FLUID, ELECTOROLYTES ACID BASE BALANCE

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Body water 55 – 75% of Body mass .

average ~ 60%

2/3 Intracellular

1/3 Extra cellular

Extracellular Fluid

Extravascular 75% (Interstitial fluid).

Intervascular 25% (Plasma fluid).

S. Osmolality

no. of solute particles (osmoles) per unite volume of water.

serum osmolality is high in deydration serum osmolality is low in overhydration

S. Osmolality

2 x Na(MEq) + BUN + glucose (265 – 285 mosmol /L).

normal serum Na: 140 mEq normal BUN: 4-6 mmol/L If BUN is high, then patient is dehydrated or has renal failure normal Glucose: 4-6 mmol hypoglycemia: <4 hyperglycemia: >6

S. Osmolality maintained by :

- Kidney function (diluting or concentrating urine)

 ADH. retain water to the body
 Aldosterone. retain Na to the body
- Thirst center.

when the serum osmolality is high, we get thirsty

Principles of management of Dehydration

- Maintenance Daily requirement.
- **Deficit** volume & Electrolyte loss.
- Replenish abnormal ongoing losses.

Maintenance :

Sensible + Insensible fluid loss

urine and feces



Insensible water loss (30ml / kg / day):

Skin.Respiration.

Sensible loss:

OUrinary.

if a patient has for example oligouria, anuria or renal failure we don't replace his sensible loss, cuz he is unable to urinate! we only replace him for the insensible fluid loss because if you replace his sensible fluid loss, he will become over hydrated, -> pulmonary edema -> death

Physiological requirement of maintenance increase in:

Fever (10% for each 1C[°]).

if the maintaninance for a baby is 100/kg/day, and the baby has fever 38 C. we should increase his maintainance by 10%, becomes 110/kg/day

• Physical activity.

Physiological requirement decrease in :

- Anuria ARF.
- Oliguria ARF.
- Congestive heart failure.
- SIADH (\uparrow ADH \rightarrow H₂O retention).
 - Meningitis. kids who have meningitis we give them only 2/3 of maintainance
 Head trauma.

Maintenance requirement of water :

- 1st 10kg of weight 100ml / kg / day (4ml / kg / hour).
- 2nd 10 20kg of weight
- > 20kg of weight

A child who is age is 9 months, his weight: 10 kg, going to MRI under general anesthesia tomorrow, and has to be NPO. how much fluid will you give him? 100 x 10 = 1000 ml 1000 ml / day 41.67 ml / h Order: Please give the patient 41.67 ml / h over the coming 24 hours and keep the patient NPO

20ml / kg / day (1ml / kg / hour).

50ml / kg / day (2ml / kg / hour).

A child who is age is 9 months, his weight: 15 kg, going to MRI under general anesthesia tomorrow, and has to be NPO. how much fluid will you give him? (4×10) + (2×5) = 50 ml 50 ml / h 1250 ml / h 1250 ml / day Order: Please give the patient 50 ml / h over the coming 24 hours and keep the patient NPO A child who is age is 9 months, his weight: 25 kg, going to MRI under general anesthesia tomorrow, and has to be NPO. how much fluid will you give him? $(4 \times 10) + (2 \times 10) + (1 \times 5) = 65 \text{ ml}$ 65 ml / h

Order: Please give the patient 65 ml / h over the coming 24 hours and keep the patient NPO Maintenance requirement of electrolyte:

- Na : 3 MEq / kg / day.
- K : 2 MEq / kg / day.
- Cl : 5 MEq / kg / day.

Deficit :

• ↑ Loss.

- Intake.
- Severity in infants :
 - $\circ~$ Mild 5% of body wt.
 - Moderate 10% of body wt.
 - Severe > 15% of body wt.

Assessment of deficit :

- 1. Weight change. "the ideal way"
- 2. Physical signs :
 - Pulse.
 - **BP**.
 - Behavior.
 - Skin turgor.
 - Ant. Fontan.
 - Tears. while crying
 - Capillary refill.

compare his weight know to his normal weight if he's 9 kg in ER, and 2 weeks ago he was 10 kg, how much deficit? he lost 1 kg form 10 kg. so it is 10% (moderatley dehydrated) if he's 9.5 kg in ER, and 2 weeks ago he was 10 kg, how much deficit? 5% (mild) if he's 8.5 kg in ER, and 2 weeks ago he was 10 kg, how much deficit? 15% (severe)

Deficit calculation :

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1kg = 1L ( 1000gm = 1000ml ).
5% deficit = 50ml/kg.
10% deficit = 100ml/kg.
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15% deficit = 150ml/kg.
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A patient weight is 10 kg, he lost 10% of his weight (moderate dehydration). How will you calculate his fluid deficit? 10% x 10 = 100 ml / kg 100 x 10 kg = 1000 ml

if the deficit was 7% 7% x 10 = 70 ml / kg 70 x 10 kg = 700 ml If we are giving a patient maintainance only, we give him equally over 24 hours

but if we are giving a patient maintainance and deficit (the patient is dehydrated) we don't give him equally. give half of the fluid in the first 8 hours and the other half over 16 hors in case of normal or low Na

Example of Deficit : Wt of infant 10kg Deficit 10% $10 \text{kg} \times 100 \text{ml} = 1000 \text{ml}.$

Example of total water requirement :

Wt of infant 10kg Deficit 10%

- Maintenance 100ml x 10kg = 1000ml
- . Deficit 100ml x 10kg =1000ml

• Total = 2000ml

in this example assuming he has normal or low NA, we give the first 1000 ml over 8 hours -> 1000/8 = 125 ml / h over the coming 8 hours the remaining 1000 ml are given over the 16 hours -> 1000/16 = 62 ml / h over the comin 16 hours

Types of Dehydration (Depend on Na level):

- Isotonic (Normal Na).
- Hypotonic (Low Na).
- Hypertonic (High Na).

Replacement of fluids :



Deficit + maintenance $\frac{1}{2} + \frac{1}{2}$ 8 + 16hrs

Hypertonic

- 1st 24hrs maintenance + ½ deficit
- 2nd 24hrs

maintenance + ½ deficit

Case 1:

Wt 10kg Deficit 10% Na 135 or/130 MEq/L

- A. Maintenance 10kg x 100ml = 1000ml
- B. Deficit 1kg = 1000ml

(↓10% →100ml)

10kg x 100ml = 1000ml

C. Total

Maintenance + Deficit

1000ml + 1000ml

= 2000ml

1st 8hrs 1000ml / 8hrs = 125ml / hr.

2nd 16hrs 1000ml / 16hrs = 62.5ml per hour.



Wt 10kg Deficit 10% Na 155 MEq/L

1st day Maintenance + ½ deficit 1000ml + 500ml = 1500ml/24hrs = 62.5 ml/hr.

2nd day 1000ml + 500ml =1500ml /24hrs = 62.5 ml/hr.

other than the maintainance and deficit, if patient has NGT for example, we should calculate how much aspirated fluid through the NGT and add it to the fluid requirement

Electrolyte Sodium :

normal Na level is 135-14 average is 140

- 1) Na < 135 MEq/L (Hyponatremia)
- A. Urine Na < 20 MEq/L
 - Dehydration most common cause
- g.I loss.Skin loss (C.F).

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• Dilution

- SIADH.CHF.
- B. Urine NA > 20 MEq/L major problem in the tubules
 - Diuretics.furosemide
 - Salt wasting nephropathy.
 - Deficiency / Resistant to aldosterone

Management : Nacl (if loss)

2) Na > 145 MEq/L (Hypernatremia)

• Loss of $H_2O > Na$ (e.g viral g.E).

moslt due to gastroenteritis due to Rota virus

Management : slow fluid replacement .

\mathbf{K}^+

• Acidosis : shift k outside cells .

acidosis -> hyperkalemia

• Alkalosis : shift k inside cells.

alkalosis -> hypokalemia

Hypokalemia (K< 3.5 mmol/L):

GI loss

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vomiting \rightarrow loss of HCI.

as in pyloric stenosis: hypokalemia and metabolic alkalosis

- Renal loss
 - Diuretics. loss of Na and K
 - 1 aldosterone.
 - Bartter syndrome (tubulopathy). genetic determined they have high K in urine
 - Systemic Alkalosis.

Management : KCL replacement .

Hyperkalemia (K> 5.5 mmol/L):

- dangerous as it might cause arrhthmia
- Pseudohyperkalemia. most common, due to use of tourniquet while drawing blood sample
- Metabolic Acidosis.
- Tissue catabolism.
- · Renal failure.
- Hypoaldosteronism.
- Diuretics(spironolactone).
- ACEi . like captopril, decreases GFR, يعطي فشل كلوي كاذب, decrease K excretion and hyperkalemia, when stopped, K level will improve

Management : Insulin , glucose, ca gluconate , alkali

to avoid hypoglycemia due to insulin

sodium bicarbonate, to enhance K shifting to inside the cells

Acid – Base Balance: PH ~ 7.35. ^{Examles:} > 7.4 -> acidosis < 7.34 -> alkalosis PH maintained by :

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- \circ Extracellular buffer (HCO₃ from kidney).
- Intracellular buffer (Protein, Po4).
- \circ Alveolar regulation of PCO₂.

Metabolic acidosis :

(PH < 7.34 HCO₃ < 18).

• Anion gab (maintained by renal excretion of anions).

I.e calculate the unmeasured anions.

1) MA with normal AG :

• Loss of HCO $_3$

 \odot GI (loss of HCO₃). like diarrhea

 \odot Renal (failure to reabsorb HCO_3) .

e.g RTA. or bartter syndrome

2) MA with high AG

- Exogenous like aspirin toxicity
- Over production of acid (↑anions of these acids).

e.g

• **DM**.

 \circ Lactic acidosis (sepsis , inborn error of metabolism).

• Renal failure Po₄ (buffer).

Management : NaHCO 3 .

Metabolic Alkalosis:

- Gain of base (alkali Administration).
- Loss of acid (loss of HCI). pyloric stenosis
- · Cl resistant alkalosis e.g Bartter syndrome.
- · 2° Hyperaldosteronism.

Management : Treat underlying condition.

Respiratory alkalosis is very rare in children

Thank you