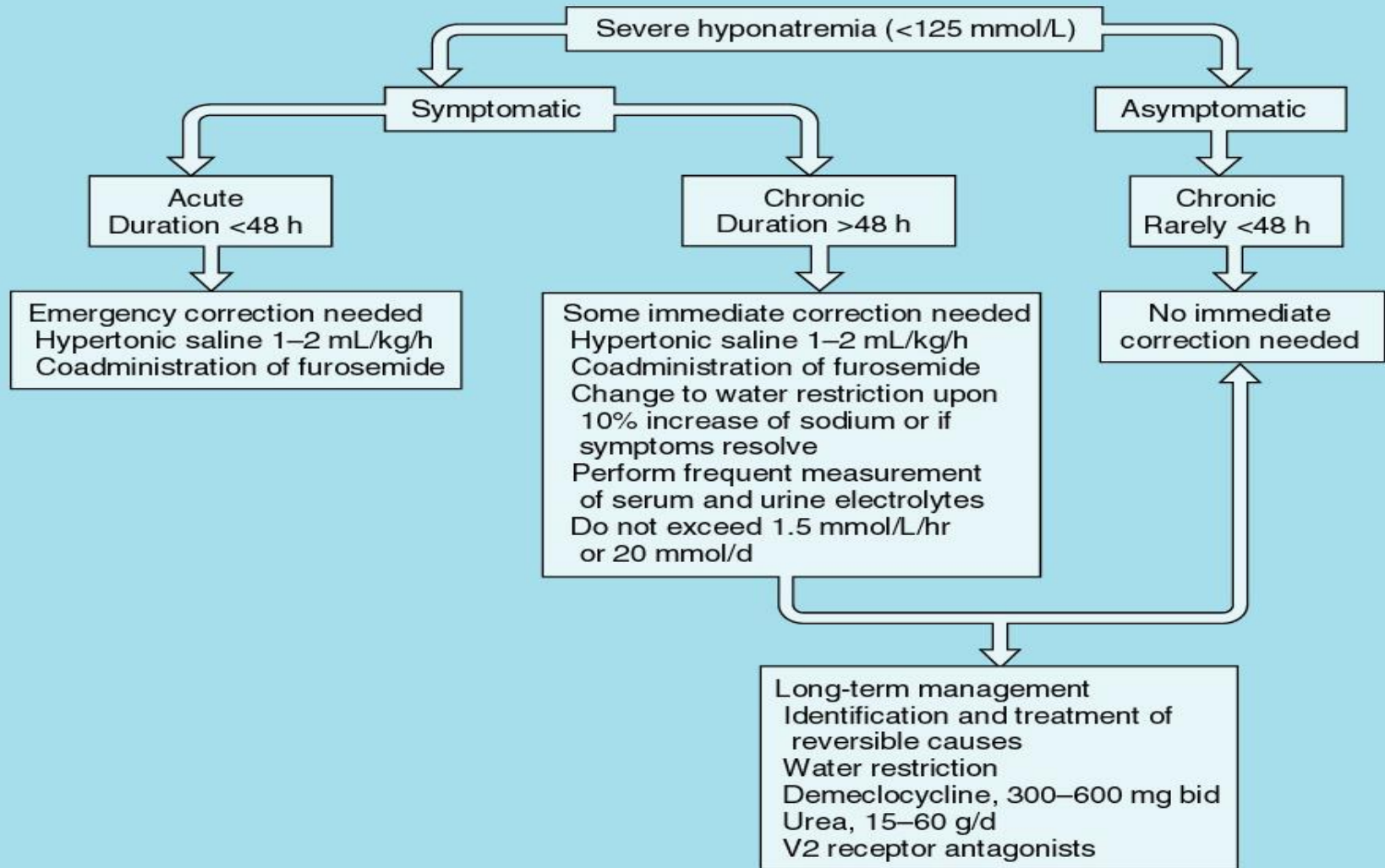
■ Supplemental criteria

1. Serum uric acid <0.24 mmol/l (<4 mg/dl)
2. Serum urea <3.6 mmol/l (<21.6 mg/dl)
3. Failure to correct hyponatremia after 0.9% saline infusion
4. Fractional sodium excretion $>0.5\%$
5. Fractional urea excretion $>55\%$
6. Fractional uric acid excretion $>12\%$
7. Correction of hyponatremia through fluid restriction

SIADH

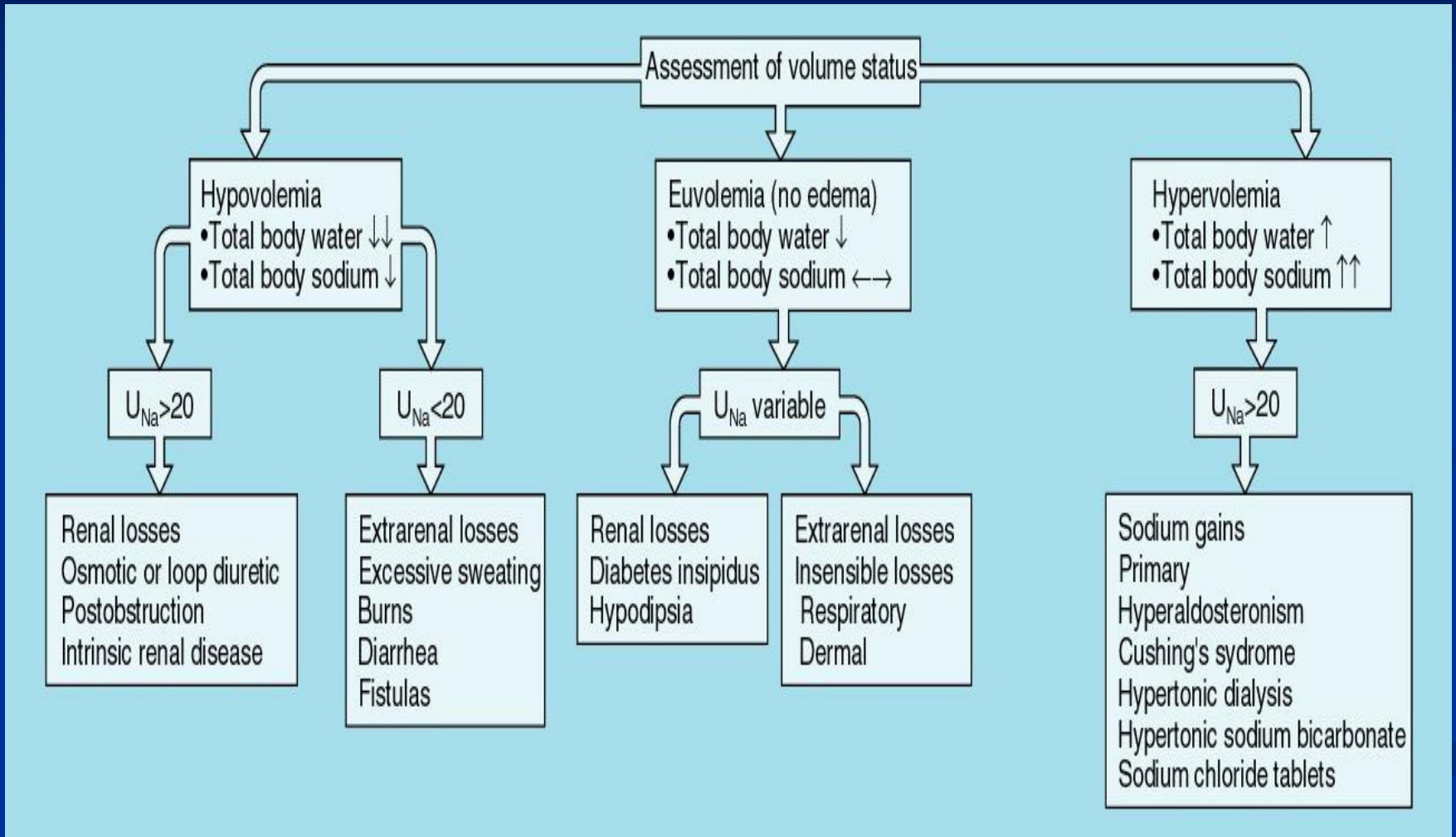
- H: Hypoosmolar Hyponatremia ($\text{Posm} < 275 \text{ mOsm/Kg H}_2\text{O}$)
- I: Inappropriate urine concentration ($\text{Uosm} > 100 \text{ mOsm/Kg H}_2\text{O}$)
- V: Euvolemia, No diuretic use
- E: Endocrine = normal Thyroid, adrenal and renal function
- Hypouricemia ($< 238 \text{ mcmmol/L}$) and low Urea ($< 3.5 \text{ mmol/L}$)

Hyponatremia: Treatment

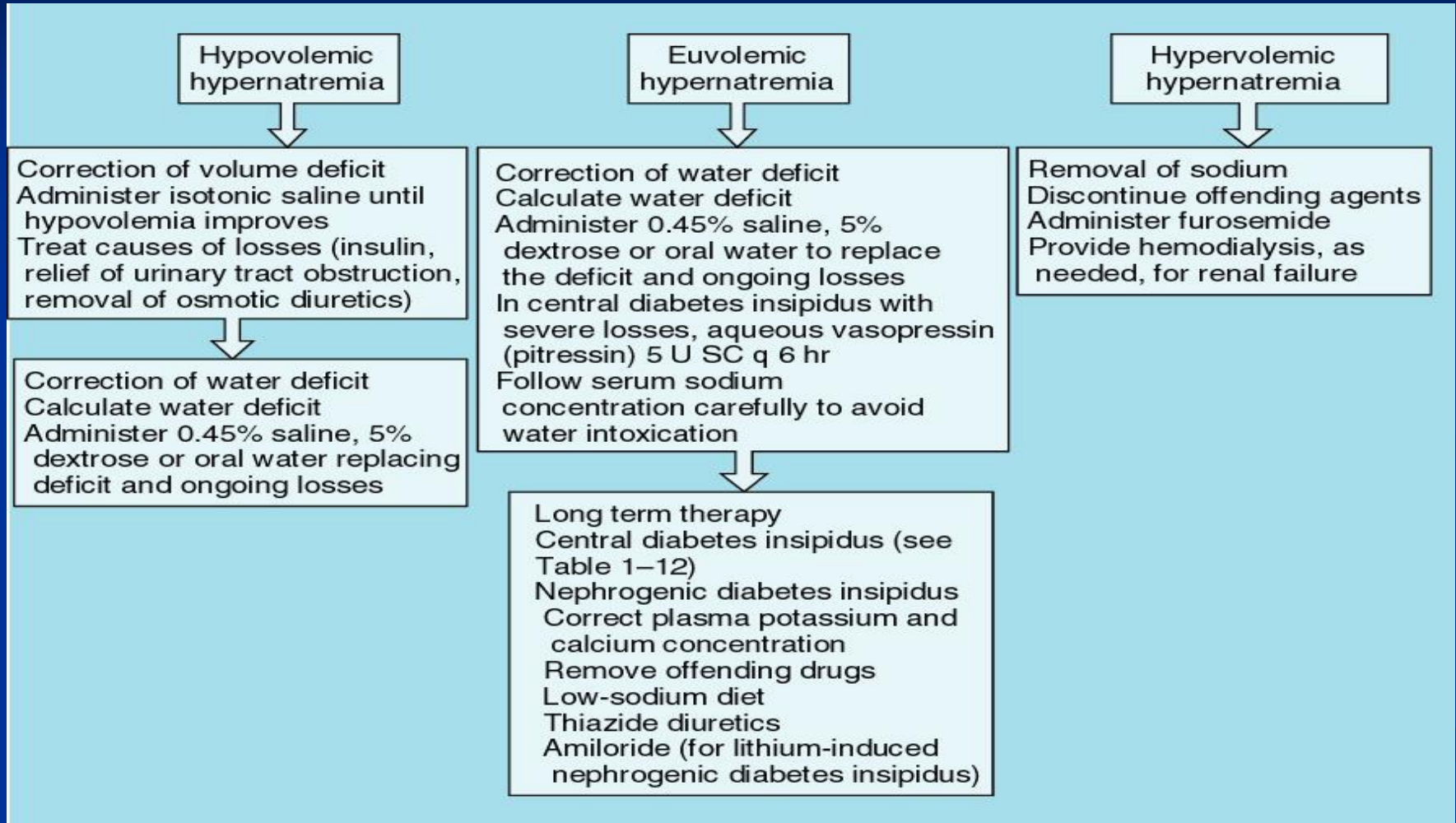


Hypernatremia

Hypernatremia: Causes



Hypernatremia: Treatment



Water Deficit Calculation:

- Current Total Body Water = $0.6 \times \text{Current Body Weight}$
- Current TBW x Current $[\text{Na}^+]$ = Target TBW x Target $[\text{Na}^+]$
- Target TBW – Current TBW = Water Deficit
- Ongoing loss
- IVF: type and rate
- Reassessment

Questions????